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1 **Algal Evolution in Relation to Atmospheric CO₂: Carboxylases, Carbon Concentrating**
2 **Mechanisms and Carbon Oxidation Cycles**

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15 **Abstract.**

16 Oxygenic photosynthesis evolved at least 2.4 Ga ago; all oxygenic organisms use the ribulose
17 biphosphate carboxylase-oxygenase (Rubisco) - photosynthetic carbon reduction cycle
18 rather than one of the five other known pathways of autotrophic CO₂ assimilation. The high
19 CO₂ and (initially) O₂-free conditions permitted use of a Rubisco with a high maximum
20 specific reaction rate. As CO₂ decreased and O₂ increased Rubisco oxygenase activity
21 increased and 2-phosphoglycolate was produced, with evolution of pathways recycling this
22 inhibitory product to sugar phosphates. Changed atmospheric composition also selected for
23 Rubiscos with higher CO₂ affinity and CO₂/O₂ selectivity correlated with decreased CO₂-
24 saturated catalytic capacity and/or for CO₂ concentrating mechanisms (CCMs). These
25 changes increase the energy, nitrogen, phosphorus, iron, zinc and manganese cost of
26 producing and operating Rubisco - PCRC while biosphere oxygenation decreased the
27 availability of nitrogen, phosphorus and iron. The majority of algae today have CCMs; the
28 timing of their origins is unclear. If CCMs evolved in a low CO₂ episode followed by one or
29 more lengthy high CO₂ episodes, CCM retention could involve a combination of
30 environmental factors known to favour CCM retention in extant organisms which also occur
31 in a warmer high-CO₂ ocean. More investigations, including studies of genetic adaptation, are
32 needed.

33 **Keywords** CO₂ concentrating mechanism – combined nitrogen – inorganic carbon – iron –
34 mixing depth – phosphorus – photorespiration - photosynthetically active radiation - Rubisco
35 – temperature – UVA – UVB

36 **Introduction**

37 Algae are oxygenic photosynthetic organisms other than embryophytic plants, and by this
38 definition include the cyanobacteria as well as a wide range of eukaryotic lineages.
39 Cyanobacteria, as indicated by the occurrence of oxygenic photosynthesis, evolved at least
40 2.4 Ga ago, although fossil (including chemical biomarker) evidence for cyanobacteria does
41 not go back beyond 2.1 Ga ago. [1]. Eukaryotic algae have occurred for not less than 1.2 Ga
42 ago [2-4] and from freshwaters, and possibly lake margins, since 1.1 Ga ago [5] (Table 1).
43 Since 2.4 Ga ago the biosphere has become increasingly oxygenated [1] reflecting the
44 colonization of the oceans by cyanobacteria after their origin in freshwater habitats, with a
45 corresponding increase in the capacity of these organisms to have global biogeochemical
46 influence [6, 7] (Table 1). A significant increase in oxygen, with oxygenation of the deep
47 ocean, occurred in the Neoproterozoic 0.54 – 1 Ga ago [8] with variations in the Phanerozoic
48 including the highest known level in the Permo-Carboniferous glaciation [9]. Carbon dioxide
49 varied with a general downward trend with minima generally related to glaciations [10, 11].
50 CO₂ was relatively constant at about 23 (range of estimates 10 – 100) times the present level
51 between 2.5 and 1.8 Ga ago, with a very significant decrease between 1.8 and 1.1 Ga ago
52 [12], further variations in the Neoproterozoic [13-15] and relatively well established changes
53 in Phanerozoic with minima in the Permo-Carboniferous and in the Pleistocene glaciations
54 [9, 16]. In this article we consider how these environmental changes have influenced algal
55 evolution, both through direct effects of the concentrations of CO₂ and O₂ on photosynthesis
56 and related metabolism, and through indirect effects of changes in temperature (CO₂) and the
57 availability of combined nitrogen, phosphorus and iron (CO₂ and O₂).

58 **Autotrophic Carboxylases**

59 Six pathways of autotrophic CO₂ fixation are known in extant organisms, including ribulose
60 biphosphate carboxylase-oxygenase carboxylase activity in the photosynthetic carbon
61 reduction cycle (Rubisco – PCRC), using CO₂ as the inorganic carbon species assimilated,
62 which is at the core of inorganic carbon assimilation in extant oxygenic photosynthetic
63 organisms [17-22]. These pathways are summarised in Table 2 with respect to their
64 stoichiometric requirement for reductant and ATP, their affinities for inorganic carbon
65 expressed in terms of the half-saturation value for CO₂ and the influence of oxygen on their
66 functioning.

67 Converting one CO₂ to the oxidation-reduction level of carbohydrates (CH₂O) requires four
68 reducing equivalents at, or lower than, the midpoint redox potential of the NADPH.NADP⁺
69 couple. The values given in Table 2 indicate this minimal stoichiometry, assuming no redox
70 side reactions or futile cycles. An example of a side reaction for the Rubisco – PCRC
71 pathway is the Rubisco oxygenase activity in the photosynthetic carbon oxidation cycles(s)
72 (Rubisco-PCOC) which occurs at a relatively low [CO₂]/[O₂] ratio. All of the autotrophic
73 CO₂ fixation pathways also require ATP (Table 2), with a range of stoichiometries from one
74 to three ATPs for CO₂ converted to carbohydrate. The Rubisco – PCRC pathway has the

75 equal highest energy cost of converting CO₂ to carbohydrate, with the further energy cost of
76 the side-reaction of the Rubisco – PCOC at low [CO₂]/[O₂] ratios.

77 The Rubisco – PCRC pathway seems more appropriate to CO₂ fixation at the present
78 atmospheric CO₂ level when the half-saturation value for CO₂ is considered. The forms of
79 Rubisco with the highest affinities for CO₂ (Form IB from some algae and vascular plants
80 relying on CO₂ diffusion to Rubisco; Form ID from other algae) have half-saturation values
81 for CO₂ almost as low as those of two other pathways, while the values for the other three
82 pathways are considerably higher (Table 2). The final criterion in Table 1 is the effect of
83 oxygen on the pathways. While the Rubisco – PCRC pathway is competitively (with CO₂)
84 inhibited by O₂, the enzymes of this pathway are not damaged by O₂: some of the other
85 pathways have one or more enzymes which are subject to irreversible inhibition by O₂ (Table
86 2). Not shown in Table 2, for lack of accurate information, is the resource cost of
87 synthesising the enzymic machinery needed to fix one mole of CO₂ per second from the
88 present atmospheric CO₂ concentration. This value is a function of the stoichiometry of the
89 enzymic protein components in the pathway, with that of the carboxylase(s) set by their CO₂
90 affinity, and the M_r (relative molecular mass) values of the enzymes [17, 18]. Here the
91 relatively low specific reaction rate and high M_r of Rubisco would tend to make the Rubisco
92 – PCRC pathway expensive at present CO₂ concentrations, even without the O₂ effect, though
93 some other pathways are probably at least as expensive as a result of the low CO₂ affinity of
94 their carboxylases and the consequent large quantity of carboxylase needed to fix one mole of
95 CO₂ per second from the present atmospheric CO₂ concentration.

96 The occurrence of Rubisco – PCRC as the core carboxylase in oxygenic photosynthetic
97 organisms can be related to opportunity and functionality. By opportunity is meant the
98 occurrence of the pathway at the time (about 2.4 Ga ago) at which the earliest evidence for
99 oxygenic photosynthetic organisms is known. The Rubisco – PCRC pathway originated
100 before oxygenic photosynthesis evolved (see below). By functionality is meant the
101 carboxylase CO₂ affinity and carboxylase M_r values as determinants of the quantity of
102 carboxylase (mass of protein) needed to fix one mole of CO₂ per second from the current
103 atmosphere, the extent to which O₂ competitively inhibits or damages the enzymes of the
104 pathway, and the ATP cost of the pathway per CO₂ assimilated (Table 2). The CO₂ affinity
105 criterion apparently rules out three of the five pathways (two of which are also very oxygen-
106 sensitive), leaving the 3-hydroxypropionate pathway and the Rubisco – PCRC pathway.
107 While not oxygen inhibited in the manner found for Rubisco, the 3-hydroxypropionate
108 pathway may be sufficiently O₂ sensitive to restrict its functionality in oxygenic organisms
109 once O₂ had begun to accumulate in the part of the biosphere occupied by oxygenic
110 photosynthetic organisms (Tables 1 and 2). In such ways can the role of the Rubisco – PCRC
111 pathway in all known oxygenic photosynthetic organisms be rationalised.

112 **Rubisco carboxylase activity and the PCRC**

113 Rubisco evolved before the origin of oxygenic photosynthesis [23-25]. The Rubisco gene
114 family not only contains the Forms I, II and III Rubiscos which catalyse the Rubisco

115 carboxylase and Rubisco oxygenase reactions, but also the Form IV Rubisco-Like Protein
116 (RLP) which does not catalyse the typical Rubisco reactions and which is involved in
117 methionine salvage in some Bacteria [23-25]. Molecular phylogenetic analysis [23-25]
118 suggests that an ancestral Form III Rubisco arose in a methanogen (i.e. a member of the
119 Archaea) and gave rise, by two vertical transmissions, to all other Form III Rubiscos and to
120 Form IV. Horizontal gene transfer then moved Form III and Form IV Rubiscos to an
121 ancestral bacterium where Form III gave rise to the ancestors of Forms I and II, and hence by
122 vertical transmission to the Form IV RLP of Bacilli and to Forms I, II and IV of an ancestral
123 proteobacterium.

124 Vertical descent and further diversification gave rise to the Forms IA, IC, ID and II, and the
125 Form IV RLP in extant Proteobacteria. Horizontal gene transfer from the Proteobacteria
126 transferred Form IA to cyanobacteria where Form IB evolved from Form IA [23-26].
127 However, there is also evidence that extant α -cyanobacteria with Form IA Rubisco (α -
128 cyanobacteria) acquired their Form IA Rubisco (and other α -carboxysomal proteins) from a
129 proteobacterium, displacing the Form IB Rubisco and associated β -carboxysomal proteins
130 [27-32]. Endosymbiosis of a heterocystous β -cyanobacterium [33] gave rise to the plastids of
131 the Archaeplastida (= Plantae) where the Form IB Rubisco of the endosymbiont was
132 retained by glaucocystophyte and green algae and hence by vertical transfer to embryophytic
133 ('higher') plants, and was moved by secondary endosymbiosis to the (rhizarian)
134 chlorarachniophytes and the (excavate-discicristate) euglenoids. Horizontal gene transfer
135 from a proteobacterium accounts for the presence of Form ID Rubisco in red algae and in the
136 chromistan algae (ochrista or heterokontophytes, cryptophytes and haptophytes) whose
137 plastids arose by secondary endosymbiosis of a red alga. The peridinin-containing
138 dinoflagellate and basal apicomplexans (represented today by the photosynthetically
139 competent *Chromera velia* and an as yet un-named close relative) also obtained their plastids
140 by secondary endosymbiosis from a red alga, but horizontal transfer of Form II Rubisco from
141 a proteobacterium replaced the Form ID Rubisco [34]. Other dinoflagellates have other forms
142 of Rubisco as a result of tertiary endosymbioses. Finally, a second primary endosymbiosis
143 involving an α -cyanobacterium with Form IA Rubisco accounts for the presence of plastids
144 and Form IA Rubisco in the (rhizarian) euglyphid amoeba *Paulinella* [30, 31].

145

146

147

148 Some of the enzymes of the PCRC of vascular plants are derived from cyanobacteria in the
149 primary endosymbiosis leading to the Archaeplastida (= Plantae), while others are of host
150 origin [35]. This work has now been extended to include the glaucocystophyte and red algal
151 members of the Archaeplastida [36] and the diatoms whose plastids arose from secondary
152 endosymbiosis of a red alga [37] following an earlier secondary endosymbiosis of a green
153 alga [38]. There are also differences among algae in the regulation of enzymes of the PCRC
154 [39]. Further work is needed to establish the relevance, if any, of atmospheric changes to

155 these differences in the regulation of enzymes of the PCRC [39, 40], and also the absence of
156 Rubisco activase from algae with the Form ID Rubisco in the few cases examined [37].

157 It is likely that the earliest Rubiscos in chemolithotrophic and anoxygenic photosynthetic
158 Archaea and Bacteria were operating in a high CO₂ environment and, because there was no
159 oxygenic photosynthesis, in the essential absence of O₂. Such an environment would have
160 provided little or no selective pressure for a high CO₂ affinity or a high CO₂/O₂ selectivity
161 relative to a high CO₂-saturated maximum catalytic rate, using the mechanistic arguments of
162 [41] that a high maximum catalytic rate is incompatible with high CO₂ affinity and higher
163 CO₂/O₂ selectivity.

164 The early organisms using Rubisco as a carboxylase would, then, be able to use diffusive
165 entry of CO₂, with none of the resource costs associated with Rubisco operating at below the
166 CO₂-saturated rate in the presence of O₂ when Rubisco oxygenase activity is expressed. The
167 additional resource costs can be considered as capital (synthetic) costs and running costs. The
168 capital costs are those of synthesising additional Rubisco enzyme per cell if the per cell rate
169 of CO₂ assimilation is to be retained, as well as the costs of making the enzymes and
170 transporters related to the operation of a PCOC and/or a CO₂ concentrating mechanism
171 (CCM), and include energy, carbon and nitrogen as well as phosphorus for any additional
172 ribosomes that are required [42]. The running costs are in terms of energy, both for
173 synthesising 2-phosphoglycolate and metabolising it back to sugar through a PCOC and for
174 operating a CCM.

175 The early Rubisco-containing organisms could still permit CO₂ saturation of a high specific
176 reaction rate Rubisco, with corresponding savings of resources in constructing the enzymic
177 machinery able to assimilate one mole of CO₂ per second. In other words, the conditions
178 described would give the lowest possible energy, carbon and nitrogen costs of producing the
179 amount of Rubisco capable of fixing one mole of CO₂ per second, and also give the lowest
180 possible energy cost of operating Rubisco, i.e. 2 NADPH and 3 ATP per CO₂ converted to
181 carbohydrate. A correlated saving associated with this minimal requirement for NADPH and
182 ATP in autotrophic CO₂ assimilation is in the quantity of redox and ATP synthesis machinery
183 needed for NADPH and ATP production. Minimizing the protein requirements for autotrophy
184 also decreases the phosphorus requirement for mRNA in ribosomes and in tRNA and mRNA
185 needed to produce the Rubisco and associated enzymes. These characteristics would
186 minimize the nitrogen needed to produce the machinery associated with a given rate of CO₂
187 fixation, and thus the phosphorus in RNA needed to synthesise the proteins [42]. Costs in
188 energy, nitrogen, phosphorus, iron and manganese will be considered below in the context of
189 decreasing CO₂ and increasing O₂ [42-50].

190 The mechanistically constrained co-variation in Rubisco kinetics mentioned above, i.e. a high
191 CO₂-saturated specific reaction rate correlated with low CO₂ affinity and CO₂/O₂ selectivity
192 and vice versa [41] has parallels with the growth strategies of vascular plants [51] and of
193 algae [52] according to the Competitive – Stress-Tolerant – Ruderal (CSR) paradigm, for
194 which there are also mechanistic bases in terms of having high rates of metabolism and

195 growth in ruderals and lower rates but perhaps a more effective use of resources in stress-
196 tolerators [53-55].

197 **Oxygen accumulation, Rubisco oxygenase and the metabolism of phosphoglycolate**

198 The build-up of O₂ has had many effects on algal evolution, permitting respiration and, via
199 the occurrence of stratospheric ozone, a decreased UVB flux, and production of reactive
200 oxygen species from O₂ rather than UVB action on cell constituents [55, 56]. The presence of
201 O₂ does not, however, inhibit O₂ production by oxygenic photosynthetic organisms [57, 58].
202 The accumulation of O₂ in the habitats of oxygenic photosynthetic organisms using Rubisco
203 as their autotrophic carboxylase permits Rubisco oxygenase activity to occur, provided the
204 CO₂ concentration is below saturation for Rubisco. Such decreased CO₂ concentrations,
205 combined with at least local O₂ accumulation, probably occurred about 2 Ga ago, when there
206 is evidence of an ice age extending to low palaeolatitudes [12, 59].

207 The product of the Rubisco oxygenase activity is, as well as one 3-phosphoglycerate per O₂
208 consumed, one 2-phosphoglycolate. In addition to sequestering the often limiting resource
209 phosphorus if 2-phosphoglycolate continues to accumulate, 2-phosphoglycolate is also an
210 inhibitor of some reactions involving phosphate esters, including some in the PCRC [60].
211 Accordingly, all organisms using Rubisco in the presence of O₂, i.e. oxygenic
212 photolithotrophs and some chemolithotrophs, have 2-phosphoglycolate phosphatase [60, 61].
213 This enzyme would not have been needed to deal with 2-phosphoglycolate from Rubisco
214 oxygenase activity in anoxygenic photosynthetic proteobacteria, or in any anaerobic
215 chemolithotrophs, using the Rubisco-PCR pathway before build-up of O₂ in the biosphere. 2-
216 phosphoglycolate phosphatase also occurs in some non-autotrophic bacteria which have no
217 Rubisco, where it is thought to be involved in some forms of DNA repair [62, 63]. Since
218 DNA damage and its repair must have occurred before oxygenic photosynthesis, the 2-
219 phosphoglycolate phosphatase in oxygenic photosynthetic organisms could have been
220 recruited from bacteria lacking autotrophic 2-phosphoglycolate synthesis. However, the 2-
221 phosphoglycolate phosphatase from eukaryotes does not seem to have been derived from the
222 2-phospho-glycolate phosphatase of cyanobacteria [64].

223 The organic carbon product of 2-phosphoglycolate phosphatase is glycolate. This can be
224 excreted, with loss from the organism of the energy and carbon used in its synthesis.
225 Alternatively, glycolate can be salvaged by metabolism to 3-phosphoglycerate, and hence
226 triose phosphate, which occur in the PCRC, albeit with the input of energy and the release of
227 CO₂ [61]. The cyanobacteria have two variants on pathways converting glycolate to 3-
228 phosphoglycerate.

229 One means of converting glycolate to 3-phosphoglycerate is the pathway via glycine and
230 serine, with recycling of ammonia, as in the classic PCOC of embryophytic plants and at least
231 some eukaryotic algae [61, 64-68]. The pathway through glycine and serine seems to have
232 been gained by eukaryotic photosynthetic organisms during the primary endosymbiosis
233 yielding the plastids of the Archaeplastida (= Plantae), although some of the genes in
234 eukaryotes came from α -proteobacteria rather than cyanobacteria [64]. The β -cyanobacterial

235 plastids ancestor gave rise to the glycolate oxidase, glycerate kinase and hydroxypyruvate
236 reductase of algae and embryophytes, while serine hydroxymethyl transferase and the L, P
237 and T subunits of glycine decarboxylase came from α -proteobacteria by horizontal gene
238 transfer [64]. The origin of the other eukaryotic PCOC genes, i.e. those encoding the H
239 subunit of glycine decarboxylase, glutamate-glyoxylate aminotransferase and serine-
240 glyoxylate aminotransferase, has not yet been established [64].

241 The other pathway from glycolate to 3-phosphoglycerate involves tartronic semialdehyde,
242 and is called the tartronic semialdehyde pathway by phycologists; bacteriologists call it the
243 glycerate pathway, even though glycerate is also an intermediate of the PCOC [61, 64-66].
244 Parts of the PCOC (glycerate dehydrogenase, serine transaminases, serine
245 hydroxymethyltransferase) could have been recruited from core metabolism synthesising
246 serine and glycine from glycolytic intermediates, while others (glycolate
247 dehydrogenase/oxidase, glycine decarboxylase, glycerate kinase) have no known roles other
248 than in the metabolism of glycolate to PCOC intermediates [69, 70].

249 It seems likely that at least one of the metabolic pathways from glycolate to 3-
250 phosphoglycerate evolved in oxygenic photosynthetic organisms relying on diffusive CO₂
251 entry before CCMs evolved. Not only is there at least a minimal flux through Rubisco
252 oxygenase and thence to intermediates of a glycolate metabolism pathway despite high levels
253 of expression of CCMs [61, 65, 66, 68, 71], but elimination of the pathways of glycolate
254 metabolism is fatal to the organism [61, 66]. Previous misgivings [67, 68] about the
255 occurrence of the complete PCOC in diatoms have now been largely overcome, although
256 there are still doubts as to the glycerate kinase step [37, 64].

257 The changes to CO₂ fixation in oxygenic photolithotrophs in relation to decreasing CO₂ and,
258 especially, increasing O₂ is part of a wider range of resource cost increases as the biosphere
259 becomes oxygenated. Falkowski and Godfrey [48] point out that not only is Rubisco
260 impacted by increasing O₂ with decreasing CO₂, but that the potential for oxygen damage to
261 nitrogenase becomes manifest, and the very source of the O₂, the reaction centre of
262 photosystem II, is itself subject to photodamage both directly through excitation energy
263 transfer to the reaction centre but also indirectly through the accumulated O₂ forming reactive
264 O₂ species (see [42]). As Raven [42, 44-46] points out, the effects on Rubisco demand
265 additional nitrogen in the enzyme itself and in related enzymes, more iron and manganese in
266 additional thylakoid redox agents, and more phosphorus in the RNA needed to make the
267 additional protein if the rate of photosynthesis is to be maintained. More energy input as
268 NADPH and ATP is also needed to run CO₂ assimilation [42, 45, 46]. For oxygen damage to
269 nitrogenase, there is generally synthesis of 'reserve' reserve nitrogenase in addition to what is
270 needed in the absence of oxygen damage to satisfy the combined nitrogen requirements of
271 cell growth. Synthesis of the 'reserve' nitrogenase requires the nitrogen and energy needed
272 for the synthesis of any protein, but also the iron and (in almost all cases) molybdenum used
273 in the nitrogenase cofactors. When oxygen damage does occur and reserve nitrogenase is used
274 catalytically, more energy (but not nitrogen, iron and molybdenum) is needed to synthesise
275 nitrogenase to replace what is damaged. In both cases, the production of more nitrogenase

276 than would be required in the absence of oxygen involves the use of more phosphorus for the
277 RNA required for the additional protein synthesis. In the case of photoinhibition, more
278 nitrogen and energy is needed to synthesise reserve photosystem II reaction centres, more
279 energy is needed to synthesise replacement photosystem II reaction centres, and more
280 phosphorus is needed for the RNAs needed for the extra protein synthesis. A further aspect of
281 damage to proteins by O₂ concerns the absence of any core autotrophic CO₂ assimilation
282 pathway other than Rubisco – PCRC from oxygenic photosynthetic organisms. In addition to
283 the CO₂ affinity problems outlined for some of the alternative pathways, there would also be
284 the requirement for additional resources (PAR, nitrogen and phosphorus) to make and use
285 additional RNA needed for the additional resynthesis of O₂-damaged protein (see [42] and
286 above, for other cases).

287 Compounding this need for additional nitrogen and phosphorus per unit CO₂ or N₂
288 assimilated and photons used in photochemistry is the effect of increased O₂ on the
289 availability of nitrogen and phosphorus. Falkowski and Godfrey ([48]) point out that
290 oxygenation of the biosphere not only decreases the potential for diazotrophy, but allows
291 nitrifying microbes to convert NH₄⁺ to NO₃⁻, which in hypoxic or anoxic microhabitats can be
292 denitrified to produce N₂O and N₂. This nitrification – denitrification sequence decreases the
293 availability of combined nitrogen to non-diazotrophic primary producers. In the case of
294 phosphorus, the availability of O₂ converts Fe(II) to oxidised iron (Fe(III)), which binds
295 phosphate and thus decreases global phosphorus availability [72, 73]. Phosphorus is one of
296 the biogeochemical regulators of the O₂ content of the atmosphere [72, 73]. This topic will be
297 returned to below in the context of ocean deoxygenation as a function of increases CO₂ and
298 temperature.

299 **Introduction to CCMs**

300 Diffusive entry of CO₂ to Rubisco was presumably the ancestral mechanism of autotrophic
301 CO₂ assimilation in oxygenic photosynthetic organisms. Entry of CO₂ to Rubisco by diffusion
302 is found today in the majority, by species number and contribution to global primary
303 productivity, of terrestrial oxygenic photosynthetic organisms, but in a minority of oxygenic
304 photosynthetic organisms in aquatic environments where photolithotrophs with CCMs
305 predominate [10, 43, 55, 74-80]: see Table 1. The references just cited show that CCMs are
306 very widely distributed among algae, both phylogenetically and geographically, although
307 they seem to be absent from chrysophycean and synurophycean algae [81]. The mechanistic,
308 including molecular, details of the CCMs of cyanobacteria are now known [27-29, 32, 82].
309 The CCMs of eukaryotic algae are less clearly understood at both the molecular and the
310 mechanistic levels, although they are clearly polyphyletic[10, 11, 43, 55, 80, 83, 84].

311 As to the evolutionary origin of CCMs, the selective factors were presumably decreasing CO₂
312 and increasing O₂. The variability of other gases over the last 2.4 Ga suggests there were
313 several periods at which CCMs could have been resource-effective (energy, nitrogen,
314 phosphorus, iron, zinc, manganese) alternatives to diffusive CO₂ entry to Rubisco with
315 attendant high activity of Rubisco oxygenase and expression of high levels of enzymes

316 converting 2-phosphoglycolate to sugar phosphate [11, 43-47, 50, 85, 86]. We shall return to
317 the question of the timing(s) of the origin(s) of CCMs and how, if they originated early, they
318 were retained through intervening high-CO₂ episodes. First, we consider what the transition
319 from diffusive CO₂ entry to CCM-based delivery of CO₂ to Rubisco involves.

320 **The functioning of CCMs in comparison with the diffusive entry of CO₂**

321 One factor in the evolution of a CCM in an organism previously relying on diffusive CO₂
322 movement from the medium to Rubisco as CO₂ availability decreases is the kinetics of the
323 Rubisco used by the organism. A Rubisco with a high CO₂-saturated catalytic capacity, and a
324 correspondingly low CO₂ affinity and CO₂/O₂ selectivity, such as occurs in the Form IA and
325 Form IB Rubiscos of extant cyanobacteria, would be expected to present a stronger
326 evolutionary case for a CCM at a given CO₂ concentration and CO₂/O₂ ratio than would
327 eukaryotic Form IB and Form ID Rubisco with higher CO₂ affinity and CO₂/O₂ selectivity but
328 a lower CO₂-saturated maximum catalytic rate [41]. Young et al. ([87]) have used molecular
329 phylogenetic evidence to show that there was positive selection of the Form ID Rubiscos of
330 in some eukaryotes which correspond to low CO₂ and low CO₂/O₂ episodes in the geological
331 record, and that these episodes of positive selection could have corresponded to the time of
332 evolution of CCMs. To be effective, the CCM must maintain a higher CO₂ concentration at
333 the site of Rubisco than would be possible by CO₂ diffusion alone [11].

334 The essential component of a CCM is the accumulation of CO₂ in the compartment
335 containing Rubisco to a higher steady-state concentration than occurs in the growth medium,
336 and hence even higher than the steady-state concentration near Rubisco which could occur
337 with diffusive CO₂ entry. For algae, one mechanism of accumulation could involve C₄-like
338 photosynthetic metabolism with an ATP-dependent (C₃ + C₁) carboxylation in the cytosol,
339 using HCO₃⁻ obtained directly or indirectly (via CO₂ entry from the medium followed by
340 carbonic anhydrase catalysis) from the medium, followed by a (C₄-C₁) decarboxylation in the
341 chloroplast [84]. The alternative mechanisms do not involve the inorganic carbon transferred
342 from the medium to Rubisco forming organic intermediates. These alternative mechanisms
343 involve transmembrane active transport mechanisms which move an inorganic carbon species
344 (CO₂ or HCO₃⁻), or H⁺, against a free energy gradient. Such transporters could be (and have
345 been) derived from transporter gene families by change of specificity of the transported
346 substrate (to CO₂ or HCO₃⁻), with changes in regulation and, perhaps, changes in intracellular
347 targeting [27-29, 32]. An exception to the need for active transport across a membrane is CO₂
348 use in the cyanobacterial CCMs, where diffusive CO₂ entry across the plasmalemma is
349 followed by energized conversion to HCO₃⁻ by the NAD(P)H – PQ oxidoreductase of the
350 thylakoid membrane [27-29, 32]. Here a high CO₂ permeability of the plasmalemma is
351 required. Such a high membrane permeability to CO₂ is needed for diffusive CO₂ entry all the
352 way to Rubisco in organisms lacking a CCM, and (as noted above) in CO₂ ‘active transport’
353 in cyanobacteria. A high CO₂ permeability is also necessary in organisms with a CCM
354 mechanism involving CO₂ production from HCO₃⁻ in a compartment acidified by a H⁺ pump
355 followed by transmembrane movement of CO₂ into the compartment containing Rubisco
356 [80]. The energized conversion of CO₂ to HCO₃⁻ in the cyanobacterial cytosol could increase

357 the chance of any CO₂ that leaks out of the carboxysomes being trapped as HCO₃⁻ in the
358 cytosol. In all other cases a CCM is most energetically efficient with minimal CO₂ flux from
359 the compartment in which it is accumulated back to the medium, i.e. with very low
360 membrane permeabilities to CO₂.

361 The energetic savings that could result from a low CO₂ permeability of the plasmalemma, and
362 of the inner plastid envelope (if that is the membrane involved in active transport of inorganic
363 carbon), in eukaryotes with CCMs based on active transport of an inorganic carbon species
364 would, if verified, suggest a phylogenetic, and in many cases an acclimatory (changes from
365 growth in high to low CO₂) decrease in CO₂ permeability. This question has been addressed
366 by a number of workers (e.g. [83, 88]), who found relatively high CO₂ permeability in the
367 eukaryotic algal membranes examined, regardless of whether the algae had been cultured in
368 high (CCM repressed) or low (CCM expressed) inorganic carbon concentrations. The
369 permeability values for CO₂ of the plasmalemma of high-CO₂ and low-CO₂-grown cells of
370 *Chlamydomonas reinhardtii* range from 0.76 – 1.8.10⁻³ cm s⁻¹ [88], while for four species of
371 diatom the range is 15 – 56.10⁻³ cm s⁻¹ [83], with the range of values probably related to
372 methodological as well as phylogenetic differences. While models for CCMs in diatoms
373 consistent with the available data show relatively modest energy losses during CCM
374 operation, they do involve constraints such as the membrane(s) at which active inorganic
375 carbon transport occurs, and the chemical species involved in this active transport [83]. The
376 same argument applies to the protein shell of the carboxysome, for which a restriction on CO₂
377 diffusion, but not on the diffusion of anionic substrates (HCO₃⁻ and ribulose-1,5-bisphosphate)
378 and product (3-phosphoglycerate) has been demonstrated [89].

379 Most CCMs also involve one or more carbonic anhydrase enzymes: an exception is a C₄
380 mechanism which involves, as indicated above, HCO₃⁻ entry to the cytosol, (C₃ + C₁)
381 carboxylation using HCO₃⁻, and (C₄–C₁) decarboxylation in the plastid stroma with CO₂ as
382 the inorganic carbon product [68, 71, 80, 84, 90]. All other well-investigated CCMs seem to
383 involve ‘normal’ carbonic anhydrases, i.e. those catalysing the equilibration of CO₂ and
384 HCO₃⁻ [43, 80, 91]: this is the case for ‘active CO₂ influx’ in cyanobacteria which involves a
385 carbonic anhydrase in the carboxysome as well as the energized conversion of CO₂ to HCO₃⁻
386 at the thylakoid membrane, which is effectively a unidirectional carbonic anhydrase [32].

387 The functioning of CCMs is influenced by a number of factors other than the availability of
388 inorganic carbon (and, in some cases, O₂), e.g. photosynthetically active radiation, UV-B
389 radiation, the form and concentration of combined nitrogen, the phosphate concentration and
390 the iron concentration [42, 43, 50, 85]. The influence of these factors on the expression and
391 functioning of CCMs is presented in Table 3 [42, 43, 50]. There are also predicted effects of
392 expression of CCMs rather than reliance on diffusive entry of CO₂ on the resource costs of
393 synthesis of the photosynthetic apparatus, and of its operation; these are discussed below. We
394 next discuss the possible influence of these interactions on the evolution of CCMs, their
395 retention through any high-CO₂ episodes between their origin in a low CO₂ habitat and today,
396 and their fate in a future higher CO₂ and warmer world.

397 **The origins of CCMs.**

398 The ‘why’ of the origin of CCMs presumably concerns the occurrence of low CO₂, both in
399 absolute terms and in relation to O₂, which was indicated above as requiring additional
400 protein (hence RNA and phosphorus) in more Rubisco and in enzymes metabolising 2-
401 phosphoglycolate to sugar phosphate, as well as additional energy input per net CO₂
402 assimilated by diffusive CO₂ entry of CO₂ to Rubisco and metabolism of 2-phosphoglycolate.
403 Depending on the circumstances, e.g. the form of Rubisco present and the environmental
404 conditions as well as the mechanism of the particular CCM [10, 11, 44-47, 50, 55, 67, 83,
405 85], a CCM could require less energy, nitrogen, phosphorus, zinc and iron for its synthesis,
406 and less energy for its operation, than diffusive CO₂ entry with metabolism of 2-
407 phosphoglycolate (Table 2). Other factors that could have influenced the evolution of CCMs
408 include the decreasing UV-B flux with the increased stratospheric O₃ resulting from the build
409 up of O₂, which is itself an influence on the evolution of CCMs [55]. UV-B radiation causes
410 damage to Rubisco and to Photosystem II, but has less effect on Photosystem I. The limited
411 data available suggest that UV-B has little effect on CCM activity in the green alga
412 *Dunaliella tertiolecta* [92] but elevated CO₂ can increase the sensitivity of microalgae to UV-
413 B [93, 94]. There is no information about the possibility of a differential impact of UV-B on
414 the various forms of Rubisco, though it would be interesting to know if changes in UV-B in
415 the past relate to the evolution of different Rubiscos.

416 The ‘how’ of the origin of CCMs concerns the ancestry of the various components of the
417 pathway. For the active transport components H⁺ pumps are ubiquitous and anion
418 transporters/pumps (hence HCO₃⁻ transporters/pumps) are also widespread, as could be the
419 ancestors of CO₂ pumps [10, 11, 27-29, 32, 55, 80, 82, 84]. Facilitators of downhill
420 transmembrane CO₂ transport, yielding permeabilities in excess of those due to the lipid
421 phase alone, are required for cyanobacterial active transporters and for the mechanism
422 involving an acidified compartment generating CO₂ from HCO₃⁻ with subsequent
423 transmembrane CO₂ diffusion to Rubisco. Such facilitators would presumably have originally
424 been components of the diffusive pathway for CO₂ from the medium to Rubisco. Carbonic
425 anhydrases could have had a number of roles prior to their co-option into CCMs, including
426 that of facilitating diffusion of CO₂ (as HCO₃⁻) in the diffusive entry of CO₂ from the medium
427 to Rubisco. Cyanobacterial carboxysomes are part of a larger family of bacterial micro-
428 compartments [95]. This brief view may be over-optimistic as to the ease of co-opting
429 existing mechanisms into CCMs [10], and does not address the origin of the eukaryotic
430 pyrenoid [96]; however, it does indicate some possibilities.

431 There are a number of options as to the ‘when’ of the evolution of CCMs. We initially
432 consider times of relatively low CO₂, based on low palaeotemperatures (with the requirement
433 that greenhouse gases corrected for the faint young sun) or on biogeochemical or biological
434 proxies. Glacial/low CO₂ episodes occurred 2.4-2.1 Ga ago and at 750, 650 and 320-270 Ma
435 ago, as well as the Pleistocene (last 2.4 Ma) years [11]. All but the earliest of these times
436 would have been relevant to at least some of the eukaryotic as well as cyanobacterial
437 oxygenic photosynthetic organisms (Table 1). There is no direct fossil evidence as to the

438 origin of CCMs, and little help from molecular clocks [11, 43, 55], although recent work by
439 Young et al. ([87]) shows episodes of positive selection of Form ID Rubisco in diatoms and
440 haptophytes which correspond to low CO₂ episodes and hence possibly relate to the origin of
441 CCMs. Assuming, as seems very likely, that cyanobacterial and at least some algal CCMs
442 evolved before the Pleistocene, they would have had to have survived intervening period(s)
443 of higher CO₂ and higher temperatures. The mechanisms of retention of CCMs in these
444 apparently unfavourable environments is now considered in the context of the response of
445 present day CCMs to such environments.

446 **Retention of CCMs in high CO₂ episodes**

447 It is possible to argue for long-lasting low-CO₂ micro-habitats, e.g. in benthic microbial mats,
448 including stromatolites, where inorganic carbon diffusion from the bulk medium into the cells
449 is restricted by thick diffusion boundary layers [55]. Biogeochemically more important are
450 planktonic habitats where such low CO₂ refuges are less plausible. CCM retention is
451 considered in the context of present work on the response of phytoplankton to current and
452 expected environmental change, and especially increasing CO₂ and the associated warming.

453 More widespread (planktonic) retention of CCMs in relatively high CO₂ concentrations has
454 been argued by [11] in the context of what might happen with increasing CO₂ and
455 temperature over the next several decades. The argument here is that the increased buoyancy
456 of warmer surface waters will lead to a shoaling of the thermocline which will decrease
457 fluxes of combined nitrogen and of phosphorus from the deeper ocean where mineralisation
458 of sinking particulate organic matter occurs which have been modelled as decreasing global
459 marine primary productivity [97-99]. This decreased nutrient supply, with the increased mean
460 flux of photosynthetically active radiation (and UV-B) incident on the cells in the shallower
461 upper mixed layer will, on the basis of observations on extant phytoplankton (Table 3),
462 favour retention of CCMs despite higher CO₂ concentrations [11].

463 A further relevant consideration which was not mentioned by Raven et al. ([11]) is that a
464 warmer upper mixed layer has decreased oxygen solubility. For the same rate of net oxygen
465 production at two temperatures there will be greater degree of oxygen supersaturation and
466 hence a greater loss of oxygen to the atmosphere at the higher temperature. Together with the
467 decreased solute transfer between the upper mixed layer and deeper ocean there is less
468 transfer of oxygen below the thermocline [100-102]. The widespread, but not universal,
469 decrease in calcification by calcified plankton [103-105], and the much smaller effect on
470 silicification by diatoms [106] in a higher-CO₂, warmer ocean means less ballasting of
471 sinking particulate organic matter, hence slower sinking and more microbial mineralisation
472 just below the thermocline ([107] cf. [108]). While this higher nutrient concentration just
473 below the thermocline might be expected to partly offset the lower rate constant for nutrient
474 transfer to the upper mixed layer, another factor must be considered. The combination of
475 more microbial respiration and increased oxygen supply can lead to hypoxia and even anoxia
476 in certain sub-surface waters, with implications for loss of the nitrate and nitrite forms of
477 combined nitrogen produced from organic matter by mineralisation and nitrification in less

478 deoxygenated zones followed by denitrification or the anammox reaction in more
479 deoxygenated places [102], with a decrease in the nitrogen:phosphorus ratio in the nutrients
480 reaching the upper mixed layer. Although, in the long term (thousands of years and more), a
481 warmer world would heat the ocean interior as well as the upper mixed layer and potentially
482 decrease the extent to which the thermocline shoals, there is a well-established correlation of
483 wide-spread deep-ocean anoxia (and even euxinia) with warmer, high CO₂ episodes in Earth
484 history [109], so at least a decreased upward flux of combined nitrogen across the
485 thermocline would continue, favouring retention of CCMs.

486 These arguments are based on the response of extant algae to the changes occurring, and
487 predicted, in their environment as a result of increased CO₂ and temperature, with
488 downstream effects on the marine and inland water inorganic system, the mixed layer depth
489 and water body oxygenation. This can be used to inform us of how CCMs were retained in
490 past episodes of high CO₂. Dealing first with the organisms studied, almost all of the work
491 has been carried out with organisms which have only been exposed to the increased CO₂ and
492 associated changes in temperature and the availability of other resources (Table 3) for time
493 periods of days to weeks. This time period allows 1 – 100 generation, meaning that only
494 regulatory (altering the existing proteome by post-translational modification, and changes in
495 the metabolome) and acclimatory (altering the expressed proteome based on the existing
496 genome) [110] responses of extant algae can be expressed. In very few cases has relevant
497 evolutionary evidence been sought in laboratory experiments for increased CO₂ [110-117]
498 and, using different methods, higher temperatures [118]. An example of where evolution has
499 been unable to cope with natural CO₂ enrichment present for several decades concerns
500 calcified red algae growing on seagrass leaves near an underwater vent in the Mediterranean
501 [119]. Even for studies of regulation and acclimation there can be problems with the length
502 of time that an alga has been in culture and frequently exposed to CO₂, nitrogen, phosphorus,
503 PAR, UV and temperature which has little relevance to their natural environment [120].
504 Furthermore, there is a relative lack of studies in which changes in CO₂ have been combined
505 with other relevant environmental changes: Table 3 and [11] analyse, and give references to
506 the important work in which such interactions have been studied. Finally, there are some
507 relatively poorly constrained factors of ocean chemistry in the past [56, 121-125], and until
508 recently little consensus as to the appropriate methodology [105, 126-129] for laboratory and
509 mesocosm experimentation on increased CO₂.

510 Despite these reservations, which also apply to other models of past and future atmosphere –
511 ocean – organism interactions, the suggestions of Raven et al. ([11]) provide a lead into
512 further studies in how CCMs could be retained in lengthy episodes between shorter low-CO₂
513 episodes. Any retention of CCMs in high CO₂ episodes would be a further complication in
514 the use of stable carbon isotope ratios of phytoplankton-derived organic carbon from marine
515 sediments as a palaeobarometer for CO₂, since most marine phytoplankton today have CCMs
516 [43, 74, 130, 131]. Organisms lacking CCMs, e.g. terrestrial liverworts and mosses, do not
517 suffer from this problem when used in palaeobarometry of CO₂ [132].

518 **Conclusions**

519 The changing CO₂ and O₂ concentration over the last 2.4 Ga have had significant effects on
520 the physiology and ecology of cyanobacteria and algae. From the presumed ancestral
521 diffusive CO₂ entry to Rubisco all extant cyanobacteria have CCMs, Rubiscos with high CO₂
522 saturated catalytic activity and low CO₂ affinity and CO₂/O₂ selectivity, and an essential role
523 for the capacity to convert the 2-phosphoglycolate formed as a very small fraction of the total
524 carbon flux into triose phosphates. Most eukaryotic algae have CCMs: a greater fraction have
525 CCMs in the sea than in freshwaters, and there is no strong relationship to water
526 temperatures. The evolution of CCMs can apparently be related to decreased CO₂ availability
527 and to the presence of oxygen, modulated by the kinetics of the form of Rubisco in the
528 organisms, with some components of the CCMs adapted in evolution from the roles in other
529 pathways. The retention of CCMs during the high CO₂ episodes predominant through Earth
530 history could have been related in part to the interaction of CCM expression with other
531 environmental factors which change in high CO₂ water bodies.

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546

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1000 **Table 1**

1001 Inorganic carbon acquisition characteristics of cyanobacteria and algae related to earliest
 1002 known occurrence of the taxon^{1,2}

Taxon	Occurrence of CCM in Extant organisms	Oldest Known fossil	References
Cyanobacteria	CCM ubiquitous	2.15 Ga (bio-markers). 2.45 Ga (O ₂)	[1, 6, 7, 133]
Chlorophyta: Prasinophyceae	CCM ubiquitous	1.3Ga	[134, 135]
Chlorophyta: Chlorophyceae	CCM present in all?	(450 Ma) ³²	[135, 136]
Chlorophyta: Trebouxiophyceae	CCM present or absent	450 Ma	[135-139]
Chlorophyta: Ulvophyceae	CCM usually present; absent in some; C ₄ in one	540 Ma	[135, 136, 140]
Streptophyta: Charophyceae	CCM present in all?	450 Ma	[136, 140]
Streptophyta: Embryophytes	CCM usually absent; pyrenoid-based CCM is some anthocero-phytes, C ₄ or CAM in some freshwater tracheophytes, CCMs not involving C ₄ and CAM in some freshwater and all marine tracheophytes	475 Ma	[140-142]
Rhodophyta: Bangiophyceae	CCM in all?	1.2 Ga	[2]
Rhodophyta: Florideophyceae	CCM in many, absent from some marine, many freshwater	600 Ma	[143, 144]
Ochista: Bacillariophyceae	CCM in all?	120 Ma	[145, 146]
Ochista: Fucophyceae	CCM in all?	(570 Ma?)	[144, 147]
Ochista: Tribophyceae	CCM in all?	600 Ma	[3, 148]
Ochista: Chrysophyceae and Synurophyceae	CCM absent in all	?	[81]

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1005 Footnotes

1006 ¹General references on earliest known occurrence of algae: [140, 149, 150]

1007 ²References on presence or absence of CCMs: [11, 78-81, 96]

1008 ³Based on the finding of Trebouxiophyceae in the Ordovician, and branching order of the Chlorophyceae, Trebouxiophyceae and
 1009 Ulvophyceae from molecular phylogenetics [136].

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1015 **Table 2**

1016 Energy (NADPH and ATP) stoichiometry, affinity for inorganic carbon expressed as the half-
 1017 saturation concentration for CO₂, competitive inhibition by O₂ and damage by O₂ for six autotrophic
 1018 inorganic carbon assimilation pathways. Based on [17-22].

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Pathway from inorganic carbon to carbohydrate	NAD(P)H per CO ₂	ATP Per CO ₂	K _{1/2} CO ₂ mmol m ⁻³	O ₂ competitive inhibition	O ₂ damage to one or more enzymes
Rubisco-PCRC, saturating CO ₂ no O ₂	2	3	≥10	Yes	No
Reverse TCAC	2	1.67	>1,500	No	Yes ¹
3-HO-Propionate	2	2	10	No	No ²
3-HO-Propionate-4-HO-Butyrate	2	3	>2,000	No	O ₂ -insensitive pathway in Some organisms living in microaerobic habitats
Dicarboxylate-4-HO-Butyrate	2	2.67	>2,000	No	No; some organisms live in microaerophilic habitats
Wood-Ljungdahl pathway	2	1	40,000 ²	No	Yes

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1021 Footnotes

1022 ¹The reverse TCAC can occur in thermophilic aerobic chemolithotrophs as a result of low O₂
 1023 solubility and high respiratory rates maintaining a low O₂ concentration inside the cells, and
 1024 expression of an O₂-insensitive version of the 2-oxoglutarate-ferredoxin oxidoreductase which has t
 1025 least a five-fold lower specific activity than the O₂-sensitive enzyme [20].

1026 ²The most oxygen-sensitive enzyme, methylmalonyol-CoA mutase, can be assayed and even purified
 1027 at atmospheric equilibrium O₂ concentrations, it may not be sufficient O₂-tolerant to function in
 1028 illuminated cells of oxygenic photosynthetic organisms [20].

1029 ³Although acetogens live in habitats with higher CO₂ concentrations than correspond to atmospheric
 1030 equilibrium [151], the *in vitro* K_{1/2} value cited does seem very high. .

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1037 **Table 3**

1038 Influence of environmental conditions on the expression of CCMs and the resource cost of CCM
 1039 operation versus diffusive entry of CO₂

Factor	Change to algal environment caused by variation in the factor	Effects on expression of CCMs and on their affinity for CO ₂	Predicted effect of CCM expression on resources costs of synthesis and (for PAR) operation of the photosynthetic apparatus
CO ₂	Increase in CO ₂ in essentially all environments, although less predictable effect in freshwaters which can be out of equilibrium with the atmosphere	Decreased inorganic carbon affinity with growth at high CO ₂ ; can be a switch to diffusive CO ₂ entry in some eukaryotes	No general effect on carbon cost of synthesis of the photosynthetic apparatus
Temperature	Increase in temperature in all environments	Prediction of increased CCM expression, or increased fraction of organisms with CCMs, at higher temperatures as a result of lower CO ₂ solubility, is not uniformly supported by the available data.	Inapplicable: temperature (Boltzmann energy) is not resource consumed in the synthesis of the photosynthetic apparatus, in its operation or in its maintenance
PAR	Increase in PAR in pelagic planktonic environments	Decreased inorganic carbon affinity with growth at low PAR	Possible decreased energy needed in synthesis of photosynthetic apparatus which uses a CCM. Energy saving in operation of a CCM if there is low CO ₂ leakage and a low CO ₂ affinity and low CO ₂ :O ₂ selectivity
Nitrogen	Decrease in combined nitrogen in upper mixed layer of lotic environments	Generally increased inorganic carbon affinity with growth at low NO ₃ ⁻ . One example each of decreased carbon affinity with growth at lowest NO ₃ ⁻ concentration tested, and with growth over entire NH ₄ ⁺ range tested.	Decreased nitrogen cost of synthesis of the photosynthetic apparatus incorporating a CCM if the savings in the synthesis of smaller amounts per cell of Rubisco and of the PCOC enzymes are not offset by the nitrogen cost of the synthesis of CCM components No consumption of nitrogen in the operation of the CCM
Phosphorus	Decrease in phosphate in upper mixed layer of lotic environments	Two examples of increased inorganic carbon affinity, two examples of decreased inorganic carbon affinity, with growth at low phosphate	Where there is a decreased protein content in the photosynthetic apparatus there could be a corresponding decrease in the need for phosphorus needed to synthesise the photosynthetic apparatus as a result of decreased requirement for RNA
Iron	Probable decrease in iron in upper mixed layer of lotic environments	One example of increased inorganic carbon affinity with growth at low iron, one example of no effect	Decreased Fe content of the photosynthetic apparatus if the decreased requirement for NADPH in the near absence of Rubisco oxygenase activity and the PCOC, with correspondingly lower content of non-cyclic electron transport components, is not offset by the additional thylakoid components needed for the additional ATP requirement for CCM operation, especially if this additional ATP is made using cyclic photophosphorylation using only Photosystem I with its high Fe content
Zinc	As for iron	No direct measurements	Variable predictions of relative zinc requirements depending on CCM mechanism
UVA	Increase in UVA in lotic planktonic environments, but decrease with higher concentration of DOC*	No data	Not applicable
UVB	Increase in UVB in lotic planktonic environments, but decrease with higher concentration of DOC*	Variable responses of CCMs with increased UVB flux for growth.	Not applicable

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1042 Footnote to Table 2

1043 Effects on CCMs of environmental factors, and the direction of change of these environmental factors in algal
1044 and aquatic plant habitats with global environmental change between icehouse episodes. Modified from [11].
1045 Further details and references are given in the text and in [11, 77-79, 85, 86, 90, 152-160].

1046 Predicted resource costs of synthesising and operating a photosynthetic apparatus using a CCM relative to one
1047 relying on entry of CO₂ by diffusion [42, 44-47, 49, 67, 85, 161].

1048 *DOC = Dissolved Organic Carbon

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