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Advancing understanding of in-river phosphorus dynamics using an advection-dispersion model (ADModel-P)

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Abstract

The objectives of the present research are (1) to predict phosphorus compounds transport along river stretches at high spatio-temporal resolution by developing an original approach based on advection-dispersion modelling (ADModel-P); (2) to advance the understanding of in-stream phosphorus transformation processes, and (3) to explore their relation to controlling factors (water temperature, seasonality and water flow). For a case study of the River Swale (UK) modelling results, in agreement with results based on experimental data, show that resuspension is the largest contributor to the variability of organic phosphorus, while adsorption-desorption are the largest contributors to the variability of soluble reactive phosphorus. Additionally, simulations reveal that conversion of inorganic to organic forms is important. In-channel sinks appear more important than sources for Soluble Reactive Phosphorus (SRP) during 80% of the time, while there is no clear evidence that Organic Phosphorus (OP) sinks or sources are dominant except the beginning of spring (around 20% of the total time). The findings are valuable because they advance knowledge regarding: (1) which in-stream processes are important; (2) values associated to transformation rates for mineralization, sedimentation, resuspension, uptake, adsorption - desorption; (3) at which times rivers are net sources or sinks and which source/sink processes might be dominant. ADModel-P is a robust model which has the benefits of simple field data requirements (compared to more complex models) and less assumptions (e.g. compared to simple models assuming perfect mixing in reaches) but without the drawbacks of lack of process representation

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26 to enable confidence in predictions to change. There is extensive scope for transferability to other rivers:
27 rate constants can be estimated from easily attainable information on water temperature, seasonality and
28 water flow.

29 **Keywords:** phosphorus in-stream transformations; advection dispersion pollutant transport model;
30 mineralization; sedimentation - resuspension; uptake; adsorption - desorption.

31 **1 Introduction**

32 Phosphorus (P) compounds are among the most important freshwater nutrients of critical concern.
33 The increased attention they receive currently at global level is motivated by their key role in the
34 agriculture (e.g. fertilizers) and in the ecosystem dynamics (e.g. as limiting nutrients for algae
35 production). Increased loads of phosphorus to waterbodies is usually the primary driver for the
36 eutrophication of freshwater bodies (Charlton et al., 2018; Colborne et al., 2019; Zinabu et al., 2018).
37 The European Environment Agency's databases on the status and quality of Europe's water resources
38 (Waterbase) shows evidence that the average concentration of phosphates in European rivers is larger
39 during the period 2000–2017 compared to the period 1992–2017, even if a decrease of 1.6 % per year is
40 indicated over the last two-three decades (EEA, 2019), mainly attributed to the improved waste water
41 treatment technologies. Among river water bodies assessed by the Environment Agency in England,
42 55% were at less than good ecological status with respect to P concentrations in 2016 (Environment
43 Agency, 2018), while an earlier report indicates the figure to be just 45% (Environment Agency, 2012).
44 Concern surrounds persistence of elevated levels due to release of “legacy” P held in riverbed sediments
45 that offset benefits of reduced pollutant load from the land and other sources such as improved waste
46 water treatment. Moreover, recent research shows that P concentrations are and will remain high enough
47 to fail P standards until 2050 (Charlton et al., 2018; Jarvie et al., 2018; Hutchins et al., 2016; Ockenden
48 et al., 2016) or are greater than ecologically limiting thresholds (Riley et al., 2018) and emphasise the
49 increasing risk of eutrophication. Dangers, associated principally to P compounds, are long-established
50 and widely investigated for rivers in England and worldwide (Harrison et al., 2019a and 2019b; Charlton

51 et al., 2018; Zelenakova et al., 2018; Enea et al., 2017; Ji, 2017; Hutchins et al., 2016; Koraqi et al.,
52 2016; Romanescu et al., 2016; Iordache and Dunea, 2013).

53 In this context, appropriate countermeasures to water quality degradation, with respect to P, should
54 be underpinned by deep understanding of the behaviour and fate of P species and on the prediction of
55 its future trends. Recent work underlines the important role of water quality models which include P
56 concentrations (Riley et al., 2018; Tuo et al., 2015; Charlton et al., 2018), to support high quality decision
57 making and water management. The mathematical modelling of water quality facilitates both
58 understanding of phenomena and its prediction, even in cases where the availability of experimental data
59 is limited (Ani et al., 2009). Such modelling can be carried out at different levels of magnitude and detail,
60 starting from the small scale (hundreds of meters of river stretch) to the global scale (see the models
61 discussed by Harrison et al., 2019b), depending on the area of interest, its particularities and the needs.
62 With respect to the resolution in terms of space and time, the spectrum of approaches extends between:
63 (A) catchment scale modelling (describing a catchment or multiple catchments, not necessarily high-
64 resolution in terms of space and time), mainly based on usual long-term monitoring data (Keraga et al.,
65 2019; Srinivas and Singh, 2018); and (B) modelling of river stretches based on high resolution data
66 (multiple monitoring points along the stretch, where measurements of channel characteristics,
67 hydrodynamic data and water quality data are needed), such as ADModel for nitrogen (Ani et al., 2011).

68 In terms of advancing system understanding, river P transport studies of both types have proven
69 valuable in many respects: (a) facilitate understanding of catchment scale processes and patterns (Lupi
70 et al., 2019; Wade et al., 2002) including use of storm-event data (Ramos et al., 2015); (b) understand
71 nutrient (including P) interactions between water and sediment phases (Wijesiri et al., 2019; Yuan et al.,
72 2019; Tye et al., 2016) or the contribution of streambanks to in river P load (Fox et al., 2016); (c) assess
73 catchment scale ecological quality status (Charlton et al., 2018); (d) make correlations between in-river
74 compounds (Wade et al., 2002; Neal et al., 2010); (e) explore relations between chemical species
75 concentration (including P) and different influences, such as water flow, seasonality, temperature, certain
76 types of sources (Charlton et al., 2018; Ockenden et al., 2016; Cooper et al., 2015; Wade et al., 2002);

77 and (f) identification and/or understanding of P pollution sources (alone or along with other nutrients),
78 (Colborne et al., 2019; Liberoff et al., 2019; Records et al., 2016; Ramos et al., 2015).

79 During recent years river stretch studies (type B) including high resolution measurements of P
80 species and advective–dispersive non-conservative transport have become less common. This is most
81 probably due to the difficulties and great costs associated to gathering the needed high resolution data
82 (which despite recent advances (Halliday et al., 2015; Rode et al., 2016) is still not commonplace), and
83 also a change in focus towards studies covering larger spatial scales. Valuable recent studies include
84 those of Dou et al. (2018) and Nguyen et al. (2018), while earlier studies are a lot more abundant, as
85 shown by Mateus et al. (2018), Gao and Li (2014), Wang et al. (2013), Tsakiris and Alexakis (2012) or
86 Ani et al. (2010).

87 For effective basin-scale management it is important to maintain small-scale modelling development
88 to inform appropriate and sufficiently accurate process representation in larger-scale models (Tang et
89 al., 2019). The extent to which in-channel transformations are important in determining catchment scale
90 fluxes is unclear. Conservative or black box modelling approaches with simple dependencies on
91 residence time (Zhang et al., 2017) or sedimentation (Jackson-Blake et al., 2017) may provide sufficient
92 representation, but this merits further attention. In this context, ADMModel, a model based upon detailed
93 representation of pollutant advection and dispersion has been successfully developed for simulation of
94 nitrogen compounds (Ani et al., 2011), and performs comparably to a contrasting but extensively tested
95 model (QUESTOR: Boorman 2003a, 2003b) that assumes complete mixing in asset of reaches. The
96 advection-dispersion approach was calibrated and evaluated by considering the transport parameters
97 (e.g. dispersion coefficients, convective velocity) and associated nutrient nitrogen transformations:
98 nitrate and ammonium. Given the need to better understand phosphorus dynamics in rivers it is important
99 to build on the approach by developing from first principles a compatible approach for phosphorus. The
100 potential that would then arise for an integrated approach covering multiple nutrient determinands is of
101 much value.

102 The River Swale in north-east England has been the focus or a part of a wide range of studies on
103 phosphorus: (i) speciation in the main channel based on weekly river monitoring data (Jarvie et al., 1998)
104 including the influence of various types of wastewater (Neal et al., 2010); (ii) influence of point and
105 diffuse inputs on the River Swale and its major tributaries (House and Warwick, 1998a); (iii) modelling
106 of point and diffuse sources in three river catchments, using load appointment and export coefficient
107 approaches, focusing on Total Phosphorus, TP (Bowes et al., 2008); (iv) estimation of mass balances,
108 seasonally-variable losses and gains of nutrients along reaches under differing flow conditions including
109 stable low flows and overbank floods (House and Warwick, 1998b; Bowes and House, 2001; Bowes et
110 al., 2003; Bowes et al., 2008). These lines of evidence have supported the existence of groundwater
111 inputs of varying importance depending on discharge levels; (v) phosphorus–discharge hysteresis during
112 storm events (Bowes et al., 2005; House and Warwick, 1998a); (vi) sediment transport and interactions
113 (Kim et al., 2006; Smith et al., 2003; Owens and Walling, 2003; House and Warwick, 1999; House et
114 al., 1998).

115 Despite a breadth of process-based understanding in the Swale, a detailed modelling approach, as
116 adopted in ADModel-P, is lacking. Findings from process-based studies will be revisited later to provide
117 invaluable context for a modelling study on the P transformations.

118 Therefore the objectives of the present paper are (1) to offer a new validated detailed modelling tool
119 (ADModel-P) for the prediction of P compounds transport along a stretch of River Swale at sub-daily
120 resolution, based on principles of advection and dispersion previously applied for nitrogen simulation;
121 (2) to advance the understanding and significance of in-stream P transformation processes, (3) to explore
122 their relation to controlling factors, namely water temperature, seasonality and water flow, and (4) offer
123 a P compounds transformations model (using field data easy to collect) transferable to other case studies.

124 In terms of river pollutant transport the present research contributes to:

- 125 i. the detailed representation of phenomena which empowers good prediction efficiency of
126 concentrations (and in the same time does not make it a too complex model);

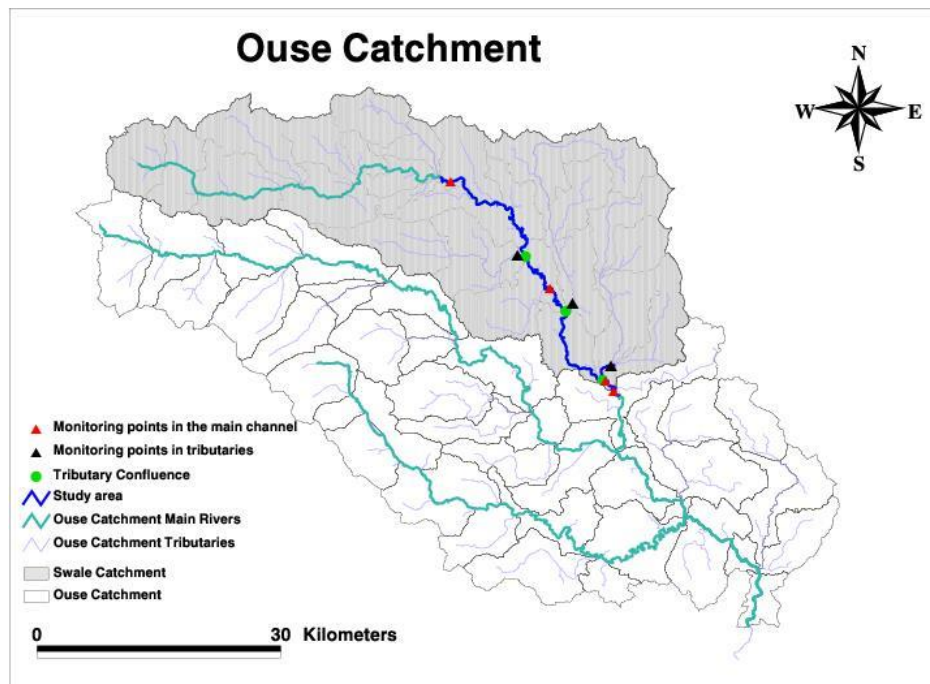
- 127 ii. the use of easily attainable field data (water flow, water temperature and seasonality) to
128 generate empirical relations to express the dynamics of process rates. This is in contrast to
129 existing mechanistic models which require detailed characterisations and extensive
130 assumptions of channel properties potentially making applications complex and challenging;
- 131 iii. the original perspective of approaching correlations between phosphorus compounds
132 transformation rates and resulting in-stream concentrations on one side and their main
133 controlling factors (sources of P, water flow, water temperature and seasonality) on the other
134 side.

135 Section 2 of the paper describes the study area and field data. Section 3 describes the model
136 background and the development process. Section 4 describes the transformations model development
137 (the context of P compounds transformations and their representation in existing models) and describes
138 the model in detail. Section 5 comprises results and discussions, starting with results of the new
139 developed transformations model discussed in conjunction to previous findings (sub-section 5.1). In the
140 next sub-section (5.2) the results of ADModel-P for concentration prediction are presented and analysed.
141 The original perspective of approaching correlations between controlling factors, P transformations and
142 concentrations follows (sub-section 5.3), while sub-section 5.4 places the findings in the context of other
143 river P models.

144 **2 Study area and field data**

145 The investigated area of River Swale (further referred to also as the river stretch or the study area) is
146 54km long, delimited at the upstream end by Catterick (National Grid Reference, NGR, SE225994508)
147 and at the downstream end by Crakehill (NGR SE426734). It includes three major tributaries (River
148 Wiske, Bedale Beck and Cod Beck) and 15 minor tributaries. This River Swale Catchment is a part of
149 the greater River Ouse Catchment, presented in Fig. 1.

150



151
152 **Fig. 1. The map of the study area and the Swale Catchment in the greater Ouse Catchment.**
153

154 The monitoring data used by House and Warwick (1998a and 1998b) and Bowes and House (2001)
155 along with additional field data will be used. The focus is on data from the late 1990s, which provides a
156 valuable perspective on model development as it covers a period where in general in-river legacy stores
157 had not yet started to be depleted. This is likely still the case in sensitive environments where large point
158 sources (as targeted by EU Urban Wastewater Treatment Directive legislation) are absent.

159 The field data include measurements at the monitoring stations (for the channel width, water depth,
160 water flow rate and P species concentrations) and estimations based on GIS (for the river bed slope)
161 corresponding to the main stream of the River Swale and the three major tributaries. The ten intensive
162 monitoring campaigns carried out at temporal resolution up to 3 hours are briefly described in Table 1.

Campaign number	Campaign timing	Total duration [h]	Water flow range [m ³ /s]	SRP range [mg/L]	OP range [mg/L]	Water temperature range [°C]	Seasonality factor range
C1.	September 1994	43	0.80 - 47.80	0.01 - 1.71	0.01 - 0.33	8.4 - 12.4	0.048 - 0.049
C2.	February 1995	40	0.80 - 41.50	0.01 - 0.47	0.03 - 0.56	4.0 - 7.1	0.377 - 0.381
C3.	October 1995	107	0.17 - 5.12	0.03 - 3.24	0.00 - 2.21	6.0 - 10.0	0.009 - 0.011
C4.	February 1996	117	2.66 - 57.70	0.01 - 0.38	0.00 - 0.50	0.7 - 4.7	0.349 - 0.371
C5.	April 1996	108	0.43 - 16.60	0.02 - 1.42	0.00 - 1.01	7.0 - 11.0	0.830 - 0.857
C6.	March 1998	190	0.53 - 196.67	0.01 - 0.97	0.00 - 6.34	5.0 - 9.0	0.414 - 0.464
C7.	July 1998	190	0.30 - 53.31	0.02 - 1.72	0.00 - 2.93	15.0 - 19.0	0.463 - 0.582
C8.	October 1998	190	0.22 - 29.67	0.02 - 2.37	0.00 - 0.82	10.2 - 14.0	0.019 - 0.026
C9.	July 1999	264	0.57 - 30.65	0.02 - 1.15	0.00 - 0.53	15.0 - 19.0	0.445 - 0.610
C10.	February 2000	336	0.55 - 90.39	0.02 - 1.09	0.00 - 1.02	2.9 - 7.0	0.287 - 0.368

Table 1 Description of field data employed for ADModel-P (SRP, Soluble Reactive Phosphorus; OP, Organic Phosphorus). Ranges regard values measured in the main channel and in the three major tributaries. The significance of the seasonality factor is explained in section 4.2.

Different environmental conditions are reflected, as data has been collected during a wide range of events from low discharge (e.g. event C3 in October 1995) to very high discharge during storms (e.g. event C6 in March 1998). The water flow graphs for campaigns C1 to C10 are available in Fig. D. 1 and Fig. D. 2. During these events the concentration of P components in the River Swale is heavily influenced by sinks and sources, which can be represented under two labels (1) in river processes (denominated also transformations), which cause consumption or generation of a certain chemical species and (2) discharges of effluents from outside the river (denominated also pollution sources), which introduce significant amounts of phosphorous presented under different forms (Neal et al., 2010; Withers and Jarvie, 2008). For River Swale there are point and non-point (diffuse) phosphorous sources (Bowes et al., 2008). Point sources (mainly Sewage Treatment Works, STWs) provide a continuous load of phosphorus to the river and strongly influence SRP, mainly during low flow. Their influence is less significant during high flows due to dilution. Non-point sources (agricultural and farming activities) predominate mainly during high flows (confirmed also by House and Warwick, 1998a), being associated with runoff and soil erosion and are the dominant sources of soluble unreactive phosphorus (SUP) (increasing TDP and consequently TP) and particulate phosphorus (PP) (increasing TP). Pollution sources are included in ADModel-P, as inputs, but their detailed investigation is out of the scope of the present paper. In contrast, in-river P transformations are comprehensively discussed.

The concentrations field data (measurements of phosphorus species concentrations) at the River Swale monitoring stations has been made public on HydroShare (Hutchins and Timis, 2020). This data

186 is part of a larger freely available accredited dataset by Leach et al. (2013), including nutrient
187 concentrations also for the River Swale. The water flow data is available online in the UK National River
188 Flow Archive (link available in the references list) at the specific locations for the two sites: the upstream
189 end of the investigated river stretch (station number 27090) and at the downstream end of the river stretch
190 (station number 27071).

191 **3 Methodology**

192 *3.1 Model background*

193 ADModel-P is based on the one-dimensional (1D) advection-dispersion equation (ADE) for mass
194 transport along distributed parameters systems (such as the rivers).

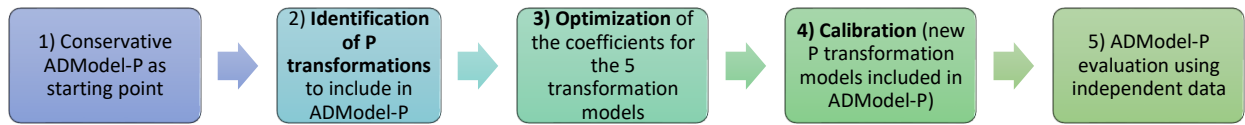
$$195 \quad \frac{\partial c}{\partial t} = \frac{\partial}{\partial x} \left(D \frac{\partial c}{\partial x} \right) - \frac{\partial (cV)}{\partial x} + S \pm T$$

196 The equation describes the evolution in time (t [s]) of the concentration (c [mg/L]) of a chemical
197 species along the river reach (x [m]), while the pollutant is discharged from sources (S [mg/s L]) and it
198 is consumed or produced in the river trough processes expressed by the rate of transformation (T [mg/s
199 L]). The concentration dynamics is also affected by the convective transport velocity (V [m/s]) and the
200 longitudinal dispersion coefficient (D [m²/s]).

201 For the development of ADModel-P an analytical solution of the 1D ADE suitable to describe the
202 pollutant dynamics during the continuous point discharge (Pujol and Sanchez-Cabeza, 2000) has been
203 implemented in MATLAB together with a model for the transformations. The appropriate description
204 of transformation processed during the in-stream transport is the main challenge related to the ADE
205 based approach.

206 *3.2 The steps to calibrate and verify the model*

207 The dynamic representation of transformation rates in ADModel-P has been guided by identifying
208 semi-empirical models to estimate the rates for each transformation process. A stepwise process was
209 implemented (Fig. 2).



211
212 **Fig. 2. Steps for ADMModel-P calibration and evaluation for P compounds**
213

214 1) The start point was a conservative advection-dispersion model with no transformations.

215 2) Representing the phosphorous transformations, which involves two sub-steps identifying (i)
216 which transformations to include and (ii) suitable mathematical relationships (initial form, to be refined
217 later during preliminary calibration) to represent their controlling factors. A robust pollutant transport
218 model needs identification of optimal complexity, as explored by Jackson-Blake et al. (2017) who sought
219 opportunity to simplify the complex INCA-P model. For ADMModel-P the identified transformations
220 model (see section 4) involves 5 processes each expressed mathematically with empirical rate
221 coefficients for calibration.

222 3) Optimisation of the empirical rate coefficients (the eight M and R described in Table 3). The
223 MATLAB built-in optimization function, *fmincon*, has been employed as basis for the optimization
224 framework developed for ADMModel-P. The framework key components are: (1) the objective function
225 which calculates the difference between the two types of concentration vectors (estimated vs.
226 measurements for both SRP and OP) based on the widely-used Nash Sutcliffe model efficiency
227 coefficient (NSE) (Nash and Sutcliffe, 1970); (2) the non-linear constraints function with respect to the
228 SRP and OP concentrations, in order not to allow negative values of the estimated concentrations; (3)
229 lower and upper limits for the decision variables (M and R coefficients) and (4) the starting point for
230 each decision variable. It is crucial how the options of the optimization function (*optimoptions*) are
231 defined, because different kind of settings related to issues such as the optimization algorithm, the
232 tolerances or the search stopping criteria could lead to very different solutions. The eight decision
233 variables are modified automatically by the optimization algorithm at each computation cycle to achieve
234 a minimum value of the objective function (corresponding to the maximum of the NSE coefficient) while

235 meeting the non-linear constraints, the tolerances and the limits for the decision variables. The initial
236 values of the rate coefficients employed by ADModel-P as start point for the optimization have been
237 estimated based on QUESTOR rates for mineralization and sedimentation (transformations included in
238 QUESTOR, as employed by Hutchins et al. (2010)), while for the other transformations the initial rate
239 coefficients are zero. Observed data for P compounds are taken from 8 of the 10 calibration data sets
240 excluding two campaigns to be used later for model evaluation.

241 4) The optimum forms of the models for transformations have been further introduced in
242 ADModel-P in order to finalize its calibration. Following implementation in ADModel-P, applications
243 demonstrate model performance against experimental data for the eight model development campaigns.

244 5) ADModel-P was later verified using independent data from the remaining two monitoring
245 campaigns.

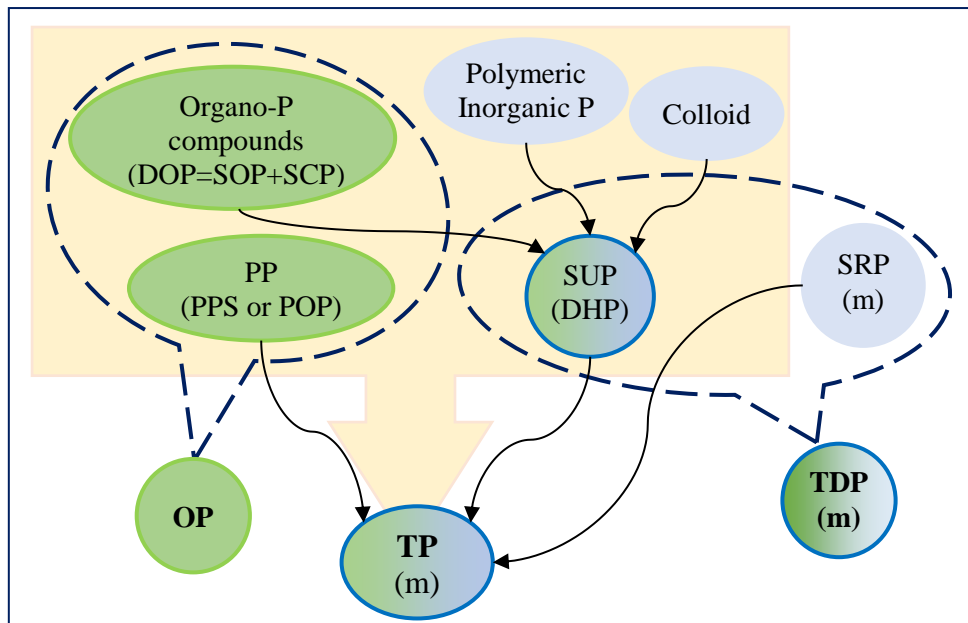
246 Multiple runs of preliminary model calibration followed by evaluation (steps 3 to 5) have been
247 carried out employing each time two different independent campaigns for the model evaluation. The
248 selection of the two evaluation campaigns has been done fulfilling two main rules: (1) evaluation events
249 should take place in different times of the year, in order to ensure variability over seasonality; and (2)
250 evaluation events should take place at different flow regime. Otherwise, the combinations are random.
251 The purpose of these preliminary runs was to identify the trends of transformation processes in relation
252 to controlling factors, in order to validate the most suitable form of the equations for the process rates
253 (initially identified in step 2 above and in final form as described in section 4.2) and the appropriate
254 settings for the optimization framework (step 3 above). The results in terms of goodness of fit associated
255 to the preliminary runs are presented in Table B. 1 (in Appendix B). Further, the finalised identification
256 of equations and optimization settings has been implemented in ADModel, which then underwent a final
257 round of calibration and evaluation employing a chosen data splitting option (#2 in Table B. 1 in
258 Appendix B).

259 ADModel-P has been made public on HydroShare (Timis, 2021).

260 **4 The identification of the phosphorus transformations model to employ**

261 *4.1 Phosphorus components and transformations*

262 Phosphorus is present in multiple constituents of the river ecosystem (algae, zooplankton and aquatic
263 animals; water; river channel, e.g. sediments, river banks soil) and under multiple forms (Ji, 2017; Chau,
264 2005; Loucks and van Beek, 2017). Terminology has not always been consistent with different
265 appellations or assumptions used and wider terms often introduced (e.g. organic P and particulate P used
266 as interchangeable), hence these are comprehensively related in tabular (Table A. 1 in Appendix A) and
267 graphical (Fig. 3) as employed in the literature (Charlton et al., 2018; Dou et al., 2018; Ji, 2017; Tye et
268 al., 2016; Tuo et al., 2015; Neal et al., 2010; Bowes et al., 2005 and 2003; House, 2003; Wade et al.,
269 2002; House and Warwick, 1998a; Bowie et al., 1985).



270
271 **Fig. 3. In-stream phosphorus compounds and their general relations, not involving processes. Acronyms and**
272 **explanations are available in Table A. 1 in Appendix A (m = measurable, P, Phosphorus; DOP, Dissolved Organic**
273 **P; SOP, Soluble Organic P; SCP, Soluble Complex P; PPS, POP or PP, Particulate P; OP, Organic P; TP, Total P;**
274 **SUP or DHP, Soluble Unreactive P; SRP, Soluble Reactive P; TDP, Total Dissolved P).**
275

276 Visual representation of processes (associated to models and not only) typically uses one of three
277 different approaches, depending on the purpose of the case study: (1) a “generalist” approach employing
278 an all-inclusive P term contained in phases (e.g. P in water and P in sediments, as in Wijesiri et al.,
279 2019); (2) simple representations of the phosphorus cycle, usually involving one type of organic P (under
280 different designations, e.g. OP, PP) and one type of inorganic P (under different designations, e.g. SRP,

281 IP), (Hutchins et al., 2010, a QUESTOR application including the Swale, and also Srinivas and Singh,
 282 2018; Jackson-Blake et al., 2017; Cooper et al., 2015; Lindenschmidt et al., 2007); and (3) more complex
 283 representations, involving more components and multiple interdependencies (Dou et al., 2018; Alam
 284 and Dutta, 2016; Kim et al., 2006; Loucks and van Beek, 2017; Bowie, 1985). Robson (2014) states that
 285 “catchment - river models are almost universally on the simpler side with respect to their treatment of
 286 in-stream processes”, sometimes only considering TP and either representing sedimentation alone, or
 287 not representing in-stream processes thereby assuming that nutrient sources from outside the river are
 288 determinant on the in-river nutrient load. Of the post-2003 river models reviewed by Robson (2014),
 289 74% include less than 3 biogeochemical or ecological processes; and Robson (2014) also states that
 290 ecological models for rivers often do not consider phosphorus at all.

291 There are however a wealth of models taking into account more than one phosphorus species and
 292 multiple in-stream processes. The processes either act as sinks or sources for each category of
 293 compounds (Table 2). A sub-set of the processes in Table 2 are chosen later for specific modelling of P
 294 compounds transport by ADModel-P in River Swale.

<p style="text-align: center;">Dissolved organic P</p> <ul style="list-style-type: none"> + excretion by living organisms + respiration of living organisms + plant decay + decomposition of detritus + chemical transformations + release from sediments - chemical processes: mineralization, hydrolysis - sedimentation 	<p style="text-align: center;">Dissolved inorganic P</p> <ul style="list-style-type: none"> + excretion by living organisms + chemical processes: mineralization of OP, hydrolysis of DOP, oxidation of PP + decomposition of detritus + release from sediments - uptake by plants - sorption by sediments - chemical processes
<p style="text-align: center;">Particulate organic P</p> <ul style="list-style-type: none"> + excretion by living organisms + plankton mortality + zooplankton grazing + decomposition + resuspension - sedimentation - chemical processes: acid oxidation 	<p style="text-align: center;">Sediment P</p> <ul style="list-style-type: none"> + detritus settling + algal settling + sedimentation of PP - sediment decomposition and release - resuspension - chemical processes: mineralization

295 **Table 2 Main transformations governing different types of P forms (P, Phosphorus; OP, Organic Phosphorus; DOP,**
 296 **Dissolved Organic Phosphorus; PP, Particulate Phosphorus): (+) is associated to formation/ sources, while (-) to**
 297 **consumption / sinks**
 298

299 *4.2 Phosphorus transformations included in ADModel-P*

300 As stated earlier, P transformation are important in the Swale. House and Warwick (1998b) identified
301 large losses of SRP associated with uptake by the river bed, suspended sediments, macrophytes and
302 algae. Mass balance calculations for TDP, SRP and TP (House 2003) determined the relative importance
303 of riverine processes in controlling P concentrations, and discriminated those in terms of whether they
304 were chemically or physically driven. Bowes et al. (2003; 2005) conclude that the interaction with the
305 river bed and sediments (e.g. mineralisation) is likely significant within the River Swale and that
306 sedimentation and resuspension should be considered by models.

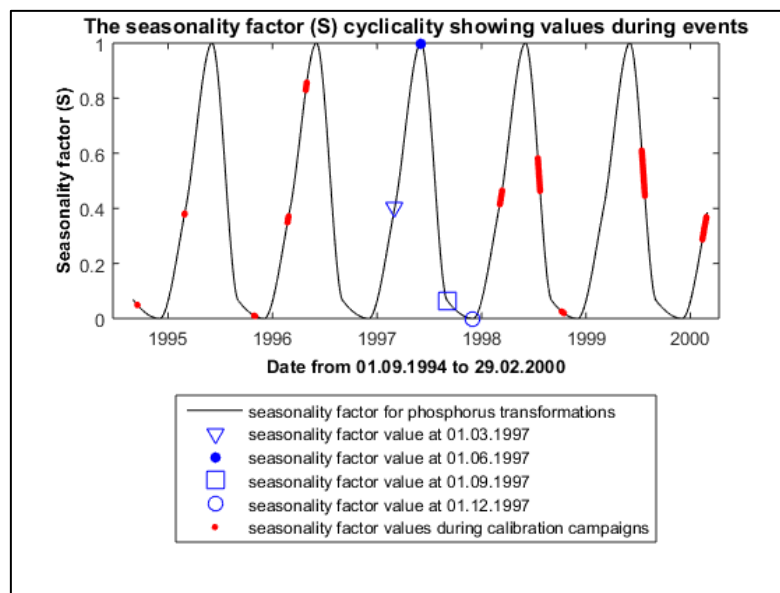
307 Five categories of in stream P transformations were identified as being important: mineralization,
308 uptake, adsorption-desorption, sedimentation, and resuspension. These are deemed responsible for the
309 loss and/or accumulation of SRP and OP during the transport. The rates of the chemical processes are
310 considered as having first order kinetics, while the physical processes are zero order kinetics. They are
311 mainly influenced by the water temperature (T), water flow (Q) and seasonality. Each rate depends on a
312 particular transformation rate constant (k) as shown in Table 3, where the form of the ADModel-P
313 transformations model equations is presented.

Process	Details
Mineralization	$r_{\min} = k_{\min} c_{OP}$, where $k_{\min} = M_{\min} \log(1 + T)$ Sink of OP, source of SRP, accounting for the conversion of organophosphates and inorganic polyphosphates to SRP.
Sedimentation	$r_{\text{sed}} = k_{\text{sed}} c_{OP}$, where $k_{\text{sed}} = \frac{M_{\text{sed}}}{(1+Q)}$ Sink of OP, including the sorption to sediments, sedimentation of PP.
Resuspension	$r_{\text{res}} = k_{\text{res}}$, where $k_{\text{res}} = M_{\text{res}} \log(1 + Q) + R_{\text{res}} (1 + S)$ Source of OP, including the resuspension of fine sediment, mobilisation of material from the river-bed, in-wash from the floodplain, desorption of P from sediments, resuspension of detritus from the bed sediment followed by the formation of TDP through the decomposition of plants and detritus. The respiration as a source is included here as well.
Uptake	$r_{\text{upt}} = k_{\text{upt}}$, where $k_{\text{upt}} = M_{\text{upt}} \log\left(1 + \frac{1}{1+Q}\right) + R_{\text{upt}} \log(1 + S)$ Sink of SRP, accounting for the biological uptake by macrophytes, phytoplankton and benthic algae.
Adsorption - Desorption	$r_{\text{ads-des}} = k_{\text{ads-des}} c_{\text{SRP}}$, where $k_{\text{ads-des}} = M_{\text{ads-des}} \left(1.5 + \log\left(\frac{1}{1+T}\right)\right) + R_{\text{ads-des}}$ Sink of SRP, including the sorption of SRP to sediments and to the soil and also the precipitation reactions of SRP. Source of SRP including SRP release from sediments. The resulting value shows the net adsorption as difference between adsorption and desorption.
Where r is the transformation process rate [mg/L s]; T is the measured water temperature [°C]; Q is the water flow [m ³ /s]; S is the seasonality factor; c_{SRP} and c_{OP} [mg/L] are the estimated values of concentrations of SRP and OP; k are the transformation rate constants for the processes of uptake k_{upt} [mg/L day], adsorption - desorption $k_{\text{ads-des}}$ [1/day], mineralization k_{\min} [1/day], sedimentation k_{sed} [1/day], and resuspension k_{res} [mg/L day]; M and R are constants assigned with specific values for each transformation during the model calibration.	

314 **Table 3 Details on the transformations model developed for ADModel-P (OP, Organic Phosphorus; SRP, Soluble**
315 **Reactive Phosphorus).**
316

317 The processes included in the SRP uptake by macrophytes, phytoplankton and benthic algae and also
318 those under the formation of TDP through the decomposition of plants and detritus are likely to vary
319 substantially on a seasonal basis. Across the seasons, the aggregated impacts of growth and then decay
320 of primary producers reflect an integration of climate-driven factors on a longer timescale than specific
321 diurnal environmental fluctuations. Therefore, in addition to applying water temperature controls on
322 these transformation rates we invoke seasonal variability into the rate parameters defining sources and
323 sinks. We consider two key variables (the Julian Day number (JDN) and the factor of seasonality or
324 seasonality factor (S)), quantified in the context of the monitoring campaigns (Fig. 4). The JDN allows
325 the representation of multiple events in the time succession they happen and facilitates the

326 implementation of continuous functions to describe the cyclical influence of seasonality along the
 327 monitoring campaigns. A non-dimensional scalar determined by a function based on the sine of the JDN
 328 defines S (see Fig. 3), which takes values between 0 (beginning of winter) and 1 (beginning of summer)
 329 (with an example provided for 1997). This cyclical variation of S represents the continuous periodicity
 330 in the intensity of processes fluctuating cyclically during the year. For example, in the models for
 331 resuspension and uptake the intra-annual variability is expected to be strongly determined by S, process
 332 rates being expected to increase during spring (Angert et al., 2011). The conjunctive use of JDN and S
 333 results in a continuous function describing seasonality in transformation rates models. S is included
 334 separately for each transformation rate constant as it may have differing significance.



335
 336 **Fig. 4. Representation of the seasonality factor continuity and the values assigned to the monitoring campaigns. x-**
 337 **axis labels correspond to the 1st Jan for each year.**
 338

339 5 Results and discussions

340 5.1 Transformations model

341 Values are shown for the M and R coefficients identified during optimization for the calculation of
 342 each transformation rate (as in equations in Table 3) and the resulting values of rate coefficients (Table
 343 4). Resuspension is the largest contributor to the variability of OP, while adsorption – desorption are the
 344 largest contributors to the variability of SRP (in Table 4 positive values mean net sink).

Transformation	M	R	Process indicators	Calibration			Evaluation		
				Min	Aver	Max	Min	Ave	Max
Mineralization	1.137	NA	k, day ⁻¹	0.60	2.47	3.41	1.83	3.21	3.41
			rate, g m ⁻³ day ⁻¹	0.03	0.37	2.39	0.11	0.16	0.38
			flux at M4, kg day ⁻¹	802.95			224.28		
Sedimentation	49.415	NA	k, day ⁻¹	0.25	4.06	14.00	1.16	4.36	7.62
			rate, g m ⁻³ day ⁻¹	0.02	0.57	1.89	0.07	0.25	0.34
			flux at M4, kg day ⁻¹	551.16			304.32		
Resuspension	0.315	-0.204	k, g m ⁻¹ day ⁻¹	0.19	0.60	1.37	0.28	0.48	0.90
			rate, g m ⁻³ day ⁻¹	0.19	0.54	1.37	0.28	0.42	0.90
			flux at M4, kg day ⁻¹	1120.25			590.72		
Uptake	0.757	0.382	k, g m ⁻³ day ⁻¹	0.03	0.17	0.34	0.14	0.22	0.28
			rate, g m ⁻³ day ⁻¹	0.03	0.18	0.34	0.14	0.24	0.28
			flux at M4, kg day ⁻¹	225.17			281.50		
Adsorption – Desorption	3.795	19.064	k, day ⁻¹	13.39	16.51	22.74	13.39	14.04	18.65
			rate, g m ⁻³ day ⁻¹	0.00	2.11	3.80	0.39	1.76	3.14
			flux at M4, kg day ⁻¹	2815.09			2077.20		
NA - not applicable; M and R are transformations model coefficients; k – rate coefficients; Max (maximum), Min (minimum), Ave (mean) values are calculated based on the instantaneous values from dynamic series corresponding to each transformation process during calibration and evaluation).									

345 **Table 4 Transformation model results obtained during optimization.**
346

347 Rate coefficients vary during the events (as illustrated in Fig. C.1 in Appendix C) and their variation
348 is influenced by the change of controlling factors, as illustrated in Appendix C: Fig. C.2 for the water
349 flow, Fig. C.3 for the water temperature and Fig. C.4 for the seasonality. Figures show relations to be
350 expected from the basics of the transformations model in Table 3 and reveal agreement with previous
351 findings. Resuspension, followed by mineralization influence the most organic species, while for
352 inorganic species adsorption-desorption processes predominate, as found by House (2003). Highest
353 adsorption rates occur at lowest temperatures (campaign #4: average of 2.7 °C), whilst lowest rates (i.e.
354 highest desorption) occur during campaign # 7 (average of 17 °C), findings in agreement with the
355 knowledge that higher water temperatures facilitate anoxic conditions which enhance the release of P
356 from sediments (Han et al., 2022; Mbabazi et al., 2019; House, 2003). There is a positive relationship
357 between water temperature and mineralization rate (as also found by Yuan et al., 2019). Its variability
358 within campaigns is caused by diurnal variation of water temperature (Caissie et al., 2013). Resuspension
359 (which is the largest contributor to the variability of OP) shows greatest relative variability at low flows,
360 as observed by Yuan et al. (2019) and House (2003). Highest rates of resuspension are associated with
361 higher flow and moderate S (seasonality: i.e. early spring). Uptake has the lowest values (Table 4,
362 varying substantially with seasonality showing highest rates in summer when flows are also low. SRP

363 losses due to uptake are much lower than those due to other sinks, as also found by House (2003) and
364 Jarvie et al. (2005) in nutrient-rich rivers.

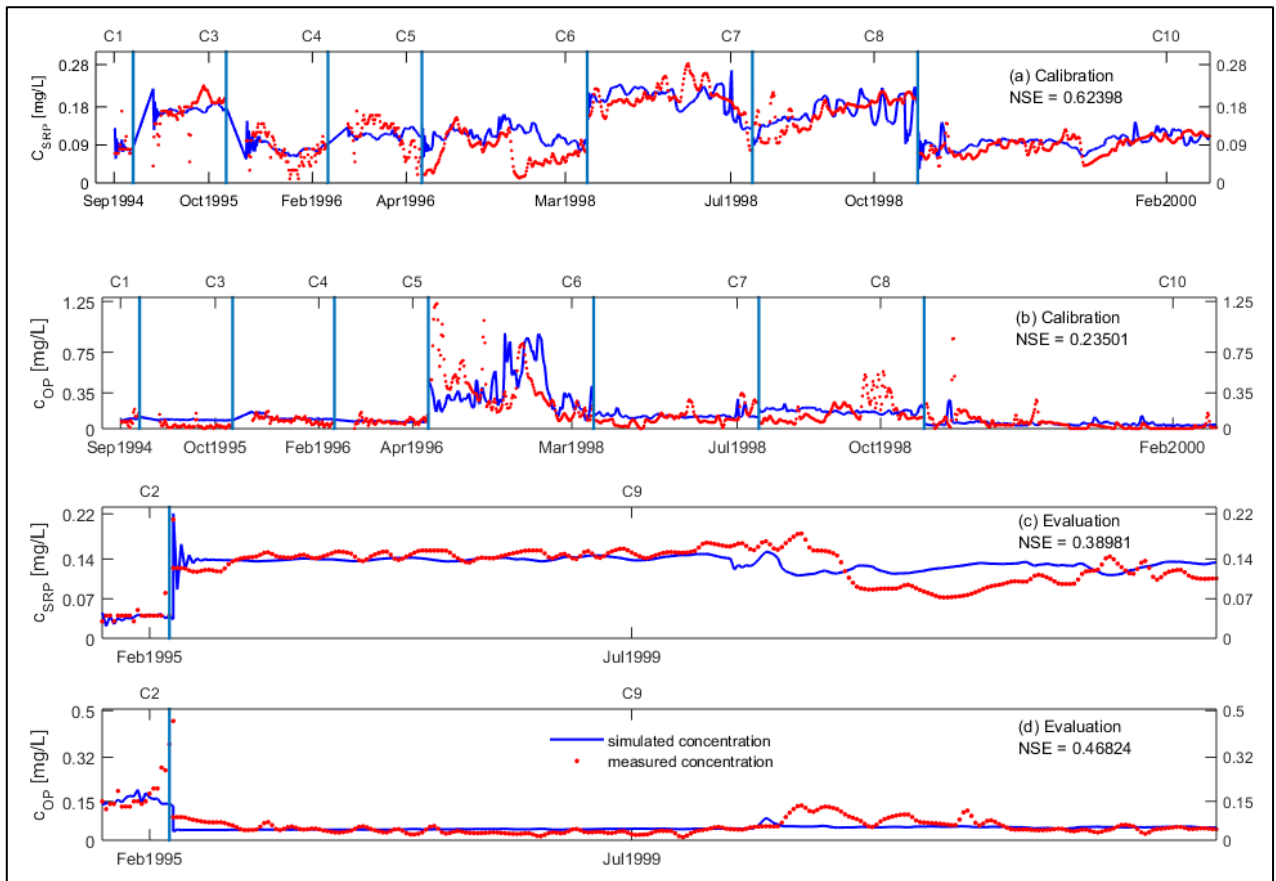
365 *5.2 ADMoDel-P calibration and evaluation results*

366 The ADMoDel-P inputs consist of the dynamic profiles of water flow and SRP and OP concentrations
367 (1) at the upstream end (acting as boundary condition), (2) at the tributaries and (3) at the actual pollution
368 sources along the stretch. The outputs consist of the dynamic profiles of concentrations of SRP and OP
369 at the downstream end (labelled M4).

370 Outputs for all 8 development and calibration campaigns and the 2 evaluation campaigns (#2 and #9,
371 kept independent during all earlier stages) are presented (Fig. 5a-d, Fig. E. 1 and Fig. E. 2). It has been
372 decided to simulate all campaigns together as a continuous event (Fig. 5a-d), using concatenated data
373 from individual campaigns (C1 to C5 in Fig. E. 1 and C6 to C10 in Fig. E. 2) in order to test the capability
374 of ADMoDel to cater for very rapid changes in time at campaign boundaries, as controlling factors and
375 other variables change between campaigns.

376 It is demonstrated that evaluation campaigns encompass a representative range of conditions with
377 respect to: (1) water flow, storm during campaign #2 and normal flow during campaign #9; (2) water
378 temperature, mean of 6 °C and 17 °C for campaigns #2 and #9 respectively; and (3) seasonality, late
379 winter during campaign #2 and mid-summer during campaign #9. Results show that simulations follow
380 the general trend of measurements for both SRP and OP, despite underestimation of peaks in case of
381 events #3, #5 and #7 for the calibration of SRP, events #6 and #8 for the calibration of OP and event #2
382 for the evaluation of OP. On the other hand, the lowest concentrations of OP are slightly overestimated
383 while the lowest concentrations of SRP are overestimated on few occasions.

384



385
 386 **Fig. 5. Dynamic profiles of concentration during calibration (subplots a and b) and evaluation (subplots c and d)**
 387 **presented at hourly resolution at the downstream end of the river stretch. The time x-axes are not continuous. Data**
 388 **from the individual campaigns C1 to C10 carried out during the specified months (Table 1) has been concatenated**
 389 **during simulations. The vertical lines mark the boundaries between C1 to C10.**
 390

391 The ADModel-P performance has been evaluated using the multiple criteria, as shown in Table 5,
 392 where values for the non-conservative runs (employing the transformations model) are included. The
 393 criteria equations are available in Table B. 2. Additionally, the conservative runs have been assessed
 394 with the NSE, and values are -1.65 for the calibration of SRP, -4.77 for the evaluation of SRP,
 395 respectively 0.13 and 0.35 for OP. In all occasions, as expected, due to the high importance of
 396 transformations, the non-conservative version of the model behaves better compared to the conservative
 397 version.

398

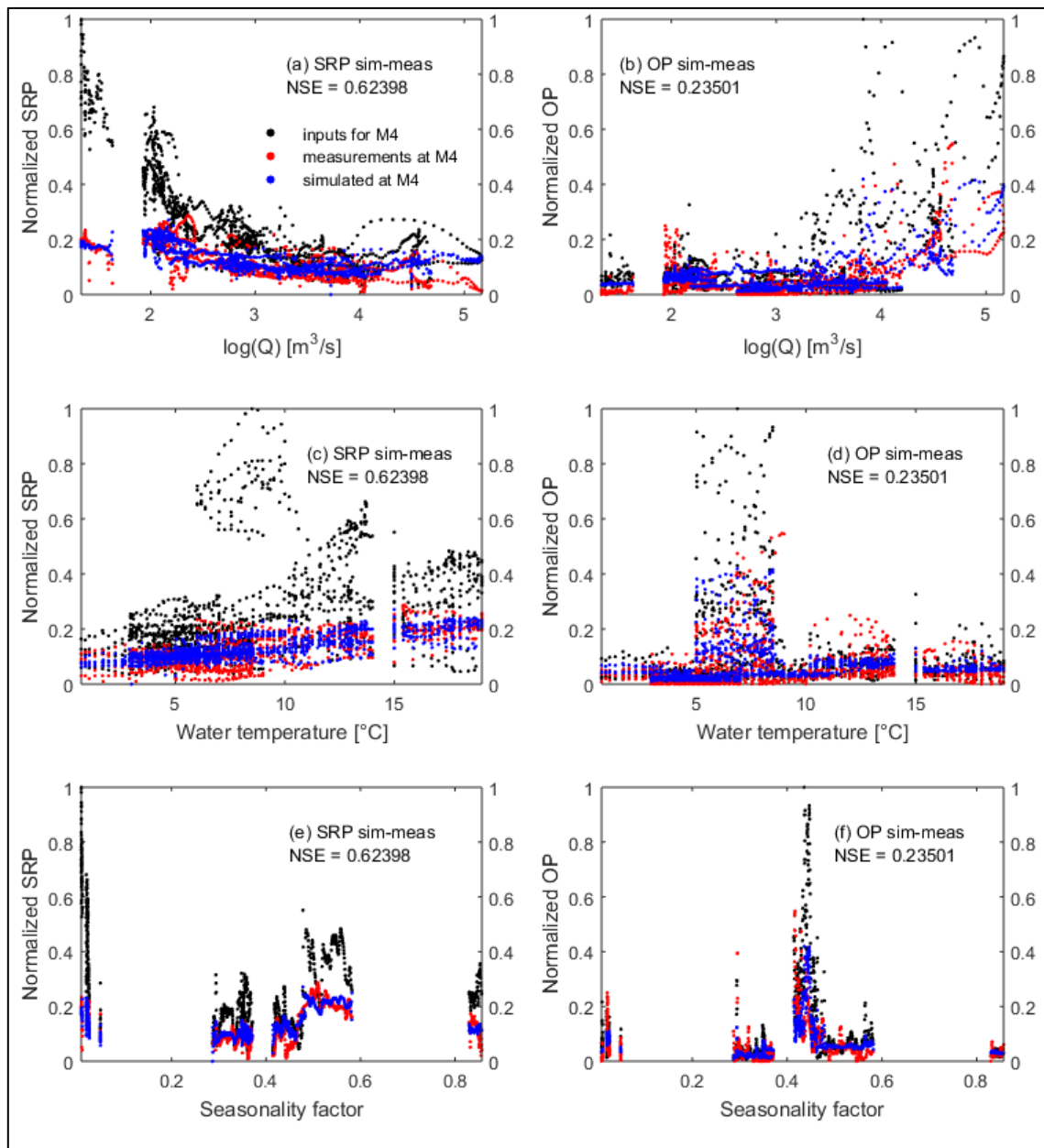
Measured concentration	Unit of measure	Calibration		Evaluation		
		SRP	OP	SRP	OP	
Minimum	mg/L	0.01	0.00	0.03	0.04	
Average	mg/L	0.32	0.38	0.38	0.16	
Maximum	mg/L	0.87	3.76	0.57	0.46	
Standard deviation	mg/L	0.05	0.17	0.03	0.05	
Model evaluation criteria	Range	Optimal value	Calibration		Evaluation	
			SRP	OP	SRP	OP
Nash Sutcliffe efficiency (NSE)	$-\infty$ to 1	1	0.62	0.24	0.39	0.47
Percent bias (PBIAS)	$-\infty$ to ∞	0	-6.78	-6.92	-0.99	11.24
Root mean square error ratio to observations standard deviation (RSR)	0 to ∞	0	0.61	0.87	0.78	0.73
Kling-Gupta Efficiency (KGE)	$-\infty$ to 1	1	0.71	0.53	0.57	0.47
Modified coefficient of determination (bR^2)	0 to 1	1	0.61	0.31	0.43	0.44

399 **Table 5 The range of measured concentrations and the goodness of fit associated to ADModel-P runs at hourly**
400 **resolution during calibration and evaluation.**

401
402 Moriasi et al. (2007) discuss models for the streamflow and transport of sediments, nitrogen and
403 phosphorus at larger time steps (mostly daily and monthly) in relation to NSE (range from $-\infty$ to 1), RSR
404 (range from 0 to large positive values) and PBIAS and find that the “minimally acceptable” model
405 performance is indicated by NSE > 0.00, while satisfactory simulations are indicated by NSE > 0.50
406 together with RSR \leq 0.70 and PBIAS < $\pm 70\%$ (for nitrogen and phosphorus). The results for detailed
407 hourly simulations of ADModel-P achieve NSE values close to the thresholds of 0.50 (NSE) and
408 respectively 0.70 (RSR) for the calibration of SRP and the evaluation of OP, while for the evaluation of
409 SRP and calibration of OP the values are in the acceptable limits for the model performance. The low
410 values of PBIAS in Table 5 indicate good simulations (according to Gupta et al., 1999 and Moriasi et
411 al., 2007), with small overestimation bias for SRP during calibration and evaluation (the negative values
412 of PBIAS) and for OP during calibration, while there is underestimation bias for OP during evaluation
413 (the positive value of PBIAS). The KGE values between 0.47 and 0.71 indicate good results of
414 ADModel, if taking into account the limit of KGE > 0.3 or NSE > 0.5 mentioned by Knoben et al. (2019)
415 in relation to hydrological models. Positive values of KGE indicate good model performance, while
416 negative KGE values indicate bad model performance (Knoben, 2019; Gupta et al., 2009). The bR^2
417 values (Krause et al., 2005) are between 0.31 and 0.61 when the intercept is zero and gradients between
418 0.94 and 1.15. Further on, the performance of ADModel is discussed in Section 5.4 in a wider context,
419 providing a comparison with other models.

420 5.3 Wider correlations of SRP and OP dynamics to controlling factors, using ADMoDel-P

421 In order to understand the underlying phenomena behind these results, correlate it to the modelled
422 transformation rate constants and their controlling factors, and understand in which situations the river
423 is acting as a source or sink Fig. 6 is offered. There are 3 different series of concentration data provided
424 in the same graph: (1) the total value of sources influencing the concentration at M4 (i.e. inputs for M4
425 which comprise pollution sources, tributaries and the concentration at the upstream end of the stretch,
426 labelled M1); (2) the measurements of concentration at M4 and (3) the simulated concentration at M4.



427
428 **Fig. 6** Normalized values of SRP and OP concentrations (observed, calculated sources and simulated) against water
429 flow (subplots a and b), water temperature (subplots c and d) and seasonality factor (subplots e and f) during the 8
430 development and calibration campaigns. Normalized values obtained by dividing the actual value of the variable in
431 the dynamic series to the maximum value of all three series.

432

433 With respect to water flow (Fig. 6a-b) it is observed that relationships between inputs and outputs
434 are similar below $6 \text{ m}^3/\text{s}$ (i.e. $\log(Q)$ of $1.79 \text{ m}^3/\text{s}$) and above $105 \text{ m}^3/\text{s}$ (i.e. $\log(Q)$ of $4.65 \text{ m}^3/\text{s}$), where
435 consumption along the stretch occurs for both SRP and OP. At low flows for OP this is attributed to
436 sedimentation, although there is a tendency to overestimate OP concentrations. The sinks of SRP seem
437 to be prominent along the stretch on the entire flow range, except between $22 \text{ m}^3/\text{s}$ and $55 \text{ m}^3/\text{s}$ (i.e.
438 $\log(Q)$ of 3.09 to $4.01 \text{ m}^3/\text{s}$), where there are alternating periods of net sources and sinks (Fig. 6a). For
439 mid-range flows between $6 \text{ m}^3/\text{s}$ and $105 \text{ m}^3/\text{s}$ the flow range where the model simulates SRP and OP
440 observations most closely, for OP, there are alternating periods of net sources and sinks. At flows above
441 $105 \text{ m}^3/\text{s}$ both SRP and OP concentrations are generally overestimated. The implication could be: (1)
442 errors in representation of dilution processes at high flows; (2) additional sinks along the stretch at high
443 flows which are unaccounted for; (3) sinks included in ADModel-P are sufficient but are underestimated
444 at high flows. Refinements increasing the rates of existing sinks may be incorporated if supported by
445 field evidence of other affected in-river conditions (e.g. biotic indicators), but the additional complexity
446 may lead to a less reliable and over parameterised model overall, despite the improved performance.

447 With respect to water temperature, sinks of SRP are prevalent throughout apart from some occasions
448 above $16 \text{ }^\circ\text{C}$ (Fig. 6c), which may be due to desorption (Fig C.3b). OP sinks and sources alternate along
449 the stretch on the entire temperature range, except between $5 \text{ }^\circ\text{C}$ to $10 \text{ }^\circ\text{C}$ where additional sinks are
450 evident. In this interval, ADModel-P gives better OP predictions compared to temperatures above $12 \text{ }^\circ\text{C}$
451 when overestimation occurs in multiple occasions.

452 *5.4 With respect to seasonality, there is evidence of SRP sinks at almost all times, except*
453 *the S range 0.45 to 0.50 (corresponding to mid spring) where there is evidence of*
454 *SRP sources (Fig. 6e). ADModel-P gives fair predictions, except at S values around*
455 *0.45, 0.50 and 0.84 (corresponding to mid-end spring) where SRP is often*
456 *overestimated, despite the large simulated uptake values. There is no clear evidence*

457 *that additional OP sinks or sources are dominant at certain times (Fig. 6f), with the*
458 *exception of the S range 0.45 to 0.60, where in-stream sinks of OP are evident.*
459 *Overestimations of OP occur between spring and mid-summer (S above*
460 *approximately 0.49), while underestimations occur at the beginning of winter (S*
461 *around 0.02). It is suspected that accuracy at different seasons may be improved by*
462 *an improved biotic processes representation (taking into account additional*
463 *controlling parameters, such as solar radiation or algae), since there is an*
464 *overestimation of both SRP and OP in spring when biological processes are most*
465 *significant. Putting findings in the context of other river phosphorus models*

466 ADModel-P represents a phenomena based simulation of phosphorus dynamics in rivers. It differs
467 from other widely available models in that it uses information on river flow, water temperature and
468 seasonality to generate empirical relations to summarise the variation in process rates. Constants (M and
469 R values) are estimated and applied to these functions to generate rate constants (k) values. The rate
470 constants, coupled with zero or first order dependency on the source P pool, define variation in fluxes
471 of phosphorus. In contrast deterministic models such as INCA-P (Jackson-Blake et al., 2016), SWAT
472 (Neitsch et al., 2011), CE-QUAL-W2 (Cole and Wells 2006) and EFDC (Wu and Xu 2011) directly use
473 theoretical relationships along a spectrum of complexity in hydrobiochemical dynamics (for example
474 isotherm kinetics, particulate settling, biotic uptake related to primary productivity) and require a number
475 of additional measurements or assumptions to define these fluxes.

476 The modelling in the present study demonstrates that if Q, T and S are known then considerable
477 model skill (NSE=0.24-0.62), can be achieved in simulating sub-daily dynamics of OP and SRP in the
478 River Swale system. Evidence from Wellen et al. (2015) suggests that despite a relatively small number
479 of parameters for optimisation (8 empirical coefficients) this compares favourably to other studies at
480 sub-daily resolution. Sub-daily studies are scarce (approximately 2% of studies). Of the studies reviewed

481 by Wellen et al. (2015) few appear to address identification of in-channel processes reflecting a trend
482 towards whole-basin large-scale system assessment, as identified in Section 1 above.

483 It is instructive to compare ADModel-P process representation and performance with the Simply-P
484 model (Jackson-Blake et al., 2017) which represents a simplification of INCA-P, having an order of
485 magnitude fewer calibration parameters and achieved without a substantial sacrificing model
486 performance. River particulate phosphorus in Simply P is entirely controlled by particulate behaviour
487 (sedimentation and resuspension). Additionally for dissolved P, although substantial retention has been
488 identified from controlled channel experiments (House et al., 1995), the Simply-P model omits biotic
489 uptake and net adsorption, treating it as a conservative substance within the channel. Jackson-Blake et
490 al. (2017) evaluated the model comparison in a small Scottish rural catchment otherwise similar to the
491 Swale. Short residence time gives limited opportunity for riverine processes to become a significant part
492 of catchment mass-balance. In identifying the importance of net sedimentation and the lack of influence
493 of biotic uptake, the present study corroborates Jackson-Blake et al. (2017), although contrastingly
494 identifies importance of desorption and conversion of inorganic to organic forms to be important.
495 Performance statistics for the two models are broadly similar.

496 **6 Conclusions**

497 ADModel-P is a powerful simulation tool capable to make detailed dynamic predictions, with
498 reasonable accuracy for the case of River Swale. The complexity of phenomena involving P compounds
499 during their transport along the river has been treated with special care during our demarche of finding
500 the optimum representation of transformations included in ADModel-P. Five transformations have been
501 considered (mineralization, sedimentation, resuspension, uptake and adsorption - desorption) and
502 thereby estimated using empirical coefficients catering for a wide range of conditions with respect to
503 water flow, water temperatures and seasonality. The values identified for the coefficients and
504 consequently for the process rates reveal that interaction with sediments and transformation of inorganic
505 P to organic P species are the dominant processes, while the uptake has the lowest influence on inorganic
506 P variability. Simulation results are in agreement with results based on experimental data showing

507 interactions between water and sediments to be highly significant. Overall, sinks of phosphorus appear
508 to be more important than sources during most of the events studied. Results suggest adsorption-
509 desorption processes are likely to predominate. The rates of interaction with the sediments are highly
510 influenced by the water flow (via the sedimentation and resuspension rates) by the water temperature
511 (via adsorption-desorption rate), and by seasonality (via resuspension rate). The conversion of inorganic
512 to organic forms of P, as well as the mineralization of OP to SRP are influenced by water temperature.
513 The biological uptake of SRP is influenced by the water flow and seasonality.

514 Model skill has been discussed in the context of the transformation rate coefficients and their main
515 controlling factors. Visual correlations have been offered as support for discussions. Such information
516 can be of additional use as support for other projects related to the identification and formulation of
517 transformation models for different case studies. Work to further improve performance of ADModel-P,
518 may include the estimation of uptake using chlorophyll measurements, or adding new pollutant species
519 for which field data is available (e.g. silica).

520 Models such as ADModel-P are needed to support water quality management in all rivers. These
521 include those less polluted river networks such as the Ouse of which the Swale is a part, which has been
522 evaluated as oligotrophic/mesotrophic by Hutchins et al. (2010), but with risk of degenerating to a
523 mesotrophic/eutrophic system by 2080. The wider implications from this point of view are the options
524 to adapt ADModel-P to other monitored river stretches over different time-scales which may have levels
525 of accrual or depletion of legacy phosphorus contrasting to those in the Swale in the 1990s. Studies using
526 ADModel to identify change in process rate parameters through a set of sequential applications at
527 intervals before and after undergoing mandatory upgrades in wastewater treatment would improve
528 mechanistic understanding of how river systems respond to interventions and the extent to which legacy
529 effects weaken the expected benefits. The model structure is based on general principles of flow routing
530 pollutant transport and phosphorus transformations and whilst site-specific calibration is necessary it is
531 transferable to any other situation. The approach described here, whereby process rates are estimated

532 based on readily-available hydro-climatic and seasonal information, can also be used as a means to
533 constrain analogous parameters used in other models.

534 **7 Acknowledgements**

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537 basin. The Environment Agency and Yorkshire Water provided data for model testing.

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- 794

795 **Appendix A**

796 **Table A. 1 Phosphorus main forms/ compounds in the river.**

P form/ components	Short explanation
SRP	Measurable component.
Soluble Reactive Phosphorus	Routinely termed also: Phosphates; Orthophosphates; Ortho-phosphorus; Bioavailable Phosphorus, Dissolved Inorganic Phosphorus (DIP), Dissolved Reactive Phosphorus (DRP), Inorganic Phosphorus (IP), Molybdate Reactive Phosphorus (MRP), Filterable Reactive Phosphorus (FRP). The labile phosphorus component in water. The SRP fraction is biologically available. Largely equivalent to Total Reactive Phosphorus (TRP), the most relevant component when assessing the eutrophication risk.
SUP	Sometimes referred to as Dissolved Hydrolysable Phosphorus (DHP).
Soluble Unreactive Phosphorus	The non-labile component of the TDP. It represents a mixture of polymeric inorganic phosphorus, colloidal material and organo-phosphorus compounds: Soluble Organic Phosphorus (SOP) and Soluble Complex Phosphorus (SCP). Calculated (not measured) as TDP – SRP. SOP and SCP compose the Dissolved Organic Phosphorus (DOP) denomination used often as equivalent to SUP.
TDP	Measurable component.
Total Dissolved Phosphorus	Composed of the labile (SRP) and no-labile (SUP) components in water, presented as soluble inorganic and soluble organic P. Only a part of the dissolved organic P is considered immediately available.
TP	Measurable component.
Total Phosphorus	Composed of SUP (DHP), SRP and PP. In particular cases, where it is assumed that most of the dissolved TP is unreactive, TP consists of SUP and PP.
PP	PP (PPS or POP) is the more available component of the suspended sediments that breaks down with acid oxidation to generate SRP.
Particulate Phosphorus	It contains organic phosphorus. Calculated (not measured) as PP = TP – TDP or PP = TP – SUP (in particular cases, where it is assumed that most of the dissolved TP is unreactive).

797

798 **Appendix B**

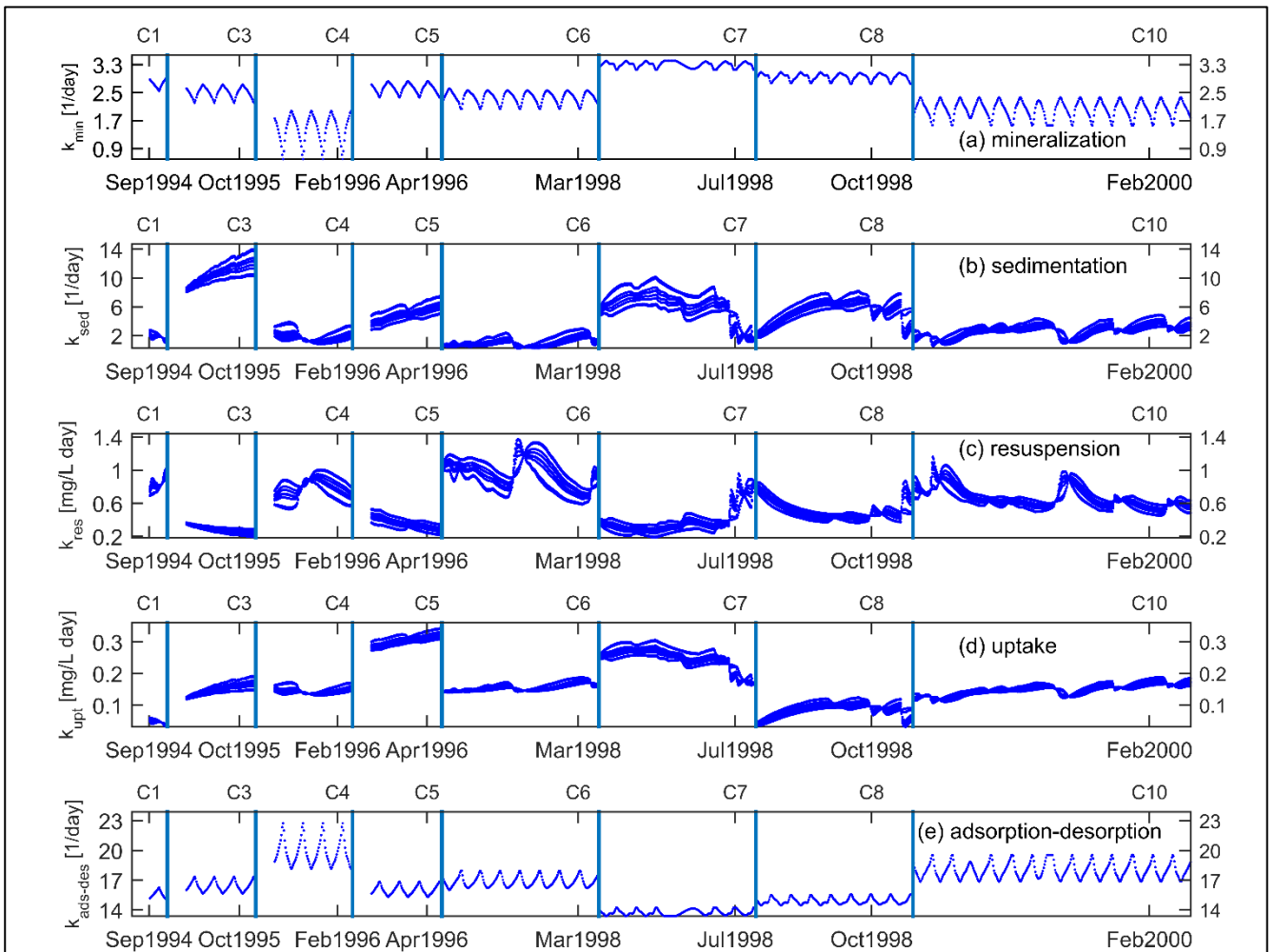
799 **Table B. 1 Goodness of fit associated to non-conservative ADModel-P runs in different options of splitting the data**
800 **for preliminary calibration and evaluation (NSE – Nash Sutcliffe model efficiency criteria)**

Run number	Campaigns used for evaluation	NS during calibration runs		NS during evaluation runs	
		SRP	OP	SRP	OP
1.	1, 10	0.45	0.21	-0.36	-0.02
2.	2, 9	0.44	0.20	-0.24	0.34
3.	2, 8	0.49	0.25	0.56	-0.18
4.	3, 6	0.41	0.09	0.16	-0.02
5.	3, 10	0.41	0.21	0.69	-0.32
6.	4, 7	0.52	0.23	-0.29	-0.41
7.	5, 10	0.50	0.21	-0.26	-0.21
8.	6, 7	0.48	0.06	0.14	-0.01
9.	6, 8	0.46	-0.12	-0.11	-0.18
10.	7, 10	0.48	0.22	0.26	-0.03
11.	8, 10	0.47	0.24	0.75	0.18
12.	9, 10	0.45	0.13	0.48	-0.02

801

802 **Table B. 2 Criteria employed for the assessment of ADModel-P performance**

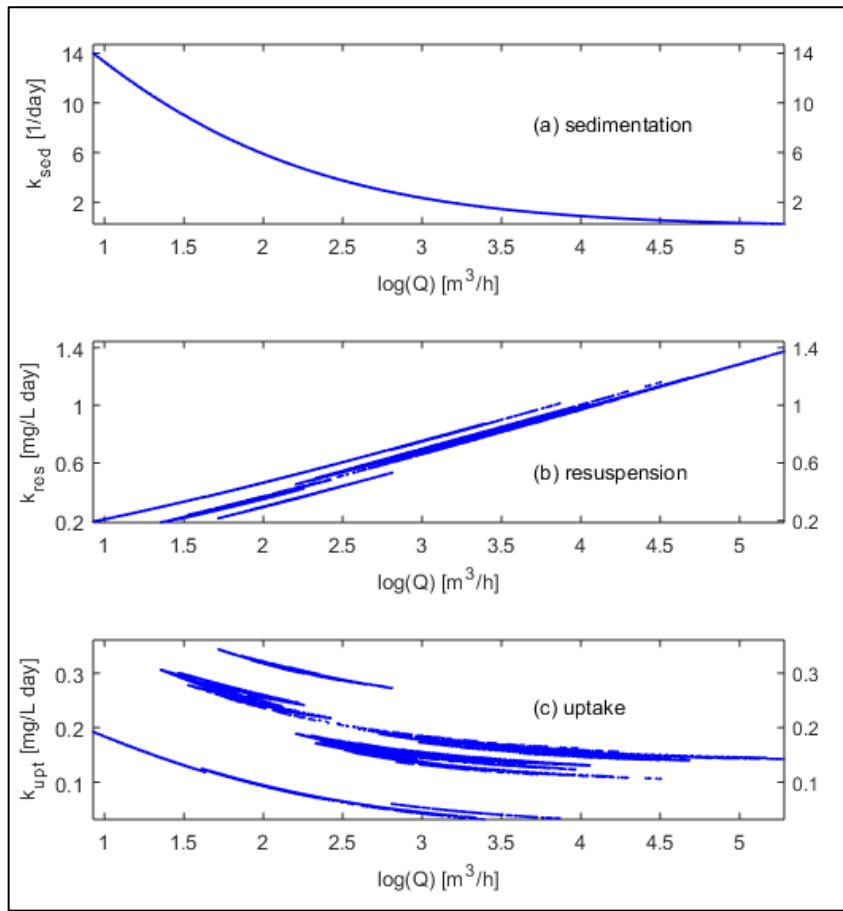
Model evaluation criteria	Equation	References
Nash Sutcliffe efficiency (NSE)	$NSE = 1 - \frac{\sum_{i=1}^n (O_i - P_i)^2}{\sum_{i=1}^n (O_i - \bar{O})^2}$ <p>n is the total number of samples, i the individual samples, O the observations, P the predictions, and \bar{O} the mean of observations.</p>	Nash and Sutcliffe (1970) Moriassi et al. (2017) Gupta et al. (2009)
Percent bias (PBIAS)	$PBIAS = \frac{\sum_{i=1}^n (O_i - P_i)}{\sum_{i=1}^n O_i} 100$	Moriassi et al. (2007) Gupta et al. (1999)
Root mean square error ratio to observations standard deviation (RSR)	$RSR = \frac{\sqrt{\sum_{i=1}^n (O_i - P_i)^2}}{\sqrt{\sum_{i=1}^n (O_i - \bar{O})^2}}$	Moriassi et al. (2007)
Kling-Gupta Efficiency (KGE)	$KGE = 1 - \sqrt{(r - 1)^2 + \left(\frac{\sigma_P}{\sigma_O} - 1\right)^2 + \left(\frac{\bar{P}}{\bar{O}} - 1\right)^2}$ $r = \frac{Cov_{PO}}{\sigma_P \sigma_O}$ <p>r is the linear correlation coefficient between predictions and observations, Cov_{PO} the covariance between the predictions and observed values, σ_P and σ_O the standard deviations of predictions and observations, and \bar{P} the mean of predictions.</p>	Knoben et al. (2019) Gupta et al. (2009)
Modified coefficient of determination (bR^2)	$bR^2 = \begin{cases} b R^2 & \text{for } b \leq 1 \\ b ^{-1}R^2 & \text{for } b > 1 \end{cases}$ $R^2 = \left(\frac{\sqrt{\sum_{i=1}^n (O_i - \bar{O})(P_i - \bar{P})}}{\sqrt{\sum_{i=1}^n (O_i - \bar{O})^2} \sqrt{\sum_{i=1}^n (P_i - \bar{P})^2}} \right)^2$ <p>b is the slope of the regression line between predictions and observations when the intercept is zero or very close to zero, R^2 the coefficient of determination defined as the square of the coefficient of correlation according to Bravais-Pearson.</p>	Krause et al. (2005)



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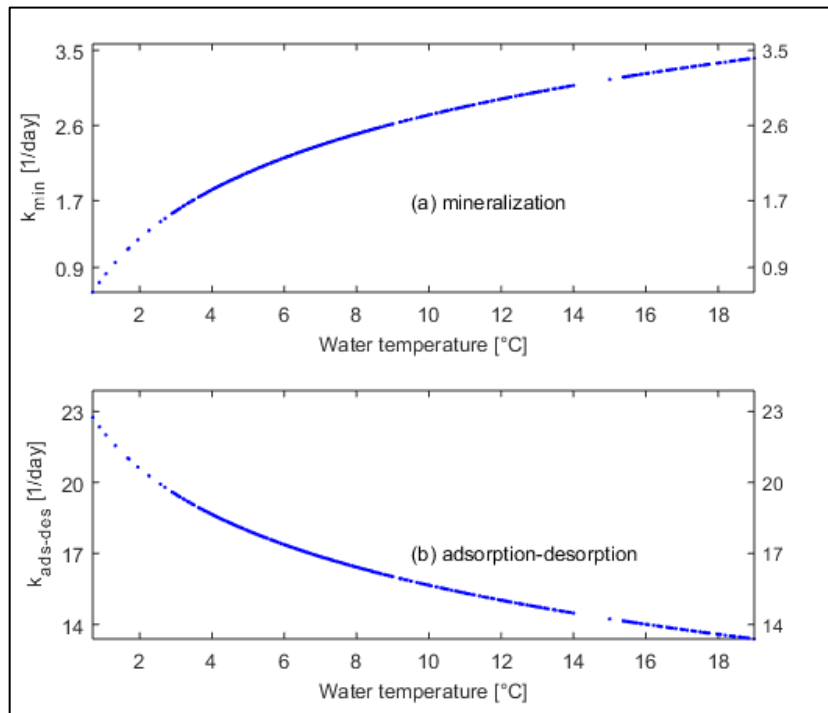
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Fig. C.1 Dynamic evolution of transformations rate constants used by ADModel-P during calibration (subplot a - mineralization, subplot b - sedimentation, subplot c - resuspension, subplot d - uptake, subplot e – adsorption-desorption). The time x-axes are not continuous. Data from the individual campaigns C1 to C10 carried out during the specified months (Table 1) has been concatenated during simulations. The vertical lines mark the boundaries between C1 to C10.



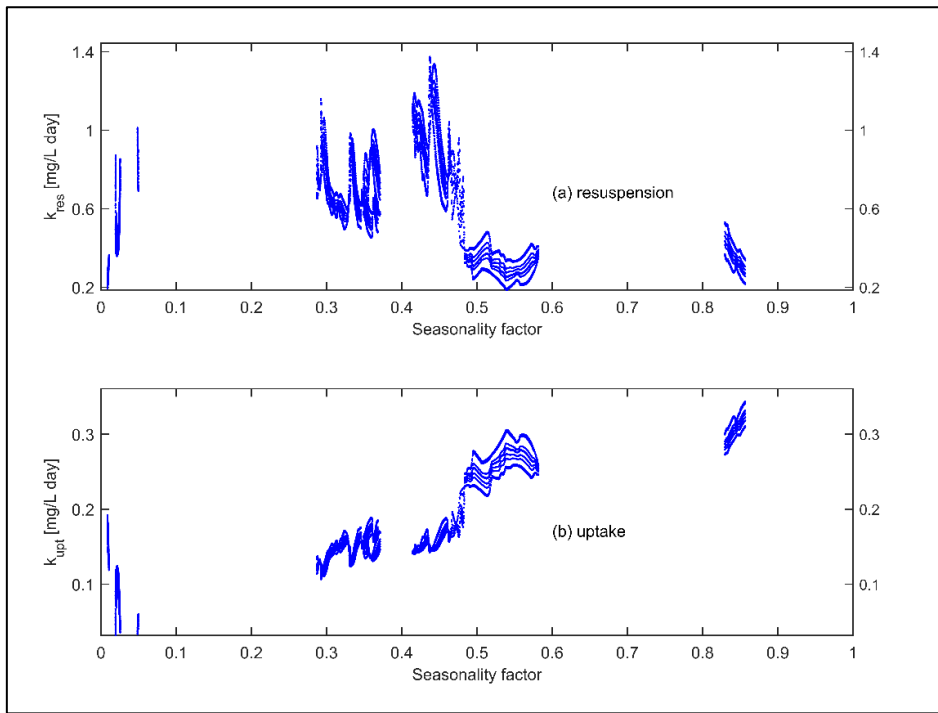
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Fig. C.2 The dependence on water flow of the transformation rate constants used by ADModel-P during calibration (subplot a - sedimentation, subplot b - resuspension, subplot c - uptake).



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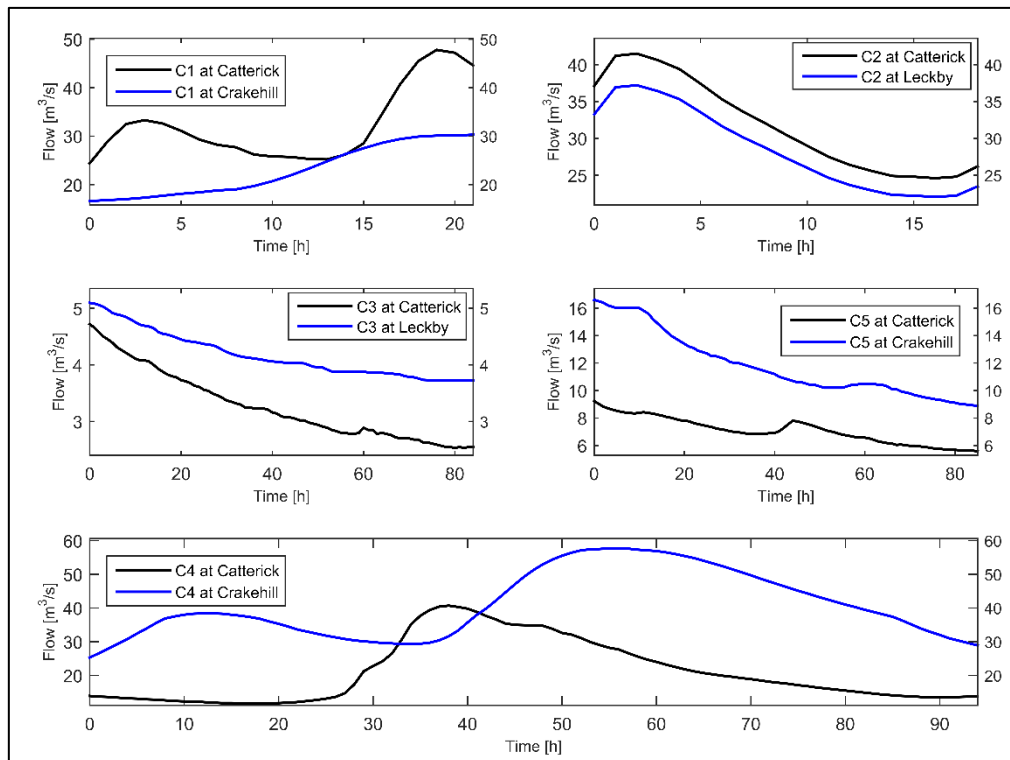
Fig. C.3 The dependence on water temperature of the transformation rate constants used by ADModel-P during calibration (subplot a - mineralization, subplot b – adsorption-desorption).



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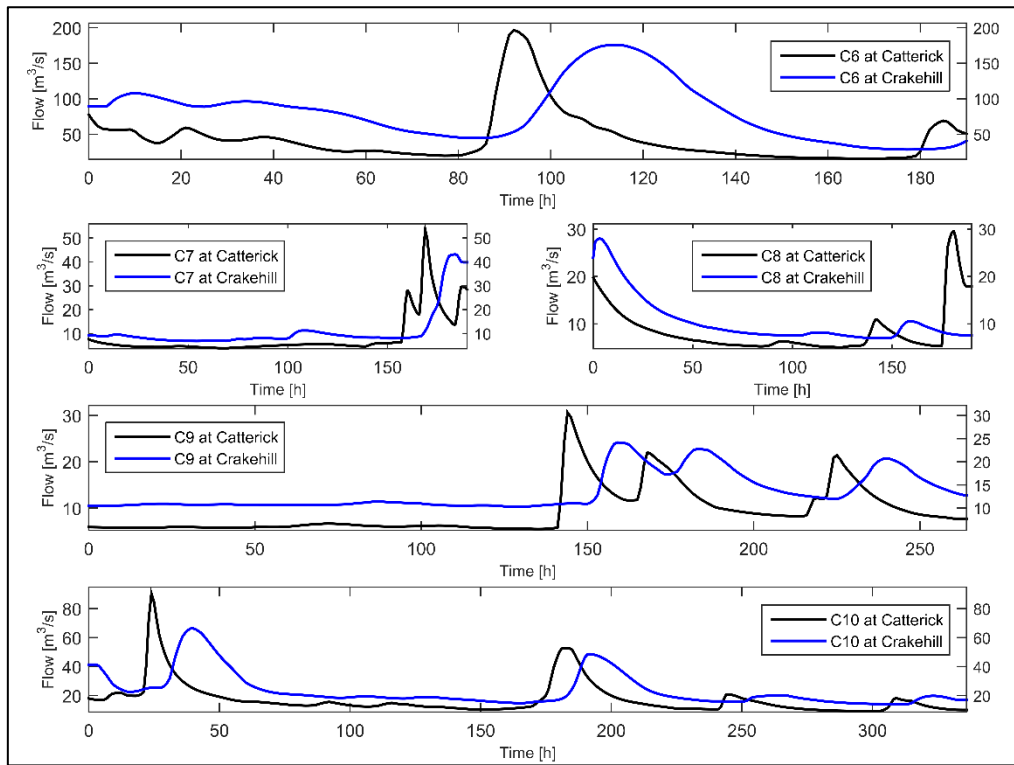
Fig. C.4 The dependence on the seasonality factor of the transformation rate constants used by ADModel-P during calibration (subplot a - resuspension, subplot b - uptake).

824 **Appendix D**



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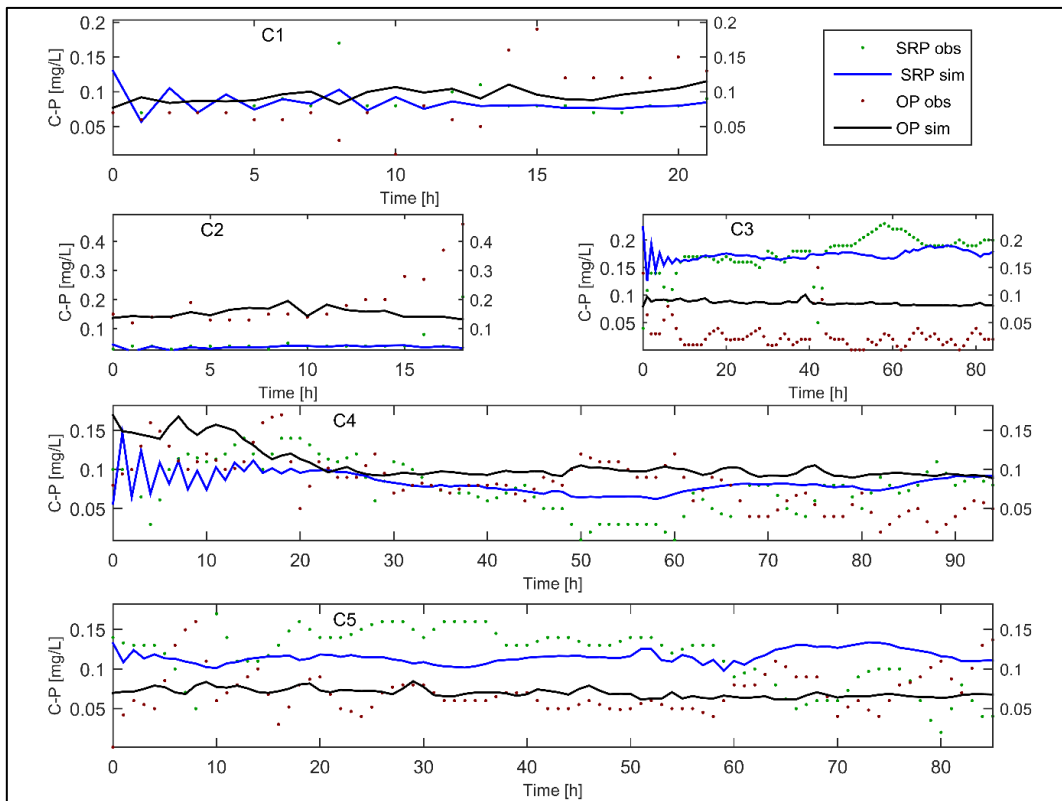
Fig. D. 1 The dynamics of water flow at the upper and lower end of the stretch during the monitoring campaigns C1 to C5.



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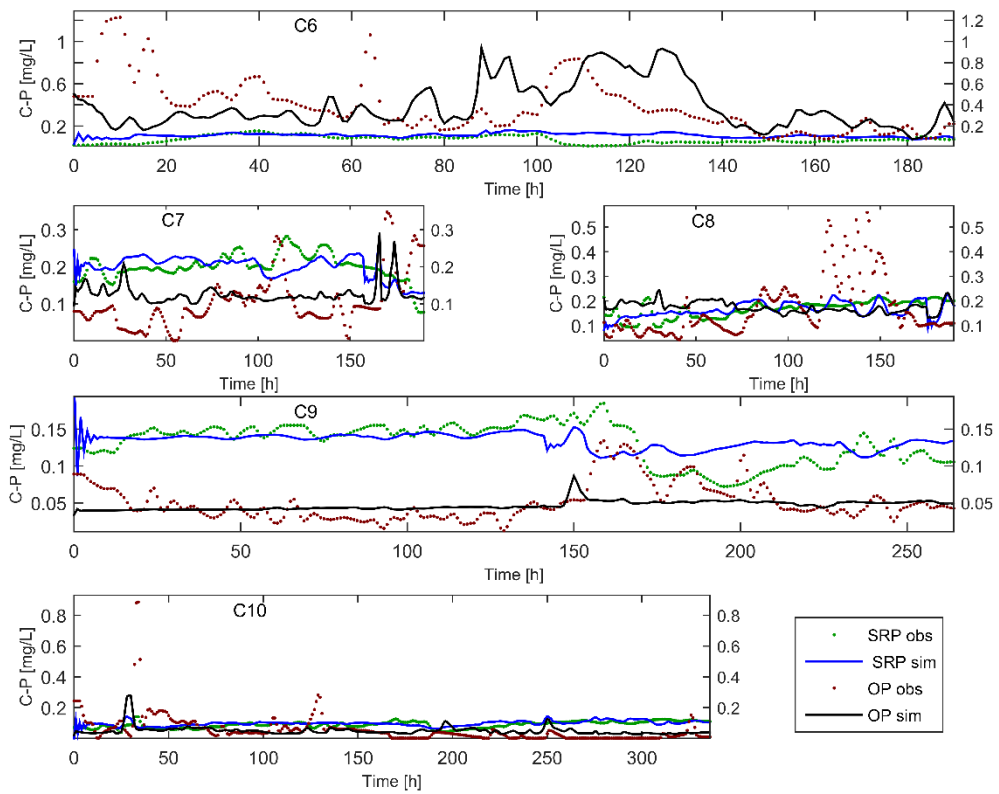
Fig. D. 2 The dynamics of water flow at the upper and lower end of the stretch during the monitoring campaigns C6 to C10.

833 **Appendix E**



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Fig. E. 1 Dynamic profiles of the observed (obs) and simulated (sim) concentration (C-P [mg/L]) of SRP and OP during campaigns C1 to C5 presented at hourly resolution at the downstream end of the river stretch.



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Fig. E. 2 Dynamic profiles of the observed (obs) and simulated (sim) concentration (C-P [mg/L]) of SRP and OP during campaigns C6 to C10 presented at hourly resolution at the downstream end of the river stretch.