

Research Article

Dispersion and Intersection of Hydrothermal Plumes in the Manus Back-Arc Basin, Western Pacific

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The composition of hydrothermal plumes reflects the physical and chemical characteristics of seafloor hydrothermal fluids, which in turn reflects the host rock and subseafloor reaction conditions as well as the water column processes that act to alter the plumes as they disperse and age. Here, we show that the turbidity, current, pH value, dissolved Fe (dFe), and dissolved Mn (dMn) compositions of hydrothermal plumes can be used to understand the spatial distribution and source of hydrothermal systems in the submarine geological environment. Data were obtained from 18 hydrocast stations, among which the water column samples were collected at 8 stations during the MANUS cruise of R/V KEXUE in 2015. The results showed that the Satanic Mills plume and Fenway plume rose approximately 140 m and 220 m above the seafloor, respectively. In the Satanic Mills plume, dFe remained longer than dMn during lateral plume dispersal. There was a clear intersection of the Satanic Mills plume and Fenway plume between 1625 m and 1550 m in the PACMANUS hydrothermal field, and the varied dispersion trends of the mixed plumes were affected by current velocities at different depths. The physical and chemical properties of the seawater columns in the Manus Basin were affected by the input of high-Mn, high-Fe, and low-Mg vent fluids. The turbidity and dFe, dMn, and dissolved Mg concentrations in the sections of the plumes proximal to the Satanic Mills, Fenway, and Desmos vent sites were generally higher (turbidity, Mn, and Fe) and lower (Mg) than those in the sections of the plumes that were more distal from the vent sites. This implied that the plumes proximal to their vent fluid sources, which were interpreted to have relatively young ages, dispersed chemically over time, and their concentrations became more similar to those of the plumes that were more distal from their vent fluid sources.

1. Introduction

Hydrothermal plumes represent a significant dispersal mechanism for chemicals released from seafloor vents to the oceans. They are also of interest to geochemists because they can be exploited to detect and locate new hydrothermal fields [1]. Hydrothermal circulation in young oceanic crust involves cold seawater that penetrates the seafloor and reacts with the underlying rocks, where it undergoes chemical and thermal modification and is converted into hydrothermal fluid enriched in dissolved Fe (dFe) and dissolved Mn (dMn) that is often one million times more concentrated than the background deep ocean concentrations [1–4]. As soon as they emerge at seafloor hydrothermal vents, the fluids begin the processes of mixing and chemical exchange with ambient seawater and oxidation, dispersion, and settling



FIGURE 1: (a) Tectonic map of the Manus Basin (modified from Binns et al. 1995). (b) Location of hydrothermal fields in the eastern Manus back-arc basin (bathymetric map and data were obtained from http://www.geomapapp.org/index.htm). (c) Location of CTD stations in the Satanic Mills and Fenway vent sites in the PACMANUS hydrothermal field (modified from [34]). (d) Location of CTD stations in the Desmos hydrothermal field (modified from [44]).

of particles by physical processes [1, 5-11]. The resulting hydrothermal plumes can be traced over tens to thousands of kilometers using tracers, such as dMn, dFe, particulates, and isotopes such as ³He [12, 13]. Total dFe is expected to rapidly precipitate from hydrothermal plumes in solid phases as sulfides or oxyhydroxides [11, 14]. Moreover, the long-distance transport of hydrothermally sourced dFe has been identified in the Pacific Ocean [15]. The stabilization mechanisms for dFe include the formation of small inorganic nanoparticles in the colloidal size fraction [16-18], complexation by organic ligands [19-21], and reversibility of exchange onto slowly sinking particles [12] that protect dFe from precipitation and gravitational settling. By contrast, the dMn within hydrothermal plumes typically exhibits much slower oxidation kinetics and remains predominantly in dissolved form at the time of emplacement in the nonbuoyant plume [1]. As such, the plumes carry with them a record of the source fluids, reaction conditions below the seafloor, and thermal energy fluxes produced at the vents. However, the plumes proximal to their vent fluid sources are formed later and are younger than those far from their vent fluid sources.

Since 1990, studies on seafloor hydrothermal activities in the eastern Manus back-arc basin have shown widespread occurrence of hydrothermal plumes. For example, during the 21st cruise of R/V *Akademik Mstislav Keldysh*, all seawater column casts in the Manus Basin showed elevated turbidity anomalies (up to 1880 nephels at a depth of 1600 m) and positive anomalies of dMn (up to 43.5 nmol/kg at a depth of 1975 m) [22]. Later, during November and December of that year, large-scale triple-layered buoyant plumes were identified from water column anomalies of dMn (up to 97 nmol/kg at a depth of 1732 m) and pH (as low as 7.55 at a depth of 1700 m) in the eastern Manus Basin [23]. Furthermore, during the KX08-973 cruise of R/V *KEXUE YIHAO* in 2008, buoyant plumes were identified at 1.8 km north-northwest of the Desmos caldera (dMn of up to 79.3 nM at a depth of 1641 m), above the North Susu Knolls (dMn of up to 36.3 nM at a depth of 1492 m), and 1.3 km to the east of the Suzette hydrothermal field (dMn of up to 110.8 nM at a depth of 1500 m) [24].

Despite the progress made by these studies, little is known about the influence of vent fluids on the chemical compositions of seawater in the Manus Basin. In this study, we determined the turbidity, current velocity, pH, dFe, and dMn distributions of water columns in the PAC-MANUS and Desmos hydrothermal fields in an effort to understand how they vary and the relationships between dFe, dMn, physical properties, and currents in hydrothermal plumes.

2. Geological Setting

The eastern Manus Basin is the youngest spreading area in the Manus Basin, and it is experiencing extension within a remnant Eocene-Oligocene island-arc crust spreading at a rate of 13.5–14.5 cm/y [25] (Figure 1(a)). There are five main hydrothermal fields in the Manus Basin, namely, the PAC-MANUS field, Desmos caldera (23 km east of PACMANUS), Susu Knolls field (40 km east of PACMANUS), Solwara 12 (3.7087°S, 151.8833°E), and Vienna Woods (3.1645°S, 150.2795°E) [24, 26–31] (Figure 1(b)).

The PACMANUS field, which is located on the Pual Ridge, is notable for its distinctly siliceous volcanic host rocks (rhyodacite) and features various sulfides and Fe-Si-Mn oxyhydroxides [26, 32]. The geochemical characteristics of PAC-MANUS vent fluids, such as negative δD_{H_2O} and $\delta^{34}S_{H_2S}$ values, low pH values (at 25°C), high F concentration, and high CO₂ content, reflect magmatic water input, magma degassing, and seawater ingress into the upflow zone [29]. In addition, the ⁸⁷Sr/⁸⁶Sr and $\delta^{34}S$ signatures of anhydrite recovered at Roman Ruins and Snowcap and the multiple -S isotope compositions of hydrothermal vent fluid H₂S and coexisting chalcopyrite in this field suggest that a complex interplay exists between hydrothermal fluids, magmatic fluids, and seawater in the PACMANUS hydrothermal system [30, 33].

Satanic Mills is an active hydrothermal field with numerous isolated discharge sites at 1695-1675 m. The active vents emit predominantly black smoker fluids from clusters of numerous branched, thin (maximum of 10 cm) chimneys. The distance between the northern and southernmost chimney clusters is approximately 100 m. The east-west dimension of the field has a maximum width of 40 m [34]. The Fenway hydrothermal field is situated on the southeast flank of the Pual Ridge in a depression surrounded by steep (30°) slopes to the west, northwest, and north. Fenway consists of four clusters of hydrothermal vents and a central black smoker complex (Big Papi) at a depth of approximately 1715 m. In 2006, the Big Papi mound was the most vigorously venting black smoker site of PACMANUS. In 2011, the discharge of black smoker fluids declined. Diffuse venting is prominent in the immediate vicinity of Big Papi, which is manifested by shimmering water streaming up from the surrounding anhydrite sand and sediment [34].

The Desmos caldera, which is hosted by a more mafic basaltic andesite host rock [35], is slightly elongated in the north-northwest direction with dimensions of 1.5 km by 2.0 km and a deep circular depression of approximately 150–250 m [36]. The white smoker fluids rich in sulfate with a low pH of 0.95 to 2.10 and temperatures of 70° C to 120°C were discovered at the northwest terrace of the Desmos caldera during the 1995 Manus Flux cruise, which was named the Onsen hydrothermal site [28, 31, 36]. Similar to the North Su fluid, abundant CO₂, SO₂, and excess F in the Onsen fluid indicate magma degassing; the alteration assemblage of basaltic andesite from the Desmos caldera is thought to be formed by the interaction of basaltic andesite and hot acidic fluid (up to 340°C) originating from the mixing of magmatic fluid and seawater [37].

3. Sampling and Analytical Methods

3.1. Sample and Physical Data Collection. During the MANUS cruise of R/V KEXUE in 2015, 18 hydrocast stations were set up in the water column above the PACMANUS (9 stations) and Desmos (9 stations) hydrothermal fields, among which the seawater and hydrothermal plume samples were collected at 8 stations (Figures 1(c) and 1(d)). We followed the methods of Zeng et al. [38]. The seawater and hydrothermal plume samples were collected at different depths with a Conductivity-Temperature-Depth (CTD) alu-

minum rosette (Seabird) containing twenty-four 10 L Niskin bottles. To ensure the highest possible purity of the samples, the Niskin bottles were cleaned thoroughly using 1% v/v HNO₃ solution and Milli-Q water before usage.

Upon recovery of the Niskin bottles, 30 cm long Teflon tubes were inserted through the outlet at the bottom of the Niskin bottles, and the first 200 ml of the liquid was discarded. The remaining aqueous samples were collected in 5L precleaned high-density polyethylene bottles. The precleaning method is described in detail by Cutter et al. [39].

During the operation of each station, data were collected throughout the seawater column with an SBE 911plus CTD system (Seabird) coupled to a Seapoint turbidity meter and a lowered acoustic Doppler current profiler (LADCP). The probes were obtained from the manufacturers and calibrated by the National Center of Ocean Standards and Metrology in July 2013. The measurement accuracies were $\pm 0.001^{\circ}$ C for temperature, ± 0.0003 S/m for conductivity, $\pm 0.015\%$ of the full-scale range for pressure, $\pm 0.0002^{\circ}$ C, ± 0.0003 S/m, $\pm 0.0015\%$ of the water velocity relative to LADCP), and $\pm 2^{\circ}5'$ for direction, with resolutions of $\pm 0.0002^{\circ}$ C, ± 0.0003 S/m, $\pm 0.0015\%$ of the full-scale range, 0.001 m/s, and 0.01° , respectively. The turbidity sensitivity was 200 mV/FTU (100x gain; range of 25 FTU) [38].

3.2. Analytical Methods. The pH of each aqueous sample was determined by a portable pH meter (JENCO 6010; resolution of 0.01 with automatic temperature compensation) immediately after collection at approximately 25°C. The pH meter was calibrated with buffer solutions of pH 4.00 (0.050 mol/L of potassium hydrogen phthalate) and 6.86 (0.025 mol/L of mixed phosphate). The aqueous samples were filtered through a 47 mm Merck Millipore 0.10 μ m nitrocellulose membrane into 1 L Nalgene polypropylene bottles (previously soaked in 1:1 HNO₃ for 48 h and washed to neutral pH with deionized water and ultrapure water) within the ship laboratory immediately after collection from the Niskin bottles [38]. Filtered water samples were acidified to a pH of 1.8 using 2 M ultrapure HNO₃ (J. T. Baker), capped tightly, and resealed with a Parafilm sealing membrane.

For the determination of Na, Mg, Ca, and K, the filtered and acidified samples were diluted 10 times by pure water and the concentrations of dissolved Na, Mg, Ca, and K in the aqueous samples were measured by inductively coupled plasma optical emission spectrometry (PE 2100DV) with a precision greater than ±5% at the Shandong Institute of Geophysical and Geochemical Exploration [38]. For the determination of Cl and SO_4^{2-} , the samples were diluted 200 times by pure water and the Cl and SO_4^{2-} concentrations were measured by ion chromatography (ICS-1100) with an anion exchange resin column (Dionex AS19) rinsed with a solution of 1.8 mmol/L of Na₂CO₃ and 1.7 mmol/L of NaHCO₃ at a rinsing rate of 1 mL/min and precision of $\pm 3\%$ [38]. The accuracy was controlled by the recovery; for Na, Mg, Ca, K, Cl, and SO_4^{2-} , the recovery rates were 99.5%, 99.6%, 99.8%, 99.7%, 99.7%, and 100.1%, respectively.

The dFe and dMn contents of the seawater and plume samples were determined by inductively coupled plasma sector field mass spectrometry (Element, Thermo Scientific) at ALS Scandinavia AB, Luleå, Sweden, in accordance with the method described by Rodushkin and Ruth [40]. Reference materials NASS-6 (North Atlantic Seawater) and CASS-5 (Nearshore Seawater) from the National Research Council of Canada were used to evaluate the accuracy of the dFe and dMn analyses, and accuracies and precision (n = 5) greater than 5% were achieved for both [38].

4. Results

4.1. Element Concentrations in Seawater Columns. The concentrations of major components (Ca, Mg, K, Na, Cl, and SO₄²⁻) in the Satanic Mills and Fenway vent fluids [29] were different from those of the bottom seawater. Their profiles in the water column samples collected above the PACMANUS and Desmos fields exhibited slight variation (Tables 1 and 2). The highest dissolved Ca (10.2 mmol/kg) concentrations were observed in the background seawater samples (depth shallower than 800 m) from the Desmos field, and dissolved Na (432.0 mmol/kg) concentrations were observed in the seawater column samples (depth deeper than 1200 m) in the Fenway vent site from the PACMANUS field (Tables 1 and 2). However, the concentrations of dissolved K (9.72–11.5 mmol/kg), SO_4^{2-} (28.3– 30.9 mmol/kg), dFe (0.010-0.133 mmol/kg), and dMn (0.001–0.539 mmol/kg) and the pH of the seawater profiles in the PACMANUS and Desmos fields were significantly more variable and greater than those of ambient seawater, except for pH, which was lower (7.49 to 8.14) (Tables 1 and 2; Figures 2 and 3). At stations P-W07, P-W02, P-W01, and P-W04 in the PACMANUS field, from a depth of 1200 m to the bottom, the turbidity and dFe increased relative to that in the seawater shallower than 1200 m, and pH showed a decreasing trend with the lowest value of 7.49 observed at 1665 m at P-W04. However, the concentration of dMn began to increase from different depths at these four stations (Figure 2).

4.2. Physical and Chemical Characteristics of Hydrothermal Plumes. Station P-W02 was located at the Satanic Mills vent site, and station P-W04 was located at the Fenway vent site (Figure 1(c)). As the hydrothermal plumes above the Satanic Mills (P-W02) and Fenway (P-W04) vent sites dispersed laterally, the anomalous turbidity layer also dispersed (Figure 2), which was identical to that in the middle plume observed by Gamo et al. [23] during the second cruise of the AQUARIUS Expedition (KH-90-3) in 1990. The turbidity anomaly was 400-500 m thick with values 2-5 times greater than those of the background seawater, which was similar to that observed in the Manus Basin in 1990 [22]. At P-W02, the maximum turbidity (0.475 NTU), dMn (0.533 μ mol/kg), and dFe (0.058 μ mol/kg) anomalies were found at 1550 m, and at P-W04, the maximum turbidity (0.484 NTU), dMn (0.539 µmol/kg), and dFe $(0.133 \,\mu \text{mol/kg})$ anomalies were found at 1526 m (Table 1; Figure 2). According to the anomalous layer with the maximum dMn concentration, the Satanic Mills plume rose approximately 140 m above the seafloor, and the Fenway

plume rose approximately 220 m above the seafloor (Figure 2). Moreover, the Satanic Mills plume and Fenway plume intersected each other at a depth between 1625 m and 1550 m (Figure 4).

At stations D-W09, D-W02, D-W01, and D-W04 in the Desmos field, from a depth of 1400 m to the bottom, the turbidity and dFe concentration increased relative to those of the seawater shallower than 1400 m, and the pH showed a decreasing trend with the lowest value of 7.51 observed at depth of 1824 m at D-W01 and 1840 m at D-W04; however, the dMn concentration began to increase from a depth of 1500 m (Figure 3). Stations D-W01 and D-W04 were located at the inner southeast flank wall of the Desmos caldera (Figure 1(d)) where plumes were identified by anomalies of turbidity, dMn, and dFe (Figure 3). At station D-W01, the anomalous layer with the maximum dMn concentration $(0.101 \,\mu \text{mol/kg})$ was found at a depth of 1768 m, which was deeper than the deep plume (approximately 1700 m) observed by Gamo et al. [23] in 1990. The maximum dFe $(0.076 \,\mu \text{mol/kg})$ anomaly was found at a depth of 1800 m, thereby indicating that the plume was 110-140 m above the seafloor. At station D-W04, the maximum dMn $(0.158 \,\mu \text{mol/kg})$ and dFe $(0.084 \,\mu \text{mol/kg})$ anomalies were found at a depth of 1840 m (Table 2; Figure 3), so the plume rose approximately 50 m above the seafloor. Moreover, the significant anomalous centers of turbidity at D-W01 and D-W04 appeared alternately at different depths (Figure 5).

However, the turbidity (up to 0.484 FTU and 0.292 FTU), dFe (up to 0.133 μ mol/kg and 0.084 μ mol/kg), and dMn (up to $0.539 \,\mu \text{mol/kg}$ and $0.158 \,\mu \text{mol/kg}$) concentrations of the P-W04, P-W02, D-W01, and D-W04 hydrothermal plumes near the Satanic Mills, Fenway, and Desmos vent sites were generally higher than those of the P-W07 and D-W09 plumes far from these vent sites (Figures 1 and 2), and the dissolved Mg (49.5 mmol/kg and 49.3 mmol/kg) and Ca (9.12 mmol/kg and 9.13 mmol/kg) concentrations of the P-W04, P-W02, D-W01, and D-W04 plumes near the vent sites were generally lower than those of the plumes far from these vent sites in the PACMANUS and Desmos fields (Tables 1 and 2). Furthermore, the dissolved Mg, Cl, dFe, and dMn concentrations of the hydrothermal plumes in the PACMANUS field near the Satanic Mills and Fenway vent sites were generally higher than those of the plumes near the Desmos vent site, and the SO₄²⁻ concentrations of the plumes near the Satanic Mills and Fenway vent sites were generally lower than those of the plumes near the Desmos vent site (Tables 1 and 2; Figures 2 and 3).

5. Discussion

5.1. Turbidity, dMn, and dFe in Hydrothermal Plumes. Hydrothermal plumes can be distinguished by anomalies relative to the background seawater values, including elevated dFe and dMn concentrations and turbidity. In the PACMANUS hydrothermal field, the maximum temperatures of the vent fluids of the Satanic Mills and Fenway vent sites were 345°C and 313°C in 2011, respectively [34]. Above these vent sites, plumes were detected with high dMn concentrations that were 150–200 times that of the background

Sample	P								ζ	2-	ţ	
	Ueptn (m)	Velocity (m/s)	Turbidity (NTU)	pHv 25°C	Ca (mmol/kg)	Mg (mmol/kg)	K (mmol/kg)	Na (mmol/kg)	د (mmol/kg)	ь04 ⁻ (mmol/kg)	Fe (µmol/kg)	Mn (µmol/kg)
004'S, 151°40.23	17'E; deptl	h: 1719 m)										
P-W07-1	1629	0.089	0.089	7.64	9.57	50.7	10.5	505	543	30.2	0.028	0.142
P-W07-4	1601	0.096	0.254	7.61	9.49	50.8	10.5	496	548	29.9	0.027	0.288
P-W07-5	1575	0.067	0.085	7.64	9.52	50.5	10.8	507	545	29.7	0.017	0.014
P-W07-6	1551	0.074	0.089	7.68	9.55	51.2	10.9	494	551	29.8	0.024	0.010
P-W07-7	1525	0.098	0.09	7.69	9.58	51.1	10.9	478	521	30.3	0.028	0.011
P-W07-8	1500	0.085	0.092	7.68	9.50	50.8	10.5	466	535	29.2	0.017	0.010
P-W07-9	1475	0.059	0.087	7.64	9.41	51.2	10.9	483	544	29.8	0.044	0.012
P-W07-10	1451	0.059	0.094	7.66	9.52	51.0	11.0	481	550	29.9	0.021	0.00
P-W07-11	1425	0.095	0.077	7.67	9.65	51.9	10.9	479	559	29.7	0.020	0.007
P-W07-12	1400	0.081	0.065	7.66	9.65	50.8	10.5	473	535	30.3	0.026	0.008
P-W07-13	1351	0.020	0.055	7.67	9.77	52.4	10.9	491	555	29.2	0.011	0.007
P-W07-14	1300	0.078	0.041	7.68	9.67	50.0	10.8	481	534	29.2	0.016	0.006
P-W07-15	1251	0.038	0.037	7.67	9.59	50.8	10.6	471	541	29.2	0.011	0.005
P-W07-16	1201	0.085	0.037	7.67	9.42	49.8	10.5	473	551	29.0	0.011	0.005
P-W07-17	1001	0.043	0.032	7.70	9.52	51.0	11.2	482	534	29.5	0.017	0.002
P-W07-20	800	0.080	0.038	7.69	9.36	51.0	10.2	435	548	30.0	0.012	0.002
P-W07-21	601	0.140	0.034	7.74	9.47	50.3	10.5	488	551	29.1	0.014	0.001
P-W07-22	400	0.077	0.031	7.81	9.46	50.9	10.1	469	553	29.8	0.017	0.002
P-W07-23	200	0.221	0.028	7.96	9.82	51.9	10.4	482	555	30.3	0.013	0.002
P-W07-24	100	0.200	0.054	8.11	9.53	51.7	10.3	487	547	30.7	0.011	0.002
;4'S, 151°40.284	t'E; depth:	1690 m)										
P-W02-1	1641	0.103	0.294	7.63	9.72	52.2	11.0	501	534	28.9	0.054	0.335
P-W02-4	1626	0.107	0.329	7.63	9.63	51.9	10.8	487	538	28.6	0.064	0.360
P-W02-5	1600	0.121	0.361	7.61	9.53	51.6	10.4	453	548	29.5	0.032	0.384
P-W02-6	1575	0.128	0.352	7.63	9.37	51.2	11.0	487	538	29.9	0.050	0.386
P-W02-7	1551	0.100	0.475	7.63	9.51	52.4	10.9	474	527	29.5	0.058	0.533
P-W02-8	1525	0.088	0.103	7.67	9.43	52.4	10.7	480	531	28.8	0.023	0.014
P-W02-9	1500	0.120	0.096	7.62	9.58	52.1	10.5	474	540	28.8	0.030	0.013
P-W02-10	1475	0.094	0.077	7.67	9.38	51.1	10.6	468	538	29.2	0.032	0.012
P-W02-11	1451	0.097	0.072	7.67	9.44	52.4	10.8	475	549	30.2	0.034	0.010
P-W02-12	1426	0.115	0.063	7.67	9.43	50.9	10.6	478	548	28.8	0.027	0.009
P-W02-13	1401	0.077	0.062	7.64	9.49	51.3	10.7	477	555	29.9	0.028	0.009
P-W02-14	1351	0.107	0.052	7.67	9.43	51.7	10.6	474	557	29.1	0.026	0.008

Geofluids

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Station	Sample	Depth (m)	Velocity (m/s)	Turbidity (NTU)	pHv 25 °C	Ca (mmol/kg)	Mg (mmol/kg)	K (mmol/kg)	Na (mmol/kg)	Cl (mmol/kg)	SO ₄ ²⁻ (mmol/kg)	Fe (µmol/kg)	$\mathop{\rm Mn}_{(\mu {\rm mol/kg})}$
	P-W02-15	1301	0.120	0.043	7.67	9.61	52.5	10.6	476	544	29.3	0.034	0.007
	P-W02-16	1201	0.163	0.035	7.65	9.49	51.8	10.5	472	544	29.5	0.017	0.007
	P-W02-17	666	0.087	0.04	7.68	9.61	52.6	10.8	479	549	29.2	0.021	0.004
	P-W02-20	800	0.120	0.028	7.71	9.44	52.5	10.6	473	552	29.3	0.020	0.002
	P-W02-21	600	0.285	0.027	7.74	9.84	51.4	10.2	465	541	29.6	0.026	0.003
	P-W02-22	400	0.038	0.027	7.81	9.57	51.6	10.4	471	539	29.9	0.019	0.003
	P-W02-23	200	0.128	0.025	7.98	9.53	52.2	10.7	490	546	29.9	0.018	0.004
	P-W02-24	101	0.219	0.031	8.13	9.61	52.4	10.8	492	560	29.9	0.017	0.002
P-W01 (3°43.64]	1'S, 151°40.314	'E; depth:	1695 m)										
	P-W01-1	1625	0.119	0.254	7.53	9.63	51.0	10.6	476	546	30.1	0.067	0.251
	P-W01-2	1601	0.082	0.172	7.60	9.63	52.3	10.2	468	543	29.5	0.040	0.089
	P-W01-3	1575	0.060	0.096	7.62	9.71	51.1	11.0	492	534	29.8	0.038	0.012
	P-W01-4	1551	0.086	0.095	7.60	9.41	51.2	10.9	485	539	29.4	0.027	0.012
	P-W01-5	1525	0.102	0.086	7.62	9.37	49.7	10.9	490	543	29.3	0.026	0.011
	P-W01-6	1500	0.124	0.089	7.64	9.33	49.9	10.5	472	551	29.0	0.027	0.013
	P-W01-7	1476	0.132	0.096	7.61	9.30	49.6	10.9	485	539	29.8	0.023	0.013
	P-W01-8	1451	0.105	0.083	7.59	9.25	49.9	10.1	478	548	28.4	0.030	0.013
	P-W01-9	1426	0.094	0.106	7.67	9.12	50.4	10.5	479	550	29.7	0.036	0.013
	P-W01-10	1401	0.107	0.058	7.63	9.59	52.9	11.2	506	557	30.0	0.028	0.012
	P-W01-11	1376	0.089	0.079	7.64	9.52	51.0	10.5	465	539	30.0	0.027	0.00
	P-W01-12	1351	0.048	0.054	7.66	9.29	51.4	6.6	474	557	29.3	0.026	0.008
	P-W01-13	1326	0.060	0.048	7.68	9.34	51.8	10.5	493	553	28.9	0.105	0.00
	P-W01-14	1301	0.092	0.046	7.65	9.34	51.1	10.2	450	553	28.9	0.037	0.009
	P-W01-15	1276	0.114	0.044	7.65	9.33	52.0	10.2	472	554	29.3	0.029	0.008
	P-W01-16	1251	0.095	0.04	7.65	9.49	52.0	10.5	475	556	29.9	0.029	0.008
	P-W01-17	1226	0.139	0.039	7.62	9.29	50.4	10.6	481	553	29.5	0.033	0.008
	P-W01-18	1201	0.166	0.041	7.67	9.49	51.5	10.1	439	556	30.0	0.024	0.005
	P-W01-19	1000	0.057	0.032	7.65	9.41	51.0	10.8	472	558	29.4	0.030	0.005
	P-W01-20	800	0.122	0.033	7.68	9.45	51.6	10.6	477	554	29.5	0.026	0.069
	P-W01-21	600	0.106	0.028	7.73	9.43	50.3	11.0	478	549	28.3	0.039	0.015
	P-W01-22	400	0.130	0.031	7.82	9.28	51.6	11.1	491	553	29.7	0.026	0.003
	P-W01-23	201	0.166	0.032	7.99	9.74	53.1	11.4	506	545	30.1	0.045	0.003
	P-W01-24	100	0.104	0.034	7.99	9.53	52.0	10.9	481	544	30.6	0.035	0.004

TABLE 1: Continued.

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Station	Sample	Depth (m)	Velocity (m/s)	Turbidity (NTU)	pHv 25 °C	Ca (mmol/kg)	Mg (mmol/kg)	K (mmol/kg)	Na (mmol/kg)	Cl (mmol/kg)	SO ₄ ²⁻ (mmol/kg)	Fe (µmol/kg)	Mn (µmol/kg)
P-W04 (3°43.67	'76'S, 151°40.34	4'E; depth	:1749m)										
	P-W04-1	1665	0.071	0.13	7.49	9.30	51.0	10.5	467	547	30.0	0.034	0.170
	P-W04-4	1650	0.026	0.165	7.61	9.33	50.2	10.8	481	542	30.0	0.025	0.184
	P-W04-5	1625	0.041	0.22	7.62	9.22	50.4	10.9	492	548	29.5	0.025	0.258
	P-W04-6	1600	0.031	0.346	7.60	9.21	50.0	10.6	478	545	29.4	0.039	0.348
	P-W04-7	1576	0.026	0.403	7.60	9.12	49.5	10.6	468	538	29.5	0.063	0.486
	P-W04-8	1550	0.010	0.367	7.58	9.18	50.8	11.0	479	551	29.5	0.043	0.433
	P-W04-9	1526	0.009	0.484	7.55	9.23	51.3	10.5	472	538	29.7	0.133	0.539
	P-W04-10	1500	0.045	0.451	7.58	9.18	50.4	10.5	470	528	29.9	0.029	0.466
	P-W04-11	1475	0.048	0.094	7.61	9.37	49.8	10.8	483	552	30.1	0.045	0.059
	P-W04-12	1450	0.091	0.083	7.63	9.26	49.7	10.7	479	545	29.6	0.057	0.017
	P-W04-13	1426	0.091	0.067	7.65	9.59	52.1	10.8	482	544	30.1	0.016	0.010
	P-W04-14	1401	0.096	0.059	7.66	9.59	51.0	10.8	471	555	29.5	0.069	0.021
	P-W04-15	1301	0.050	0.066	7.66	9.36	50.2	10.5	465	547	30.3	0.012	0.006
	P-W04-16	1201	0.072	0.036	7.63	9.45	49.5	10.7	432	537	29.7	0.024	0.006
	P-W04-17	666	0.118	0.032	7.64	9.32	49.8	10.2	474	539	29.7	0.018	0.002
	P-W04-20	801	0.077	0.027	7.70	9.42	51.1	10.6	485	538	29.3	0.015	0.002
	P-W04-21	601	0.157	0.043	7.75	9.49	50.2	10.4	502	527	29.5	0.010	0.001
	P-W04-22	401	0.057	0.031	7.78	9.45	50.9	10.4	499	531	30.2	0.011	0.001
	P-W04-23	201	0.246	0.028	7.95	9.66	52.1	10.9	545	548	29.5	0.013	0.001
	P-W04-24	100	0.243	0.038	8.13	9.66	52.2	11.3	533	551	30.0	0.010	0.002
Satanic Mills ^a		1695			2.59	11.2	12.5	52.4	401	503	4	1663	1869
Fenway ^a		1715			2.90	19.8	9.46	65.8	420	576	4.3	9761	3142
Bottom					7 80 ^b	10.5	57 4	06.6	471	540	787	0.001	0.005
seawater ^c					00.1		1.70	0	1 /1		1.01	100.0	0000
^a Satanic Mills and	Fenway average	vent fluid di	ata from Reeve	es et al. [29]. ^b	Average pl	H value from D	ouville et al. [45]. ^c Bottom seaw	ater data from S	eewald et al. [31	<u>.</u>		

Geofluids

TABLE 1: Continued.

Mn (µmol/kg)		0.085	0.077	0.087	0.048	0.057	0.057	0.050	0.047	0.051	0.056	0.051	0.044	0.070	0.039	0.035	0.036	0.037	0.040	0.042	0.039		0.068	0.062	0.057	0.069	0.067	0.059	0.067	0.057	0.094	0.096	0.097	0.079
Fe (µmol/kg)		0.039	0.017	0.015	0.016	0.017	0.013	0.016	0.013	0.016	0.014	0.012	0.016	0.015	0.012	0.010	0.024	0.010	0.013	0.017	0.011		0.023	0.025	0.022	0.022	0.021	0.017	0.066	0.018	0.023	0.018	0.053	0.020
SO ₄ ²⁻ (mmol/kg)		30.2	29.6	29.5	29.7	30.1	30.8	30.9	30.4	30.5	30.6	30.3	30.9	30.1	30.5	30.8	30.0	30.2	30.7	30.8	30.7		29.5	29.3	29.6	29.2	29.1	29.4	29.7	29.2	29.3	30.2	29.6	29.8
Cl (mmol/kg)		541	534	541	540	538	548	541	551	546	539	549	546	533	558	550	531	541	548	545	538		549	546	529	557	554	545	538	556	534	544	553	542
Na (mmol/kg)		461	479	475	481	484	458	463	460	469	490	492	477	480	472	468	484	471	471	488	468		500	492	454	475	476	454	445	438	435	493	501	459
K (mmol/kg)		10.4	10.6	11.2	11.0	10.5	10.6	10.7	10.9	10.4	11.1	11.0	10.7	10.9	10.4	10.3	11.1	10.8	10.7	11.1	10.6		10.9	11.0	10.8	10.6	10.5	10.2	9.7	10.3	10.4	10.3	11.3	10.6
Mg (mmol/kg)		50.8	49.8	49.9	50.9	50.7	48.6	49.9	50.2	49.5	51.8	51.3	51.0	51.7	49.6	49.9	50.7	49.6	49.9	50.3	51.0		51.2	50.6	51.3	51.6	50.2	50.0	50.8	50.1	50.6	50.9	51.3	51.0
Ca (mmol/kg)		9.46	9.31	9.55	9.68	9.43	9.49	99.66	9.68	9.41	9.85	9.58	9.53	9.83	9.17	9.36	9.38	9.26	9.02	9.61	9.28		9.71	9.67	9.76	9.71	9.64	9.63	9.64	9.84	10.02	9.81	9.92	9.87
pH 25 °C		7.59	7.65	7.67	7.64	7.66	7.68	7.66	7.67	7.67	7.67	7.69	7.67	7.64	7.70	7.71	7.73	7.78	7.82	7.97	8.13		7.64	7.65	7.63	7.66	7.64	7.69	7.67	7.72	7.65	7.64	7.67	7.64
Turbidity (NTU)		0.106	0.085	0.089	0.084	0.089	0.091	0.069	0.085	0.056	0.051	0.053	0.049	0.101	0.046	0.041	0.038	0.036	0.042	0.038	0.054		0.093	090.0	0.052	0.071	0.051	0.045	0.037	0.039	0.054	0.093	0.210	0.059
Velocity (m/s)	: 2089 m)	0.068	0.067	0.077	0.080	0.074	0.080	0.067	0.055	0.048	0.058	0.063	0.078	0.103	0.129	0.034	0.028	0.077	0.046	0.172	0.172	: 1959 m)	0.041	0.040	0.046	0.039	0.081	0.087	0.097	0.091	0.067	0.067	0.050	0.080
Depth (m)	'E; depth	1982	1951	1926	1900	1876	1850	1825	1800	1775	1750	1725	1700	1600	1300	1001	800	600	401	200	100	'E; depth	1949	1926	1900	1875	1850	1825	1800	1775	1751	1725	1700	1600
Sample	.464'S, 151°52.494	D-W09-1	D-W09-4	D-W09-5	D-W09-6	D-W09-7	D-W09-8	D-W09-9	D-W09-10	D-W09-11	D-W09-12	D-W09-13	D-W09-14	D-W09-15	D-W09-16	D-W09-17	D-W09-20	D-W09-21	D-W09-22	D-W09-23	D-W09-24	2178'S, 151°52.584	D-W02-1	D-W02-4	D-W02-5	D-W02-6	D-W02-7	D-W02-8	D-W02-9	D-W02-10	D-W02-11	D-W02-12	D-W02-13	D-W02-14
Station	D-W09 (3°42.1																					D-W02 (3°42.2												

TABLE 2: Measured compositions of water column samples from the Desmos hydrothermal field.

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Sample	Depth (m)	Velocity (m/s)	Turbidity (NTU)	PH 25 °C	Ca (mmol/kg)	Mg (mmol/kg)	K (mmol/kg)	Na (mmol/kg)	Cl (mmol/kg)	SO4 ²⁻ (mmol/kg)	Fe (µmol/kg)	Mn (µmol/kg)
D-W02-15	1501	0.175	0.082	7.66	69.6	51.0	10.2	471	535	29.8	0.014	0.058
D-W02-16	1300	0.182	0.037	7.64	9.74	51.5	11.1	460	548	30.0	0.019	0.059
D-W02-17	1004	0.146	0.033	7.66	9.74	52.4	11.1	475	539	29.3	0.013	0.055
D-W02-20	800	0.156	0.027	7.67	9.40	50.0	10.4	477	550	30.0	0.015	0.049
D-W02-21	601	0.273	0.029	7.73	9.48	50.4	11.0	483	548	29.4	0.012	0.045
D-W02-22	400	0.256	0.038	7.80	9.32	49.4	10.2	472	533	29.7	0.014	0.056
D-W02-23	200	0.280	0.030	7.93	9.62	50.3	10.3	461	553	30.3	0.011	0.059
D-W02-24	100	0.317	0.050	8.12	9.65	51.2	10.5	484	541	30.3	0.011	0.058
2334'S, 151°52.63.	2'E; depth	ı: 1912 m)										
D-W01-1	1904		0.130	7.68	9.79	51.8	10.4	479	544	30.0	0.018	0.029
D-W01-2	1875	0.068	0.139	7.67	9.74	50.6	10.8	485	541	29.8	0.019	0.023
D-W01-3	1851	0.068	0.241	7.66	9.61	51.3	10.3	483	545	29.4	0.028	0.030
D-W01-4	1824	0.022	0.325	7.51	9.71	51.9	10.7	443	533	29.7	0.047	0.087
D-W01-5	1800	0.036	0.223	7.55	9.64	51.6	10.2	485	541	29.5	0.076	0.087
D-W01-6	1768	0.031	0.251	7.55	9.80	51.6	10.7	511	551	29.7	0.061	0.101
D-W01-7	1750	0.012	0.167	7.65	9.73	49.9	10.6	488	541	29.7	0.024	0.025
D-W01-8	1726	0.032	0.360	7.66	9.74	50.6	10.6	475	549	29.9	0.023	0.040
D-W01-9	1701	0.011	0.077	7.67	9.78	50.9	10.5	477	547	29.7	0.019	0.026
D-W01-10	1676	0.023	0.040	7.66	9.94	51.2	10.9	492	544	29.7	0.017	0.023
D-W01-11	1651	0.035	0.297	7.65	9.70	51.3	10.8	495	548	29.6	0.020	0.046
D-W01-12	1626	0.066	0.313	7.66	9.37	50.4	10.9	486	543	29.6	0.025	0.045
D-W01-13	1600	0.071	0.076	7.69	9.52	49.3	10.8	480	553	29.6	0.018	0.037
D-W01-14	1550	0.054	0.063	7.67	9.62	50.7	10.8	490	547	29.4	0.017	0.028
D-W01-15	1500	0.031	0.076	7.69	9.72	50.6	10.9	505	535	29.8	0.013	0.010
D-W01-16	1450	0.057	0.063	7.69	9.37	50.2	10.8	487	546	29.8	0.013	0.010
D-W01-17	1400	0.040	0.042	7.71	9.85	51.2	10.9	503	551	29.2	0.010	0.007
D-W01-18	1200	0.180	0.030	7.72	9.64	50.8	10.4	479	549	29.6	0.013	0.008
D-W01-19	1000	0.123	0.031	7.71	9.87	52.1	10.7	502	534	29.4	0.020	0.003
D-W01-20	800	0.071	0.030	7.71	10.2	51.8	11.5	525	553	28.9	0.017	0.008
D-W01-21	600	0.081	0.030	7.79	9.75	50.7	10.7	480	546	29.7	0.023	0.003
D-W01-22	400	0.064	0.031	7.84	9.53	50.7	10.4	485	543	29.7	0.017	0.006
D-W01-23	188	0.104	0.029	8.04	10.1	52.5	10.9	509	548	30.9	0.016	0.002
D-W01-24	101	0.095		8.14	10.0	51.6	11.0	502	543	30.7	0.014	0.015

TABLE 2: Continued.

Geofluids

Station	Sample	Depth (m)	Velocity (m/s)	Turbidity (NTU)	pH 25 °C	Ca (mmol/kg)	Mg (mmol/kg)	K (mmol/kg)	Na (mmol/kg)	Cl (mmol/kg)	SO4 ²⁻ (mmol/kg)	Fe (µmol/kg)	Mn (µmol/kg)
D-W04 (3°42.25	05'S, 151°52.68.	22'E; depti	h: 1892 m)										
	D-W04-1	1840	0.148	0.276	7.51	9.51	49.6	10.5	475	538	30.2	0.084	0.158
	D-W04-4	1825	0.142	0.082	7.61	9.53	50.8	10.9	498	548	30.4	0.030	0.117
	D-W04-5	1800	0.162	0.292	7.57	9.44	50.0	10.7	491	548	29.7	0.050	0.150
	D-W04-6	1775	0.097	0.072	7.61	9.35	51.1	10.5	476	541	28.7	0.034	0.079
	D-W04-7	1750	0.087	0.197	7.60	9.35	50.2	10.5	488	551	30.4	0.047	0.092
	D-W04-8	1725	0.091	0.279	7.61	9.43	51.8	10.6	472	548	29.7	0.044	0.115
	D-W04-9	1700	0.070	0.079	7.66	9.35	50.9	10.4	491	551	29.7	0.026	0.100
	D-W04-10	1675	0.077	0.071	7.62	9.18	50.5	9.9	455	551	29.8	0.016	060.0
	D-W04-11	1650	0.075	0.190	7.66	9.13	50.4	10.5	477	545	29.6	0.017	0.068
	D-W04-12	1625	0.075	0.071	7.62	9.16	49.5	9.8	461	541	30.1	0.014	0.078
	D-W04-13	1600	0.071	0.093	7.64	9.36	50.5	9.9	453	548	29.7	0.014	0.071
	D-W04-14	1550	0.033	0.064	7.66	9.33	50.0	10.0	451	541	29.3	0.015	0.064
	D-W04-15	1500	0.025	0.064	7.70	9.43	49.9	10.2	452	535	29.5	0.014	0.056
	D-W04-16	1300	0.054	0.036	7.67	9.40	50.1	11.3	489	543	29.5	0.011	0.047
	D-W04-17	866	0.116	0.031	7.74	9.81	49.7	10.9	465	543	29.2	0.012	0.050
	D-W04-20	800	0.115	0.025	7.70	9.43	50.5	10.8	486	534	29.2	0.011	0.048
	D-W04-21	601	0.103	0.026	7.73	9.43	49.4	11.0	466	534	29.3	0.011	0.045
	D-W04-22	399	0.086	0.030	7.85	9.59	49.9	10.9	476	548	29.7	0.011	0.045
	D-W04-23	200	0.116	0.029	7.96	9.39	49.8	10.9	463	555	29.9	0.011	0.055
	D-W04-24	100	0.184	0.040	8.11	9.13	50.1	11.3	472	558	29.9	0.012	0.064
$\mathrm{Desmos}^{\mathrm{a}}$					1.23	10.7	48.0	8.70	414	503	89.4	7745	40
Bottom seawater ^a					7.80 ^b	10.5	52.4	06.6	471	540	28.2	0.001	0.005
^a Desmos average v	ent fluid data and	l bottom sea	awater data fro	om Seewald et	al. [31]. ^b	Average pH valı	ue from Douvill	e et al. [45].					

TABLE 2: Continued.



FIGURE 2: (a) Turbidity, (b) pH, (c) Mn, and (d) Fe anomalies in the water columns at the PACMANUS hydrothermal field (stations P-W07, P-W02, P-W01, and P-W04). The seawater composition (vertical dashed line) was obtained from Seewald et al. [31] and Douville et al. [45].

seawater (Figure 2). The maximum dMn concentration for the PACMANUS plume (539 nmol/kg) was higher than those seen elsewhere on fast (16–194 nmol/kg, East Pacific Rise) and ultraslow (0.47–10.41 nmol/kg, Southwest Indian Ridge) spreading ridges [41, 42]. The turbidity anomaly was 400–500 m thick with values 2–5 times greater than that of the background seawater, which was similar to that observed in the Manus Basin in 1990 [22]. However, based on our samples, two anomalous centers were observed corresponding to the positions of the Satanic Mills and Fenway vent sites, and a continuous turbidity profile showed multipeak anomalies at different depths (Figure 2). In the Desmos field, a 600 m thick turbidity anomaly with values 2–4 times greater than that of the background seawater was discovered above the bottom (Figure 3), which was at the same depth (1700 m) as the deep plume found by Japanese scientists in 1990 [23], thereby implying that the plume in the Desmos field has existed for more than 25 y. In the PACMANUS and Desmos fields, higher turbidity, lower pH, and higher dFe and dMn concentrations in the anomalous layer in the seawater columns were consistent with the levels of these elements in their source vent fluids in the area ([29, 31, 43]; Tables 1 and 2; Figures 2–5).



FIGURE 3: (a) Turbidity, (b) pH, (c) Mn, and (d) Fe anomalies in the water columns at the Desmos hydrothermal field (stations D-W09, D-W02, D-W01, and D-W04). The seawater composition (vertical dashed line) was obtained from Seewald et al. [31] and Douville et al. [45].

However, at the Satanic Mills and Fenway vent sites, the dFe_{fluid}/dFe_{maximum in plume} (approximately $2-7 \times 10^4$) was one order of magnitude higher than the dMn_{fluid}/ dMn_{maximum in plume} ($3-6 \times 10^3$), which was attributed to the rapid precipitation of dFe from the buoyant hydrothermal plume [1, 14]; thus, more dFe was removed than dMn at the initial stage. However, the relative reductions of dFe and dMn were different from those of the initial stage when the plume dispersed laterally. In the Satanic Mills field, at the maximum Mn anomaly layer (depth of 1550 m), the plume dispersed in the northwest direction according to the turbidity contour map, the current direction (Figures 4(d) and 4(f)), and the dMn and dFe concentrations at P-W02 (0.533μ mol/kg of dMn and 0.058μ mol/kg of dFe) and P-W07 (0.010μ mol/kg of dMn and 0.024μ mol/kg of dFe, northwest of P-W02). At this layer, from near the center of the plume (P-W02) to a distal location (P-W07), the relative reduction of dFe was 59%, which was lower than that of dMn (98%), thereby suggesting that dFe was removed more slowly than dMn when the plume dispersed laterally, which might have been related to the formation of small inorganic nanoparticles [16–18], complexation by organic ligands [19–21], and reversibility of exchange onto slowly sinking particles [12].





FIGURE 4: Turbidity contour and current velocity map of the PACMANUS hydrothermal field.





0.14

D-W06











FIGURE 5: Turbidity contour and current velocity map of the Desmos hydrothermal field.

3.7015°S

5.2. Distinguishing Fluid Sources by Their Chemical Components in Hydrothermal Plumes. The dissolved Mg, Cl, dFe, and dMn concentrations in the hydrothermal plumes above the PACMANUS field, near the Satanic Mills (station P-W02) and Fenway (station P-W04) vent sites, were generally higher than those in the plumes near the Desmos vent site (stations D-W01 and D-W04), whereas the SO_4^{2-} concentrations near the PACMANUS field vent sites were generally lower than those in the plumes immediately above the Desmos field vent site (Tables 1 and 2; Figures 2 and 3), thereby suggesting that the extent of dilution of the hydrothermal fluid with seawater was weaker in the PACMANUS field than in the Desmos field. These differences were also consistent with the differences in the dissolved Mg, Cl, dFe, dMn, and SO_4^{2-} concentrations of the vent fluids between the PACMANUS and Desmos fields [27, 43], in which the maximum SO_4^{2-} concentration (147 mmol/kg; [31]) of the vent fluids was observed in the Desmos field and the maximum dFe (13400 µmol/kg) and dMn (4540 µmol/kg) concentrations of the vent fluids were observed in the Fenway vent site of the PACMANUS field ([27, 29, 43]; Tables 1 and 2). This showed that variations in the major components of anomalous layers in the seawater columns in the Manus Basin are controlled by chemical variations in the intensity and position of the sources of the vent fluids [27, 43]. All these findings suggest that the fluid sources of hydrothermal plumes in the PACMANUS field have the characteristics of higher dissolved Cl, dFe, and dMn and lower SO₄²⁻ abundance compared with those of the Desmos field and imply that the difference in the plume chemistry may reflect the differences in the vent fluid chemistry.

5.3. Dispersion of Hydrothermal Plumes. The physical and chemical properties of seawater columns in the Manus Basin are affected by the input of high-Mn, high-Fe, and low-Mg vent fluids. The turbidity (up to 0.484 FTU and 0.292 FTU), dFe (up to 0.133 μ mol/kg and 0.084 μ mol/kg), and dMn (up to 0.539 µmol/kg and 0.158 µmol/kg) concentrations in the hydrothermal plumes near the Satanic Mills (P-W02), Fenway (P-W04), and Desmos (D-W01 and D-W04) vent sites were generally higher than those of the P-W07 and D-W09 plumes, while the dissolved Mg concentration (49.5 mmol/kg and 49.4 mmol/kg) was generally lower than those of the P-W07 and D-W09 plumes far from the vent sites in the PACMANUS and Desmos fields (Tables 1 and 2), thereby implying that the P-W04, P-W02, D-W01, and D-W04 plumes are located near their vent fluid sources and are relatively young and the P-W07 and D-W09 plumes are located far from fluid sources and are older (Figures 1(c) and 1(d)). These patterns were also consistent with the differences in the dMn and dFe concentrations of the vent fluids between the PACMANUS (1840-4221 µmol/kg of Mn; 648.8-6468 µmol/kg of Fe) and Desmos (Mn 40–110 μ mol/kg of Mn; 10–12400 μ mol/kg of Fe) fields ([27, 43]; Tables 1 and 2), thereby implying that the high-Mn content in the vent fluids and the low bottom current velocity led to the higher Mn concentration in the plume than that in the ambient seawater even after dilution. However, the maximum dMn concentration (approximately 110 nmol/kg) of the Desmos hydrothermal plume was lower than that of the PACMANUS plume, which was caused by the high-temperature (up to 358° C) hydrothermal activity that discharges mostly Mn-rich fluids and the lower bottom current (<0.18 m/s) in the PACMANUS field (Figure 4) [28, 31, 36], and higher than that (18.2–45.5 nmol/kg) of the Lau Basin [22].

However, the turbidity contour maps revealed two distinct anomaly centers corresponding to the Satanic Mills site (P-W02) and Fenway site (P-W04), and the two plumes intersected with each other at a depth of 1625 m to 1550 m (Figures 4(a)-4(d)). At a depth of 1625 m and 1620 m, the mixed plume showed a southeastward dispersion trend controlled primarily by the current at P-W01, and the current velocities of the Satanic Mills plume and Fenway plume were too low to affect the dispersion (Figures 4(a), 4(b), and 4(f)). At a depth of 1605 m, the Satanic Mills plume was affected by a significant northwestward current, which caused the mixed plume to extend along the northwest direction (Figures 4(c)and 4(f)). At a depth of 1550 m, the southeastward current at P-W07 obstructed the northwestward dispersion of the Satanic Mills plume (Figures 4(d) and 4(f)). At a depth of 1525 m, the Satanic Mills plume disappeared and the Fenway plume seemed to be influenced by a nearby southeastward current (Figure 4(e)).

The turbidity anomalies observed at P-W01, which is located between P-W02 and P-W04, were the result of the intersection of the Satanic Mills plume and Fenway plume at 1625 m, 1620 m, and 1605 m. At a depth of 1625 m, approximately 31% of the Satanic Mills plume intersected with 69% of the Fenway plume according to the turbidity of these plumes (0.329 NTU at P-W02, 0.254 NTU at P-W01, and 0.220 NTU at P-W04). At a depth of 1620 m, approximately 78% of the Satanic Mills plume intersected with 22% of the Fenway plume (0.326 NTU at P-W02, 0.300 NTU at P-W01, and 0.208 NTU at P-W04), and at a depth of 1605 m, approximately 0.7% of the Satanic Mills plume intersected with 99.3% of the Fenway plume (0.365 NTU at P-W02, 0.339 NTU at P-W01, and 0.337 NTU at P-W04). The contribution of the Fenway plume to the turbidity at P-W01 decreased from 1625 m to 1620 m and then increased, which was consistent with the turbidity variation of the Fenway plume, thereby suggesting that the influences on the turbidity at P-W01 by the Fenway plume were more notable than those by the Satanic Mills plume. This might have been due to the shorter distance between P-W01 and the Fenway plume than that between P-W01 and the Satanic Mills plume.

6. Conclusions

The hydrothermal plumes were distinguished by higher turbidity and dFe and dMn concentrations in the seawater columns, a rise of approximately 140–220 m, and lateral dispersal of hundreds of meters in the PACMANUS and Desmos hydrothermal fields. The Satanic Mills plume and Fenway plume were identified at a depth between 1625 m and 1525 m with maximum dMn concentrations of 0.533 μ mol/kg and 0.539 μ mol/kg and maximum dFe concentrations of $0.058 \,\mu \text{mol/kg}$ and $0.133 \,\mu \text{mol/kg}$, respectively. Although the dilution of dFe was one order of magnitude greater than that of dMn, dFe was removed more slowly than dMn during the lateral plume dispersal. The Satanic Mills plume and Fenway plume intersected with each other at a depth deeper than 1550 m, and the direction of dispersion of the mixed plume was affected by the nearby current.

Turbidity, dFe, and dMn anomalies higher than those in the other layers were observed in the seawater columns of the PACMANUS and Desmos fields, all of which indicated high-Mn and high-Fe fluid input. However, the turbidity and dFe and dMn concentrations of the plumes in the PACMANUS field were higher than those in the Desmos plumes, thereby implying that the influence of the Fe and Mn concentrations in the vent fluids on the hydrothermal plumes in the PAC-MANUS field was stronger than that in the Desmos field.

Compared with those of hydrothermal plumes in the Desmos field, the vent fluid sources of plumes in the PAC-MANUS field had the characteristics of higher dissolved Cl, dFe, and dMn and lower $SO_4^{2^-}$ abundance, and the plumes near the vent fluid sources were younger than those far from the fluid sources.

Data Availability

The underlying data related to our submission is available on request from zgzeng@ms.qdio.ac.cn.

Additional Points

Highlights. Dissolved Fe is removed more slowly than dissolved Mn during lateral plume dispersal. There is a clear visible intersection of plumes in the PACMANUS hydrothermal field. The influence of Fe and Mn in vent fluids on plume is weakened as the distance from the vent increased. Varied dispersion trends of the mixed plumes are affected by current velocities.

Conflicts of Interest

The authors declare that they have no conflicts of interest.

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