1	Platinum demand and potential bottlenecks in the global green transition: A
2	dynamic material flow analysis
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# 9 ABSTRACT

10 Platinum, as a key catalytic material, is important for the global green transition due both to its 11 current main use in autocatalysts and its increasing use in emerging and renewable energy 12 technologies such as fuel cells and electrolysers. In this study, we developed a dynamic material 13 flow analysis model to characterize the global platinum cycle between 1975 and 2016 and to 14 develop scenarios for future global platinum demand to 2050. Our results show that the 15 autocatalyst and jewellery uses represent the most primary platinum use and possess the highest 16 platinum stocks in use by 2016; however, when closed loop recycling is considered, the gross 17 platinum demand from the glass industry would be the largest. Many socioeconomic (e.g., 18 population and car ownership) and technological (e.g., engine and energy technologies) factors 19 will affect the future demand for platinum in a global green transition. Our analysis concludes 20 that, only in high demand scenarios and when fuel cell market penetration is high compared to 21 the expected, the aggregate demand to 2050 will exceed the 2016 global platinum reserves. 22 Improving the end-of-life collection and recycling rates would be important to address potential

- 23 future supply risks due to geopolitical reasons. These demand scenarios and further mapping of
- 24 the global platinum value chain can help inform government and industry policies on
- 25 transportation and energy transition, platinum supply risk mitigation, and recycling capacity
- 26 planning and technology development.
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- 29
- 30 **TOC**



### 40 1. INTRODUCTION

41 The post-Paris Agreement era continues to face a daunting and urgent dual challenge of 42 satisfying growing societal demand (e.g., transportation) and significantly curbing greenhouse 43 gas (GHG) emissions. The emerging and renewable energy technologies, such as wind turbines, 44 solar panels, and battery and fuel cell vehicles, are widely recognized as necessary for the global green transition in a climate-constrained world<sup>1</sup>. However, such a transition requires a large 45 46 number of critical and precious metals, such as rare earth elements (REEs), lithium<sup>2</sup>, cobalt<sup>3</sup>, and 47 platinum group metals (PGMs)<sup>4</sup> amongst others. This consequently raises questions of resource availability issues for large scale implementation of these technologies<sup>5</sup> due to the highly 48 49 concentrated geographical distribution<sup>6</sup> of these critical materials, lack of effective substitutes<sup>4</sup>, and political instability in some producing countries<sup>7</sup>. Therefore, an understanding of future 50 51 demand and potential supply bottlenecks for these critical materials will be important for 52 government and industry decision making on sustainable transportation and energy transition, 53 supply risk mitigation, and technology development in the long term. 54

Platinum is an example of a critical<sup>8</sup>, yet unevenly distributed (91.3 % of reserves in South 55 Africa<sup>9</sup>) metal, for the global green transition. Platinum is used widely in a variety of industrial, 56 57 chemical, medical, and environmental applications that are indispensable for our sustainable 58 societal transition. The transportation sector, in particular, as one of the key intervention but 59 difficult sectors to develop climate-friendly technologies due to the fuel energy density it 60 requires, relies significantly on platinum. At present, the majority of internal combustion engine 61 (ICE) vehicles need platinum-based autocatalysts to convert the harmful chemicals released from 62 the engine exhaust into less detrimental products such as carbon dioxide, molecular nitrogen, and

63	water vapour. In the future, sustainable and renewable fuels, in parallel with efficient battery
64	technologies, will be key <sup>10</sup> for a low carbon transportation transition (especially for long-haul
65	and heavy-duty vehicles, ships, and aviation where batteries cannot cover the required range),
66	and production of these fuels often requires platinum as a catalyst. For example, polymer
67	electrolyte membrane electrolysers (PEMEL) for hydrogen and further electrofuel (mixtures of
68	hydrogen from electrolysis and carbon from sustainable sources such as biogas or the air to form
69	a gas or liquid fuel) production and proton-exchange membrane fuel cells (PEMFC) currently
70	depend significantly on platinum as a catalytic material <sup>6,11</sup> .

71

Several governments have already announced plans to ban ICE vehicles<sup>12</sup>, indicating a 72 significant societal commitment towards gradually phasing them out over the coming decades. 73 74 Meanwhile, it is expected that the demand for fuel cell vehicles and other new applications for platinum (e.g., in electrolysers) will increase in a green transition. It is, therefore, important to 75 76 understand the implications of such a transition for platinum demand, recycling opportunities 77 and challenges, and related supply risk and environmental impact mitigation.

78

79 Earlier studies have characterized the platinum cycle and demand-supply balance at different 80 scales and from various points of view. Most of these studies focused on the characterization of 81 the historical and/or current platinum cycle and especially its end use in different sectors, at the country (e.g., Germany<sup>13</sup> and Japan<sup>6</sup>), regional (e.g., EU<sup>14</sup>), or global<sup>4</sup> scale. These studies are 82 based mainly on a material flow analysis (MFA) approach, but other methods such as input-83 output analysis<sup>15</sup> and system dynamics<sup>16</sup> were used as well. They provided an overview of the 84 85 platinum flow and efficiency throughout our economy. A few studies have further modeled the

future demand of platinum in major end use sectors based on scenarios and compared the availability of PGM reserves with the automotive sector's demand<sup>17,18</sup> or the increasing demand of fuel cells in the future<sup>17,18,20,21,22</sup>. However, there are still two major knowledge gaps that need to be addressed.

- Most studies did not consider the varying end of life (EoL) management and recycling
   routes (e.g., open or closed loop recycling) of different platinum end-use products in
   detail (except very few static or country-level studies<sup>13,4</sup>), which would have a significant
   impact on the platinum cycle<sup>13</sup> and future primary platinum demand.
- The future platinum cycle and demand scenarios developed in the literature often do not consider in details the temporal dynamics of market transformation and technology penetration for all end use sectors. In particular, the assumptions on fuel cell penetration and technology efficiency (platinum intensity) are often outdated<sup>21,22</sup>, and the increasing demand of electrolysers towards global renewable energy transition has seldom been considered in previous studies.
- 100

101 The main reason that previous studies excluded electrolysis is that the only commercial type of 102 alkaline electrolysis does not use platinum. However, a shift towards polymer electrolyte 103 membrane (PEM) electrolysis is expected<sup>11</sup> (for example, it is better suited for flexible operation 104 time), and it is, therefore, necessary to include electrolysers in future scenarios. Electrolysers will 105 be of high importance in a future energy system reliant on hydrogen, the source of which in a 106 renewable system is limited to electrolysers and biomass. Hydrogen use is not limited to fuel 107 cells, if the hydrogen is mixed with CO<sub>2</sub> from biogas, cement kilns, or the air to form methane.

108 Methane can, find use in gas motors to back up the electricity grid, or as feedstock in a gas to 109 liquid plant and then as a fuel in transportation in the engines that we have today.

110

111 Robust platinum demand scenarios would help understand the quantity, quality, and dynamics of scrap availability, and thus opportunities and barriers for recycling<sup>23</sup> (e.g., investment in 112 recycling infrastructure and technologies<sup>24</sup> based on a different framework condition of platinum 113 cycle in the future) and consequent environmental implications<sup>25,14</sup>. For example, Alonso et al. 114 115 analyzed declining ore grades and the environmental implications of mining PGMs<sup>17</sup> based on 116 mass flow scenarios. These scenario results, together with consideration of the geopolitical 117 conditions<sup>4</sup> and opportunities for future additional primary production, will help identify supply 118 risk mitigation strategies and plan future recycling capacities.

119

The aim of this study is to address the abovementioned knowledge gaps using a dynamic MFA model, for the gross demand of platinum in a global green transition up to 2050. All end-use sectors have been considered (including fuel cells and electrolysers) in the future demand scenarios and the EoL management and recycling routes have been investigated in details. We have then discussed the implications of the consequent demand in terms of potential supply constraints and opportunities and challenges of recycling.

### 127 2. MATERIALS AND METHOD

### 128 **2.1 System definition**

The system definition of the global platinum cycle is shown in Figure 1. The major life cycle
stages of the platinum cycle include: extraction and production; processing and manufacturing;
use; and end of life management.

132

133 Platinum can be produced via either primary (Processes 1, 2, and 3) or secondary (Processes 20, 134 21, and 23) routes. After production, platinum in the form of powder and bars is used in different 135 applications, after undergoing the stages of manufacturing of intermediate or final products 136 (Processes 6-10) and then the use stage (Processes 11-19). The largest current market demand for 137 platinum comes from autocatalysts (Process 16) and jewellery (Process 19). Large gross demand 138 comes from the glass industry as well, where platinum is used to coat ovens and other equipment for glass production<sup>26</sup>. Two sectors with a potentially large demand for platinum in the future are 139 140 electrolysers (Process 14) and fuel cells (Process 15; only fuel cell vehicles are considered, as we 141 assume that stationary and portable fuel cell power will have a negligible role in future energy 142 systems due to efficiency and cost). Electrolysers produce hydrogen powered by electricity, 143 which in the most sustainable way is generated from renewable sources (e.g., wind power). A 144 fuel cell is the reverse technology of an electrolyser, which consumes hydrogen to produce 145 electricity and water. The investment sector (Process 5) refers to platinum demand in the 146 financial sector and commodity market, through purchase of bars and coins, and is best 147 considered as storage of platinum without loss.

The last phase is end of life management where the platinum is either returned to the market, for use in a manufacturing process, or lost to landfills or the environment (Processes 20-23). Two distinct platinum recycling routes were differentiated:

In closed loop recycling, the platinum-containing product (and the embodied platinum) is
 owned by the industry through all life stages<sup>27</sup>. This means that, in general, the same amount
 of recycled platinum as was delivered to end of life management are returned to the
 manufacturing processes within the same sector, minus very small losses in refining.
 Currently platinum recycling from the glass, petroleum refining, chemical production, and

157 electrolysers sectors are examples of closed loop recycling.

In open loop recycling, the recycled platinum is sold on the global market, and may
 subsequently be used in any sectors that use platinum. Currently, the jewellery, electrical
 products, autocatalysts and fuel cells in transportation units, and all other applications are all
 recycled in open loops.

162

163 Losses of platinum occur throughout the entire life cycle starting from extraction and production. For losses in manufacturing and use, we assumed (based on literature<sup>4,28,29</sup>) that manufacturing 164 165 losses are considered only for jewelry, electrical, and other products, because platinum in these 166 sectors is handled in a hard solid form, and losses in the use stage (e.g., dissipation) are mainly 167 found in the catalyst based products and the glass industry. The losses in end of life management 168 vary depending on the end-use sector and the stage in the recycling chain. The largest losses in 169 end of life management are in the open loop recycling end-use sectors where collection (e.g., of 170 electrical products) and the existence of less professional operators (e.g., in autocatalyst pre-

- 171 processing/decanning) can be a challenge. While losses in the final stage of the recycling chain,
- 172 refining, are very low and already near thermodynamic limits.



Figure 1. System definition of the global anthropogenic platinum cycle. Methods to derive
different stocks and flows are visualized by colors of arrows and stock boxes, which is further
detailed below and in the in the Supporting Information (SI). The processes are numbered (1-23)

- so that it is easier to refer to flows using symbols; for example, F<sub>6-11</sub> represents the flow from
  Process 6 (coating ovens) to Process 11 (glass production).
- 179

# 180 **2.2 Simulation of historical stocks and flows**

We used a dynamic MFA approach<sup>30</sup> for quantifying the global historical platinum cycle from 181 182 1975 to 2016. In such an approach, the stock in use and waste flow generation are simulated by 183 the historical apparent consumption of platinum by end-use sector and their corresponding 184 lifetime (e.g., waste streams from use phase to end of life management, see Equation (1)). The 185 other upstream and downstream flows are subsequently calculated by either transfer coefficients 186 (e.g., all loss flows), statistics (e.g., all flows from the market to manufacturing), or the mass 187 balance principle (e.g., all flows from manufacturing to use phase), as detailed in Figure 1 for an 188 example of the glass production (Process 11) sector (the same applies for all other sectors; see 189 details in Table S1 in the SI).

190 
$$F_{11-20,n} = \sum_{i=n-2\mu}^{n} (F_{6-11,i \text{ to } n} - F_{11-0,i \text{ to } n}) * \left(\frac{1}{\sigma * \sqrt{2 * \pi}} * e^{-\left(\frac{1}{2}\right) * \left(\frac{X-\mu}{\sigma}\right)^2}\right)$$
(1)

191 in which  $F_{11-20,n}$  represents the waste flow generation, n is the year in focus,  $\sigma$  is the standard 192 deviation,  $\mu$  is the mean lifetime of the product, and X represents age of the cohort.

193

The primary source of platinum consumption data is the market purchase estimates by sector reported by Johnson Matthey<sup>31</sup>. These data are directly used for the open loop recycling sectors as their corresponding apparent consumption, whereas they are further adjusted by adding up simulated internal recycling flows as the apparent consumption of the closed loop recycling sectors.

200 We assumed the platinum stock in use before 1975 is negligible, due to data gaps. The impact of 201 this assumption on current stocks and flows in 2016 is limited because the lifetime of many 202 platinum end-use products is shorter than 20 years. The impact on jewellery (with a lifetime of 203 35 years or more) and close loop recycling sectors is higher (especially for jewellery stock 204 estimation in the 1970s and 1980s) than that of the open loop recycling sectors, because the 205 platinum input that occurred 40 years ago may still be recycled within closed loop sectors with 206 high recycling rates. 207 The lifetimes for each end-use sector<sup>13,4</sup> and transfer coefficients of the manufacturing, 208 collection, recycling, concentration, smelting, and refining processes <sup>13,4,28,32,33,34,29</sup> are based on 209 210 literature and detailed in the SI. 211 212 2.3 Scenario setting and simulation of future stocks and flows 213 A prospective top-down, stock-driven approach is used to quantify the future global platinum

cycle. In other words, we model the future demand of platinum by end-use sector and other flows
(e.g., recycling, losses, production, and extraction) based on the services provided by their
corresponding in-use stocks, which are assumed based on their historical patterns and the need of

a global green transition. Sensitivity analysis was also conducted to evaluate the impact of key

218 model parameters, in addition to the various scenarios developed.

219

The levels of detail for the future stock growth assumptions vary by sector, based on the relative importance of that sector in the total use and our understanding of influencing parameters. We have grouped them into three categories as shown in Figure 2 and explained below.



225 Figure 2. Conceptual model framework for the stock driven approach to quantify the 226 future global platinum cycle, with details on stock estimations and assumptions made for 227 the three different sectors (transportation, electrolysers, and all other sectors) and how 228 they are used for quantification of recycling, collection, production, and losses. Individual 229 graph legends: a) Platinum content over time in fuel cell, gasoline, and diesel cars with fast, 230 medium, and slow technological development; b) Market shares of different car engines over 231 time in the fuel cell scenario; c) The fast, medium, and slow development of global car

ownership; d) Platinum content in electrolysers over time; e) Estimated installed capacity of
electrolysers by 2050 in the 16 Danish energy scenarios (details in the SI), with red, yellow, and
green boxes representing our high, medium, and low scenarios; and f) Example of jewellery
future platinum stock on an aggregated per capita level in high, medium and low scenarios.

236

### 237 2.3.1 Transportation

238 The transportation sector is modelled in a detailed bottom-up approach. Autocatalyst currently 239 represents the largest end use of platinum, and, together with fuel cells, the transportation sector 240 is likely to remain the prominent use in the future. The factors included in the platinum stock 241 (f(x), x for a specific year) assumptions of the transportation sector are vehicle ownership per 242 thousand persons (VO), from the International Organization of Motor Vehicle Manufacturers 243 (2016) (OICA), population forecast (P) from the United Nations (2017), platinum intensity (PtI) 244 in all type of vehicles, and market penetration (MP) of different types of vehicles, as shown in 245 Equation (2):

246 
$$f(x) = (VO_x * P_x) * (PtI_{diesel,x} * MP_{diesel,x}) + (VO_x * P_x) * (PtI_{gasoline,x} * MP_{gasoline,x})$$
247 
$$+ (VO_x * P_x) * (PtI_{fuel cell,x} * MP_{fuel cell,x})$$
(2)

248

The current platinum intensity in diesel and gasoline cars is calculated based on the PGM content of autocatalysts<sup>37</sup> and the platinum content of PGMs<sup>38</sup> (ending up as 3.75 grams and 0.32 grams, respectively, per diesel and gasoline car). The future platinum intensity in cars depends on future emissions regulation and price of PGMs, and varies from 1.5 to 3 grams and from 0.1 to 0.3 grams in 2050 in our scenarios per diesel and gasoline car, respectively. The platinum content in a single fuel cell is assumed to be 20 grams per car at present<sup>39</sup>, which is representative of the

- lowest levels demonstrated today. Depending on which future scenario may prevail, this mayvary from 0 to 8 grams in 2050 in our scenarios (see figure 2a).
- 257
- 258 We assume that the global average vehicle ownership would follow an S-curve and saturate at
- the current level of Europe (471 cars/1000 persons) by 2050 as the baseline scenario. A  $\pm$  50%
- 260 difference is considered in the high (706 cars/1000 persons, close to the current level of USA)
- and low (235 cars /1000 persons) saturation level scenarios at 2050 (see figure 2c).
- 262
- 263 The market penetration of different vehicle engine types is key to platinum demand scenarios.
- We simulated the market penetration as an S-curve from 2016 to 2050 and have considered three
- scenarios for the year 2050 market share, inspired by the International Energy Agency<sup>40,1,41,42</sup>, 16
- 266 Danish energy scenarios (detailed in the SI), and a study conducted by Precious Metal Watch<sup>37</sup>.
- 267 i. A Reference (REF) scenario assuming that the car market share remains the same as of
  268 today (20% diesel, 79% gasoline, and 1% battery cars);
- 269 ii. A fuel cell scenario assumed a high share of 30% for fuel cell cars (the remaining: 5%
- diesel, 45% gasoline, and 20% battery cars); and
- 271 iii. A battery scenario with 80% of battery cars in the market (the remaining: 5% diesel and
  272 15% gasoline cars). The fuel cell scenario is shown as an example in Figure 2b.
- 273

## 274 2.3.2 Electrolysers

- Figure 2d shows the development of platinum intensity in electrolysers. The starting point in
- 276 2016 is set as 1 kg/MW installed capacity, which among others<sup>43</sup> is based on a real world project

"DuraPEM"<sup>44</sup>. The ending point in 2050 on 0.25 kg/MW is estimated based on an industry
report<sup>45</sup> and two journal articles<sup>46,47</sup> on the future technology prospect.

279

The demand for electrolysers depends fundamentally on the future energy system. There are currently no global demand forecasts for the capacity of electrolysers. Denmark has undertaken however many detailed analysis of the paths towards a full sustainable energy transition, which takes into consideration the role of electrolysers, and is therefore chosen as the reference point for predicting the future global demand for electrolysers.

285

286 The installed capacity of electrolysers to 2050, as predicted by 16 different Danish energy 287 system analyses, forms the basis (see Figure 2e and details in the SI) for our high, medium, and 288 low scenarios (red, yellow, and green bars, respectively, in Figure 2e). We have extrapolated the 289 Danish demand of platinum for electrolysers to all other countries and derived a global total, based on a factor of population (2016 to 2050, from the United Nations forecast)<sup>36</sup> and a scaling 290 indicator of technology readiness index of a country (from the World Bank)<sup>48</sup>. This index 291 292 consists of, for example, companies spending on R&D, the creativity of its scientific community, 293 and personal computer and internet penetration rates. We assumed that the relationship between 294 each country and Denmark's technology index level stays the same from 2005 to 2050. Such an 295 extrapolated global total was deemed a good proxy, given the data gaps. Also, the Danish 296 electrolyser forecasts are based on energy scenarios that compare the need for electrolysis with 297 the need for biomass input with a global limit. When compared with the scenario in the 298 Hydrogen Roadmap Europe<sup>49</sup>, which estimates an electrolyser installed capacity of about

800,000 MW (assuming a capacity factor of 50 % and an efficiency of 65 %), our extrapolated
European total capacity falls in the same range (about a quarter higher).

301

## 302 **2.3.3** All other sectors except transportation and electrolysers

303 The platinum use in all other sectors, including the chemical, glass, petroleum refining,

304 investment, medical, electrical, and other sectors, is considered on an aggregated level as per

305 capita stock of each sector. We assumed a medium 2050 saturation level for the global per capita

306 stock of each sector based on their historical patterns (with  $\pm 30\%$  as high and low scenarios) and

307 modelled the stock growth from 2016 to 2050 as an S-curve. The global population forecast is

308 based on the United Nations projections<sup>36</sup>. More details are provided in the SI.

309

### 310 **2.4 Reserves and supply concentration**

311 Global reserves of PGMs, defined as that part of the total resources in the Earth's crust that are economically and technically feasible to extract with prices and technology at the time of 312 313 determination,<sup>50</sup> is estimated by United States Geological Survey (USGS) to be around 69,000 tonnes of PGMs equivalent in 2016. Nassar et al <sup>51</sup> have collected platinum content data for 21 314 315 PGM mines (detailed in Figure S1 in the SI), which varies from 37.1% to 57.6% with an average 316 of 51.3% This equates to a global platinum reserve from 25,599 tonnes to 40,917 tonnes with an 317 average of 35,379 tonnes in 2016, based on the assumption that those 21 mines are representative 318 of the global platinum reserves in terms of PGMs ore grade.

320 We have further applied the widely used Herfindahl–Hirschman Index (HHI) to determine the 321 production concentration based on the percentages of production/refining per country in the 322 world's total  $(x_i)$ , as shown in Equation (3): (3)

323

 $HHI = x_1^2 + x_2^2 + x_3^2 \dots + x_n^2$ 

Recycling is important for platinum supply. Therefore we have included recycling data<sup>31</sup> in the 324 325 HHI calculation, which means open loop recycling flows that usually take place close to the 326 demand, were subtracted.

327

#### **3. RESULTS AND DISCUSSION** 328

#### 329 3.1 The historical and present global platinum cycle.

330 Figure 3 shows results of the global stocks and flows of platinum in 2016. It is shown that the 331 autocatalyst sector uses the most primary platinum (103.6 tonnes), which is not surprising, 332 because almost all new cars with combustion engines have an autocatalyst. However, when 333 closed loop recycling is taken into account, the total secondary platinum supply exceeds the 334 primary supply, and the glass industry would have the largest gross use of platinum (105.1 335 tonnes). It should be noted that these large closed loop recycling flows are usually not shown in 336 the industry's market statistics (e.g., from Johnson Matthey), because such platinum flows are 337 not counted as market transactions. However, considering their significant size and relatively 338 short lifetime, it is important to understand these flows in order to better characterize future 339 platinum demand and in particular their consequent economic and environmental implications 340 (e.g., short lifetime and thus rapid circulation of platinum in closed loop recycling may result in 341 more energy use for maintaining the same amount of platinum products in use).

342





- gross demand (including close loop recycling flows) for platinum by sector, and (b) losses of
- platinum by life cycle stage and (c) by end-use sector. The "stock change" is the balance
- difference of the market process.

350 Due to increasing levels of global car ownership, there are 1,364 tonnes of platinum accumulated 351 in the current global car fleet in 2016. The jewellery sector, unexpectedly, was simulated to have 352 the largest stock of platinum (1,865 tonnes) mainly because of their long lifetime (on average 35 353 years<sup>4</sup>). The investment and glass industry follow with the third (330 tonnes) and fourth (209 354 tonnes) largest platinum stock in use, though at a much lower level. All sectors, except the 355 electrical product sector, show an increasing trend in per capita platinum stock since 2006. The 356 declining trend in platinum per capita in the electrical product sector may be related to the 357 general decline in the content of other precious metals in electrical products.<sup>52</sup>

358

359 The total loss of platinum in the whole life cycle added up to 111.9 tonnes in 2016, or around 360 half of the global amount mined (231.5 tonnes). After ore processing to produce a Pt concentrate 361 (32% of total loss), the second largest loss (29% of total) is found at the end of life collection, in 362 which the transportation (autocatalysts) sector contributed the most. A comparison of our 363 modelled results with historical data on the quantity of platinum recycled from autocatalysts (see 364 SI Table S8 and S9 for details) showed that collection rate of 70% provided a good fit between 365 the modelled and historical data. This collection rate reflects the fact that EoL cars may be 366 exported to less developed countries, where proper or sufficient infrastructure and systems are not yet available<sup>29</sup> and that these flows are not easily tracked and recorded in market statistics. 367 368 The manufacturing and use phases together contribute to the third largest loss (17% of total), 369 which is in line with the results of an earlier study focusing on anthropogenic losses of platinum 370 in 2010<sup>53</sup>. All losses occuring in EoL management are relatively small compared to the overall 371 use of platinum, largely due to the high value of platinum, which incentivizes high recycling 372 rates. Even with the relatively high collection and recycling rate the accumulative loss from 1975

373 to 2016 from collection and recycling in the autocatalysts sector adds up to more than 450 374 tonnes, which is the highest of all other sectors (around three times higher than the second largest 375 sector, jewellery). Our results for the 2016 global platinum cycle in Figure 2 are generally in line 376 with the 2010 global platinum cycle modelled by Nedal<sup>4</sup>. Both studies concluded that the closed 377 loop recycling flows are almost as big as the primary use of platinum and that the autocatalyst 378 and jewellery sectors have the largest in-use stocks of platinum. When compared with a study for the 2004 European PGM cycle<sup>14</sup>, the relative relationships among flows and sectors (for the 379 same year 2004) are also in good agreement. Another study<sup>54</sup> describing the European platinum 380 381 cycle for the year 2012 simulates the European net addition to stock at 12 tonnes and platinum 382 stock at 710 tonnes, giving a ratio of 1.7%. This is lower than our result (3%, with net addition to 383 stock at 111 tones and total stock at 3,753 tonnes in 2012) for the global platinum cycle. Such a 384 lower ratio on the global level is expected, because many developing countries are still building 385 up stocks, whereas Europe already has mature stocks in many sectors. Moreover, the referenced study<sup>54</sup> showed that 20% of platinum input (from both primary and open loop recycling) comes 386 387 from post-consumer functional recycling; whereas this percentage in our global model, without 388 considering closed loop recycling, is 26%. We presumed this indicates the significant role of 389 trade on a regional/country level.

390

### **391 3.2 Primary global demand of platinum and global platinum cycle by 2050.**

392 Figure 4 presents how global aggregated values of primary demand for platinum are influenced 393 by different key parameters, in the REF, fuel cell (the high demand scenario), and battery (the 394 low demand scenario) scenarios. The thick green line in the middle of each scenario shows the 395 starting point, where all parameters are set to medium. The first line above represents high

396 population and the first line below represents low population, both with all other parameters set 397 to medium. By changing one parameter after another, continuing upwards, we move towards the 398 most aggressive case in all three scenarios and continuing downwards towards the least



399 aggressive case, to show the potential range of future platinum demand.

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400

Low numbers of cars per person



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Figure 4. Accumulative global primary demand for platinum by 2050, with absolute 402 403 sensitivities when changing one parameter after another in the three scenarios a) REF, b) 404 fuel cell, and c) battery, d) a comparison of all three scenarios with the same Y-axis, and e) 405 the global platinum cycle with accumulated flows from 2017 to 2050 and stocks in 2050 in 406 the case of middle fuel cell scenario. In a), b), and c) the thick green line in the middle 407 represents the aggregated demand, with all parameters set to medium. The first line above shows 408 the result when the parameter population is set to high; the second line above shows the result 409 when the two parameters, population (the parameter that was changed in the first line) and per 410 capita stock of platinum in other sectors than transportation and electrolysers, are set to high; and 411 so forth. This allows us to see the combined effect of changing several parameters. The three

(high, average and low) estimated global reserves of platinum in 2016 (see definition in section
2.4 and discussion in section 3.3.1) are only in the high demand fuel cell scenarios exceeded by
the demand.

415

416 The total aggregated demand for platinum from 2016 to 2050 varies between 5 kt (the least 417 aggressive battery scenario) to 51.4 kt (the most aggressive fuel cell scenario). The medium 418 cases in the three scenarios are 9.8 kt for the battery scenario, 13.1 kt for the REF scenario, and 419 24.6 kt for the fuel cell scenario. The main demand in the medium REF scenario is found in 420 autocatalysts, followed closely by electrolysers which together account for above 63% of the 421 primary demand. In the medium battery scenario, this percentage falls to 52% due to a lower 422 demand for autocatalyst. The main demand in the medium fuel cell scenario is not surprisingly fuel cells, which represents 58% of the primary demand (more than three time as much as 423 424 electrolysers).

425

426 It can be seen that population change affects all sectors in all the three scenarios (since it is the 427 first parameter we set to change in different scenarios), but to what extent the other parameters 428 affect the demand of platinum in different sectors varies in different scenarios. For example, the 429 impact of changing car ownership and platinum intensity in cars (due to engine technology 430 improvements) is larger in the fuel cell scenarios than in the battery scenarios, because fuel cells 431 in 2016 have a much higher platinum intensity than gasoline, diesel, and battery cars. It should 432 be noted that, if we change the order of parameter change in the scenarios in Figure 4, the 433 aggregated effects on platinum demand would still be the same but the contribution of each 434 parameter may be different.

436	The impact of two parameters, demand for electrolysers and per capita platinum stock in other
437	sectors than transportation and electrolysers, is the same in all the three scenarios. However,
438	there is a difference between changing the demand of electrolysers from medium to low than
439	from medium to high, since the parameter population is changed before the demand of
440	electrolysers. The percentage of demand increase is larger than that of demand decrease, because
441	of the population factor used in electrolysers demand calculation.
442	
443	Figure 3 shows that the current stock of platinum is 1364 tonnes in autocatalysts (the only
444	considerable use in the transportation sector in 2016) and this stock changes to between 300 and
445	6427 tonnes in our scenarios with global car ownership rising to the European 2016 levels in
446	2050 and the gradual change of engine technologies. For example, in the REF scenario in which
447	99% of the cars still run on gasoline and diesel, the stock of platinum in the transportation sector
448	will change to 800-6500 tonnes (due to technological development, population, and car
449	ownership increase). Such an increase is even more marked (2500-24400 tonnes) in all versions
450	of fuel cell scenarios, because the platinum demand increase due to increasing fuel cell
451	penetration will make up for, and eventually exceed, the decrease as a result of the phase-out of
452	gasoline and diesel engines. It can also be seen in Figure 4b that several of the scenarios has a
453	declining demand in the years close to 2050, which relates mainly to technological development
454	that decreases platinum content in fuel cells.

However, the introduction of electric cars may result in a major decrease of platinum demand. Inall versions of the battery scenarios (with 80% of electric cars), the platinum stock in the

transportation sector will decrease, with the lowest reaching around half (700 tonnes) of the
current level. This renders more secondary resources available from autocatalysts recycling for
use in electrolysers, whose demand in high (4000 tonnes), medium (2100 tonnes), and low (750
tonnes) scenarios is higher than the decrease in platinum used in combustion engine cars. It is
worth mentioning that, while the battery scenarios may not cause supply shortages for platinum,
they may do so for other metals, such as cobalt and lithium.<sup>55</sup>

464

465 Figure 4(e) shows, an example of the medium fuel cell scenario, an overview of the future global 466 platinum cycle with accumulated flows from 2017 to 2050 and stocks in 2050. When compared to 467 the current situation (Figure 3), the most notable changes are the large inflows into electrolysis 468 and fuel cells and their large stocks-in-use built up in the next decades. The stock of autocatalyst 469 and electrical products will decrease, as the accumulated outflows are larger than the accumulated 470 inflows. The autocatalyst sector has dropped from the largest consumption of primary platinum in 471 2016 to only the fourth after fuel cells, electrolysers, and jewellery in 2050. The largest loss occurs 472 in concentration just as in 2016, but the collection of fuel cells is now the sector with the second 473 largest loss (see Figure S12 in the SI).

474

The sensitivity analysis results with each key model parameter changed one at a time are shown for a case of the medium fuel cell scenario in Figures S14 and S15. The sensitivity analysis investigates the importance of each parameter on the accumulative primary demand. Figure S14 shows that lifetime has by far the largest influence on the accumulative demand ( $a \pm 10\%$  change leads to above 900 tonnes in the change of primary platinum demand). Extraction loss rate follows with the second largest impact, while other parameters such as manufacturing loss, use

481 phase loss, collection efficiency, and recycling loss have a very small effect on the final primary 482 demand, because these loss rates are already low at present. For those key parameters included in 483 the scenarios in Figure 4 (Figure S15), cars ownership and technological development in 484 transportation are found to have much larger impact on accumulative primary platinum demand 485 than other parameters such as population, per capita platinum stock in other sectors, and demand 486 of electrolysers in 2050.

487

### 488 **3.3 Discussion on potential bottleneck of platinum supply and mitigation strategies.**

**3.3.1 Geological availability.** When the global demand of platinum in different scenarios (see Figure 4) is compared with the 2016 estimation of reserves, it can be concluded that, except in the most aggressive fuel cell scenarios, the future platinum supply will not face geological constraints. Even if the lowest estimate on platinum reserves at 25,599 tonnes was used for comparison, it is only the demand of the 5 highest fuel cell scenarios (out of 11 fuel cell scenarios) that ends above the 2016 estimated reserves.

495

This comparison does not capture the dynamic nature of reserves, which vary in response to changes in many factors, e.g. economic (e.g., an increase in willingness to pay more for platinum makes other extraction methods and/or sources economically viable), technological (e.g., which facilitates new discoveries or mining at greater depths), and regulatory (e.g., the rezoning of land that makes mining possible or impossible).

501

502 The prediction of how reserves will change in the future is out of the scope of this study.

503 Nevertheless, we may consider how the future reserves might evolve based on the evolution of

504 historical reserves. One way would be to consider the historical relationship between annual 505 reserves change and production in the past two decades, is 0.75:1 from 1997 to 2016, meaning 506 that for every 4 tonnes of platinum is mined, the reserves increase by 3 tonnes of platinum. 507 Another way is to forecast is the future development of reserves follows the historical average 508 development on 138 tonnes reserve growth per year. In the second case, the global platinum 509 reserves will increase by 4700 tonnes by 2050 to 41000 tonnes (with the average PGMs ore 510 grade). This kind of evaluation of the dynamic nature of reserves is required for the long term. In 511 the short to medium term, however, capacity and co-production rates should be used as 512 indicators instead.

513

514 3.3.2 Geopolitical risks and supply constraints. Rather than geological availability, 515 geopolitical risks and price fluctuation are more likely to cause future constraints of global 516 platinum supply, especially in the short and medium term. Since South Africa hosts more than 517 90% of the presently known reserves, the geopolitical risk to supply constraints is considered 518 high. If calculated based on the percentages of production/refining per country (without 519 recycling), the HHI value of platinum is already above 5000, much higher than the criticality 520 threshold of 2500<sup>56</sup>. Supply shortages and consequent price peaks due to geopolitical risks have already been seen with cobalt, palladium, and rare earth metals in recent decades<sup>57</sup>. Platinum has 521 522 also faced similar threats in recent years, most notably with the striking of 70,000 South African 523 platinum workers in 2014-2015, which cut off ~40% of global platinum production.

524

525 There are additional factors which may add to the potential supply bottleneck, such as declining 526 ore grade (e.g., the mines in South Africa have experienced declining ore grade throughout their

527 lifetime<sup>17</sup>) and considerable share of co-production. About 18% of the platinum currently mined
528 is mined as co-production<sup>4</sup>, e.g. palladium and platinum are mined as co-products of nickel in
529 Russia.

530

3.3.3 Mitigation strategies and opportunities and challenges for substitution and recycling.
The consequence of a shortage in production of platinum may be temporarily mitigated by
depletion of stocks at the mines and in the investment sector. Long-term mitigation strategies
include exploration for new deposits to diversify the supply base, better use of the resource
potential of mine waste, improved co-production rates, better collection in the open loop
recycling sectors, spreading the use of best available practices and technologies in recycling,
substitution, and demand management.

538

539 The most likely substitutes for platinum in many applications are metals with similar properties 540 to platinum, in particular the other five platinum group metals (ruthenium, rhodium, palladium, 541 osmium, and iridium), nickel, cobalt, and gold. However, all other PGMs are already used in almost all platinum containing end uses<sup>58</sup>, making any further substitution hard<sup>4</sup>. For example, 542 iridium can substitute platinum in electrolysis, but iridium is also a very limited resource<sup>59</sup>; and 543 544 the same is for palladium use in autocatalyst. Moreover, these other five PGMs are often mined 545 as co-products of platinum, which in the event of a geopolitical supply disruption makes these metals insufficient as substitutes for platinum<sup>4</sup>. Using metals outside the PGM group is possible, 546 547 but there are usually either negative technical or economic trade-offs<sup>4</sup>. Substitution can also 548 happen on the component level (e.g., Polymer Electrolyte Membrane electrolysis is replaced by

platinum free Solid Oxide Electrolysis Cells or Alkaline ones) and product level (e.g., fuel cell
cars are replaced by battery cars), which deserves closer investigation in the future.

551

552 Recycling is also an important strategy for addressing supply constraints in the long run, as it 553 generally occurs where the products are used and technologies are available (not limited to 554 geological distribution). For example, when recycling is included, the current HHI index of platinum is very close to 2500, making it borderline critical<sup>60</sup>. As platinum demand and stocks 555 556 continue to increase in the following decades, platinum scrap is expected to become more 557 available, as shown in our scenarios, indicating increasing opportunities for recycling, especially 558 in industrialized countries such as Europe and the U.S. Another advantage of recycling is the 559 environmental benefit, since the emission intensity from recycling platinum is substantially 560 lower than that from primary production.

561

562 However, several challenges are foreseen for platinum recycling in the global green transition. 563 First, the leakages of EoL products to countries with less robust recycling infrastructure and 564 technologies (e.g., in terms of export of second hand vehicles) could reduce the overall availability for recycling<sup>14</sup>. Second, the material specification and design of products is 565 566 becoming more complex in the electrical and automotive sectors, which may challenge the 567 current recycling infrastructure. Third, the high price of platinum has already incentivised the 568 optimization of recycling and manufacturing streams, which indicates that further improvements 569 in recycling are not easily achieved even if the price rises. Last but not least, the future platinum 570 scrap supply may change significantly in the future, which may require significant adaptations to 571 the recycling chain. For example, if the use of fuel cells and electrolysers increases as seen in

572 some scenarios here, the recycling landscape will shift from one that is mainly dominated by 573 autocatalyst scrap today, to one that is more diverse due to the addition of fuel cell and 574 electrolyser scrap. The technological development that lowers the platinum content in fuel cells 575 could make recycling in a few scenarios less attractive in the far future.

576

**3.3.4 Future work.** Our global scenario results indicate that there is enough platinum to supply future global demand, but the high HHI value also suggests potential supply bottlenecks at a regional or national level. A multi-regional trade-linked platinum cycle can help explore these aspects by using more specific demand assumptions by sector by region/country and by more explicit consideration of the role of international trade of raw, intermediate, final, and EoL products in the value chain. This will be addressed in our future work.

583

## 584 ASSOCIATED CONTENT

### 585 Supporting Information

586 Detailed description of the system definition, analytical solutions to the system, and data sources.

587 The Supporting Information is available free of charge on the ACS Publications website at DOI:

588 XXX...

589

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- 593 Author Contributions

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## 607 **REFERENCES**

- 608 (1) Outlook, W. E. World Energy Outlook 2018. 2018. https://doi.org/10.1787/weo-2018-en.
- 609 (2) Simon, B.; Ziemann, S.; Weil, M. Potential Metal Requirement of Active Materials in
  610 Lithium-Ion Battery Cells of Electric Vehicles and Its Impact on Reserves: Focus on
  611 Europe. *Resour. Conserv. Recycl.* 2015, 104, 300–310.
  612 https://doi.org/10.1016/j.resconrec.2015.07.011.
- 613 (3) Alonso, E. Material Scarcity from the Perspective of Manufacturing Firms : Case Studies
  614 of Platinum and Cobalt By, 2010.
- 615 (4) Nassar, N. T. Global Stocks and Flows, Losses, and Recoveries of Platinum-Group
   616 Elements. 2015.
- 617 (5) Graedel, T. E.; Harper, E. M.; Nassar, N. T.; Nuss, P.; Reck, B. K. Criticality of Metals and
   618 Metalloids. *Proc. Natl. Acad. Sci.* 2015, *112* (14), 4257–4262.
   619 https://doi.org/10.1073/pnas.1500415112.
- 620 (6) Seo, Y.; Morimoto, S. Analyzing Platinum and Palladium Consumption and Demand
  621 Forecast in Japan. *Resources* 2017, 6 (4), 61. https://doi.org/10.3390/resources6040061.
- (7) Habib, K.; Wenzel, H. Exploring Rare Earths Supply Constraints for the Emerging Clean
  Energy Technologies and the Role of Recycling. J. Clean. Prod. 2014, 84 (1), 348–359.
  https://doi.org/10.1016/j.jclepro.2014.04.035.
- 625 (8) USGS. Platinum-Group Elements Critical Mineral Resources of the United States-

- 626 Economic and Environmental Geology and Prospects for Future Supply. **2017**.
- 627 (9) USGS. Yearbook 2018 PLATINUM-GROUP METALS; 2018.
- (10) Chalk, S. G.; Miller, J. F. Key Challenges and Recent Progress in Batteries, Fuel Cells, and
  Hydrogen Storage for Clean Energy Systems. *J. Power Sources* 2006, *159* (1 SPEC. ISS.),
  73–80. https://doi.org/10.1016/j.jpowsour.2006.04.058.
- 631 (11)Bertuccioli, L.; Chan, A.; Hart, D.; Lehner, F.; Madden, B.; Eleanor Standen. Study on Electrolvsis 632 Development of Water in the EU. 2014. No. February. 633 https://doi.org/10.1146/annurev.ecolsys.110308.120159.
- 634 (12) Michael J. Coren. Nine countries say they'll ban internal combustion engines
   635 https://qz.com/1341155/nine-countries-say-they-will-ban-internal-combustion-engines 636 none-have-a-law-to-do-so/ (accessed Feb 18, 2019).
- 637 (13) Hagelüken, C. Materials Flow of Platinum Group Metals System Analysis and Measures
  638 for a Sustainable Optimisation. 27th Int. Precious Met. Conf. 2003.
- 639 (14) Saurat, M.; Bringezu, S. Platinum Group Metal Flows of Europe, Part I: Global Supply, Use
  640 in Industry, and Shifting of Environmental Impacts. J. Ind. Ecol. 2008, 12 (5–6), 754–767.
  641 https://doi.org/10.1111/j.1530-9290.2008.00087.x.
- (15) Nansai, K.; Nakajima, K.; Kagawa, S.; Kondo, Y.; Suh, S.; Shigetomi, Y.; Oshita, Y. Global
  Flows of Critical Metals Necessary for Low-Carbon Technologies: The Case of
  Neodymium, Cobalt, and Platinum. *Environ. Sci. Technol.* 2014, 48 (3), 1391–1400.
  https://doi.org/10.1021/es4033452.
- (16) Sverdrup, H. U.; Ragnarsdottir, K. V. A System Dynamics Model for Platinum Group Metal
  Supply, Market Price, Depletion of Extractable Amounts, Ore Grade, Recycling and Stocksin-Use. *Resour. Conserv. Recycl.* 2016, 114, 130–152.
  https://doi.org/10.1016/j.resconrec.2016.07.011.
- (17) Alonso, E.; Field, F. R.; Kirchain, R. E. Platinum Availability for Future Automotive
  Technologies. *Environ. Sci. Technol.* 2012, 46 (23), 12986–12993.
  https://doi.org/10.1021/es301110e.
- (18) Busch, J.; Dawson, D.; Roelich, K. Closing the Low-Carbon Material Loop Using a
  Dynamic Whole System Approach. J. Clean. Prod. 2017, 149, 751–761.
  https://doi.org/10.1016/j.jclepro.2017.02.166.
- 656 (19) Saurat, M.; Bringezu, S. Platinum Group Metal Flows of Europe, Part II Exploring the
  657 Technological and Institutional Potential for Reducing Environmental Impacts. *J. Ind. Ecol.*658 2009, *13* (3), 406–421. https://doi.org/10.1111/j.1530-9290.2008.00106.x.
- (20) Månberger, A.; Stenqvist, B. Global Metal Flows in the Renewable Energy Transition:
  Exploring the Effects of Substitutes, Technological Mix and Development. *Energy Policy*2018, 119, 226–241. https://doi.org/10.1016/J.ENPOL.2018.04.056.
- 662 (21) Sun, Y.; Delucchi, M.; Ogden, J. The Impact of Widespread Deployment of Fuel Cell
  663 Vehicles on Platinum Demand and Price. *Int. J. Hydrogen Energy* 2011, *36* (17), 11116–
  664 11127. https://doi.org/10.1016/j.ijhydene.2011.05.157.
- 665 (22) Elshkaki, A. An Analysis of Future Platinum Resources, Emissions and Waste Streams
  666 Using a System Dynamic Model of Its Intentional and Non-Intentional Fl Ows and Stocks.
  667 *Resour. Policy* 2013, *38* (3), 241–251. https://doi.org/10.1016/j.resourpol.2013.04.002.
- 668 (23) Hagelüken, C. Recycling the Platinum Group Metals : A European Perspective. 2012, No.
  669 1, 29–35.
- 670 (24) Sharma, R.; Gyergyek, S.; Andersen, S. M. Environmentally and Industrially Friendly
   671 Recycling of Platinum Nanoparticles Through Electrochemical Dissolution–

- 672 Electrodeposition in Acid-Free/Dilute Acidic Electrolytes. *ChemSusChem* 2018, *11* (21),
  673 3742–3750. https://doi.org/10.1002/cssc.201801604.
- (25) Hagelüken, C.; Buchert, M.; Ryan, P. Materials Flow of Platinum Group Metals in
  Germany. Int. J. Sustain. Manuf. 2009, 1 (3), 330–346.
  https://doi.org/10.1504/IJSM.2009.023978.
- 677 (26) Couderc, C. Platinum Group Metals in Glass Making. *Platin. Met. Rev.* 2010, 54 (3), 186–
  678 191. https://doi.org/10.1595/147106710X514012.
- (27) Deloitte Sustainability; British Geological Survey; Bureau de Recherches Géologiques et Minières; Netherlands Organisation for Applied Scientific Research. *Study on the Review of the List of Critical Raw Materials - Critical Raw Materials Factsheets*; 2017. https://doi.org/10.2873/876644.
- 683 (28) Graedel, T. E.; Allwood, J.; Birat, J. P.; Buchert, M.; Hagelüken, C.; Reck, B. K.; Sibley,
  684 S. F.; Sonnemann, G. What Do We Know about Metal Recycling Rates? *J. Ind. Ecol.* 2011,
  685 *15* (3), 355–366. https://doi.org/10.1111/j.1530-9290.2011.00342.x.
- 686 (29) Hagelüken, C.; Buchert, M.; Stahl, H. Materials Flow of Platinum Group Metals; 2005.
- (30) Liu, G.; Bangs, C. E.; Müller, D. B. Stock Dynamics and Emission Pathways of the Global
  Aluminium Cycle. *Nat. Clim. Chang.* 2012, *3* (4), 338–342.
  https://doi.org/10.1038/nclimate1698.
- 690 (31) Johnson Matthey. PGM MARKET REPORT FEBRUARY 2018 Summary of Platinum
   691 SUPPLY & amp; DEMAND IN 2017; 2018.
- (32) Vermaak, C. F. *Platinum-Group Metals: A Global Perspective.*, 1st ed.; Mintek, Randburg,
   South Africa, 1995.
- (33) Jones, R. An Overview of Southern African PGM Smelting. *Nickel Cobalt 2005 Challenges Extr. Prod. 44th Annu. Conf. Metall. Calgary, Alberta, Canada* 2005, No. 21-24 August, 147–178.
- 697 (34) Merkle, A. D.; McKenzie, R. K. W. The Mining and Beneficiation of South African PGE
  698 Ores An Overview. Geol. Geochemistry, Mineral. Miner. Benef. Platinum-gr. Elem. L.
  699 J. Cabri, Ed. 2002, 54, 793–809.
- (35) International Organization of Motor Vehicle Manufacturers. Vehicles in use | OICA
   http://www.oica.net/category/vehicles-in-use/ (accessed Aug 23, 2018).
- (36) United Nations. World Population Prospects Population Division United Nations
   https://esa.un.org/unpd/wpp/Download/Standard/Population/ (accessed Aug 23, 2018).
- (37) Boele, G.; ABN AMRO. Electric Vehicles to Result in Large Platinum and Palladium Price
   Declines. *Precious Met. Watch* 2017, No. November.
- 706 (38) Johnson Matthey. *The Components of Autocatalyst Demand*; 2013.
- Wittstock, R.; Pehlken, A.; Peñaherrera, F.; Wark, M. Assessment of the Demand for
  Critical Raw Materials for the Implementation of Fuel Cells for Stationary and Mobile
  Applications. *Cascade Use Technol. 2018* 2018, 111–121. https://doi.org/10.1007/978-3662-57886-5 14.
- (40) International Energy Agency. Energy Technology Perspectives 2017 Catalysing Energy
   Technology Transformations Together Secure Sustainable. 2017.
   https://doi.org/10.1787/energy tech-2017-en.
- 714 (41) The International Energy Agency: Technology Roadmap Electric and Plug-in Hybrid
  715 Electric Vehicles (EV/PHEV). *Encycl. Prod. Manuf. Manag.* 2009, 781–782.
  716 https://doi.org/10.1007/1-4020-0612-8\_961.
- 717 (42) June, U. The International Energy Agency: Technology Roadmap Electric and Plug-in

- 718 Hybrid Electric Vehicles. **2011**, *5* (June).
- (43) Smolinka, T. PEM Water Electrolysis Present Status of Research and Development. In
   *18th World Hydrogen Energy Conference*; 2010; p 23.
- (44) Laila Grahl-Madsen; Madeleine Odgaard; Mikkel Juul Larsen; Thomas Steenberg; Hans
  Åge Hjuler; Casper Frydendal Nørgaard; Shuang Ma Andersen; Peter Brilner Lund; Eivind
  Skou; Jens Oluf Jensen; Qingfeng Li; Lars Nilausen Cleemann; Mark Tonny Dalsgaard
  Jakobsen; *PEM Durability and Lifetime*; 2013.
- (45) Smolinka, T.; Wiebe, N.; Sterchele, P.; Palzer, A.; Lehner, F.; Jansen, M.; Kiemel, S.;
  Miehe, R.; Wahren, S.; Zimmermann, F. Studie IndWEDe -Industrialisierung Der
  Wasserelektrolyse in Deutschland: Chancen Und Herausforderungen Für Nachhaltigen
  Wasserstoff Für Verkehr, Strom Und Wärme; 2018.
- (46) Bernt, M.; Siebel, A.; Gasteiger, H. A. Analysis of Voltage Losses in PEM Water
  Electrolyzers with Low Platinum Group Metal Loadings. J. Electrochem. Soc. 2018, 165
  (5). https://doi.org/10.1149/2.0641805jes.
- (47) Carmo, M.; Fritz, D. L.; Mergel, J.; Stolten, D. A Comprehensive Review on PEM Water
  Filectrolysis. Int. J. Hydrogen Energy 2013, 38 (12), 4901–4934.
  https://doi.org/10.1016/j.ijhydene.2013.01.151.
- 735(48)The World Bank.9th pillar Technological readiness TCdata360736https://tcdata360.worldbank.org/indicators/hf74f651d?country=BRA&indicator=741&viz737=line chart&years=2007,2017 (accessed Aug 23, 2018).
- (49) FHC. Hydrogen Roadmap Europe a Sustainable Pathway for the European Energy Transition; Bietlot, 2019. https://doi.org/10.2843/249013.
- 740 (50) USGS. Appendix C Reserves and Resources. 2012, 219.
- (51) Nassar, N. T.; Graedel, T. E.; Harper, E. M. By-Product Metals Are Technologically
  Essential but Have Problematic Supply. *Sci. Adv.* 2015, No. April, 1–11.
- (52) Bangs, C.; Meskers, C.; Van Kerckhoven, T. Trends in Electronic Products the Canary in
   the Urban Mine? *Electron. Goes Green 2016*+ 2016, 1–8.
- 745 (53) Nassar, N. T. Anthropospheric Losses of Platinum Group Elements. In *Element Recovery* 746 and Sustainability; 2013; pp 185–206.
- 747 (54) Bio by Deloitte. Study on Data for a Raw Material System Analysis: Roadmap and Test of
   748 the Fully Operational MSA for Raw Materials. 2015.
- (55) Fishman, T.; Myers, R.; Rios, O.; Graedel, T. E. Implications of Emerging Vehicle
  Technologies on Rare Earth Supply and Demand in the United States. *Resources* 2018, 7
  (1), 9. https://doi.org/10.3390/resources7010009.
- (56) U.S. Department of Justice and the Federal Trade Commission. Horizontal Merger
  Guidelines (08/19/2010) | ATR | Department of Justice
  https://www.justice.gov/atr/horizontal-merger-guidelines-08192010 (accessed Feb 18, 2019).
- 756 (57) Habib, K. Critical Resources in Clean Energy Technologies and Waste Flows. 2015.
- Johnson Matthey. Market data tables http://www.platinum.matthey.com/services/market research/market-data-tables (accessed Aug 23, 2018).
- (59) Suermann, M.; Babic, U.; Büchi, F. N.; Gubler, L.; Schmidt, T. J. Critical Review—
  Identifying Critical Gaps for Polymer Electrolyte Water Electrolysis Development. J. *Electrochem. Soc.* 2017, 164 (4), F387–F399. https://doi.org/10.1149/2.1441704jes.
- (60) U.S. Department of Justice and the Federal Trade Commission. *Horizontal Merger Guidelines*; 2010.