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1 **Weathering processes, catchment geology and river management impacts on**
2 **radiogenic and stable strontium isotope compositions of Canadian boreal rivers**

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4 **14 Abstract**

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7 **15** Radiogenic ($^{87}\text{Sr}/^{86}\text{Sr}$) and stable ($\delta^{88/86}\text{Sr}$) strontium isotope compositions spanning a calendar year are
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9 **16** reported for rivers from across subarctic Canada that drain contrasting lithologies ranging from
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11 **17** Precambrian bedrock (Koksoak, Great Whale and La Grande rivers of Northern Quebec) to carbonate
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13 **18** and clastic Phanerozoic sedimentary rocks of the Western Interior Platform (Nelson River, of central and
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15 **19** western Canada). The $^{87}\text{Sr}/^{86}\text{Sr}$ isotopic compositions of the rivers reflect the underlying geology, with
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17 **20** rivers draining the Precambrian Shield having higher $^{87}\text{Sr}/^{86}\text{Sr}$ ratios (0.727-0.734) than the Phanerozoic
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19 **21** dominated Nelson River (0.713). The stable strontium isotope values ($\delta^{88/86}\text{Sr}$) range from 0.26 to 0.39
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21 **22** ‰, with the values for the Nelson River overlapping those of the other three. Rivers that have not been
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23 **23** developed for hydroelectric power show a seasonal variation in the $^{87}\text{Sr}/^{86}\text{Sr}$ ratios, whereas those that
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25 **24** have been diverted or dammed show little or no seasonal variation due to increased residence time of
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27 **25** their water in hydroelectric reservoirs. The three rivers from Northern Quebec show discrete ranges in
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29 **26** their $^{87}\text{Sr}/^{86}\text{Sr}$ and $\delta^{88/86}\text{Sr}$ isotope compositions that correlate with the Ca/Sr and Na/Sr ratios of the
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31 **27** river water. These correlations are interpreted to reflect differential chemical weathering of felsic versus
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33 **28** mafic source rocks and/or of surficial sediment vs bedrock sources.
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4 **30 1. Introduction**

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7 **31** The northern boreal rivers of Asia, Europe and North America are important contributors of fresh water
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9 **32** and organic carbon to northern seas such as the Arctic Ocean, the Baltic Sea and Hudson Bay (e.g.
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11 **33** Moore, 2003; Ingri et al., 2005; Déry et al., 2005; Kritzberg et al. 2014). Many of these boreal rivers are
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13 **34** characterized by large accumulations of snow and ice during the winter months that is released as large
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15 **35** pulses of spring melt water that ultimately flow into the northern seas and oceans. Trace element and Sr
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17 **36** isotope studies of boreal rivers and streams have shown that the injection of melt water leads to a
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19 **37** dilution of most trace elements in the river, but can also lead to a change in the source of the elements
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21 **38** during the spring flood season (Ingri et al., 2005; Land et al., 2000). This is also true for rivers draining
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23 **39** into Hudson Bay where Rosa et al. (2012) have shown that Sr and most major ions and trace elements
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25 **40** are diluted by the spring melt water pulse.

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32 **41** Trace element and radiogenic strontium isotope ($^{87}\text{Sr}/^{86}\text{Sr}$) studies are a proven resource for tracing the
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34 **42** transfer of sediment and water from the continents to the oceans. This has been illustrated in a number
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36 **43** of studies ranging from the determination of global silicate weathering rates (Millot et al., 2002;
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38 **44** Gaillardet et al. 1999) to documenting the flux of major and trace elements to the ocean on a global
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40 **45** basis (e.g., Vance et al. 2009). More recently, the stable ($\delta^{88/86}\text{Sr}$) isotope ratio of strontium has proven
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42 **46** to be a useful proxy for tracing chemical weathering (de Souza et al., 2010; Wei et al., 2013; Pearce et
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44 **47** al., 2015; Chao et al., 2015; Andrews et al. 2016; Stevenson et al. 2016). For example; Andrews et al.
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46 **48** (2016) and Stevenson et al. (2016) found significant fractionation in the $\delta^{88/86}\text{Sr}$ ratio as a result of
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48 **49** differential chemical weathering of distinct source rocks. The observed variations in $\delta^{88/86}\text{Sr}$ are
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50 **50** attributed to a combination of carbonate and silicate weathering, secondary carbonate precipitation, as
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52 **51** well as to mixing between river waters and soil/shallow aquifer waters that have been affected by
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54 **52** biological (plant) fractionation (Shalev et al., 2013a; Andrews et al., 2016). Wei et al. (2013)

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demonstrated that weathering of carbonate rocks led to low stable and radiogenic strontium isotope compositions in the Xijiang River of southeast China and that the isotope compositions increased as a result of increased weathering of silicate rocks in the rainy season. In a study of a river catchment dominated by siliciclastic sedimentary rocks in Taiwan, Chao et al. (2015) observed little or no variation in $\delta^{88/86}\text{Sr}$ over the course of a year and that the values obtained overlapped with global flux-weighted average of 0.27 ‰ (Krabenhöft et al., 2010). Stevenson et al. (2016) obtained similar results in a study of glacial outflow from the Lemon Creek Glacier in Southeast Alaska that is underlain largely by granitic rocks.

These studies suggest that rivers underlain by calcareous rocks yield the largest fractionation of $\delta^{88/86}\text{Sr}$ ratios and often portray a correlation between the stable and radiogenic strontium isotope compositions (Krabenhöft et al., 2010; Wei et al., 2013; Pearce et al., 2015). Conversely, rivers draining silicate-dominated terrains show limited variation in the stable strontium isotope composition and no correlation with the radiogenic strontium isotope composition with the exception of rivers that drain volcanic islands (Chao et al., 2015; Pearce et al., 2015; Stevenson et al., 2016). The variable stable Sr-isotope compositions of rivers draining volcanic islands were interpreted to be the result of the formation secondary minerals (Krabenhöft et al., 2010; Pearce et al., 2015).

The above studies have looked at a large number of boreal rivers, but do not include a study of boreal rivers over space and time that would provide insights to the effect of meltwater surges during the spring flood period or to the impact of river management. For example, Déry et al. (2011) documented a shift in the seasonality of Hudson Bay discharge between 1964 and 2008 due to river management. This paper attempts to document seasonal variations in the chemical weathering of silicate vs carbonate dominated catchments, as well as the impact of large hydroelectric reservoirs on Sr-isotope signatures in subarctic Canada. $^{87}\text{Sr}/^{86}\text{Sr}$ and $\delta^{88/86}\text{Sr}$ isotope data have been collected over a one-year period in

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4 76 the Koksoak, Great Whale and La Grande rivers (northern Quebec), and Nelson River (central and
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6 77 western Canada). These rivers were chosen because they have been previously well studied in terms of
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8 78 trace element data (Rosa et al. 2012) and sub-samples of the Rosa et al. (2012) study were made
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10 79 available for this study. In addition, three of these four rivers (Nelson, La Grande, Koksoak) have either
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12 80 been dammed or diverted in some way, allowing an investigation of the impact of human activity on
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14 81 geochemical fluxes.
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23 83 **2. Materials and Methods**

24 25 84 2.1 Geological Setting

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29 85 The four river catchments sampled in this study drain a combined area of more than 1.5×10^6 km² in
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31 86 northern and western Canada (Fig. 1). Three of these rivers (the Koksoak, La Grande and Great Whale)
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33 87 drain the Proterozoic granite-greenstone terrains of the Canadian Shield in Northern Quebec, while the
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35 88 fourth (the Nelson River) predominantly drains Paleozoic sediments exposed in the interior of the
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37 89 continent. All of the rivers drain a combination of tundra to boreal and temperate domains, with mean
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39 90 annual temperatures ranging from 4°C in the Canadian Prairies to -5.7°C in Kuujuaq near the outlet of
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41 91 the Koksoak River into Ungava Bay (Déry et al., 2005; Environment Canada, climatic archive database).
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46 92 The La Grande and Great Whale Rivers principally drain Archean terrains (2.9-2.65 Ga) dominated by
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48 93 intrusive granitoids and metamorphic rocks that discharge into James Bay and Hudson Bay with mean
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50 94 annual discharge rates of 3808 and 632 m³/s, respectively (Rosa et al., 2012). The Koksoak River has a
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52 95 mean annual discharge rate of 1600 m³/s (Rosa et al., 2012) and discharges into Ungava Bay. The
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54 96 drainage basin of the Koksoak River differs from the other two Quebec Rivers in that it also flows
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56 97 through extensive areas of Paleoproterozoic intrusive, volcanic (mafic) and sedimentary rocks of the
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98 New Quebec Orogen (Rosa et al., 2012). The Nelson River draws water from melting glaciers in the
99 Rocky Mountains of Western Canada (e.g., Bow River), as well as from a large portion of the interior
100 Western Platform (North and South Saskatchewan Rivers) and the western portion of the Canadian
101 Shield. These diverse waters flow into Lake Winnipeg, thence into the Nelson River and ultimately into
102 Hudson Bay. Unlike the other three rivers of this study, the larger, western portion of the Nelson River
103 drainage basin is predominately underlain by sedimentary rocks ranging from Paleozoic carbonates and
104 evaporites to poorly consolidated Cretaceous clastic sediments and unconsolidated Quaternary
105 sediments mostly deposited during the last glaciation (Klassen, 1989; Stott and Aitken 1993). As a
106 consequence the Nelson River water is significantly more alkaline than the other three rivers (Rosa et
107 al., 2012).

108 Three of the four rivers have either been dammed or diverted for hydroelectric plants. The Nelson and
109 La Grande Rivers contain one or more dams and their water discharge rates vary as a result of
110 hydroelectricity production. The headwaters of the Koksoak River have been diverted into the
111 Caniapiscau Reservoir that constitutes the head of the La Grande River, but the Koksoak River still
112 captures some of the natural hydro-climatic signals (Rosa et al. 2012) as does the Great Whale River
113 which remains undeveloped.

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115 **2.2 Sampling methods**

116 The samples used in this study are a subset of the samples collected between 2007 and 2009 as part of a
117 larger geochemical investigation on rivers feeding into Hudson Bay, James Bay and Ungava Bay. A
118 detailed description of the sampling procedures in that study can be found in Rosa et al. (2012). The
119 samples were taken at different periods of the year in order to monitor chemical and isotopic changes
120 during the summer and snowmelt seasons. In brief, all samples were collected in mid-stream at a depth

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4 121 of about 30 cm below the surface or 50-100 cm below the ice surface in the winter with the exception of
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6 122 the La Grande River that was sampled at a water sampling site at the head of the feed to the turbines of
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9 123 the LG2 hydroelectric dam. The Nelson River was sampled downstream from the Long Spruce
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11 124 Hydroelectric Station in Manitoba. Samples for the Koksoak and Great Whale Rivers were collected a
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14 125 few kilometers upstream from their outlets. The water samples were filtered through a 0.45 µm nylon
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16 126 membrane, acidified to a pH of 2 and stored at ~5°C until analysis.

19 127 **2.3 Analytical methods**

22 128 Radiogenic and stable Sr isotope analyses were conducted at the National Oceanography Centre in
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25 129 Southampton and at the Geotop laboratories at the Université du Québec à Montréal. Both laboratories
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27 130 followed the chemical and analytical procedure described in Pearce et al. (2015): aliquots of the river
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30 131 water samples that contained 1000 ng of Sr were evaporated in Teflon™ beakers, before being re-
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32 132 dissolved and evaporated in concentrated nitric acid to remove any residual organic material and to
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35 133 convert the residues to nitride salts. The salts were then re-dissolved in 2ml of 3N HNO₃ acid with half of
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37 134 the solution put aside for radiogenic isotope analysis and the other half spiked with an appropriate
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39 135 amount of a ⁸⁴Sr–⁸⁷Sr double spike. The amount of spike was optimized for a 500 ng sample in order to
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42 136 optimize error propagation during the spike deconvolution process (Krabbenhöft et al., 2009; Shalev et
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44 137 al., 2013b; Neymark et al., 2014; Pearce et al., 2015). The Sr in each fraction was then separated using
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46 138 Eichrom Sr-Spec™ resin using the elution sequence described in Pearce et al. (2015).

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52 140 technique described by Charlier et al. (2006), prior to being analyzed on a Triton™ Thermal Ionisation
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54 141 Mass Spectrometer in both Southampton and Montreal via static analysis. The samples were analyzed at
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57 142 an intensity of 8 V for 54 cycles with 10 ratios in each cycle using a 16 second integration time. A
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59 143 detailed description of the analytical routine can be found in Stevenson et al. (2014). The analyses for
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4 144 the radiogenic aliquots were normalized to the fixed $^{86}\text{Sr}/^{86}\text{Sr}$ ratio of 0.1194 (Nier, 1938), while stable Sr
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6 145 isotope ratios were determined by combining measurements on the unspiked and spiked samples via
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9 146 the Newton-Raphson iteration technique in ^{87}Sr denominator space (Pearce et al., 2015).
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12 147 Analyses of splits of the same solution of the international NBS-987 Sr standard run during the course of
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14 148 this study yielded average $^{87}\text{Sr}/^{86}\text{Sr}$ ratios of 0.710259 ± 5 (2σ ; $n = 11$; Southampton) and $0.710251 \pm$
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17 149 5 (2σ ; $n = 18$; Montreal). Analyses from both laboratories were normalized against a NBS-987 value of
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19 150 0.710255 . The calculated $\delta^{88/86}\text{Sr}$ values for NBS-987 analyses for each analytical session were averaged
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22 151 and then normalized against a true NBS-987 $\delta^{88/86}\text{Sr}$ value of zero. The resulting normalization value was
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24 152 used to correct the sample $\delta^{88/86}\text{Sr}$ values for each session. Both laboratories yielded identical $\delta^{88/86}\text{Sr}$
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27 153 values for NBS-987 over the course of the study; -0.002 ± 0.014 (2σ ; $n = 9$) at Southampton and -0.002
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29 154 ± 0.021 (2σ ; $n = 11$) for measurements at Montreal. The Sr blank at both institutions was less than 50
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32 155 pg and represents $<0.01\%$ of the total Sr in each aliquot.
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35 156 3. Results

38 157 3.1 Trace element geochemistry

41 158 The geochemical data used for this study is presented in Table A1 (supplementary data). Figure 2
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43 159 illustrates the geochemical differences between these rivers based on the samples selected for this
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46 160 study. During the sampling period, the pH of the rivers varied between 6.1 and 8.1, with the Nelson
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48 161 River being significantly more alkaline than the three northern Quebec rivers. These different pH values
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51 162 reflect the varying geological compositions of the drainage basins (Rosa et al., 2012) as $\sim 90\%$ of the
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53 163 Nelson River drainage basin is underlain by sedimentary rocks that include calcareous sediments,
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55 164 whereas the three Quebec rivers are underlain by the intrusive and volcanic igneous rocks of the
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58 165 Canadian Precambrian Shield (70-99%). These differences are also reflected in the plot of pH versus the
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7 167 Quebec rivers, the pH and the Ca content increase from the La Grande to the Great Whale then to the
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9 168 Koksoak Rivers. Figure 2b shows that the Ca/1000Sr ratio increases with increasing pH, suggesting that
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11 169 Ca increases with alkalinity at the expense of Sr. Figure 2c shows the relationship between the
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14 170 Ca/1000Sr and Mg/1000Sr ratios among the four rivers. This plot underlines the higher Mg content of
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16 171 the Koksoak River compared to those of the other four rivers likely reflects the higher proportion of
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18 172 volcanic (largely mafic in composition) and sedimentary rocks (30%) within its drainage basin (Rosa et
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20 173 al., 2012). No correction was made for atmospheric deposition (e.g., Gaillardet et al., 1999) in these
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23 174 basins, as it did not affect the observed correlations.
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30 176 3.2 Radiogenic stable Sr isotope analyses

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33 177 The stable and radiogenic isotope data for the four rivers are presented in Table 1 and plotted in Figures
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35 178 3 and 4. Figure 3a displays $\delta^{88/86}\text{Sr}$ vs $\delta^{84/86}\text{Sr}$ values of samples and of the NBS-987 standard. It illustrates
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38 179 that the values obtained at both the UQAM and Southampton laboratories conformed to the mass
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40 180 dependent fractionation line of -0.9545. Figure 3b plots the $^{87}\text{Sr}/^{86}\text{Sr}$ isotope ratios for samples analyzed
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43 181 at both UQAM and Southampton and also demonstrate that samples analyzed at both laboratories
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45 182 yielded comparable $^{87}\text{Sr}/^{86}\text{Sr}$ isotope ratios. In Figure 4, the Nelson River has the lowest $^{87}\text{Sr}/^{86}\text{Sr}$ ratios
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47 183 with very little variation (0.71295 to 0.71315) whereas the three Quebec rivers are much more
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50 184 radiogenic than the majority of rivers surveyed for stable Sr isotopes (e.g., Pearce et al. 2015;
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52 185 Krabenhof et al., 2010) but are comparable with the Ganges River of India (Pearce et al., 2015) as well
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54 186 as with rivers draining the Slave Craton of northwestern Canada (Millot et al., 2002). The Koksoak River
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57 187 has the lowest $^{87}\text{Sr}/^{86}\text{Sr}$ ratios among the Quebec rivers with values ranging between 0.72701 and
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59 188 0.72985. The La Grande River yielded $^{87}\text{Sr}/^{86}\text{Sr}$ ratios between 0.73023 and 0.73156 and the Great Whale
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4 189 River is (on average) the most radiogenic river with $^{87}\text{Sr}/^{86}\text{Sr}$ ratios varying from 0.73030 and 0.73424.

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6 190 The more radiogenic Sr isotope composition in Quebec rivers, compared to the Nelson River, reflects the
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9 191 older Precambrian crust underlying these rivers compared to the younger Phanerozoic
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11 192 sediment/carbonate-dominated rocks underlying the Nelson River.
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14 193 The stable Sr isotope composition $\delta^{88/86}\text{Sr}$ of the four rivers varies from 0.29 to 0.39 (Fig. 4), effectively
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16 194 bracketing the terrestrial silicate average of 0.30‰ (Moynier et al., 2010; Charlier et al., 2012) and the
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19 195 flux weighted global river average of 0.32 +/- 0.008 (Krabbenhöft et al., 2010; Pearce et al., 2015). The
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21 196 Nelson River shows the largest variation from 0.30 to 0.38 followed by the Great Whale River with
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24 197 values from 0.29 to 0.36. The La Grande and Koksoak rivers show more narrow ranges of 0.30 to 0.32 and
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26 198 0.37 to 0.39, respectively. All rivers yield a range of $\delta^{88/86}\text{Sr}$ values that are also overall similar to those
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29 199 derived from glacial run-off in the Alaska Panhandle (0.26 ± 0.02 to 0.40 ± 0.02 ‰; Stevenson et al. 2016)
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31 200 and to rivers draining into the Milford Sound region of Fiordland, New Zealand (Andrews et al., 2016). As
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33 201 a group, the Quebec rivers show a slight negative trend of decreasing radiogenic Sr isotope ratios with
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36 202 increasing stable Sr isotope compositions. This likely reflects the relative proportion of felsic versus
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38 203 mafic silicate rocks in the three catchment basins and is discussed in section 4.2.
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41 204 3.3. Seasonal variations in the radiogenic and stable Sr isotope ratios

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45 205 The $^{87}\text{Sr}/^{86}\text{Sr}$ and $\delta^{88/86}\text{Sr}$ isotope data are plotted against their sampling date in Figure 5. Over a
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47 206 period of six months the $^{87}\text{Sr}/^{86}\text{Sr}$ ratios of the Koksoak River decrease from a high of 0.72985 in
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50 207 February 2009 to a low of 0.72714 in July 2009 (Fig. 5a). The time series $^{87}\text{Sr}/^{86}\text{Sr}$ profile for the Great
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52 208 Whale River is more radiogenic in the winter and summer, but shows an abrupt decrease in the $^{87}\text{Sr}/^{86}\text{Sr}$
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54 209 ratio during the snowmelt period (Fig. 5b). The Nelson River yields a somewhat flat $^{87}\text{Sr}/^{86}\text{Sr}$ profile and
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57 210 the La Grande River yields a convex profile with slightly more radiogenic Sr dominating in late winter and
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59 211 through the snowmelt season (Fig. 5b),. The trends observed in the Koksoak and Great Whale Rivers are
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4 212 in the opposite direction with more radiogenic Sr isotope ratios dominating in the winter months (Fig.
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10 214 The $\delta^{88/86}\text{Sr}$ values of the Koksoak River varied little over the six month sampling period and yielded a
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12 215 flat profile (Fig. 5c). The Great Whale River time series profile (Fig. 5d) is relatively constant through the
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14 216 winter and spring flood season but increases during the summer before decreasing from 0.36 to 0.29 in
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16 217 the autumn. The La Grande River also has a relatively flat time series profile (Fig. 5d), except for a slight
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18 218 decrease in the $\delta^{88/86}\text{Sr}$ values in December, 2007. The Nelson River time series profile (Fig. 5d) increases
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20 219 slowly from the winter to the fall, but it is the only river to show a significant shift, a decrease, in the
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22 220 $\delta^{88/86}\text{Sr}$ values during the spring flood season.

27 221 **4.0 Discussion**

30 222 4.1. Temporal variations

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34 223 Seasonality has been shown to have a marked influence on the dissolved $^{87}\text{Sr}/^{86}\text{Sr}$ composition of rivers
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36 224 (Douglas et al., 2013; Voss et al., 2014). For example, Voss et al. (2014) showed that $^{87}\text{Sr}/^{86}\text{Sr}$ isotope
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38 225 ratios from the Frazer River in Western Canada varied from as low as 0.709 in the winter to as high as
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40 226 0.714 in the summer with the increase reflecting the release of radiogenic Sr from weathered clays and
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42 227 clay rich sedimentary rocks during summer months. Stevenson et al. (2016) also noted a slight increase
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44 228 in the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of water from the subglacial outflow of the Lemon Creek glacier (Juneau Ice Field,
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46 229 Alaska) during a one month period during the summer of 2012. Douglas et al (2013) documented a
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48 230 similar trend of more radiogenic Sr isotope ratios in the summer versus winter in the boreal Chena river
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50 231 of Central Alaska that they attributed to a higher proportion of silicate versus carbonate mineral
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52 232 dissolution in the summer months.
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233 Unlike the Chena and Frazer river profiles noted above, none of the four rivers sampled for this study
234 shows a consistent trend in their radiogenic Sr isotope composition over the sampled period. The
235 $^{87}\text{Sr}/^{86}\text{Sr}$ ratios of the Koksoak decrease steadily during the summer from January to July, 2009 (Fig. 5a).
236 The time series $^{87}\text{Sr}/^{86}\text{Sr}$ profile for the Great Whale River is more radiogenic in the winter and summer,
237 but shows an abrupt decrease in the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio during the snowmelt period (Fig. 5b) which suggests
238 a greater contribution from a less radiogenic source, such as carbonates, or an increased dissolution of
239 silicate minerals derived from mafic rocks. The lack of a strong time-series $^{87}\text{Sr}/^{86}\text{Sr}$ profile for the La
240 Grande and Nelson Rivers (Fig. 5b) may be related to the fact that these rivers are extensively developed
241 for hydroelectric power and that the damming of the rivers results in a longer residence time of the river
242 water that buffers seasonal gradients. For example, a study of dammed rivers in China, found that the
243 weathering rate increased with the impoundment time of the water in the reservoir (Gao et al., 2013).
244 This greater weathering could lead to an averaging out of geochemical variations on a yearly basis.
245 Another factor is that given the large size of the Nelson River drainage basin, it is unlikely that snowmelt
246 is concurrent across the basin, leading to an averaging out of the meltwater pulse over time. The trends
247 observed in the Koksoak and Great Whale Rivers are in the opposite direction with more radiogenic Sr
248 isotope ratios during winter months (Fig. 5a, b). Given their similar climatic conditions, the Sr isotope
249 composition of the Quebec rivers will reflect the Sr isotopic composition of the most easily altered and
250 eroded rock-type. In the subarctic environment of Northern Quebec, mafic rocks are abundant and
251 among the most easily altered and eroded rock-type. During the summer months and/or snowmelt
252 period, the friable mafic mixture would be prone to chemical weathering and the release of larger
253 amounts of less radiogenic Sr to the rivers resulting in a decrease in the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio as observed for
254 the Koksoak and Great Whale Rivers. Water derived from the melting of the snow in the spring is not
255 viewed as a significant contributor to the Sr budget because of the very low Sr contents in snow

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4 256 (typically less than 0.5 ppm; e.g. Simonetti et al., 2000) compared to the larger Sr reservoirs of the rocks,
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10 258 The time series variations observed in the $\delta^{88/86}\text{Sr}$ values for the four rivers are not consistently
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12 259 associated with any one season and do not correlate with time series changes observed for the
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14 260 radiogenic Sr isotopes. The $\delta^{88/86}\text{Sr}$ values for the Koksoak remain essentially constant throughout the
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17 261 year (within error, Fig 5c). The $\delta^{88/86}\text{Sr}$ values for the La Grande River show no abrupt changes but display
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19 262 a small decrease during the winter (Fig. 5d). The $\delta^{88/86}\text{Sr}$ values of the Great Whale River rise through the
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22 263 winter and spring months and decrease abruptly in the summer (Fig. 5d), but there is no abrupt change
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24 264 in the stable Sr isotope composition during the snowmelt period as seen in Figure 5b for the radiogenic
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27 265 Sr isotope compositions. The $\delta^{88/86}\text{Sr}$ values of the Nelson River show an abrupt decrease during the
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29 266 snowmelt period (April 2008) and returned to an average value during the summer (Fig. 5d). Here again,
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31 267 the abrupt decrease in the $\delta^{88/86}\text{Sr}$ values of the Nelson River is not accompanied by a corresponding
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34 268 change in the radiogenic Sr isotope composition. Thus although the $^{87}\text{Sr}/^{86}\text{Sr}$ ratios may respond to
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36 269 seasonal forcing in still "pristine" rivers (e.g., Great Whale River), the $\delta^{88/86}\text{Sr}$ values appear to be largely
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39 270 immune to the forces that act upon the radiogenic Sr isotope ratios. The $^{87}\text{Sr}/^{86}\text{Sr}$ and $\delta^{88/86}\text{Sr}$ isotope
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41 271 systems are therefore decoupled from one another in these river systems. This decoupled behavior has
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44 272 also been observed for time-series $^{87}\text{Sr}/^{86}\text{Sr}$ and $\delta^{88/86}\text{Sr}$ isotope data for the Hou-Ku River in Taiwan
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46 273 (Chao et al., 2013) and glacial out flow in Southeast Alaska (Stevenson et al., 2016), although Wei et al.
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48 274 (2013) documented a strong correlation between $^{87}\text{Sr}/^{86}\text{Sr}$ and $\delta^{88/86}\text{Sr}$ isotope compositions for the
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51 275 Xijiang River in China, that was ascribed to variable carbonate vs silicate weathering.
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54 276 In a $^{87}\text{Sr}/^{86}\text{Sr}$ isotope and geochemical study of a boreal stream environment, Land et al. (2000) divided
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56 277 water contributions to the streams into 3 compartments: deep ground water that contributed mostly
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59 278 during the late summer and fall base flow season (5-20%), soil water that dominated during the spring
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279 melt and shallow groundwater that dominates during base flow (during summer/fall). These
280 observations could explain some of the observed variations in the rivers of this study. The Nelson River
281 is distinct from the three Quebec rivers in that it drains the well-developed soils of the North American
282 Prairie grasslands. The relatively constant $^{87}\text{Sr}/^{86}\text{Sr}$ isotope ratio over the nine month sampling period
283 likely reflects the buffering of the radiogenic Sr isotope composition of these soils on the Nelson River.
284 The decrease in the stable Sr isotope composition of the Nelson river during the spring melt could
285 reflect the flushing of the soils by the spring run-off and the addition of soil water that is typically
286 depleted in ^{88}Sr (Andrews et al., 2016). In contrast, the radiogenic Sr isotope composition of the Great
287 Whale River decreases during the spring melt, but shows no change in its stable Sr isotope composition
288 during the same period. The change in source indicated by the decrease in the $^{87}\text{Sr}/^{86}\text{Sr}$ may reflect the
289 erosion of glaciofluvial and glaciomarine sediments containing younger crustal components during the
290 spring flood that form the shores of the lower reaches of the river (Paradis and Parent, 1997). The lack
291 of response in the stable Sr isotope composition suggests that the eroded Archean rocks and
292 glaciofluvial/glaciomarine sediments in Great Whale River catchment have a similar mineralogical
293 composition. The subsequent decrease in the stable Sr isotope composition of the Great Whale River in
294 the late summer may reflect the addition of groundwater that has been depleted in ^{88}Sr due to its
295 interaction with silicate rocks and/or the formation of secondary minerals (*e.g.*, clay minerals, micas).

296 4.2. Silicate vs carbonate weathering

297 Insights to the decoupled behavior of radiogenic and stable Sr isotope compositions in boreal rivers can
298 be obtained by comparing the Sr isotope compositions with the geochemistry of the rivers. There is a
299 broad negative correlation between the pH of the rivers (especially among the Quebec rivers) and the
300 radiogenic Sr isotope ratios with the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio generally increasing with decreasing pH (Fig 6a). Wei
301 et al. (2013) also noted a negative correlation between pH and the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio in a year-long

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302 observation of the Xijiang River water of Southeast China and suggested that this reflected increased
303 chemical weathering of silicate rocks under low pH conditions, yielding an increase in the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio.
304 These variations also reflect the difference in the geology of the drainage basins. The weathering of
305 granitic rocks, of greatest abundance in the Great Whale watershed (Rosa et al., 2012), yields a higher
306 abundance of acidic cations and results in the lower pH and a higher $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of Great Whale River.
307 Conversely, the weathering of mafic rocks such as basalt (Koksoak River) release greater amounts of
308 basic cations such as Mg, and less radiogenic Sr. The even higher pH and lower $^{87}\text{Sr}/^{86}\text{Sr}$ ratios of the
309 Nelson River result from the higher abundance of carbonates in the hinterland of its drainage basin.
310 The $^{87}\text{Sr}/^{86}\text{Sr}$ ratios of the three Quebec rivers show a positive correlation with the Ca/Mg ratios and
311 broadly negative correlations with the Ca/1000Sr and Ca/Na ratios (Fig. 6b-d). These correlations can
312 also result from some control by relative contributions of granitic versus mafic rocks (extrusive and
313 intrusive). Drainage basins within the Canadian Shield that contain significant amounts of mafic igneous
314 rocks (e.g., Koksoak River) would have high Ca/Na and Ca/Sr ratios, low Ca/Mg ratios and low $^{87}\text{Sr}/^{86}\text{Sr}$
315 ratios. The higher $^{87}\text{Sr}/^{86}\text{Sr}$ ratios and Ca/Mg ratios, and lower Ca/Na and Ca/Sr ratios in the La Grande
316 and Great Whale rivers reflects the greater proportion of granitic versus mafic rocks in their respective
317 watersheds.
318 In contrast to the rivers of northern Quebec, Andrews et al. (2016) documented a positive correlation
319 between the $^{87}\text{Sr}/^{86}\text{Sr}$ and Ca/Sr ratios in rivers draining into the Milford Sound of New Zealand. The
320 authors attributed their positive correlation to the mixing of radiogenic soil water with riverine water.
321 Soils are thin and poorly developed in the sub-arctic environment of northern Quebec. In the scarce and
322 thin soils that characterize these drainage basins, both the interpretation above and that of Andrews et
323 al. (2016) are consistent. In northern Quebec, the exchangeable Sr would yield $^{87}\text{Sr}/^{86}\text{Sr}$ isotope ratios
324 similar to that of the bedrock considering the reduced chemical weathering that is present in the

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325 subarctic versus temperate environment of New Zealand. The abundance of carbonate and evaporite
326 deposits in the watershed of the Nelson River results in the low Ca/Mg and higher Ca/Na and Ca/Sr
327 ratios with lower $^{87}\text{Sr}/^{86}\text{Sr}$ ratios. It is noteworthy that the territory drained by the three Quebec rivers
328 was the first area to be regionally sampled to obtain an average composition of the Earth's crust (Shaw
329 et al. 1967), and the $^{87}\text{Sr}/^{86}\text{Sr}$ isotope ratio of this composite (0.7283; McCulloch and Wasserburg 1978)
330 is comparable with the average Koksoak River $^{87}\text{Sr}/^{86}\text{Sr}$ composition of 0.7286.

331 The $\delta^{88/86}\text{Sr}$ isotope values also vary with the composition of the river waters. The pH of the Quebec
332 rivers fall into discrete ranges with minor overlaps and, generally, the higher the pH of the river, the
333 higher the stable Sr isotope composition of the river (Figure 7a). A number of studies have indicated that
334 high $\delta^{88/86}\text{Sr}$ values in rivers are the product of weathering that releases heavy ^{88}Sr from silicates while
335 the light ^{86}Sr is precipitated in secondary carbonate minerals (De Souza et al., 2012; Wei et al., 2013;
336 Pearce et al., 2015; Andrews et al., 2016; Stevenson et al., 2016). Carbonate precipitation is also greater
337 at higher pH and this would lead to higher $\delta^{88/86}\text{Sr}$ values in the river water after the carbonate
338 precipitation.

339 However, as seen in the correlations with the radiogenic Sr isotope compositions, the increase in $\delta^{88/86}\text{Sr}$
340 values with pH may also be a product of the underlying geology with the heavier values reflecting
341 weathering of the mafic rock that dominate the Koksoak drainage basin. The alteration and
342 metamorphism of mafic rocks (basalt, gabbro) are accompanied by extensive formation of serpentine,
343 talc and calcite; the formation of calcite in these assemblages likely traps the light Sr leaving the heavy
344 Sr to be released to the environment. Alternatively, the weathering and metamorphism of mafic rocks
345 releases abundant heavy Sr from silicate minerals such as pyroxene and calcic plagioclase that are less
346 resistant to alteration compared to the more resistant K-feldspar and quartz that dominate in granite

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347 domains. The Nelson River does not follow this correlation, presumably due to the buffering effect of
348 the abundant carbonate and evaporites in its drainage basin.

349 The $\delta^{88/86}\text{Sr}$ values also correlate with the Ca/Na ratios, Ca/Mg and Ca/1000Sr ratios (Fig. 7b-d). The
350 correlation with the Ca/Na ratio underlines the relationship of the $\delta^{88/86}\text{Sr}$ variation with the degree of
351 differentiation present in the bedrock underlying the drainage basins. The Quebec rivers show a
352 decrease in the $\delta^{88/86}\text{Sr}$ values with a decrease in the Ca/Na ratio. As the abundance of mafic rock in the
353 drainage basin decreases and the amount of granitic rock increases, the $\delta^{88/86}\text{Sr}$ values decrease due to a
354 lesser contribution from weathered mafic rocks. Note that the Nelson River also follows this trend. This
355 likely reflects the contribution of carbonate/evaporite vs silicate weathering products. The negative
356 trend on the Ca/Mg versus $\delta^{88/86}\text{Sr}$ diagram (Fig. 7c) could also be interpreted the same way. The high
357 Ca/Mg ratio is found among the rivers draining basins underlain by greater proportions of granitic
358 bedrocks whereas the lower Ca/Mg ratios are associated with the rivers underlain by mafic and
359 sedimentary rocks (Koksoak and Nelson Rivers). The higher $\delta^{88/86}\text{Sr}$ values associated with lower Ca/Mg
360 ratios reflect the breakdown of the mafic silicates containing heavy Sr and the addition of this heavy Sr
361 to river waters. The correlation between the Ca/1000Sr and $\delta^{88/86}\text{Sr}$ values (Fig. 7d) behaves similarly
362 with the heaviest $\delta^{88/86}\text{Sr}$ isotope values correlating with the highest Ca/Sr ratios in the Quebec rivers.
363 This reflects the higher proportion of Ca in the rivers (Koksoak) that have a greater abundance of mafic
364 rocks in their watershed.

365 4.3 Global comparisons

366 Figure 8 provides a comparison of the Ca/1000Sr ratios and the $\delta^{88/86}\text{Sr}$ values for the rivers of this study
367 with those measured in other river water studies (Chao et al., 2013; Wei et al., 2013; Pearce et al., 2015;
368 Andrews et al., 2016; Stevenson et al., 2016). The black arrows indicate the positive correlation between
369 the Ca/1000Sr ratios and $\delta^{88/86}\text{Sr}$ values as a result of carbonate and silicate weathering. This trend has

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4 370 been documented for carbonate dominated rivers in France (Pearce et al., 2015) and China (Wei et al.,
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6 371 2013) and is evident in the Upper Holly River catchment of New Zealand (Andrews et al., 2016). Low
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9 372 $\delta^{88/86}\text{Sr}$ and Ca/1000Sr ratios reflect intense carbonate weathering with the release of light Sr and large
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11 373 amounts of Sr to the rivers, while the high $\delta^{88/86}\text{Sr}$ and Ca/1000Sr ratios reflect carbonate precipitation
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13 374 and increased silicate weathering. In the absence of significant carbonate dissolution, the positive
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15 375 correlation is controlled by the silicate weathering process. Weathering of Ca-rich mafic bedrock
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17 376 releases heavy Sr to the rivers because light Sr is retained in secondary carbonate. Drainage basins that
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19 377 are underlain by more differentiated bedrock (granites) yield lower $\delta^{88/86}\text{Sr}$ values due to release of
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21 378 lighter Sr from weathered feldspars and micas. Charlier et al. (2012) documented stable Sr isotope
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23 379 values as low as 0 to -0.2 for highly differentiated volcanic tuffs and rhyolitic glasses and suggested that
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25 380 magmatic fractional crystallization could result in these low $\delta^{88/86}\text{Sr}$ values. The high Ca content of the
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27 381 modern glacier meltwater studied by Stevenson et al. (2016) likely reflects the high Ca content of the
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29 382 underlying tonalite rocks.
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36 383 Andrews et al. (2016) also found a strong correlation between $\delta^{88/86}\text{Sr}$ values and the Ca/Sr ratios in
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38 384 rivers draining into the Milford sound of New Zealand. They argued, based on the analysis of soil waters
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40 385 and leaching experiments of the bedrock, that this correlation was the result of the incorporation of
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42 386 isotopically heavy soil water (due to the uptake of light Sr by plants) in the rivers, with secondary
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44 387 mineral formation and precipitation playing a secondary role. This mixture is likely responsible for the
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46 388 negative trend indicated by the dashed line in Figure 8 that is defined by the data from the Lower
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48 389 Hollyford River (LHR) catchment of Andrews et al. (2016). Andrews and Jacobson (2017) found that the
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50 390 $\delta^{88/86}\text{Sr}$ isotopic composition of Icelandic rivers could also be ascribed to a mixture of basalt and calcite
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52 391 and atmospheric deposition. The three Quebec rivers drain an area in which soils are very thin and often
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54 392 absent and affected by permafrost in the case of the Koksoak River. Thus the uptake of isotopically light
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56 393 Sr by plants, and residual soil waters are not expected to significantly impact the composition of the
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394 Quebec rivers. Samples from the Nelson River drain a soil-rich region and define less of a positive trend
395 in Figure 8 compared to the Quebec rivers, thus plant uptake and soil waters may play a greater role in
396 defining the stable Sr isotope composition of the river. Andrews and Jacobsen (2018) found that the
397 $^{87}\text{Sr}/^{86}\text{Sr}$ and $\delta^{88/86}\text{Sr}$ isotope composition of Icelandic rivers could be modeled as a mixture of Sr derived
398 from hydrothermal calcite, atmospheric deposition and weathered basalt. A comparison with this study
399 is not easily made due to the complex geology of the drainage basins of the rivers of this study.

400 The $^{87}\text{Sr}/^{86}\text{Sr}$ and $\delta^{88/86}\text{Sr}$ isotope compositions from this study and those of rivers from a number of
401 previous studies are plotted together in Figure 4 (Andrews et al., 2016; Stevenson et al., 2016; Pearce et
402 al., 2015; Wei et al., 2013). In a study of global rivers, Pearce et al. (2015) noted a general lack of
403 correlation between the $^{87}\text{Sr}/^{86}\text{Sr}$ ratios and the $\delta^{88/86}\text{Sr}$ values of the rivers and suggested that the lack
404 of correlation of the stable Sr isotope with other isotopic systems was likely an artefact of the
405 complexity of weathering in multi-lithological catchments. This is supported by the fact that Wei et al
406 (2013) and Andrews et al. (2016) documented both positive and negative correlations in this plot
407 stemming from the competing effects of carbonate and silicate weathering (Wei et al., 2013) and mixing
408 of river water dominated by either silicate or carbonate weathering with soil water (Andrews et al.,
409 2016). The somewhat negative correlation defined by the three Quebec rivers can be interpreted within
410 this context as reflecting a higher contribution of isotopically light and radiogenic Sr released from the
411 weathering of K-feldspar-rich lithologies such as granites (Charlier et al., 2012) and TTG series that
412 dominate the Great Whale River and La Grande River catchments, versus the isotopically heavy and less
413 radiogenic Sr that is released via the weathering of pyroxenes and calcic plagioclase that are more
414 abundant in the Koksoak River catchment.

415 The watersheds of the three Quebec rivers drain an area that is dominated by silicate rocks and bereft
416 of limestone and/or dolomite. Carbonates are present in the form of secondary weathering/alteration

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4 417 minerals such as calcite, especially in terrains dominated by mafic and ultramafic rocks, as well as in
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6 418 more discrete biogenic carbonates and carbonate concretions from glaciomarine deposits (*e.g.*, Hillaire-
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8 419 Marcel, 1980a and b). It is interesting that the scope of $\delta^{88/86}\text{Sr}$ values present in the three Quebec rivers
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10 420 combined is comparable to the range of $\delta^{88/86}\text{Sr}$ values found in Nelson River that has abundant
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12 421 limestone, dolomite and evaporites in its watershed. Given that the amount of carbonate in the
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14 422 catchments of the Quebec rivers that can be weathered is extremely small and that calcite precipitation
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16 423 would be more limited in these northern latitudes, the variation in the $\delta^{88/86}\text{Sr}$ values is most likely
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18 424 controlled by silicate weathering. Neymark et al. (2014) analyzed a K-feldspar standard (SRM-607) and
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20 425 found $\delta^{88/86}\text{Sr}$ values as low as 0.18 and $^{87}\text{Sr}/^{86}\text{Sr}$ ratios as high as 1.2. The erosion of this type of material
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22 426 would contribute light Sr in weathered granitic terrains such as La Grande and Great Whale Rivers.
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24 427 Stevenson et al. (2016) found that the $\delta^{88/86}\text{Sr}$ composition of suspended sediment in subglacial outflow
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26 428 from a weathered granitic terrain yielded $\delta^{88/86}\text{Sr}$ values that were uniformly lower than the glacial
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28 429 outflow water and as low as 0.12. The authors attributed these low values to the leaching of light Sr
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30 430 from micas (biotite) and feldspars.
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42 431 43 44 45 432 **5. Conclusions**

46 433 We presented new radiogenic and Sr isotope data for water samples collected over a 6-12 month period
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48 434 for four Canadian boreal rivers. The data demonstrate that underlying geology exerts a primary
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50 435 influence on both the radiogenic and stable Sr isotope compositions. Three rivers draining the
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52 436 Precambrian Shield of Northern Quebec yield highly radiogenic $^{87}\text{Sr}/^{86}\text{Sr}$ ratios whereas the Nelson River
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54 437 that drains a catchment dominated by carbonate and clastic sedimentary of the mid-continent has much
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56 438 lower $^{87}\text{Sr}/^{86}\text{Sr}$ ratios. The $\delta^{88/86}\text{Sr}$ compositions straddle the global river average (Krabenhöft et al.,
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58 439 2010; Pearce et al., 2015), but values from the three Quebec rivers vary as a consequence of the
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440 proportion of mafic versus felsic silicate rocks within the catchment basin, with heavier $\delta^{88/86}\text{Sr}$ values
441 correlating with higher proportions of mafic rocks. This correlation is interpreted to reflect the
442 formation of secondary carbonates during the weathering of the mafic rocks that trap the lighter Sr
443 isotopes and underlines the importance of chemical weathering in a northern boreal environment.
444 Samples of the river waters collected in different seasons to monitor changes in river composition
445 during the snow melt period yielded inconsistent results for both radiogenic and stable Sr isotope
446 compositions. Only the Great Whale River showed a variation during the spring thaw with a decrease in
447 the radiogenic Sr isotope composition that was interpreted to result from increased winnowing of
448 younger glacio-fluvial glacio-marine sediments from the banks of the river. The absence of variations in
449 the other rivers may reflect the development of these rivers for hydroelectricity leading to longer
450 residence time for the water. The Great Whale River is the only river of the study that is not developed
451 or diverted for hydroelectricity. Thus, development of rivers appears to affect the isotopic composition
452 of the river in that it smooths out seasonal variations due to longer residence time of the water.

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4 **564 Figure Captions**

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7 **565** Figure 1. Locations of the watersheds sampled for this study. 1 = Koksoak River; 2 = Great Whale River; 3
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9 **566** = La Grande River, 4 = Nelson River. Modified from Rosa et al. (2012).

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13 **567** Figure 2. Examples of the Water chemistry of the rivers sampled in this study. A. Ca/Mg vs pH and b.
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15 **568** Ca/Sr vs pH. Chemical data from Rosa et al. (2012).

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19 **569** Figure 3. Figure depicts the general coherence of stable Sr isotope data to mass dependent
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21 **570** fractionation line (a.) and the similarity of the radiogenic isotope compositions obtained by the UQAM
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23 **571** and Southampton laboratories (b).

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27 **572** Figure 4. Plot of stable versus radiogenic Sr isotope compositions. The plot illustrates the radiogenic
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29 **573** nature of the three Quebec rivers with respect to the Nelson River and the majority of global river
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31 **574** waters studied to this point. See text for details. Terrestrial silicate average from Moynier et al. (2010).
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33 **575** Global river average from Krabenhof et al. 2010 and Pearce et al. (2015).

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37 **576** Figure 5. Time series radiogenic and stable isotope plots for the studied rivers. Solid symbols and large
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39 **577** X's are data from UQAM; open symbols and small X's are data from Southampton. Figures b and d share
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41 **578** the same legend. Average error for stable Sr isotope analyses indicated in figure c. See text for details..

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45 **579** Figure 6. Plots of the $^{87}\text{Sr}/^{86}\text{Sr}$ composition of the river water samples versus a. pH; b. Ca/Mg; c.
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47 **580** Ca/1000Sr and d. Ca/Na. The trends reflect the lithological control on the isotope composition and the
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49 **581** geochemistry of the water. Catchments with greater carbonate or mafic rock contents yield lower
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51 **582** $^{87}\text{Sr}/^{86}\text{Sr}$ ratios and higher Ca, Mg and pH. The red star is the Sr isotope composition and Ca/Mg ratio of
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53 **583** the Canadian Shield composite from McCulloch and Wasserburg (1978) and Shaw et al. (1967).
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584 Figure 7. Plots of $\delta^{88/86}\text{Sr}$ versus a. pH; b. Ca/Na; c. Ca/Mg; d. Ca/1000Sr. The trends reflect the chemical
585 weathering of mafic versus silicate rocks. See text for details.

586 Figure 8. A compilation of river water stable Sr isotope data versus Ca/1000Sr. The bulk of the data
587 define sub-parallel positive correlations. These correlations reflect differing degrees of carbonate and
588 silicate weathering. Some data such as that from the Milford Sound LHR catchment illustrate the effects
589 of mixing between river water and soil water (Andrews et al. (2016)).

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Figure 1

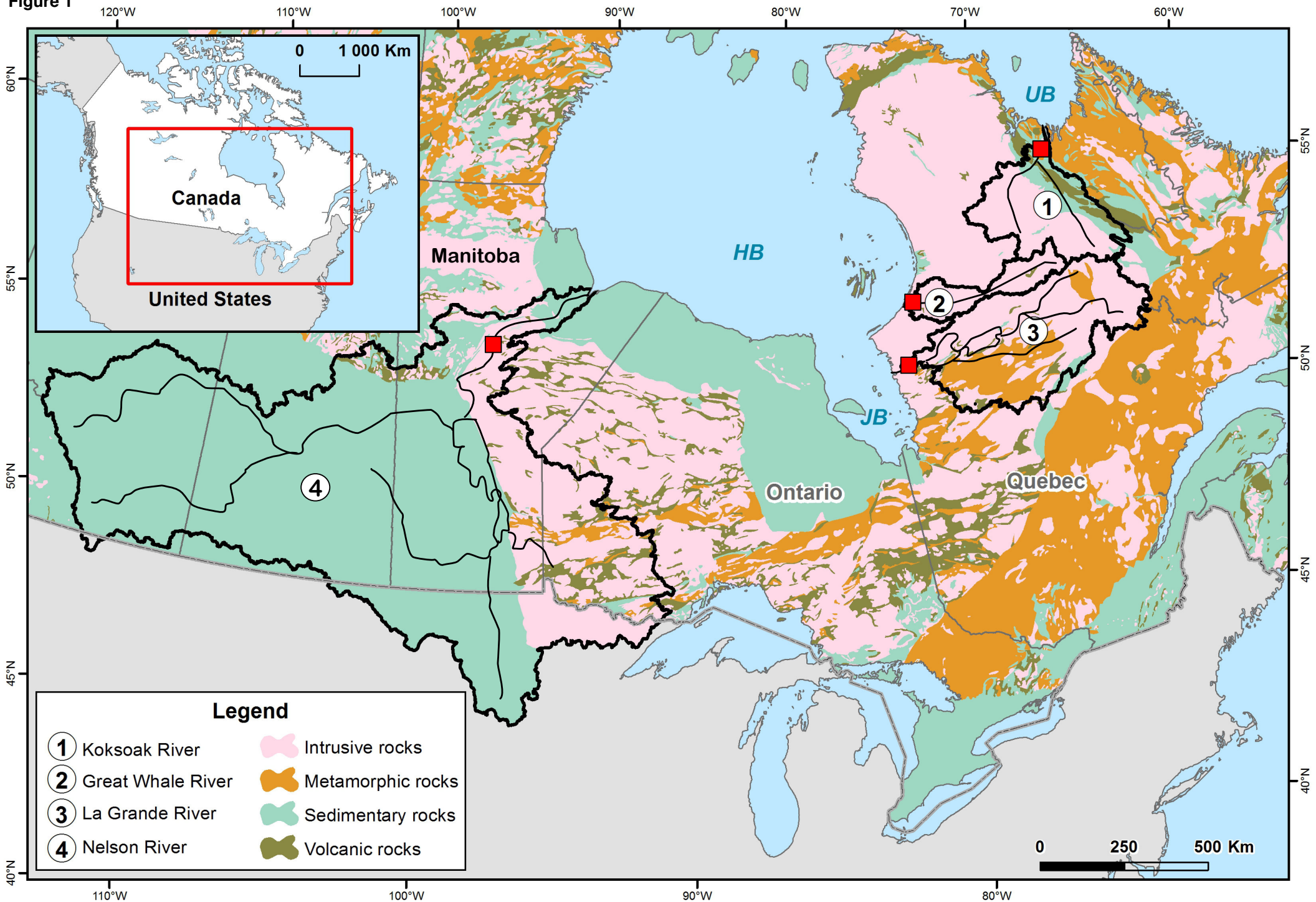


Figure 2

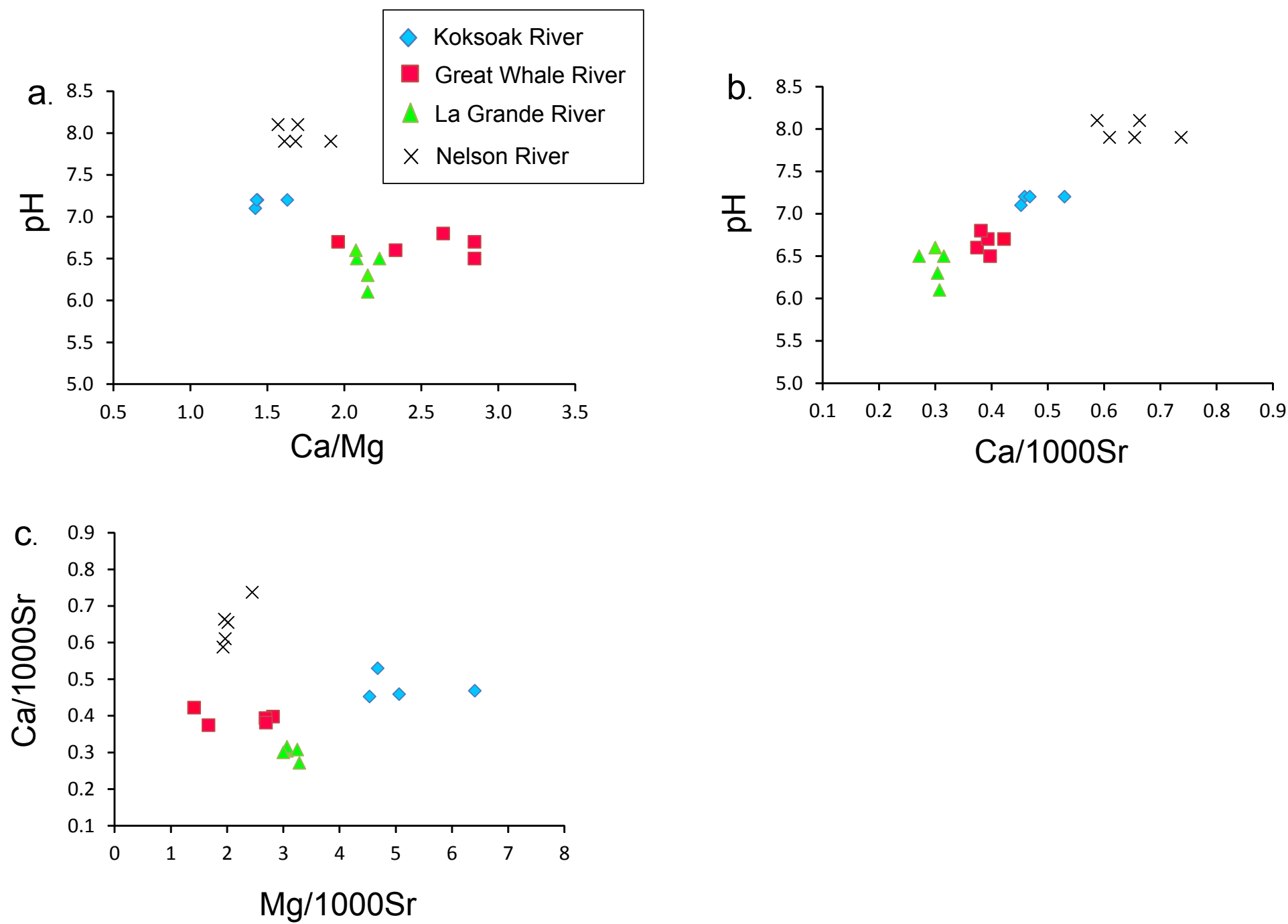


Figure 3

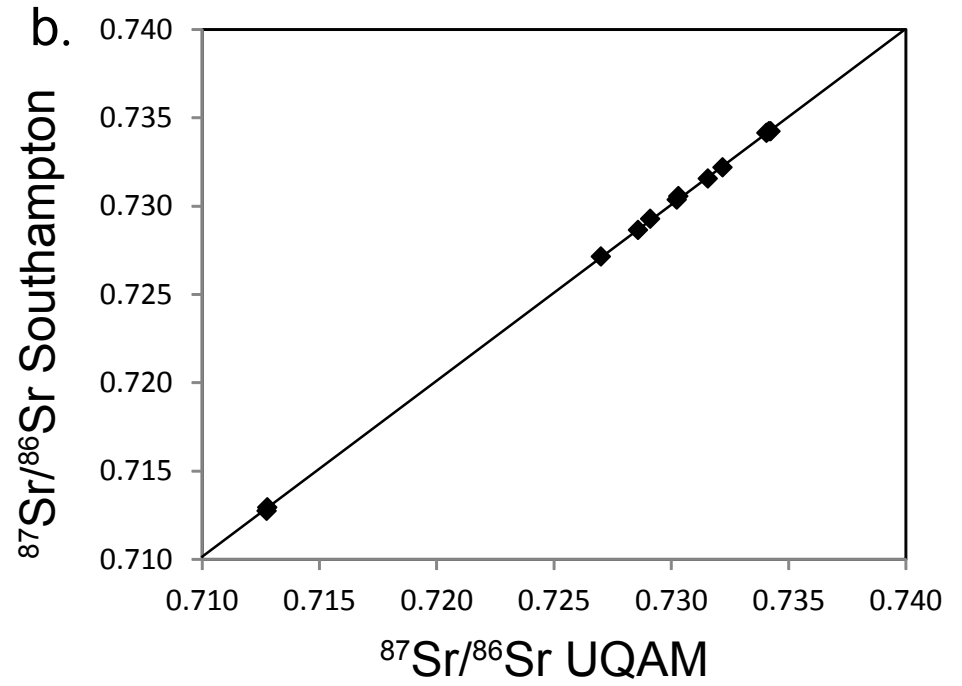
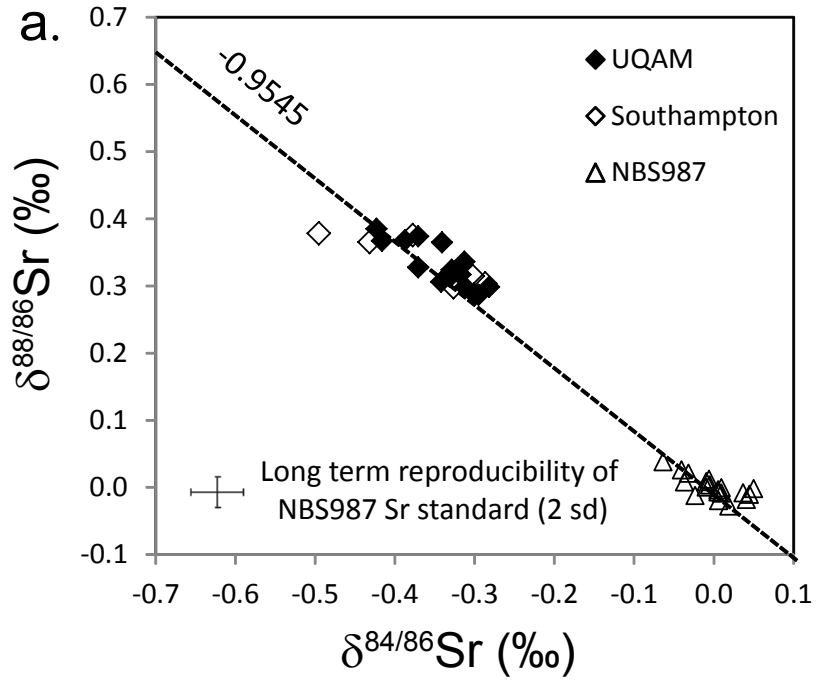


Figure 4

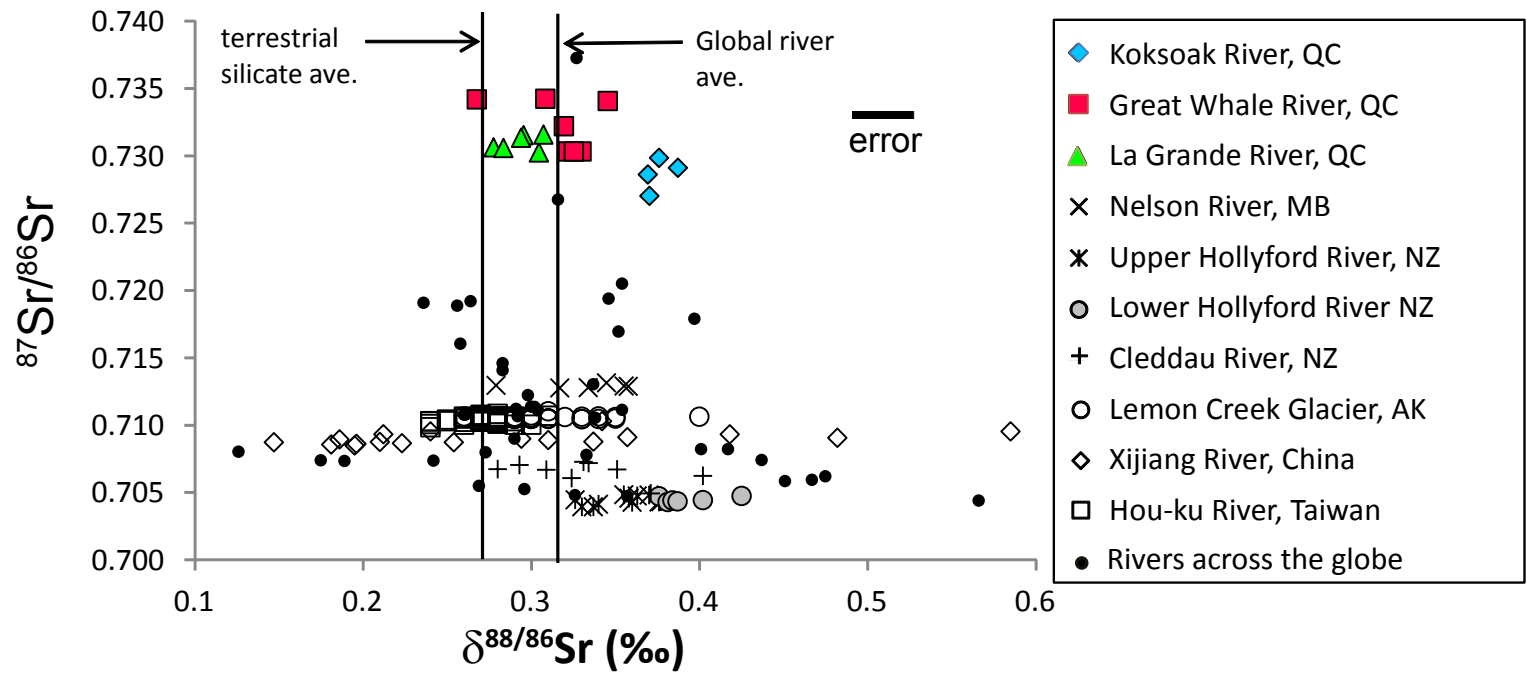


Figure 5

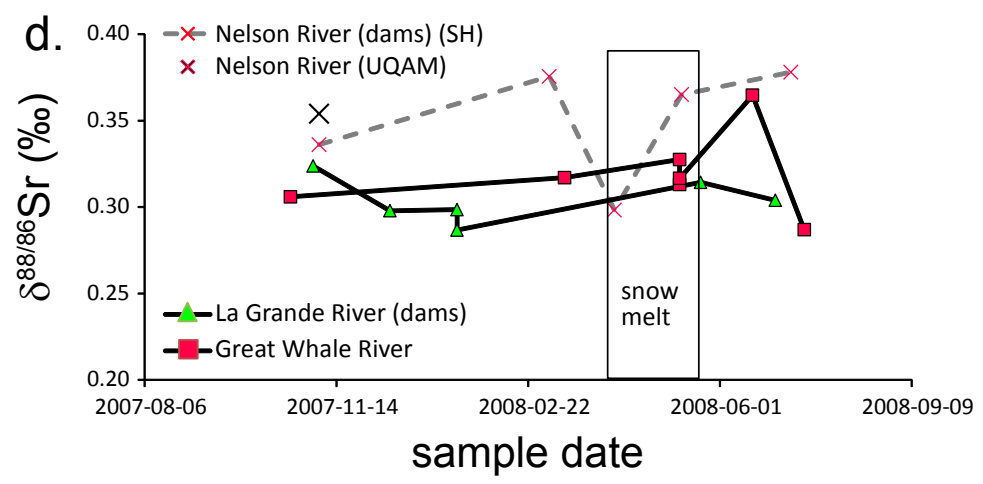
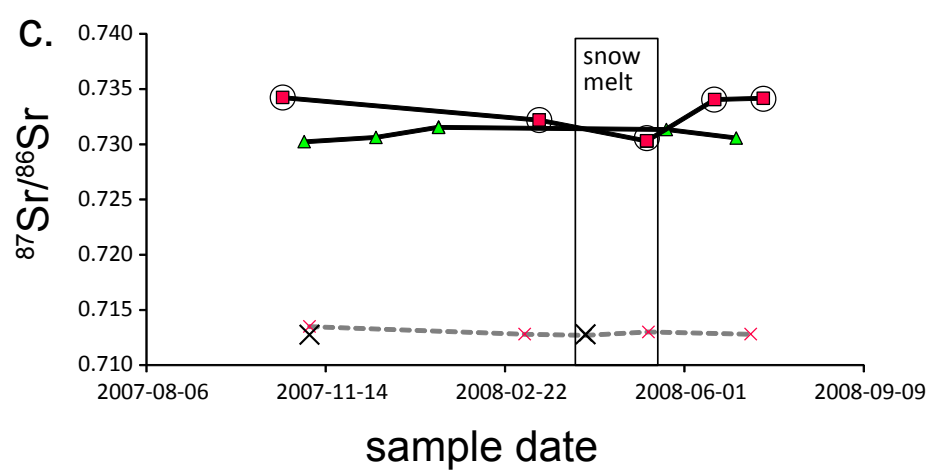
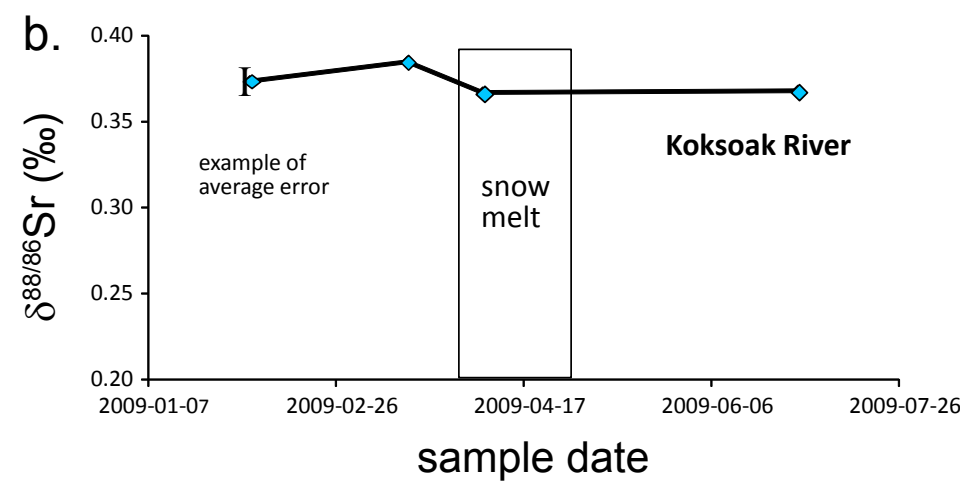
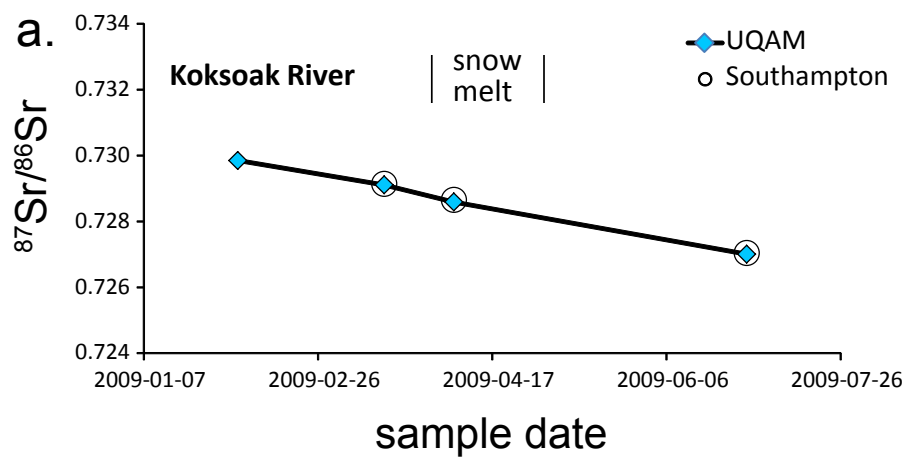


Figure 6

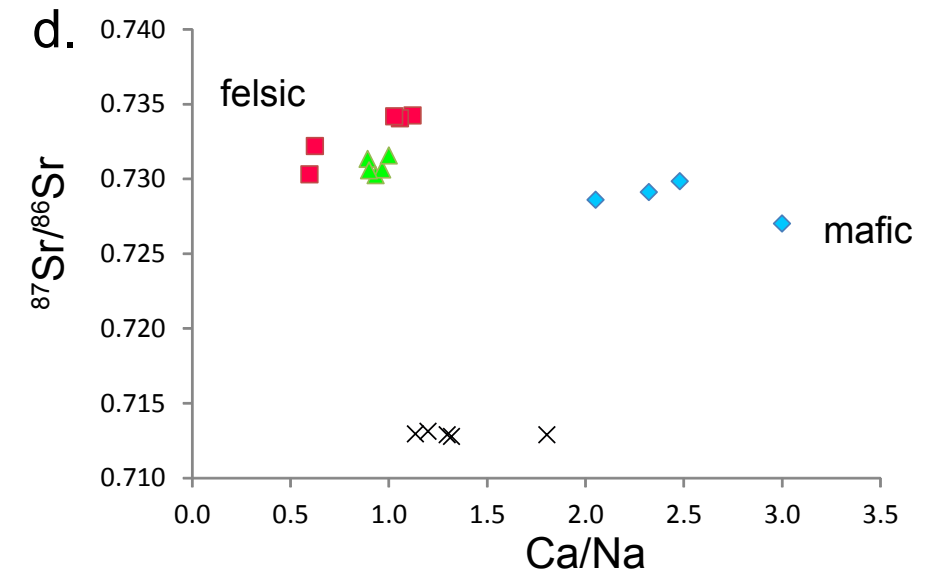
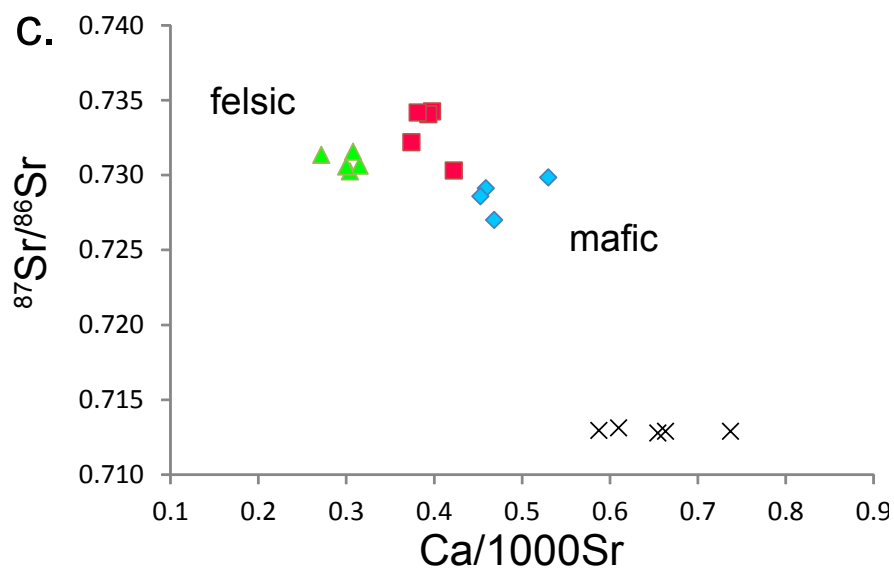
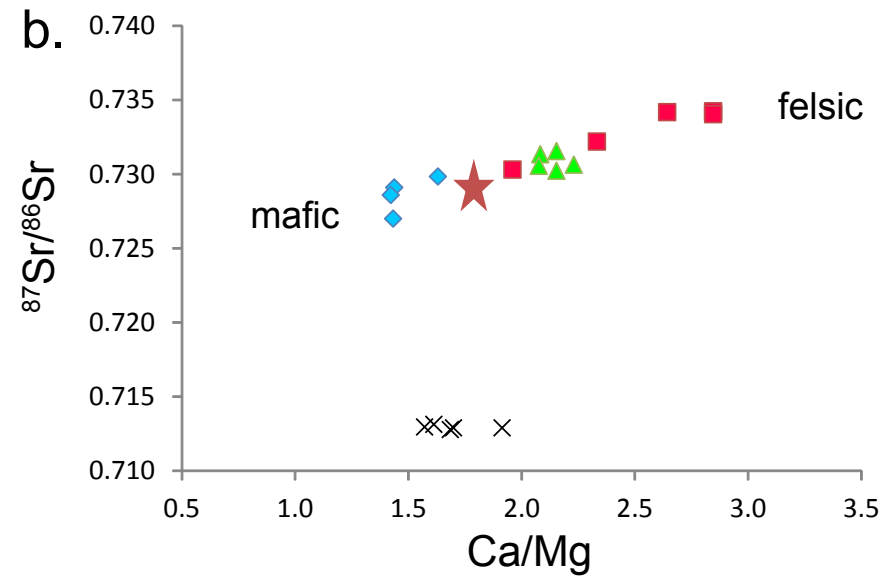
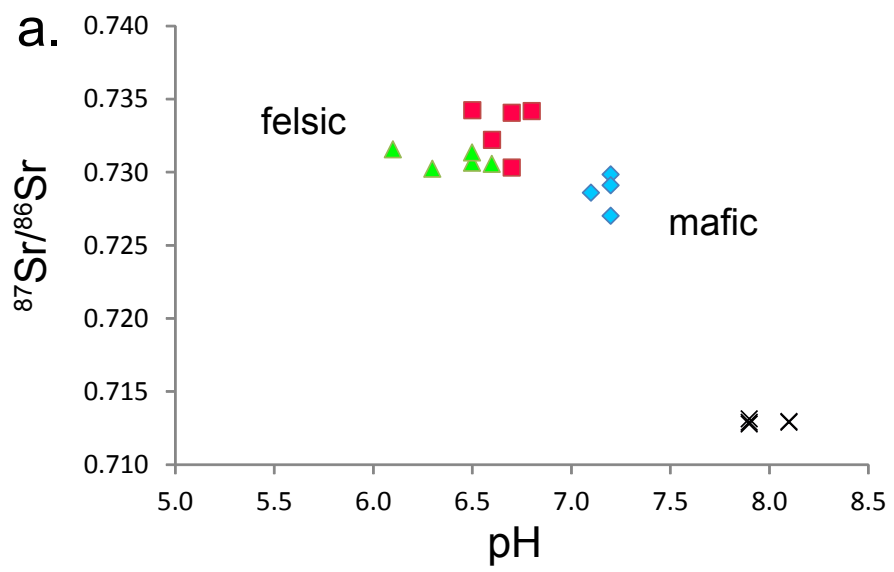


Figure 7

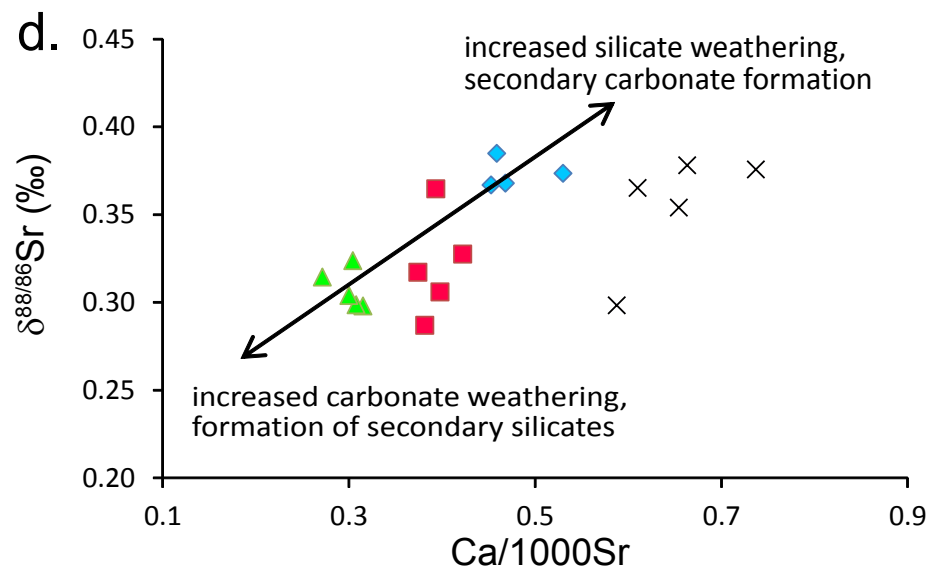
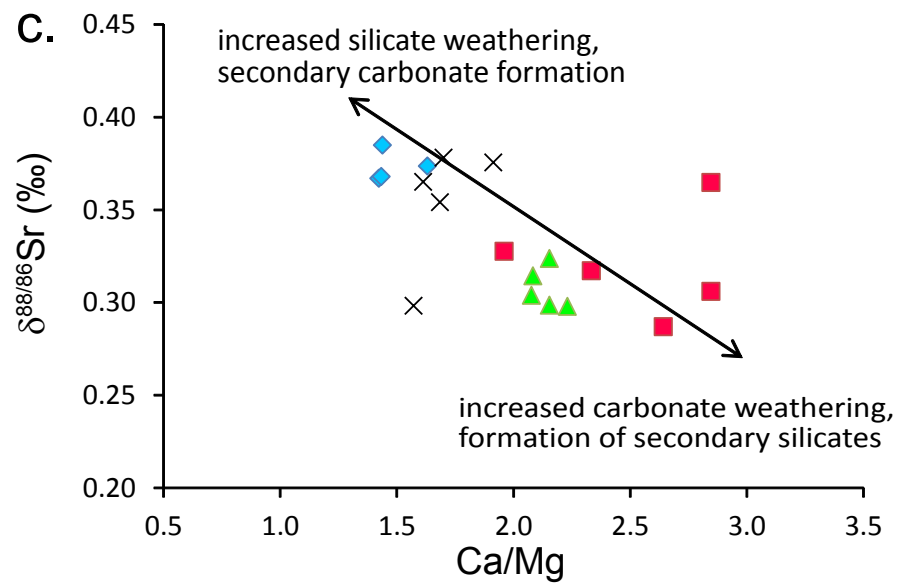
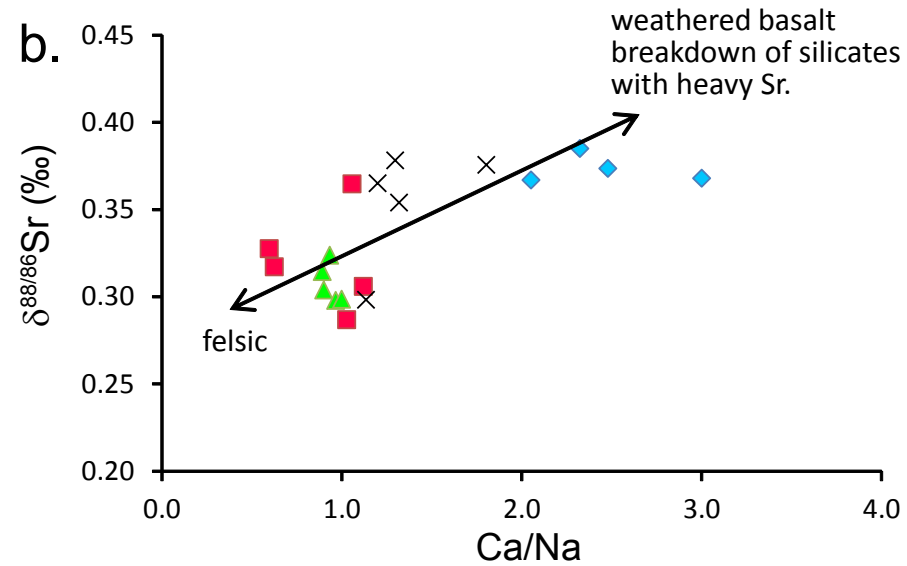
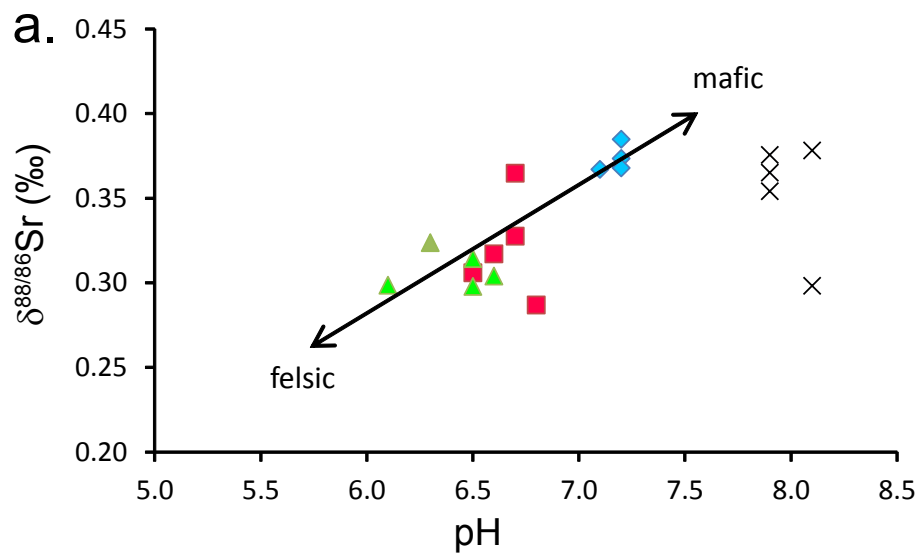


Figure 8

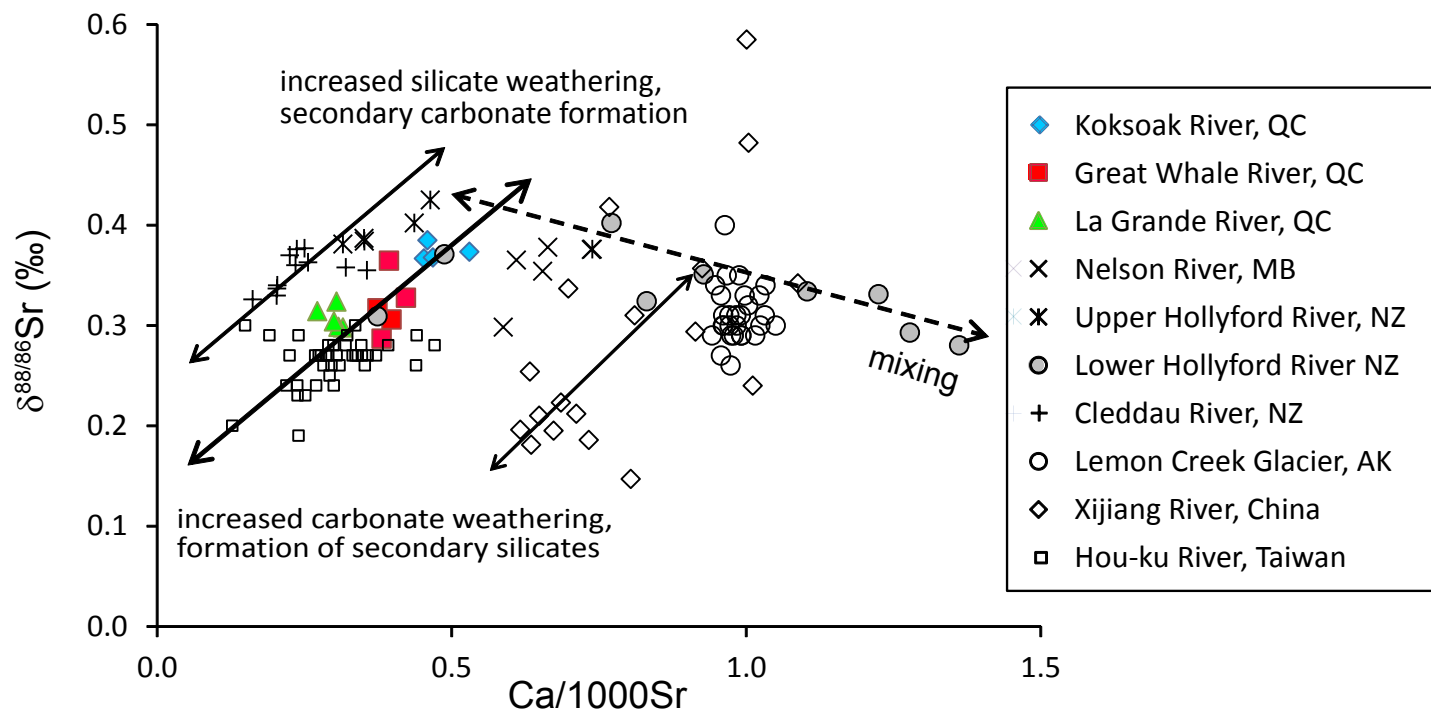


Table 1

[Click here to download Table: Table 1rev.docx](#)

Table 1. Radiogenic and stable Sr isotope data for four boreal rivers of Canada

Sample No	River	Sample date	Discharge Rate	UQAM		Southampton		UQAM		Southampton					
				$^{87}\text{Sr}/^{86}\text{Sr}$	2 σ error	$^{87}\text{Sr}/^{86}\text{Sr}$	2 σ error	$\delta^{84/86}\text{Sr}$	2 σ error	$\delta^{88/86}\text{Sr}$	2 σ error				
			m^3/s												
1	Koksoak	2009-02-03	377	0.729848	0.000004			-0.371	0.040	0.376	0.011				
2	Koksoak	2009-03-17	243	0.729107	0.000003	0.729276	0.000003	-0.423	0.034	0.387	0.014				
3	Koksoak	2009-04-06	227	0.728591	0.000003	0.728656	0.000003	-0.416	0.037	0.369	0.009				
4	Koksoak	2009-06-29	6335	0.727007	0.000003	0.727142	0.000003	-0.387	0.040	0.370	0.022				
5	Great Whale	2007-10-21	874	0.734236	0.000003	0.734236	0.000003	-0.342	0.052	0.308	0.011				
6	Great Whale	2008-03-12	179	0.732193	0.000004	0.732193	0.000004	-0.318	0.039	0.319	0.009				
7	Great Whale	2008-05-11	602	0.730302	0.000005	0.730558	0.000004	-0.371	0.051	0.330	0.018				
7	repeat	2008-05-11	602	0.730312	0.000004			-0.331	0.030	0.322	0.008				
7	Great Whale	2008-05-11	602	0.730310	0.000004			-0.318	0.034	0.325	0.008				
8	Great Whale	2008-06-18	1254	0.734051	0.000006	0.734157	0.000003	-0.341	0.054	0.345	0.012				
9	Great Whale	2008-07-15	716	0.734174	0.000005	0.734236	0.000004	-0.295	0.043	0.268	0.015				
10	La Grande	2007-11-02	n.d.i.	0.730234	0.000008	0.730363	0.000003	-0.328	0.062	0.305	0.013				
11	La Grande	2007-12-12	n.d.i.	0.730630	0.000003							-0.327	0.030	0.277	0.011
12	La Grande	2008-01-16	n.d.i.	0.731557	0.000003	0.731557	0.000003	-0.312	0.030	0.307	0.008				
12	repeat	2008-01-16	n.d.i.	0.731530	0.000003			-0.300	0.030	0.295	0.008				
13	La Grande	2008-05-22	n.d.i.	0.731341	0.000004							-0.302	0.035	0.294	0.009
14	La Grande	2008-06-30	n.d.i.	0.730567	0.000003							-0.287	0.027	0.283	0.008
15	Nelson	2008-04-07	3961	0.712783	0.000004	0.712946	0.000003	-0.282	0.044	0.279	0.023				
16	Nelson	2007-11-05	4421			0.712780	0.000003					-0.384	0.027	0.354	0.007
16	repeat	2007-11-05	4421	0.712752	0.000010			-0.313	0.094	0.317	0.019				
17	Nelson	2008-03-04	4154			0.712896	0.000003					-0.378	0.026	0.355	0.007
18	Nelson	2008-05-12	3541			0.713124	0.000003					-0.432	0.030	0.345	0.008
19	Nelson	2008-07-08	4041			0.712886	0.000003					-0.495	0.035	0.358	0.008

Latitude/longitude sampling sites for Koksoak River (58.029/68.475), Great Whale River (55.279/77.650), La Grande River (53.781/77.530) and Nelson River (56.685/93.790).

n.d.i. : discharge rates are confidential as per non-disclosure agreement with Hydro-Quebec.