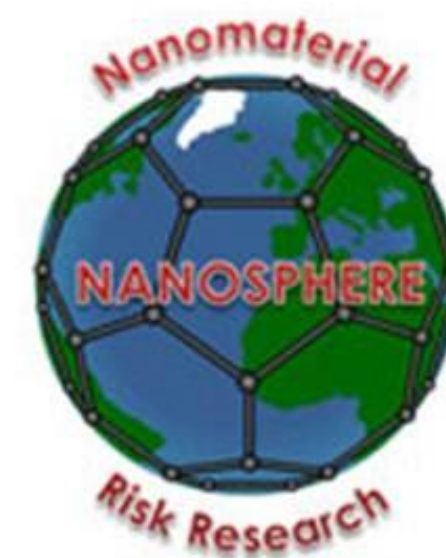




Marianne Matzke^{1,4}, Julian Gallego², Martin Hassellöv², Kerstin Jurkschat³, Thomas Backhaus⁴
¹ Centre of Ecology and Hydrology (CEH), Maclean Building, Crowmarsh Gifford, United Kingdom
² Department of Chemistry, University of Gothenburg, Sweden
³ Department of Materials, Oxford University, Begbroke Science Park, United Kingdom
⁴ Department of Biological and Environmental Sciences, University of Gothenburg, Sweden
 e-mail: martzk@ceh.ac.uk



INTRODUCTION

- ❑ Silver nanoparticles (AgNPs) are one of the particles types with the highest production volume.
- ❑ Widespread use of AgNPs will lead to environmental release and their beneficial/wanted microbiocidal effect might become problematic in the natural aquatic environment
- ❑ Especially algae & bacteria might be put at risk.
- ❑ Predicted concentrations of AgNPs in surface water range from 0.1 to 1 nmol/L (Gottschalk et al. 2009, 2010, Mueller & Nowack 2008).

AIM - to answer the following questions...

- Effects on microbial freshwater communities at environmentally relevant concentrations?
- Are there any nano-specific effects?
- Are their differences in the sensitivity of biofilm algae and bacteria?
- Toxicant induced changes in the community structure?

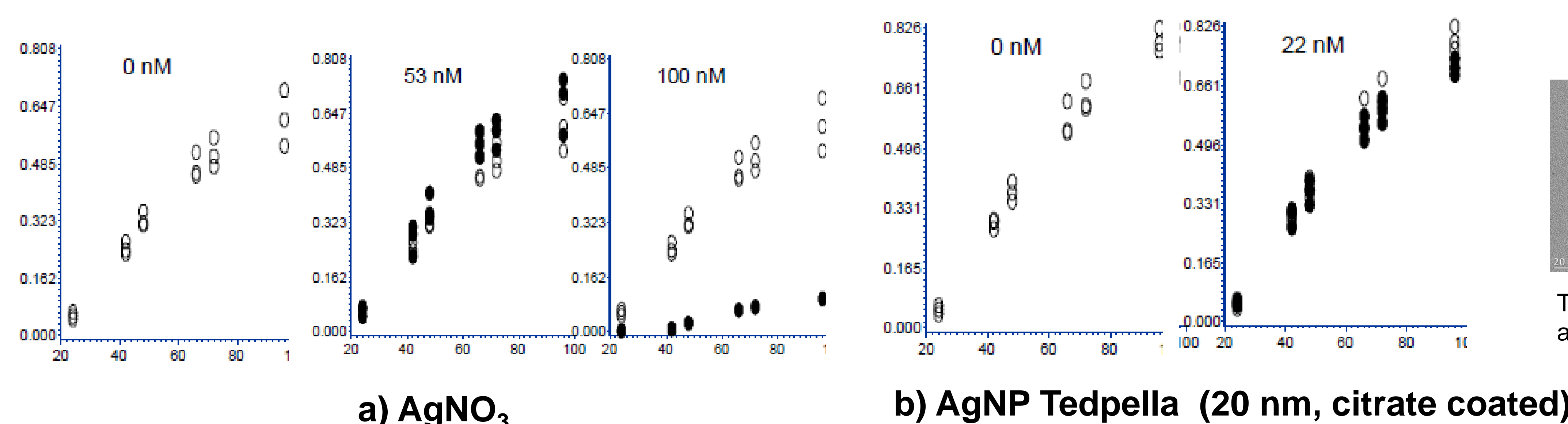
METHODS

SWIFT periphyton assay according to Porsbring et al., 2007

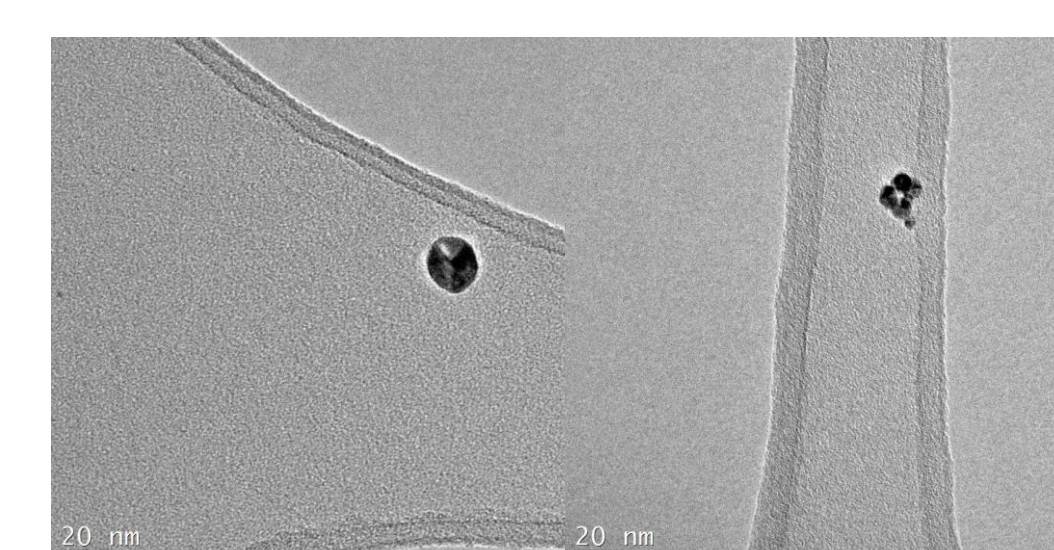
- ❑ Colonisation of glass discs by natural communities of algae and bacteria for 7-10 days in a stream.
- ❑ Exposure to the toxicant in the lab for 96 hours under controlled semi-static conditions (light:dark cycle 16:8 hours & ambient temperature).
- ❑ Structural analysis of the algal & the bacterial part of the community:
 Catabolic profiling & physiological activity of the bacteria using the Ecolog approach = ability to metabolise different carbon sources.
 HPLC pigment profiling: Chl a content as a biomass indicator & changes in pigment patterns indicating changes in species composition and interferences with specific physiological processes.
- ❑ Analytical verification of the exposures with ICP-MS/ultrafiltration (total and dissolved silver) & NTA (agglomeration, NP behaviour, size).

RESULTS

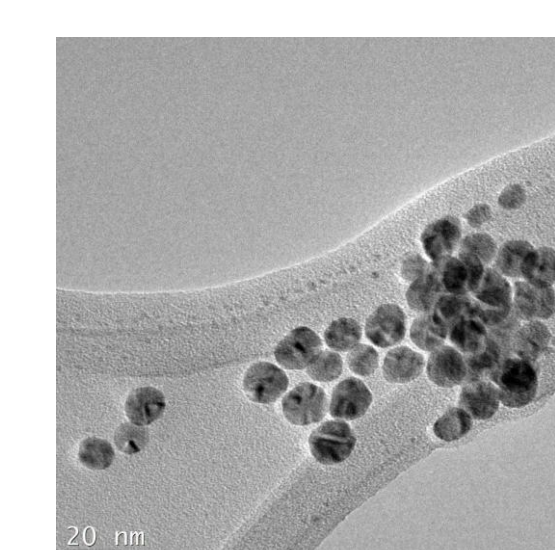
- Bacteria reacted more sensitive than algae.
- Bacterial part (effects based on average well colour development (AWC), Fig. 1):
AgNP NM-300K (20 nm) > 0.5 nM, **AgNP TP** (20 nm) > 22 nM, **AgNO₃** between 53 nM – 100 nM.
- Algal part (effects based on Chl a content, Fig. 2):
AgNO₃ effects > 150 nM, **AgNP TP** (20 nm) > 300 nM and **AgNP NM-300K**: > 1300 nM.
- Comparing AgNO₃ with AgNPs: AgNO₃ more toxic for the algae, AgNPs more toxic for the bacteria.
- Analytics suggest differences between nominal and real concentrations in terms of total and dissolved silver.



Testing of two different spherical AgNPs in reference to AgNO₃



Tedpella 20 nm, citrate coated, aqueous suspension (AgNP TP)



OECD particles NM-300K, 20 nm, aqueous suspension with stabilizers (Tween 20) (AgNP NM-300K)

CONCLUSIONS & OUTLOOK

- ✓ Bacteria more sensitive than algae with effects for the NM-300K in the range of environmental realistic concentrations.
- ✓ Analytics crucial for a correct interpretation of the ecotoxicological results— otherwise underestimation of the AgNPs effects and hazards.
- ✓ Indications for nanospecific effects: higher toxicity of the AgNPs than the AgNO₃ for the bacteria.
- ✓ Link of the ecotoxicological and analytical data (ICP-MS, NTA and Minteq modelling).

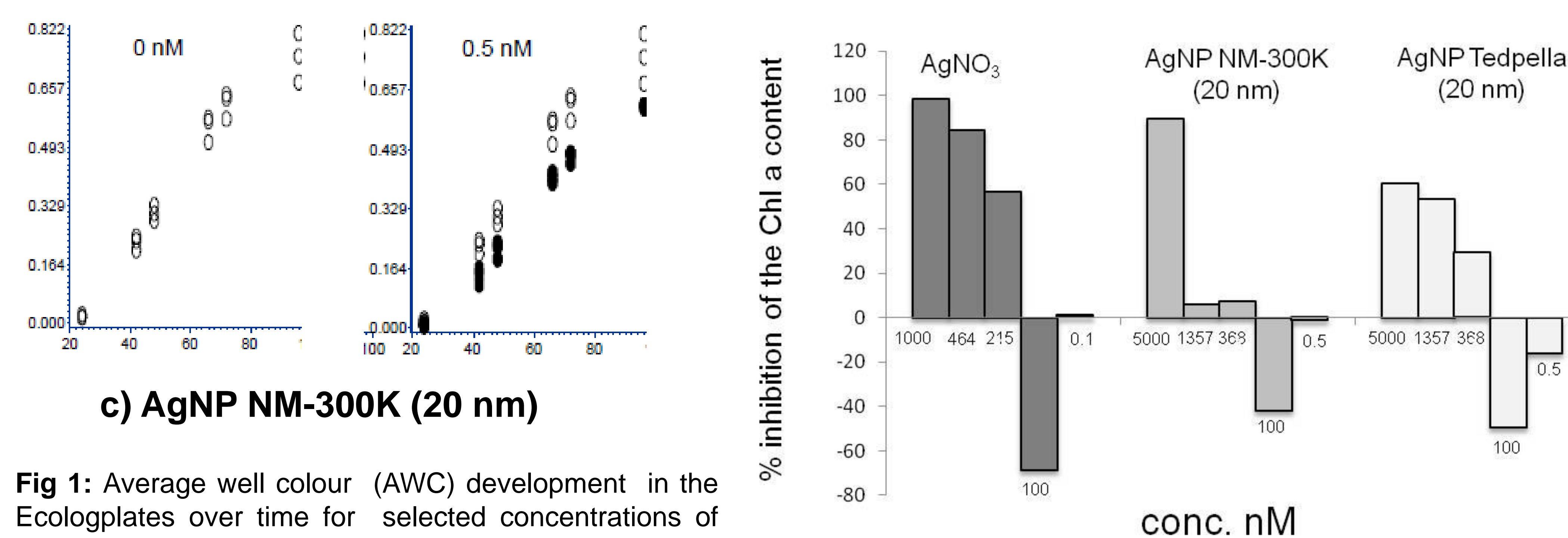
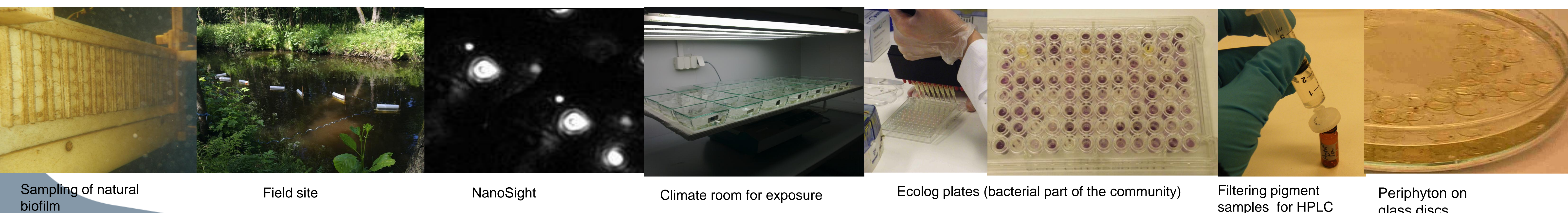


Fig. 2: % Inhibition of the Chlorophyll a (Chl a) content for 5 different concentrations of AgNO₃ and AgNPs



References

Gottschalk, F. et al., 2009. Modeled environmental concentrations of engineered nanomaterials (TiO₂, ZnO, Ag, CNT, Fullerenes) for different regions. Environmental science & technology, 43(24)
 Gottschalk, F. et al., 2010. Possibilities and limitations of modeling environmental exposure to engineered nanomaterials by probabilistic material flow analysis. Environmental toxicology and chemistry, 29(5)
 Mueller, N.C. & Nowack, B., 2008. Exposure modeling of engineered nanoparticles in the environment. Environmental science & technology, 42(12), pp.4447-53.

Acknowledgements

Funded by Nanosphere (FORMAS, The Swedish Research Council for Environment, Agricultural Sciences and Spatial Planning) and Nanofate (EU Commission, FP7)