

Centre for Ecology & Hydrology

NATURAL ENVIRONMENT RESEARCH COUNCIL

enquiries@ceh.ac.uk WWW.Ceh.ac.uk

INTRODUCTION

 \triangleright Recent studies of CINO₂, N₂O₅ PANs and trimethlyamines at background sites are important \geq The background EMEP supersite 'Auchencorth Moss' in South East Scotland routinely measures NO₂, NO, NH₃, HONO and HNO₃ in gas phase and particulate (PM₁₀ and PM₂₅) NH₄⁺ and NO₃⁻. >A study in spring 2014 aimed to:

- 1. Develop a better understanding of the N speciated budget at Auchencorth Moss (refer to reactive N cycle).
- 2. Identify potential artefacts in the routine N measurements



Location of Auchencorth Moss



Auchencorth Moss

OVERVIEW OF MEASUREMENT PERIOD

 \succ Changing air masses resulted in different N species composition.

>Periodic pollution events were observed during the study resulted in elevated N species

Example: 19 May 2014 where an observed increase in N species with the exception of NO as no data was available (highlighted on graph)

 \succ The reported average HNO₃ from TDLIF is around four times that reported by the MARGA (refer to table). Possible explanations:

- 1. MARGA suffers from loses at the inlet
- 2. TDLIF HNO₃ suffers artefacts from NO_3^- aerosol



INTERCOMPARISON STUDIES

 \geq Poor correlation between the MARGA and TDLIF for HNO₃ measurements. \succ The Thermo Scientific analyser reports higher NO₂ compared to the TDLIF most likely due to interferences at low NO₂ concentrations previously demonstrated by Steinbacher et al.(2007). NEXT STEPS OF STUDY:

Determine if the GC overestimates PANs

 \triangleright Assess the potential interference of particulate NO₃⁻ in the TDLIF measurements of HNO₃.

References: Di Carlo, P. et al. (2013) Aircraft based four-channel thermal dissociation laser induced fluorescence instrument for simultaneous measurements of NO₂, total peroxy nitrate, total alkyl nitrate, and HNO₃, Atmos. Meas. Tech., 6, 971. Jenkin and Clemitshaw (2000) Ozone and other secondary photochemical pollutants: chemical processes governing their formation in the planetary boundary layer. Atmospheric Environment. 34, 2499-2527 McFadyen, G. G., Cape, J. N., 2005. Peroxyacetyl nitrate in eastern Scotland. Sci. Total Environ. 337, 213-222 Phillips, G. J. et al. (2013) The detection of nocturnal N₂O₅ as HNO₃ by alkali- and aqueous-denuder techniques, Atmos. Meas. Tech., 6, 231-237, Steinbacher, et al. (2007) Nitrogen oxide measurements at rural sites in Switzerland: Bias of conventional measurement techniques. Journal of Geophysical Research, 112(D11), D11307.

Deriving a speciated atmospheric nitrogen budget at Auchencorth Moss, a background site in South East Scotland

M. Twigg¹, E. Aruffo², J. Kentisbeer¹, C. Malley^{1,3}, S. Leeson¹, M. Jones¹, M. Coyle¹, E. Nemitz¹, P. Di Carlo², and C. Braban¹. ¹Centre for Ecology and Hydrology, Bush Estate, Penicuik, UK. ²Center of Excellence CETEMPS, Universita' degli studi di L'Aquila, Via Vetoio, 67010 Coppito, L'Aquila, Italy, ³School of Chemistry, University of Edinburgh, Edinburgh, United Kingdom Corresponding author: sail@ceh.ac.uk



<u>S1</u>	tatistics fo	<u>r May 20</u>	14 using	hourly						
	averages from instrumentation									
		Average	σ	Max						
		(ppb)	(ppb)	(ppb)						
	TDLIF									
	HNO ₃	0.39	0.28	1.46						
	ΣANs	0.11	0.10	0.72						
	ΣPNs	0.24	0.22	2.71						
	NO ₂	0.94	0.80	7.55						
	MARGA									
	$PM_{2.5} NH_4^+$	0.99	1.12	7.88						
	$PM_{2.5} NO_3^-$	0.46	0.71	4.70						
	NH ₃	1.38	1.64	26.51						
	HNO ₃	0.09	0.08	0.55						
	HONO	0.04	0.03	0.25						
	Thermo Scientific analyser									
	NO	0.15	0.17	1.64						
	NO ₂	1.54	1.00	8.19						



>Phillips et al. (2013) provided evidence to suggest night time HNO₃ reported by MARGA may include N₂O₅, where:

 \succ This work suggests a relationship between the MARGA HNO₃ and Σ PNs measured by the TDLIF

by the MARGA

 \geq Night time HNO₃ measured by the MARGA correlates well with the additional measured HNO₃ derived from Σ PNs \succ This suggests that the HNO₃ reported by the MARGA at Auchencorth Moss may additionally contain N₂O₅, though further studies are required to confirm this

>Investigate the chemical transformations of N species at the site

Twigg, M. M., et al. (2015) Water soluble aerosols and gases at a UK background site – Part 1: Controls of PM2.5 and PM10 aerosol composition, Atmos. Chem. Phys. Discuss., 15, 3703-3743

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Μετμορε	Routine and non-routine measurements during the campaign					
IVIE I HUUS	Species measured	Instrumentation	Measurement technique	Height	Reference to	
The table lists N species measured and the instrumentation used	$PM_{10} and PM_{2.5}$: Na ⁺ , NH ₄ ⁺ , K ⁺ , Ca ²⁺ , Mg ²⁺ , Cl ⁻ , NO ₃ ⁻ , SO ₄ ²⁻ Gases: NH ₃ , HNO ₃ , HONO, HCl, SO ₂	Montior for AeRosols and Gases in Ambient air (MARGA, Metrohm Applikon, NL.)	Wet rotating annular denuder and steam jet aerosol collectors with online IC analysis	(m) 3.60	Twigg <i>, et al.</i> (2015)	
Due to issues with baseline drift in the ANNO _x the data hasn't been presented.						
PAN GC measured from 24 April 2014 to 06 May 2014	NO _{2,} total peroxy nitrate (Σ PNs), total alkyl nitrate (Σ ANs), HNO ₃	Thermal dissociation laser induced fluorescence (TDLIF)	Thermal dissociation with laser induced fluorescence detector	3.00	Di Carlo <i>, et al.</i> (2013)	
All other instrumentation operated for the length of the campaign.	NO/ NO ₂	ANNO _x analyser (CLD, 88 p, Eco Physics, AG. Switzerland)	Chemiluminesence analyser	3.00		
	NO/ NO ₂	Thermo Scientific Analyser (model 42CTL)	Chemiluminesence analyser (Molybdenum NO ₂ to NO converter)	2.02		
	PANs	GC-ECD	Online gas chromatography with electron capture detection	3.00	McFadyen and Cape (2005)	

$N_2 O_5 + H_2 O \rightarrow 2 HNO_3 = additional measured HNO_3$

- >5 consecutive nights were plotted (see LHS graph) assuming $\Sigma PNs = N_2O_5$
- \geq Molar N₂ O₅ was used to calculate the molar HNO₃, assuming a 100% capture efficiency and compared to the measured HNO₃

CONCLUSIONS:

DERIVING A SPECIATED N BUDGET

>MARGA HNO₃ may have an N₂O₅ artefact in the measurement at night NEXT STEPS OF THIS STUDY: > Determine the N species to be used to derive a N budget

>Examine the influence of long range transport of air masses on the speciated N composition at this background site





