

## Article (refereed) - postprint

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Machon, Attila; Horváth, László; Weidinger, Tamás; Grosz, Balázs; Moring, Andrea; Führer, Ernő. 2015. **Measurement and modeling of N balance between atmosphere and biosphere over a grazed grassland (Bugacpuszta) in Hungary.** *Water, Air, & Soil Pollution*, 226 (3), 27. 19, pp. [10.1007/s11270-014-2271-8](https://doi.org/10.1007/s11270-014-2271-8)

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# Measurement and modeling of N-balance between atmosphere and biosphere over a grazed grassland (Bugacpuszta) in Hungary

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**Abstract** – This work is a synthesis of a 5-year estimation of nitrogen-balance at a semi-arid, semi-natural, undisturbed grassland site (Bugac). We measured the N input of atmospheric pollutants by wet and dry deposition of gases and aerosols whilst we considered N output as NO, N<sub>2</sub>O gases volatilized from soil. Beside measurements of soil fluxes the denitrification-decomposition (DNDC) ecological model was also used and simulations were compared to and validated against the measured values. The daily flux simulations generally did not match well the measured data for N<sub>2</sub>O and NO. In most cases the mean fluxes were underestimated, though results of the comparison of monthly values suggest that model data, together with observed deposition data, are applicable to estimate the net N-balance for grasslands. The calculated yearly N-balance (net flux) between atmosphere and surface, without biological fixation and effect of grazing ranged between –9.4 and –14 kg N ha<sup>-1</sup>year<sup>-1</sup> as the sum of the measured deposition and emission terms: –11 to –15 and 0.9-2.9 kg N ha<sup>-1</sup>year<sup>-1</sup>, respectively, between 2006 and 2010. Observed and modeled soil emissions were lower by one order of magnitude than atmospheric deposition. Considering the biological nitrogen fixation and the effect of grazing (effects of both grazed plant and excreta) the net nitrogen balance varies within –6.6 and –11 kg N ha<sup>-1</sup>year<sup>-1</sup>. It seems - taken into account the high uncertainty in calculation due to the effect of grazing - that sources of nitrogen exceed the sinks; the surplus is probably mineralized in the soil.

**Keywords:** nitrogen exchange, deposition, soil emission, denitrification, DNDC model, grassland

## 1. Introduction

Among elements nitrogen (N) has one of the most complex biogeochemical cycles. It is mostly affected by human activities both directly and indirectly resulting in altered concentration, distribution and flux of reduced and oxidized N species (Erisman et al. 2011). Anthropogenic emissions of reactive N compounds have a direct and manifold impact on N cycle. The released compounds undergo numerous sequences of transformations in the atmosphere and water as well as soil ecosystems until they are immobilized or de-nitrified to nitrogen gas (N<sub>2</sub>); this system of processes is termed as N cascade (Galloway et al. 2003). The indirect impact of human activities further complicate the cycle of N species through affecting metabolic processes of animals, plants, and a large variety of microorganisms. Depending on the lifetime of various N compounds - from hour to hundred years - their environmental impacts can range from local direct damage to climate change (Galloway et al. 2003; Moldanová et al. 2011).

Due to the complexity of N cycle many transformation processes and influencing factors have not been completely explored yet. However, it is essential to monitor them to determine the rate of pollutant emissions and the harmful effects in current and future context. There are some synthesis studies of N cycles on continental and global scale (Erisman et al. 2003; 2010; Galloway et al. 2003; Vitousek et al. 1997); nevertheless, the uncertainty in estimation of N-balance and emissions from vegetation remains high because of the relatively sparse number of available, appropriate (both lab and field) measurements.

The N-balance of non-intensively managed ecosystems is dominantly determined by atmosphere-surface exchange processes. For non-fertilized grasslands, atmospheric deposition is the main source of N (Machon et al. 2011), but we also have to consider the N fixation by legumes (Ammann et al. 2009). On the other hand, a significant amount of N compounds is emitted by the biosphere. For some compounds (e.g. for ammonia) the exchange is bi-directional (e.g. Sutton et al. 2001; Zhang et al. 2010). In this case the sum of deposition and emission rates is called as net flux.

61 Several N compounds (e.g.: nitrogen dioxide (NO<sub>2</sub>), nitric acid vapor (HNO<sub>3</sub>), ammonium (NH<sub>4</sub><sup>+</sup>) and  
62 nitrate (NO<sub>3</sub><sup>-</sup>) ions in fine and coarse mode of particle phase) have only negative flux (deposition) from  
63 atmosphere to the ecosystems, i.e. they are not released by soil and vegetation (Watt et al. 2004; Wesely and  
64 Hicks 2000). Atmospheric gases and particles are deposited in two ways to the surfaces, partly by precipitation  
65 (wet deposition) during cloud formation and below cloud scavenging, and partly by turbulent flux onto the plants  
66 and soil surface (dry deposition) (Erisman et al. 2005). In general, the rate of dry are wet deposition in grasslands  
67 is in the same magnitude in Hungary (Kugler et al. 2008; Machon et al. 2011).

68 N compounds are emitted partly by soil and stomata of plants. Nitrogen has different forms in soil with  
69 wide-range of oxidation number from NH<sub>4</sub><sup>+</sup> (-3) to NO<sub>3</sub><sup>-</sup> (+5). Nitrification and denitrification processes  
70 fundamentally affect the soil N-balance (Robertson and Groffman 2007), through producing intermediate gases  
71 as nitric oxide (NO), nitrous oxide (N<sub>2</sub>O) and elemental N<sub>2</sub>. All of those can be emitted to the atmosphere. By  
72 vegetation N compounds are emitted dominantly as ammonia gas (NH<sub>3</sub>) (Horváth et al. 2005; Massad et al.  
73 2010). Ammonia emission from soil can be observed as well but only for alkaline soils. The role of NH<sub>3</sub> gas in  
74 the troposphere, the dynamics of exchange (Massad et al. 2010), as well as its role in N load and in nutrient  
75 supply and other aspects of NH<sub>3</sub> in ecosystems are well known (Sutton et al. 2008; 2011).

76 Trace gas exchange within surface-biosphere-atmosphere system strongly depends on meteorological  
77 conditions (Nagy et al. 2007), concentrations, as well as characteristics of ecosystem and soil physical, chemical  
78 and biological properties (Horváth et al. 2008; Meixner and Yang 2006; Smith et al. 2003; Sutton et al. 2011).  
79 For this reason investigation of N-exchange processes above different ecosystems is important and necessary.  
80 Over grassland ecosystems our knowledge is quite comprehensive concerning these processes; however, there are  
81 still some remaining uncertainties e.g. the contribution of N compounds to the whole N budget for a given  
82 ecosystem and how the contribution varies in response to external natural and anthropogenic drivers, such as  
83 changing climate and land management (Skiba et al. 2009).

84 Several European integrated research programs (e.g. GRAMINAE, GreenGrass, NOFRETETE and  
85 NitroEurope) (Pilegaard et al. 2006; Soussana et al. 2007; Sutton et al. 2001; 2007) have been dealing with the N  
86 turnover of various terrestrial ecosystems. N flux measurements have been carried out by different scientific  
87 communities at many different sites from different points of view (e.g. air pollution, greenhouse effect, water  
88 pollution, nutrient load, biodiversity etc.) (Hicks et al. 2011; Sutton et al. 2007; 2011; INI; and EU initiatives  
89 like NinE and COST Action 729), while there are just a sparse number of total N-balance estimations on  
90 landscape scale involving all major sources and sinks.

91 For detailed investigation of the biosphere-atmosphere exchange of different N compounds including  
92 plot measurements over different types of ecosystems and for modeling of N fluxes from plot to continental  
93 scale, the EU Framework 6<sup>th</sup> Integrated Project (NitroEurope – [www.nitroeuropa.eu](http://www.nitroeuropa.eu), Sutton et al. 2007) was  
94 started in 2006 coordinating the N researchers across Europe. One of the grassland stations with intensive  
95 measurement program of the NitroEurope network was established in central Hungary, on the Hungarian Great  
96 Plain, in Bugacpuszta.

97 Our task in the project was to determine the N-exchange between the atmosphere and semi-natural  
98 grassland in a semi-arid continental climate representative for the Hungarian Great Plain, in central Hungary,  
99 based on measurements of N-fluxes, taking into account also their dependence on climatic conditions and on the  
100 possible feedbacks to soil/vegetation dynamics. In addition, we employed the DNDC (DeNitrification  
101 DeComposition) model to estimate soil fluxes of N-gases. This paper summarizes the results of the five year  
102 measurement record of the N-exchange, taking into account all of the significant N species.

103 The current study is not the first one dealing with N-turnover above Bugacpuszta. In Horváth et al.  
104 (2010) solely the measured soil NO and N<sub>2</sub>O fluxes were published for 2002-2004. Machon et al. (2010)  
105 reported a preliminary net nitrogen balance based only on one year observation record. In Machon et al. (2011)  
106 preliminary results for the whole nitrogen balance were reported including wet and dry deposition and soil  
107 emission focusing mostly on weather induced variability of nitrogen exchange.

108 The main aim of our work was to determine the surface-atmosphere N-balance balance over the grassland  
109 on the basis of measured and/or modeled upward and downward N-fluxes. In addition, we attempted to  
110 give a rough estimation for the total nitrogen balance, including the effect of grazing.

## 112 2. Materials and methods

### 113 2.1 Site of investigations

114 The selected location was Bugacpuszta (46.69 °N, 19.60 °E, 113 m a.s.l.) in the Hungarian Great Plain, between  
115 the rivers Danube and Tisza. This semi-natural, semi-arid, sandy grassland is one of the most characteristic  
116 landscape types in Hungary; therefore, the area is protected and part of the Kiskunság National Park.

117 The climate is semi-arid temperate continental, where the mean annual temperature is 10.7 °C and the  
118 average yearly precipitation (1989–2006) is 550 mm. The region has Chernozem-type sandy soil (according to  
119  
120

121 the World Reference Base (WRB) classification, see [www.fao.org](http://www.fao.org)) with high sand (79%) and low clay (13%)  
122 contents in the upper 10 cm soil layer. The total carbon, organic carbon, total nitrogen content and pH in the soil  
123 extracted by KCl solution in the 0-10 cm layer are: 7.0% (SD=3.2); 6.3% (SD=3.2); 0.69% (SD=0.32) and 7.7  
124 (SD=0.36), respectively. The same parameters in the 0-60 cm layer are: 4.1% (SD=1.2); 3.2% (SD=0.88); 0.32%  
125 (SD=0.09) and 8.2 (SD=0.27), respectively. The area was never ploughed around at least 200 m away from the  
126 measurement plot. Except grazing the soil has been undisturbed. The plant association is semi-arid sandy  
127 grassland (*Cynodonti Festucetum pseudovinae*) dominated by *Festuca pseudovina*, *Carex stenophylla*, *Salvia*  
128 *pratensis* and *Cynodon dactylon*. During their evolution, endemic plant and animal species (e.g. Hungarian Grey  
129 Cattle) have developed extraordinary strategies to survive heat and drought. The plant community is sensitive to  
130 physical or chemical disturbances. For this reason, the area is a nature reserve and management is not allowed.  
131 The only exception is the traditional extensive grazing by a herd of the ancient Grey Cattle breed at an average  
132 grazing pressure of 0.5–0.8 stock ha<sup>-1</sup> in the grazing season (220 days in each year), which has been going on for  
133 centuries in dynamic equilibrium with the grass ecosystem (Machon et al. 2010).

## 134 2.2 DNDC (denitrification-decomposition) modeling

### 135 2.2.1 Description of the model

136  
137 Ecological models like DNDC can predict the rate of processes and fluxes in different scales. These models are  
138 also applicable to support climatic or land use scenarios for the future planning. One of the main aims of  
139 biogeochemical models is to simulate C and N exchange and cycles. DNDC, a process-based biogeochemical  
140 model (Li et al. 1992a;b) was used in this study to calculate soil fluxes of all important gaseous N forms  
141 including NO, N<sub>2</sub>O, NH<sub>3</sub>, and N<sub>2</sub> that are difficult to determine by field or laboratory measurements. Advantage  
142 of the model is the online access ([www.dndc.sr.unh.edu/](http://www.dndc.sr.unh.edu/)); it has been used by many research groups all over the  
143 world (Giltrap et al. 2010; Hsieh et al. 2005; Levy et al. 2007; Smith et al. 2010). The model has been  
144 continuously developed and modified (e.g. Brown et al. 2002; Neufeldt et al. 2006; Saggarr et al. 2007; Xu-Ri et  
145 al. 2003); therefore, a lot of experiences of operation are available in the literature.

146 Ecological drivers, like i) soil properties (soil texture, density, slope, field capacity, wilting point, clay  
147 fraction, pH, SOC, NO<sub>3</sub><sup>-</sup>, and NH<sub>4</sub><sup>+</sup> content), ii) meteorological variables (daily minimum and maximum  
148 temperature, precipitation, and global radiation), iii) vegetation characteristics (crop type, plant, and harvest time,  
149 details of crop phenology), iv) and farming management as tillage, fertilizers, manure, weed, irrigation, grazing  
150 pressure, and type (cattle, horse, sheep) with start and end date of grazing etc. as daily input parameters are  
151 required to simulate trace gas fluxes. These are mainly determined by soil climate and soil processes such as  
152 denitrification, nitrification, mineralization, decomposition etc. (Li 2007).

153 Vertical profiles of soil parameters as well as trace gas fluxes are calculated based on these input data.  
154 Firstly, DNDC predicts daily soil temperature, moisture, redox potential (E<sub>h</sub>), pH, and substrate concentration,  
155 and then uses these to drive nitrification, denitrification, CH<sub>4</sub> production/oxidation, and other relevant  
156 geochemical or biochemical reactions. Most parts of the model run at a daily time step except the soil climate  
157 and denitrification submodels which run at an hourly time step. Output parameters from the model runs are daily  
158 soil profiles of temperature, moisture, E<sub>h</sub>, pH, and concentrations of total soil organic carbon, nitrate, nitrite,  
159 ammonium, urea, ammonia, as well as daily fluxes of CO<sub>2</sub>, NO, N<sub>2</sub>O, CH<sub>4</sub>, and NH<sub>3</sub>.

160 For parameterization and validation of the model the meteorological (global radiation, min., max., and  
161 average air temperature, precipitation) and the botanical (plant species composition, areal coverage of plant,  
162 functional group e.g. legumes) datasets were gained by field observations of Szent István University (SzIU). The  
163 site-specific soil information (texture, pH, clay fraction, organic C content, bulk density etc.) was provided by  
164 the Research Institute for Soil Science and Agricultural Chemistry (RISSAC) of the Hungarian Academy of  
165 Science (HAS). Sensitivity analyses were also performed to identify main input model parameters responsible  
166 for the majority of changes in soil emission (for more details see Machon et al. 2010).

167 The model treats nitrogen inputs from atmospheric deposition, fertilizer usage as well as nitrogen  
168 fixation and accounts for soil (in)organic turnover, allowing the calculation of leaching of nitrogen as well as  
169 gaseous emissions. The model consists of two components (Li 2000).

170 The first component consists of three different submodels.

171 a) Soil climate submodel. It simulates soil temperature and moisture profiles based on soil physical  
172 properties, weather, and plant water use in one dimension. The soil is divided into horizontal layers, water fluxes  
173 and heat flows which are determined by soil texture and the gradients of soil moisture potential (for water fluxes)  
174 and soil temperature (for heat flows).

175 b) Crop growth submodel. It simulates crop biomass accumulation and partitioning of biomass into  
176 grain, stalk, and roots based on thermal degree days and daily N and water uptake. If N or water stress occurs,  
177 crop growth will be suppressed. Biomass partitioning is determined by the physiological parameters stored in the  
178 crop library files.

181 c) Decomposition sub-model. It simulates daily decomposition, ammonification, ammonia volatilization  
182 and CO<sub>2</sub> production by soil microbes. The submodel calculates turnover rates of soil organic matter at a daily time  
183 step. In the DNDC model SOC resides in four major pools: plant residues or litter, microbial biomass, humads,  
184 and passive humus. Each pool consists of one or more sub-pools with different properties. The daily  
185 decomposition rate for each sub-pool is regulated by pool size, its specific decomposition rate or fraction lost per  
186 day, soil clay content, N availability, soil temperature and moisture, and effective depth of the soil profile. The  
187 effects of cropping practices on C and N dynamics are also considered in the model. The effect of soil properties  
188 such as soil temperature, clay fraction and water content is modeled using reduction factors that constrain  
189 decomposition rate from the maximum in non-optimum conditions. Nitrogen mineralized during decomposition  
190 enters the inorganic nitrogen pool as NH<sub>4</sub><sup>+</sup>, where it accumulates, or is nitrified to NO<sub>3</sub><sup>-</sup> (with gaseous losses as  
191 NO and N<sub>2</sub>O), or is removed via plant uptake, leaching, transformation to NH<sub>3</sub> and volatilization, or adsorption  
192 onto clay minerals. Soluble carbon levels, which fuel both nitrification and denitrification, are related to the  
193 fraction of carbon released by the decomposition of litter, labile humus, and dead microbial biomass that is re-  
194 assimilated in microbial biomass each day.

195 The second component consists of three submodels as well (Li et al. 1992 a;b; Li 2000; 2007).

196 a) Nitrification submodel. It tracks growth of nitrifiers and turnover of ammonium to nitrate.  
197 Nitrification rate is calculated as a function of ammonium concentration, nitrifier population, temperature and  
198 pH.

199 b) Denitrification submodel. It operates at an hourly time step to simulate denitrification and the  
200 production of nitrite (NO<sub>2</sub><sup>-</sup>), nitric oxide (NO), nitrous oxide (N<sub>2</sub>O), and dinitrogen (N<sub>2</sub>). based on soil redox  
201 potential and dissolved organic carbon (DOC) concentration. The rates for each step in the denitrification  
202 reduction sequence are a function of soluble C, soil temperature (or E<sub>h</sub> for frozen soils), soil pH, N-substrate  
203 availability, and denitrifier biomass. As the soil dries following a rain event, the denitrifying part of each model  
204 layer decreases with soil water content. The growth and the death of denitrifier populations are simulated, which  
205 enables consumption of DOC, NO<sub>3</sub><sup>-</sup>, NO<sub>2</sub>, NO, and N<sub>2</sub>O. The hourly time-step denitrification sub-model in  
206 DNDC is activated by three conditions which increase soil moisture and/or decrease soil oxygen availability:  
207 rain or irrigation events, flooding (as in irrigated rice agriculture), and freezing temperatures. Air temperature  
208 below -5 °C is assumed to freeze the soil and thus, inhibit oxygen diffusion into the soil. An oxidation-reduction  
209 potential (E<sub>h</sub>) is calculated depending on soil organic matter content as a proxy for oxygen consumption and  
210 denitrification rate is computed by using E<sub>h</sub> as a multiplication factor. For any initiation of denitrification the  
211 initial status of the available NO<sub>3</sub><sup>-</sup> and soluble carbon pools is provided by the decomposition submodel.

212 c) Fermentation submodel.

213 Classical laws of physics, chemistry, and biology, as well as empirical equations generated from  
214 laboratory studies, have been employed in the model to parameterize each specific geochemical or biochemical  
215 reaction (Li 2000).

216 During the model calibration (according to the test runs) we modify some default settings and also  
217 redefine some preliminary field data such as atmospheric CO<sub>2</sub> and NH<sub>3</sub> concentration, atmospheric N deposition,  
218 soil texture parameters (e.g. clay fraction, SOC profile etc.), crop biomass/yield and grazing time. To adjust the  
219 fluxes of N gases there was no direct approach, so it could be done only by modifying crop or soil parameters to  
220 alter the biogeochemical processes. After the refinement with calibration all parameter were fixed and with the  
221 final setting we ran simulations (different time period than the calibration period) in plot scale and compared  
222 directly with the field measurement (measurement had also a local scale footprint).

223 The model can be run in two modes: in plot and regional scale. In the last decade this widely used  
224 model was calibrated and validated by field measurements by many research groups (e.g. Beheydt et al. 2007;  
225 Brown et al. 2002; Butterbach-Bahl et al. 2004; Cai et al. 2003; Jagadeesh et al. 2006; Pathak et al. 2005; Smith  
226 et al. 2002). The entire model links C and N biogeochemical cycles and the basic ecological drivers.

### 227 228 **2.2.2 Sensitivity test of the model**

229  
230 We ran different simulations to examine the main input parameters and the relevant processes (mentioned above)  
231 responsible for the majority of the change in soil emission. Analyses were done by varying one input factor within  
232 commonly observed range, while keeping all the other input conditions at a constant value (without exhaustive  
233 explanation the input variables can be seen in Table 1). Variation of these parameters results in slight differences  
234 in the total soil nitrogen fluxes; however, they significantly affect the share of different nitrogen compounds  
235 (NH<sub>3</sub>, NO, N<sub>2</sub>O, and NO<sub>2</sub>) in total emission (Table 1).

236 According to our results, there are relationships between emitted trace gases and meteorological  
237 conditions. In the case of temperature exponential relationship can be assumed because thermal reaction rate  
238 generally exponentially increases with the temperature. Overall, the model is sensitive to the changes of  
239 meteorological parameters. Some soil properties were also varied. Parallel simulations were performed by  
240 changing initial parameters e.g. pH, clay fraction, SOC (soil organic carbon). We concluded that the model is

241 sensitive to clay content, pH as well as microbial processes and variations in surface carbon content (Table 1).  
242 During sensitivity analysis we examined the effect of variation in land management. DNDC is able to consider  
243 various land use practices, e.g. use of organic or inorganic fertilizers, irrigation, cutting and grazing by cattle,  
244 horses, sheep, etc. We varied the number of cattle per hectare and the grazing time. Grazing seems to have a  
245 strong effect on ammonia emission, while other N-gas emissions are not significantly affected.

246 The model is also sensitive to the vegetation properties: for instance, grass classification determines the  
247 yearly amount of dry matter. Additionally, plants also affect other processes, like decomposition of organic  
248 compounds, and affect the C/N ratio or the microbial activities. The sensitivity analysis has shown the most  
249 critical input parameters which have to be measured accurately.

250

### 251 **2.3 Measurement of plant/soil C/N content**

252

253 Total carbon and nitrogen content in soil and plant samples were analyzed by EuroVector EA 3000 type elemental  
254 analyzer based on dynamic flash combustion principle. The lowest absolute detection limit is 0.2 µg in  
255 1 g sample. Analyses were performed at the laboratory of Hungarian Forest Research Institute.

256

### 257 **2.4 Measurement of wet nitrogen deposition**

258

259 Wet depositions of ammonium and nitrate ions were determined based on concentration measurements by  
260 spectrophotometry (indophenol-blue) and ion chromatography methods, respectively in daily wet-only  
261 precipitation samples. Relative error (precision) of both analytical methods was below 5% while the minimum  
262 detection limit (MDL) was 0.05 mg N L<sup>-1</sup>. The estimated bulk error of the concentration measurements and  
263 precipitation sampling was around 10%. Fluxes can be calculated as follows:

264

$$265 \quad -F_{\text{wet}} = c \times p_d, \quad (1)$$

266 where  $F_{\text{wet}}$  is the deposited nitrogen in the precipitation (mg N m<sup>-2</sup> day<sup>-1</sup>),  $c$  is the concentration of ammonium  
267 or nitrate in the precipitation (mg N L<sup>-1</sup>) and  $p_d$  is the amount of daily precipitation (L m<sup>-2</sup> day<sup>-1</sup>).

268 Meteorological parameters like precipitation, air temperature, soil temperature, and soil moisture were  
269 also measured at the station.

270

### 271 **2.5 Determination of dry nitrogen deposition based on the inferential method**

272

273 24-hour sampling by the three-stage filter pack method by EMEP (1996) was used to determine the  
274 concentrations of NH<sub>3</sub> gas, HNO<sub>3</sub> vapor, NH<sub>4</sub><sup>+</sup>, and NO<sub>3</sub><sup>-</sup> particles. Concentrations of nitrate/nitric acid and  
275 ammonium/ammonia ( $c$ ) were measured by ion chromatography and spectrophotometry (indophenol-blue  
276 method), respectively. For all components the bulk relative error (precision) of sampling and measurements was  
277 around 10% and the detection limit (MDL) was 0.1 µg N m<sup>-3</sup>. Concentration of NO<sub>2</sub> was monitored by a  
278 HORIBA APNA 350 instrument (detection limit: 0.2 µg N m<sup>-3</sup>, precision: <5%).

279 Dry deposition fluxes  $F_{\text{dry}}$  (mg N m<sup>-2</sup> day<sup>-1</sup>) of nitrogen components mentioned above were determined  
280 by the inferential method according to Eq. (2) based on the measured concentration  $c$  of the given compound (µg  
281 N m<sup>-3</sup>) using inferred deposition velocities  $v_d$  (mm s<sup>-1</sup>) from the literature (Table 2):

282

$$283 \quad -F_{\text{dry}} = f \times v_d \times c, \quad (2)$$

284

285 where  $f$  is a conversion factor among different length, mass, and time units. Deposition velocities of nitric acid  
286 vapor and ammonia gas were estimated from measurement of Horváth et al. (1992; 2005) (see Table 2) where  
287 the climate, soil characteristics and grass surface were similar as to our site. The deposition velocity of NO<sub>2</sub>  
288 above grass varies between 0.4 and 2.8 mm s<sup>-1</sup> caused by climate dependence, annual and diurnal variations at  
289 different sites. Yearly mean deposition velocity (1.35 mm s<sup>-1</sup>) was adopted and used based on the literature data  
290 of Hesterberg et al. (1996): 1.1 to 2.4 mm s<sup>-1</sup>; Horváth et al. (2005): 0.4 mm s<sup>-1</sup>; Marner and Harrison (2004):  
291 0.8 to 2.5 mm s<sup>-1</sup>; Yamulki et al. (1997): 0.67 mm s<sup>-1</sup>; and Watt et al. (2004): 2.77±0.17 mm s<sup>-1</sup>.

292 Limited number of dry deposition velocity values can be found for nitrate and ammonium particles in the  
293 literature. Mean particle diameter can be a good approximation for estimation of  $v_d$  due to the dependence of  
294 deposition velocity on particle size. As the rate of sedimentation (deposition by gravitation) is proportional to the  
295 size the ammonium has lower deposition velocity rate than nitrate, in accordance with the deposition velocities  
296 recommended by Borrell et al. (1997) and Gallagher et al. (2002) for nitrate and ammonium ions, respectively  
297 (Table 2).

298

### 299 **2.6 Soil nitrous oxide flux measurements by static chambers**

Soil N<sub>2</sub>O flux was measured by weekly samplings during non-freezing periods (between 2006 and 2010) using 8 parallel static soil chambers (A=0.25 m<sup>2</sup>; h=5 cm) (Christensen et al. 1996; Clayton et al. 1994; Horváth et al. 2006). The installation of the chambers, the sampling protocol and the concentration measurement by GC-ECD were described earlier in Machon et al. (2010). To eliminate the effect of non-linearity caused by saturation effect during sampling (Stolk et al. 2009) some pilot measurement were done to justify the linearity of concentration changes in the static chambers. Above the sandy soil linear concentration changes were observed in the first 30 minutes of the enclosure period. The relatively constant atmospheric background concentrations (320 ppb) were sampled immediately after closure. The detection limit was determined taking into account a minimum 10% change in concentration during sampling from the initial background values. According to this criterion, the calculated detection limit of fluxes is 1.3 μg N m<sup>-2</sup> h<sup>-1</sup>.

Flux was determined according to Horváth et al. (2008):

$$F_{N_2O} = \frac{2 \times \Delta C \times A_N \times V_{ch} \times 60 \times f}{V_m \times A_{ch} \times t} = 3.5 \times \Delta C \times \frac{f}{t}, \quad (3)$$

where  $F_{N_2O}$  is the flux (μg N m<sup>-2</sup> hour<sup>-1</sup>), 2 is a conversion factor from ppb unit into μg N m<sup>-3</sup> (being two N in the N<sub>2</sub>O molecule),  $\Delta C$  is the difference in mixing ratios (nL L<sup>-1</sup> or ppb) in chambers at the end and the start of samplings,  $t=30$  min: the time of samplings,  $A_N$  is the relative atomic weight of N (14 g),  $V_{ch}$  the volume of chambers (m<sup>3</sup>), 60 is the time conversion factor from minutes to hours (min/hour),  $f=1.090$  is a dimensionless correction factor taking into account the residual pressure in the evacuated tubes after evacuation,  $V_m$  is the temperature dependent molar volume of ideal gases (m<sup>3</sup>), and  $A_{ch}$  is the surface area (m<sup>2</sup>) of soil covered by chambers.

Results of the statistical analysis showed that the non-systematic bulk error (coefficient of variation) of sampling and analysis was always below 10%. It was estimated by 10-10 parallel chambers in the field using  $t=10$  samples where accumulation of soil emitted N<sub>2</sub>O is zero and the background mixing ratio is measured in chambers (~320 ppb). Taking into account that the average accumulation rate is of 30%, the estimated error is in the same magnitude for samples taken in  $t=10$ ; 20 and 30 min.

## 2.7 Soil nitric oxide flux measurements by dynamic chambers

Soil emission of NO has been determined by the dynamic chamber method as described in Horváth et al. (2006). The installation of the chambers, the sampling protocol and the concentration measurement by Horiba APNA/APOA 350 instruments were described earlier in Machon et al. (2010).

Chemical correction of rapid reaction of NO with ozone (NO+O<sub>3</sub>→NO<sub>2</sub>+O<sub>2</sub>) was taken into account. NO flux was estimated according to Meixner et al. (1997). Under steady-state conditions the mass balance equation for NO can be written as follows (the photolysis rate of NO<sub>2</sub> inside the dark chambers was estimated to be zero):

$$F_f + F_m + F_{bl} + F_{gp} = 0, \quad (4)$$

where  $F_f$  is the soil flux,  $F_m$  is the difference between fluxes entering and leaving the chamber,  $F_{bl}$  is the term for the wall effect which was negligible because of the relatively short residence time (40 s) of the gas mixture in the chamber, and  $F_{gp}$  is the loss of NO due to the chemical reaction with ozone. After solving Eq. (4) soil flux of NO ( $F_f$ , μg N m<sup>-2</sup> hour<sup>-1</sup>) can be calculated as:

$$F_f = \frac{[NO]_{out} - [NO]_{in}}{f_1} \times f_2 \times \frac{Q}{A} + k \frac{[NO]_{out} - [NO]_{in}}{f_1} \times \frac{V}{A}, \quad (5)$$

where  $Q$  is the flow rate (1.5×10<sup>-3</sup> m<sup>3</sup> minute<sup>-1</sup>),  $A$  is the area of the chamber (20×10<sup>-3</sup> m<sup>2</sup>),  $[NO]_{out}$  and  $[NO]_{in}$  are the mixing ratio of NO in the air (ppb or nL L<sup>-1</sup>) leaving and entering the chamber, respectively, while

$$f_1 = \frac{A_N}{V_t}, \quad (6)$$

where  $A_N$  is the relative atomic weight of N,  $V_t$  is the molar volume (m<sup>3</sup>) of air at the given temperature,  $f_2=60$  is the conversion term from minutes to hours (min/hour).  $V$  is the volume of the chamber (m<sup>3</sup>), and  $k$  is the reaction

354 rate constant:  $k = 1.8 \times 10^2 \times \exp(-1400/T)$  ( $\text{ppb}^{-1} \text{hour}^{-1}$  or  $\text{L nL}^{-1} \text{hour}^{-1}$ ), after Seinfeld and Pandis (1998),  
355 where  $[\text{O}_3]_{\text{out}}$  is the mixing ratio of ozone ( $\text{nL L}^{-1}$ ) leaving the chamber.

356 Soil emissions of  $\text{N}_2$  and  $\text{NH}_3$  have not been measured because of practical reasons (field measurements  
357 would have been difficult and/or expensive).

358  
359  
360

### 361 **3.1 Measured wet deposition of inorganic and organic N**

362

363 Based on the 5 year measurement average  $\text{NO}_3^-$  and  $\text{NH}_4^+$  content of the precipitation were 2.84 and 0.78  $\text{mg L}^{-1}$   
364 (0.64 and 0.61  $\text{mg N L}^{-1}$ ), respectively (Table 3). The average share of ammonium and nitrate in the total  
365 inorganic N wet deposition is balanced in the average of 5 years (51% and 49%, respectively), but ratios vary  
366 significantly in different years.

367 It has to be noted that generally not negligible amount of organic nitrogen is also deposited by the  
368 precipitation (Cape et al. 2001). The difference between total N and inorganic N concentrations gives the  
369 dissolved organic N (DON) but the approach has some analytical artifacts. DON may take approximately 20%-  
370 30% of the total N deposition in precipitation although it is generally not included in N deposition estimates  
371 (Cornell et al. 2003). Annual average of DON in precipitation correlates better with ammonium than with nitrate  
372 and has seasonal pattern suggesting an agricultural source (Cape et al. 2004).

373 In some areas (e.g. in United Kingdom) the share of organic N in precipitation can be 24-40% in the  
374 total wet deposition (Cape et al. 2004), while in other regions (e.g. Soroe in Denmark) this ratio is only 9%  
375 (Skiba et al. 2009). Within the NitroEurope Integrated Project organic nitrogen content of our precipitation  
376 samples collected between October 2008 and February 2009 were analyzed in CEH (Centre for Ecology and  
377 Hydrology, Edinburgh). Based on the results (Neil Cape, CEH, Edinburgh personal communication) the organic  
378 N fraction in Bugac samples is around 16% of the total N content in the rainfall (Fig. 1). Because of the high  
379 uncertainty mostly caused by the short sampling period this value was not considered in our N-balance  
380 estimation.

381

### 382 **3.2 Inferred dry deposition fluxes of nitrogen compounds**

383

384 The share of different nitrogen compounds in dry deposition calculated by the inferential method can be seen in  
385 Table 3. Reduced and oxidized forms contribute 55% and 45%, respectively as an average during the  
386 observation period (5 years). These results agree with other European measurement sites (Skiba et al. 2009),  
387 although they found slightly larger differences among years. Dry deposition is dominated by ammonia and nitric  
388 acid (Table 3). It can be also noticed that the contribution of the deposited N gases ( $\text{NH}_3$ ,  $\text{HNO}_3$ , and  $\text{NO}_2$ ) is 83-  
389 89%, while the aerosol particles ( $\text{NH}_4^+$ ,  $\text{NO}_3^-$ ) takes only 11-17% of the total amount of dry nitrogen downward  
390 flux. The multi-year variation in share among dry deposition of N-forms derives solely from the year-by-year  
391 variation in atmospheric concentrations of pollutants according to Eq. (2) taking into account the constant  
392 inferred dry deposition velocity.

393

### 394 **3.3 Measured soil emission flux of $\text{NO}$ and $\text{N}_2\text{O}$**

395

396 The multi-year average of the soil  $\text{N}_2\text{O}$  flux measurement (Table 3) is 0.67  $\text{kg N ha}^{-1} \text{year}^{-1}$  although there are  
397 large differences among years. This value is close to the mean emission (0.93  $\text{kg N ha}^{-1} \text{year}^{-1}$ ) determined  
398 during a survey by nine European grassland measurement sites (Flechard et al. 2007). The observed lower  $\text{N}_2\text{O}$   
399 emission levels can be explained by the differences in rainfall, by the low N-input, and by differences in water-  
400 management features of the area of our measurements.

401 Precipitation is highly responsible for changes in the microclimate of soil, influencing the metabolisms,  
402 changing the favoring circumstances to (de)nitrification processes. The optimum range of wetness for  
403 denitrification at our site is 40-50% WFPS (Machon et al. 2011). In "regular" or dry years of 2006-2009 with an  
404 average precipitation of 480 mm and a soil moisture of 32% mean soil  $\text{N}_2\text{O}$  flux was 0.38  $\text{kg N ha}^{-1} \text{year}^{-1}$  in  
405 contrast with 2010 when soil moisture was within the optimum range (43%) caused by doubled precipitation  
406 (967 mm) and the  $\text{N}_2\text{O}$  emission increased nearly by a factor of 5. As a consequence, lower soil humidity in  
407 2006-2009 influenced the soil processes through water stress and indirectly affected the nutrient uptake by plants  
408 and the cycle of N. The annual mean temperatures in 2007-2010 were 0.5 °C warmer than the long term average  
409 mainly caused by the mild winters (soil generally was not frozen).

410 There are some flux estimations by DNDC model for European grasslands in the literature. Average  
411 emission of  $\text{N}_2\text{O}$  from grasslands was estimated by DNDC at 1.0  $\text{kg N ha}^{-1} \text{year}^{-1}$  (Levy et al. 2007). This figure  
412 is higher but it is in relatively good agreement in order with our measured and annual average values, if we  
413 consider that the well aerated and dry sandy soil is not favorable for anaerobic denitrification producing  $\text{N}_2\text{O}$ .

414 The multi-year mean NO emission was around  $1.1 \text{ kg N ha}^{-1}\text{year}^{-1}$ . Soil emission of NO exceeds that of  
415  $\text{N}_2\text{O}$  in each year (4-5 times higher in the drier years 2006-2009) with the exception of highly precipitated year  
416 2010. It is in accordance with the fact that denitrification process producing  $\text{N}_2\text{O}$  is more effective in wet soils in  
417 contrast with NO production (Davidson 1991). In 2006-2009 the soil moisture content of the well-ventilated dry  
418 soil is close to the optimum for NO formation that is around at 20-30% WFPS for Bugac soil (Machon et al.  
419 2011). Due to the properties of sandy soil (i.e. easy aeration in contrast to clay rich soils), generally nitrification  
420 dominates, favoring the formation of NO. The soil becomes anaerobic after rain events, thus significant emission  
421 peaks of  $\text{N}_2\text{O}$  and  $\text{N}_2$  by denitrification can be observed or modeled only for short periods. The rate of  $\text{N}_2\text{O}/\text{N}_2$   
422 production depends on WFPS (water-filled pore space) controlled by the amount of precipitation.

423 Within the NitroEurope IP cooperation program different soils from Europe, including soil from Bugac  
424 station, were analyzed by incubation technique using oxygen and nitrogen isotopes. With this technique O-  
425 exchange between water and intermediate forms of the N-transformations during metabolism, and the change of  
426 isotope ratio could be measured (Kool et al. 2007; 2009a). By this methodology they described that  $\text{NH}_4^+$   
427 especially in our soil can be the source for nitrous oxide production instead of  $\text{NO}_3^-$  i.e. in our soil nitrifier  
428 denitrification is an alternative  $\text{N}_2\text{O}$  formation pathway (Kool et al. 2009b and 2010). This alternative metabolic  
429 process, i.e. the nitrifier denitrification of ammonia by oxidizing bacteria may control the formation of  $\text{N}_2\text{O}$ . This  
430 biochemical pathway can be a contributor to the majority of  $\text{N}_2\text{O}$  production at our measurement site, thus  $\text{N}_2\text{O}$   
431 can also be produced at lower humidity levels in soil, resulting in a secondary peak in the range of lower wetness  
432 (20-30% WFPS) (Machon et al. 2011).

433 Comparing our  $\text{N}_2\text{O}+\text{NO}$  fluxes to another NitroEurope managed grassland site in Switzerland  
434 (Ammann et al. 2009) we can see that although the main drivers of soil processes are quite different (e.g. the clay  
435 fraction is 40% there and the amount of yearly precipitation at Swiss site was two times higher) the soil fluxes of  
436  $\text{N}_2\text{O}$  and NO were at the same level ( $<1 \text{ kg N ha}^{-1}\text{year}^{-1}$ ).

### 437 **3.4 The net atmosphere-surface flux based on measurements**

438 The summary of atmospheric deposition and soil emission rates can be seen in Table 3. There are large  
439 variations in deposition and emission fluxes among years especially due to different meteorological conditions.  
440 In the year of 2010 annual precipitation was 967 mm; significantly higher than long-term (1989-2006) average  
441 (550 mm). Years 2007 and 2009 were slightly arid with 446 mm and 486 mm of precipitation, respectively.  
442 Comparing both years to the long term seasonal pattern it can be concluded that the deficit in the yearly  
443 precipitation generally occurred in spring and summer. In 2008 the yearly precipitation (567 mm) reached the  
444 regular level, whilst the annual mean temperature was a bit higher than the long term average. All of these  
445 variations in meteorological parameters show that the conditions were different in every year.

446 Analyzing the deposition dataset it is assumed that ratio of dry to wet deposition of nitrogen is  
447 influenced by the amount of yearly precipitation because of the generally observed linear relationship between  
448 wet deposition and precipitation rate. The ratio of dry to wet deposition varied within the range of 1.5-2.3  
449 between 2006 and 2010.

450 The share of different N-gas soil emissions is influenced year-by-year by different atmospheric  
451 concentrations and weather conditions such as precipitation or soil/air temperature at our measurement site. Both  
452 soil N-fluxes and ratio of  $\text{N}_2\text{O}/\text{NO}$  soil emission varies in a wide range and we can observe that meteorological  
453 conditions affect soil processes in larger extent than the rate of dry or wet deposition. Consequently, the ratio of  
454 total deposition and soil gas emission ( $\text{NO}+\text{N}_2\text{O}$ ) varies between 6 and 13 year-by-year and the magnitude of the  
455 deposition is higher approximately by one order of magnitude than soil emission. The measured N net flux  
456 between the atmosphere and the surface at the study site ranged between  $-9.4$  and  $-13.3 \text{ kg N ha}^{-1}\text{year}^{-1}$  as the  
457 sum of the measured deposition and emission terms. Depending on the year 2-10% of the deposited N returns to  
458 atmosphere in the form of greenhouse gas  $\text{N}_2\text{O}$  and 5-13% of it in the form of NO.

### 461 **3.5 Modeled soil fluxes**

462 For validation of the DNDC model we compared the simulated NO and  $\text{N}_2\text{O}$  soil flux data and C/N ratio with  
463 measured values (see Figs. 2, 3 and Tables 4, 5). At our measurement site there were  $\text{N}_2\text{O}$  flux measurements in  
464 2002-2004 in the framework of the GreenGrass project on a fortnightly sampling basis (Horváth et al. 2010). So,  
465 together with our current measurements, we can compare the simulation results with almost a whole decade of  
466 measured dataset of  $\text{N}_2\text{O}$  flux (Table 5). Comparison for a shorter period was published by Hidy et al. (2011).

467 Monthly fluxes modeled and measured by static/dynamic chambers show a slight agreement ( $r=0.74$  for  
468  $\text{N}_2\text{O}$  and  $r=0.83$  for NO at the probability level of  $p<0.01$ ). In winter months the agreement is worse caused by a  
469 number of reasons: e.g. in winter smaller number of sampling was taken due to snow or frost, or the model  
470 probably does not predict well the emissions around  $0^\circ\text{C}$  (e.g. rain or snow, soil layers are frozen or not etc).

473 The significant proportion of the annual N<sub>2</sub>O emission is produced during winter at low soil temperature. The  
474 thawing events (when the upper soil layer is not frozen) can cause large emission peaks (Müller et al. 2002).

475 It can be seen (Fig. 3) that DNDC model systematically underestimates the NO emission peaks. Though  
476 the simulation was not able to capture the measured emission peaks, trend of the monthly emissions gives a  
477 relatively good agreement. Model simulations of daily soil N<sub>2</sub>O fluxes follow well the emission immediately  
478 after the rain events. In contrast, by conducting weekly measurements we have probably missed some N<sub>2</sub>O  
479 emission peaks. Comparing the data of Table 4 it can be concluded that simulation (with a given uncertainty)  
480 gives good estimation for C/N ratios in each year.

481 Seasonal changes in N<sub>2</sub>O and NO soil fluxes can be observed both for measured and modeled values  
482 (Figs. 2, 3). In summer months the microbial activity of both nitrification and denitrification is increasing with  
483 the soil temperature. Most of the measurement period between 2006 and 2009 can be characterized by aerobic  
484 condition favoring nitrification; therefore, measured soil flux of NO was generally higher than N<sub>2</sub>O flux. It can  
485 be explained partly by the fact that the sandy soil dried out within a short time after the rain events. That is, the  
486 effective time for denitrifier bacteria community, preferring anaerobic condition, was shorter than in soils with  
487 higher water-retaining capacity. On the other hand, in extreme wet soils, loss of nitrous oxide is decreasing;  
488 reduction goes on to molecular nitrogen (N<sub>2</sub>). Weather conditions (heat or water stress) prevented keeping the  
489 optimum soil condition for nitrous oxide production in extended periods of the observation.

490 The emission peaks in the simulation correlate with temperature, and reflect the rain events. After rain  
491 the denitrification processes can produce an N<sub>2</sub>O emission peak due to the anaerobic period. In winter time, the  
492 soil surface used to be frozen for the cold period and the produced trace gases are stored in the unfrozen subsoil  
493 (Müller et al. 2002). For this reason the first thawing events results in high emission peaks for N<sub>2</sub>O (Priemé and  
494 Christensen, 2001; Müller et al. 2002). In spring of 2007 N<sub>2</sub>O emission peak was missing as we expected  
495 because the mild winter (soil was not frozen) so the phenomenon mentioned above was not occurred.

496 Soil emissions of N<sub>2</sub> and NH<sub>3</sub> have not been measured because of practical reasons. Hence, validation of  
497 the model for these parameters was not possible due to the lack of measurement. However, taking into  
498 consideration the relatively good agreement (especially in yearly rate) between modeled and measured fluxes for  
499 NO and N<sub>2</sub>O, we can make a rough estimation (assuming a similar relationship) for N<sub>2</sub> and NH<sub>3</sub> fluxes,  
500 emphasizing that uncertainty of these values are higher than for other components.

501 Based on model results it can be concluded that the rate of annual soil N emissions including all of  
502 gases (N<sub>2</sub>, NH<sub>3</sub>, NO, N<sub>2</sub>O) varies within a relatively narrow range, with an average of 2.1 (SD=0.44) kg N ha<sup>-1</sup>  
503 year<sup>-1</sup> in Bugac station. There are significant differences between the shares of different N-components in the  
504 given year, which was influenced by change in meteorological variables and soil physics.

505 It can be observed that in years with extremely dry and hot summer (2003 and 2009) the model  
506 significantly underestimated the N<sub>2</sub>O fluxes. It can be explained by the fact that biosphere is adapted to these  
507 conditions (drought-tolerant species); on the other hand, biosphere (through life processes) try to moderate the  
508 extreme environmental conditions, which is difficult to model due to the complexity of the system. All  
509 ecological models are based on simplifications therefore we can not eliminate the deviations caused by  
510 simplifications. Overall, there is a relatively good correlation between the simulated and measured fluxes.

511 It should be emphasized that the three-dimensional heterogeneity of the ecosystem (even on meter scale)  
512 was the main reason why the measurements were taken at several places simultaneously. The microbial activity  
513 often responds much quicker to the environmental changes, such as biomass, but the biomass remains an  
514 important character. The major simulated trace gas emissions and short-term events are often driven by  
515 meteorological extreme events (heat and water stress, freezing-thawing), drying/rewetting (nitrification-  
516 denitrification), and management (grazing/cutting/fertilization etc).

517 On the basis of multi-year simulation DNDC model underestimates the emissions of N<sub>2</sub>O and NO but  
518 the standard deviation of the measured data are much larger than the simulated values. The modeled annual  
519 emission level of N<sub>2</sub>O is 0.55 kg N ha<sup>-1</sup>year<sup>-1</sup> (which gives slight agreement with our measurements). This value  
520 is lower by one order of magnitude compared to the 5.6 kg N ha<sup>-1</sup>year<sup>-1</sup> average value calculated by the IPCC  
521 method determined for many European areas (Boeckx and van Cleemput 2001). In the IPCC method the  
522 cultivated arable lands are also included, where the N<sub>2</sub>O emission is generally higher caused by the crop N-  
523 fertilization.

### 524 525 **3.6 Estimated effect of leaching, biological N-fixation, and grazing**

526  
527 The surface run-off and N-leaching are strongly dependent on slope, soil type, depth of the groundwater  
528 level, precipitation etc. as ecological drivers. On a flat area surface run-off practically does not occur. The  
529 groundwater level is about 6 m deep, and the sandy surface tends to dry out quickly, thus the nitrate  
530 measurement from soil moisture were usually unsuccessful due to the soil water content in soil is often below  
531 20%, therefore the nitrate leaching is probably negligible. While in Bugac N-leaching was neglected for an

532 extensively managed (fertilized) grassland site in Swiss Central Plateau it was estimated to be  $3.5 \text{ kg N ha}^{-1}\text{year}^{-1}$   
533 in the period of 2002-2009 (Ammann et al. 2009).

534 The biological N-fixation by plants (BNF) is not negligible because of the 8-17% share of legumes  
535 observed according to our surveys in different years; however, we were not able to determine the N-fixation by  
536 direct measurement. Biological nitrogen fixation for natural grasslands ranges within  $2.3\text{-}3.1 \text{ kg N ha}^{-1}\text{year}^{-1}$   
537 estimated by Butterbach-Bahl et al. (2011). As to the N-fixation by legumes the European outlook is the follows:  
538 although the measurement of N-fixation by  $^{15}\text{N}$  isotope technique is not resolved everywhere yet, according to a  
539 preliminary study the estimated N-fixation is less than  $1\text{-}2 \text{ kg N ha}^{-1}\text{year}^{-1}$  at most of the sites of NitroEurope  
540 project (Skiba et al. 2009) except Oensingen (Switzerland) where a higher value was calculated since the  
541 legumes ratio is 48%.

542 Based on land cover ratio of the legumes (8-17%) the estimated average N-fixation by plants at Bugac  
543 site is around  $2.9 \text{ kg (1.9-3.9) N ha}^{-1}\text{year}^{-1}$  (personal communication of Christof Ammann, Agroscope  
544 Reckenholz-Tanikon ART, Swiss Federal Research Station, Zürich, Switzerland). This value was considered in  
545 our N-exchange estimation (see next chapter). Nevertheless, the main N-input of our grassland is the dry and wet  
546 depositions (altogether  $11\text{-}15 \text{ kg N ha}^{-1}\text{year}^{-1}$ ).

547 The research area is usually grazed by a herd of gray cattle for about 220 days yearly with  $0.5\text{-}0.8 \text{ LU}$   
548  $\text{ha}^{-1}$  (500-800 livestock unit in 1074 ha) grazing pressure. The Hungarian Grey Cattle breed is protected for  
549 genome conservation and propagation for other farms; hence there is no extensive meat or milk producing and  
550 consequently the number of animals is relatively constant. We estimated the output and input of nitrogen by  
551 grazing on the basis of literature and simulated data. Waldrip et al. (2013) reported an average of  $162 \text{ (SD=50) g}$   
552  $\text{N day}^{-1}\text{LU}^{-1}$  nitrogen uptake by grazing. Using this figure the N content of grazed grass is  $21.6 \pm 9 \text{ kg N ha}^{-1}$   
553  $\text{year}^{-1}$ . Based on DNDC simulations (Table 4) the annual average of the removal by grazing is  $21 \pm 5 \text{ kg N ha}^{-1}$   
554 which shows good agreement with the calculated data.

555 With grazing nitrogen temporarily leaves ecosystem and it is partly supplied back with excreta except  
556 the amount of nitrogen built into the bodies of cattle. Waldrip et al. (2013) estimated  $119 \text{ (SD=38) g N day}^{-1}$   
557  $\text{LSU}^{-1}$  excreted nitrogen. The calculated N uptake at our site using this figure is  $15.8 \pm 5 \text{ kg N ha}^{-1}\text{year}^{-1}$ . Another  
558 calculation of Skiba et al. (2009) estimated of  $16 \text{ kg N ha}^{-1}\text{year}^{-1}$  as N input by excretion at Bugac site.

559

### 560 3.7 Total N-balance including measured, modeled, and estimated data

561

562 The net surface atmosphere balance can be seen in Fig. 4. In the lack of direct measurements of soil  $\text{N}_2$  and  $\text{NH}_3$   
563 emission we can make a rough estimation for the orders of magnitude of these processes. Hence, we used the  
564 modeled emission fluxes to compare the soil emission to the N-uptake including deposition and biological  
565 fixation (Fig. 4). All of soil emission rates are derived from DNDC modeling. Total N-balance in Fig. 4 does not  
566 involve the effect of grazing and excreta.

567 For estimation of net balance of nitrogen including BNF and the effect of grazing and excreta these data  
568 were delivered as it was described in section 3.6 (Table 3). The net budget shows negative balance (input terms  
569 exceed the outputs) but we have to take into consideration the large uncertainty in estimation of the effect of  
570 grazing. We can only state with caution that the ecosystem has a surplus of nitrogen or it is close to the  
571 equilibrium. Probably the excess (if any) is mineralized and stored in the soil in inorganic form. This might be  
572 supported by the increase in total nitrogen content in the 0-30 cm layer between 2006 and 2010: 0.28; 0.33; 0.37;  
573 0.38; 0.34%. According to a previous study (Sophie Zechmeister-Boltenstern, BOKU, personal communication)  
574 the mineralization rate is  $194 \mu\text{g NH}_4^+ \text{ g}^{-1}\text{week}^{-1}$  related to dry weight of soil.

575

### 576 3.8. Climatic and ecological effects

577

578 Based on that global warming potential of  $\text{N}_2\text{O}$  is 298 times higher than that of  $\text{CO}_2$  on 100-year time  
579 horizon (IPCC 2007), both measured and simulated yearly  $\text{N}_2\text{O}$  emissions range between 71-849 and 108-409 kg  
580  $\text{CO}_2\text{-eq. ha}^{-1}\text{year}^{-1}$ , respectively during the period examined.

581 Temperature and precipitation mainly affect the formation of  $\text{N}_2\text{O}$  and  $\text{NO}$ , and the seasonal fluctuation  
582 of their soil emission. In the measurement period the soil flux of  $\text{N}_2\text{O}$  was less than  $1 \text{ kg N ha}^{-1}\text{year}^{-1}$ . The  
583 forecasted potentially drying climate in the Hungarian Great Plan can lead to reduced  $\text{N}_2\text{O}$  emission, which may  
584 mean a potential negative feedback on greenhouse effect. On the other hand, Pintér et al. (2008) and Barcza et al.  
585 (2009) argued that grass can turn into a net  $\text{CO}_2$  source in extreme dry years like 2003 and 2007 as a positive  
586 feedback for climate change. Though the impact of these two phenomena is opposite we should not neglect  
587 them, considering that the ratio and the strength of both processes are unknown. In addition, in Hungary the area  
588 of the surface covered by the temperate grass is large and will be growing with increasing aridity of climate.

589 The residence time of N in intensively managed grassland is potentially lower than at untreated  
590 grasslands. Soil nitrogen pool depends on the consumption and exchange of both living roots and bacteria which  
591 are competitors for the same nutrients. Function of plant physiology (plant N-uptake demand), plant growing,

592 microbial metabolism are highly affected by air and soil temperature and water deficit etc. Plant association of  
593 Bugac becomes open grassland if it is exposed many times to naturally induced droughts. So plants also induce  
594 effect on soil N transformation. The selection of herbivore is important, because every species have plant  
595 preferences. The optimum number of animals ( $0.5-0.8 \text{ ha}^{-1}$ ) is also required to preserve the treeless condition and  
596 the diversity of plant species through small disturbance. With overgrazing the nitrogen can be mobilized faster in  
597 the soil due to manure and urine and may lead faster biochemical processes. The compaction of the upper soil  
598 layer due to animal trampling can cause airless condition (decreasing porosity). This may increase the  
599 denitrification activity, which can lead to increased N losses.

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601

#### 602 4. Conclusions

603

604 Based on measurements and model simulations we quantified the extent of dry and wet N deposition  
605 with soil N-gas fluxes originated from microbiological processes in different years for semi-arid, semi-natural,  
606 extensively grazed grassland. The ammonia dry deposition is the main N-source (35-40%) of the area. The dry  
607 deposition of nitric acid vapor (20-25%) and wet deposition of ammonium and nitrate ions (30-35%) have also  
608 relatively large influence in the different years. The average modeled N-gas emission ( $2.1 \pm 0.4 \text{ kg N ha}^{-1} \text{ year}^{-1}$ )  
609 of the area is lower by one order than deposition rate. Both the rate of nitrogen load and soil emission are less  
610 than at other European sites due to the low atmospheric deposition, lack of intensive management, and  
611 fertilization. Bugac (nature reserved, undisturbed area) seems to be representing background levels of fluxes  
612 caused by the absence of local air pollution or N-sources (fertilizers etc). This statement is in accordance with  
613 the work of Skiba et al. (2009), where various N deposition and emission values were overviewed for different  
614 grasslands and other sites.

615 The calculated yearly N-balance (net flux) between the atmosphere and the surface ranged between  $-9.4$   
616 and  $-14 \text{ kg N ha}^{-1} \text{ year}^{-1}$  as the sum of the measured deposition and emission terms:  $-11.2$  to  $-15.1$  and  $0.9-2.9$   
617  $\text{kg N ha}^{-1} \text{ year}^{-1}$ , respectively, between 2006 and 2010 (without BNF and effect of grazing).

618 If we take into account the biological nitrogen fixation and the effect of grazing (effects of both grazed  
619 plant and excreta) the net nitrogen balance varies within  $-6.6$  and  $-11 \text{ kg N ha}^{-1} \text{ year}^{-1}$ . It seems - taken into  
620 account the high uncertainty in calculation of grazing effect - that sources of nitrogen exceed the sinks; the  
621 surplus is probably mineralized in the soil.

622 We applied and validated the DNDC model first time in Hungary (built up a database of the air, soil and  
623 other parameters and using the measured data of Bugac). Soil trace gas emissions are strongly controlled by soil  
624 organic carbon and soil mineral N-content, and by soil temperature and moisture.

625 Using the DNDC model we are able to give N-gas flux prediction for those lands where measurements  
626 are missing and we are able to simulate fluxes of parameters and soil processes where field or laboratory  
627 measurements are difficult or expensive. This provides some support for future use of the DNDC model in  
628 regional mode for scaling up the soil fluxes for different ecosystem types or give climate scenario estimation up  
629 to country scale.

630

631 **Acknowledgements**– The authors are grateful for the permission of the Management of Kiskunság National  
632 Park to conduct the studies in the territory of the park in Bugacpuszta. This work has been supported by the  
633 NitroEurope EU 6<sup>th</sup> FP, the ÉCLAIRE EU 7<sup>th</sup> FP (282910), National Research and Innovation Programs  
634 (NKFP/088/2004, NKFP6- 00028/2005 and NKFP6-00079/2005), COST Action 0804, and partly by the  
635 European Union and co-financed by the European Social Fund grant agreement no.: TAMOP 4.2.1/B-  
636 09/1/KMR-2010-0003. Authors thank to dr. Neil Cape (CEH) for organic N analyses.

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#### 639 References

640 Ammann, C., Spirig, C., Leifeld, J. and Neftel, A. (2009). Assessment of the nitrogen and carbon-balance of two  
641 managed temperate grassland fields. *Agr. Ecosyst. Environ.* 133, 150–162.

642 Barcza, Z., Haszpra, L., Somogyi, Z., Hidy, D., Lovas, K., Churkina, G. and Horváth, L. (2009). Estimation of  
643 the biospheric carbon dioxide balance of Hungary using the BIOME-BGC model. *Időjárás* 113, 203–219.

644 Beheydt, D., Boeckx, P., Sleutel, S., Li C. and van Cleemput, O. (2007) Validation of DNDC for 22 long-term  
645  $\text{N}_2\text{O}$  field emission measurements. *Atmospheric Environment* 41, 6196–6211.

646 Boeckx, P. and van Cleemput, O. (2001). Estimates of  $\text{N}_2\text{O}$  and  $\text{CH}_4$  fluxes from agricultural land in various  
647 regions of Europe. *Nutr. Cycl. Agroecosyst.* 60, 35–47.

648 Borrell, P., Bultjes, J.H., Grennfelt, P. and Hov, Ø. (eds.) (1997). Transport and chemical transformation of  
649 pollutants in the troposphere. *Photo-Oxidants, Acidification and Tools: Policy Applications of Eurotrac*  
650 *Results*, vol. 10. Springer, pp. 116.

651 Brown, L., Syed, B., Jarvis, S.C., Sneath, R.W., Phillips, V.R., Goulding, K.W.T. and Li, C. (2002).  
652 Development and application of a mechanistic model to estimate emission of nitrous oxide from UK  
653 agriculture. *Atmospheric Environment* 36, 917–928.

654 Butterbach-Bahl, K., Kesik, M., Miehle, P., Papen, H. and Li, C. (2004). Quantifying the regional source  
655 strength of N-trace gases across agricultural and forest ecosystems with process based models. *Plant Soil*.  
656 260, 311–329.

657 Butterbach-Bahl, K., Gundersen, P., Ambus, P., Augustin, J., Beier, C., Boeckx, P., Dannenmann, M., Sanchez  
658 Gimeno, B., Ibrom, A., Kiese, R., Kitzler, B., Rees R.M., Smith, K.A., Stevens, C., Vesala, T. and  
659 Zechmeister-Boltenstern, S. (2011). Nitrogen processes in terrestrial ecosystems, in: Sutton, M.A., Howard,  
660 C.M., Erisman, J.W., Billen, G., Bleeker, A., Grennfelt, P., van Grinsven, H. and Grizzetti B. (eds.), *The  
661 European Nitrogen Assessment*. Cambridge University Press. New York 2011, ISBN 978-1-107-00612-6.  
662 Chapter 6, pp. 99–125.

663 Cai, Z., Sawamoto, T., Li, C., Kang, G., Boonjawat, J., Mosier, A.R., Wassmann, R. and Tsuruta, H. (2003).  
664 Field validation of the DNDC model for greenhouse gas emissions in East Asian cropping systems. *Global  
665 Biogeochem. Cycles*. 17(4), 1107.

666 Cape, J. N., Kirika, A., Rowland, A. P. Wilson, D.R., Jickells, T.D. and Cornell, S. (2001). Organic nitrogen in  
667 precipitation: real problem or sampling artifact? *The Scientific World* 1, 230–237.

668 Cape, J.N., Anderson, M., Rowland, A.P. and Wilson, D. (2004). Organic nitrogen in precipitation across the  
669 United Kingdom. *Water Air Soil Pollut.* 4, 25–35.

670 Clayton, H., Arah, J.R.M. and Smith, K.A. (1994). Measurement of nitrous oxide emissions from fertilized  
671 grassland using closed chambers. *J. Geophys. Res.* 99, 16599–16607.

672 Christensen, S., Ambus, P., Arah, J.R.M., Clayton, H., Galle, B., Griffith, D.W.T., Hargreaves, K.J.,  
673 Klemmedtsson, L., Lind, A.-M., Maag, M., Scot, A., Skiba, U., Smith, K.A., Welling, M. and Wienhold, F.G.  
674 (1996). Nitrous oxide emission from an agricultural field: comparison between measurements by flux  
675 chamber and micro-meteorological techniques. *Atmos. Environ.* 30, 4183–4190.

676 Cornell, S. E., Jickells, T. D., Cape, J. N., Rowland, A. P. and Duce, R. A. (2003). Organic nitrogen deposition  
677 on land and coastal environments: a review of methods and data. *Atmospheric Environment* 37, 2173–2191.

678 Davidson E.A. (1991). Fluxes of nitrous oxide and nitric oxide from terrestrial ecosystems, in: J.E. Roger and  
679 W.B. Whitman (eds.) *Microbial production and consumption of greenhouse gases: Methane, nitrogen oxides  
680 and halomethanes*. Am. Soc. Microbiol. Washington DC. pp. 219-235.

681 EMEP (1996). EMEP Manual for sampling and chemical analysis. EMEP/CCC-Report 1/95, NILU, Kjeller,  
682 Norway.

683 Erisman, J. W., Grennfelt, P. and Sutton M. (2003). The European perspective on nitrogen emission and  
684 deposition. *Environ. Internat.* 29, 311–325.

685 Erisman, J.W., Vermeulen, A., Hensen, A., Flechard, C., Dammgen, U., Fowler, D., Sutton, M., Grünhage, L.  
686 and Tuovinen, J.P. (2005). Monitoring and modelling of biosphere/atmosphere exchange of gases and  
687 aerosols in Europe. *Envir. Poll.* 133, 403–413.

688 Erisman, J. W., van Grinsven, H., Grizzetti, B., Bouraoui, F., Powlson, D., Sutton, M. A., Bleeker, A. and Reis  
689 S. (2011). The European nitrogen problem in a global perspective, in: Sutton, M.A., Howard, C.M.,  
690 Erisman, J.W., Billen, G., Bleeker, A., Grennfelt, P., van Grinsven, H. and Grizzetti B. (eds.), *The European  
691 Nitrogen Assessment*. Cambridge University Press. New York 2011, ISBN 978-1-107-00612-6. Chapter 2,  
692 pp. 9–31.

693 Flechard, C. R., Ambus, P., Skiba, U., Rees, R.M., Hensen, A., van Amstel, A., van den Pol-van Dasselaar, A.,  
694 Soussana, J.-F., Jones, M., Clifton-Brown, J., Raschi, A., Horvath, L., Neftel, A., Jocher, M., Ammann, C.,  
695 Leidfield, J., Fuhrer, J., Calanca, P.L., Thalman, E., Pilegaard, K., Di Marco, C., Campbell, C., Nemitz, E.,  
696 Hargreaves, K.J., Levy, P., Ball, B.C., Jones, S., van de Bulk, W.C.M., Groot, T., Blom, M., Domingues, R.,  
697 Kasper, G., Allard, V., Ceshia, E., Cellier, P., Laville, P., Henault, C., Bizouard, F., Abdalla, M., Williams,  
698 M., Baronti, S., Berretti, F. and Grosz, B. (2007). Effects of climate and management intensity on nitrous  
699 oxide emissions in grassland systems across Europe. *Agric. Ecosys. Environ.* 121, 135–152.

700 Gallagher, M.W., Nemitz, E., Dorsey, J.R., Fowler, D., Sutton, M.A., Flynn, M. and Duyzer, J. (2002).  
701 Measurements and parameterizations of small aerosol deposition velocities to grassland, arable crops and  
702 forests: influence of surface roughness length on deposition. *J. Geophys. Res.* 107, AAC1–AAC8.

703 Galloway, J. N., Aber, J. D., Erisman, J. W., Seitzinger, S. P., Howarth, R. W., Cowling, E. B., and Cosby, B. J.  
704 (2003). The nitrogen cascade. *BioScience* 53, 341–356.

705 Giltrap, D.L., Li, C. and Sagar, S. (2010). DNDC: A process-based model of greenhouse gas fluxes from  
706 agricultural soil. *Agric. Ecosys. Environ.* 136, 292–300.

707 Hesterberg, R., Blatter, A., Fahrni, M., Rosset, M., Neftel, A., Eugster, W. and Wanner, H. (1996) Deposition of  
708 nitrogen containing compounds to an extensively managed grassland in central Switzerland. *Envir. Poll.* 30,  
709 1247–1254.

- 710 Hicks, W.K., Whitfield, C.P., Bealey, W.J. and Sutton, M.A., (eds.) (2011). Nitrogen Deposition and Natura  
711 2000: Science and practice in determining environmental impacts. COST729/Nine/ESF/CCW/JNCC/SEI  
712 Workshop Proceedings, published by COST. Available at: <http://cost729.ceh.ac.uk/n2kworkshop> ISBN 978-  
713 91-86125-23-3
- 714 Hidy, D., Machon, A., Haszpra, L., Nagy, Z., Pintér, K., Churkina, G., Grosz, B., Horváth L. and Barcza, Z.  
715 (2011). Modeling of biosphere-atmosphere exchange of greenhouse gases. Part III, Chapter 10 Grasslands,  
716 in: Haszpra, L. (ed.), Atmospheric Greenhouse Gases: The Hungarian Perspective. Springer, Dordrecht,  
717 Heilderberg, London, New York.
- 718 Horváth, L., Bozó, L., Haszpra, L., Kopacz, J., Molnár, Á., Práger, T. and Weidinger, T. (1992). Gradient  
719 measurement of air-soil exchange of gases, in: Schwartz, S.E. and Slinn, W.G.N. (eds.), Precipitation  
720 Scavenging and Atmosphere–Surface Exchange, vol. 2, Hemisphere Publishing Corporation, Washington,  
721 Philadelphia, London, pp. 637–648.
- 722 Horváth, L., Asztalos, M., Führer, E., Mészáros, R. and Weidinger, T. (2005). Measurement of ammonia exchange  
723 over grassland in the Hungarian Great Plain. *Agric. Forest Meteorol.*, 130, 282–298.
- 724 Horváth, L., Führer, E. and Lajtha, K. (2006). Nitric oxide and nitrous oxide emission from Hungarian forest  
725 soils; linked with atmospheric N-deposition. *Atmos. Environ.* 40, 7786–7795.
- 726 Horváth, L., Grosz, B., Machon, A., Balogh, J., Pintér, K. and Czóbel, Sz. (2008). Influence of soil type on N<sub>2</sub>O  
727 and CH<sub>4</sub> soil fluxes in Hungarian grasslands. *Commun. Ecol.* 9 (Suppl.), 75–80.
- 728 Horváth, L., Grosz, B., Machon, A., Tuba, Z., Nagy, Z., Czóbel, Sz., Balogh, J., Péli, E., Fóti, Sz., Weidinger,  
729 T., Pintér, K. and Führer, E. (2010). Estimation of nitrous oxide emission from Hungarian semi-arid sandy  
730 and loess grasslands; effect of grazing, irrigation and application of fertilizer. *Agric. Ecosys. Environ.* 139,  
731 255–263.
- 732 Hsieh, C., Leahy, P., Kiely, G. and Li, C. (2005). The effect of future climate perturbations on N<sub>2</sub>O emissions  
733 from fertilized humid grassland. *Nutrient Cycling Agroecosys.* 73, 15–23.
- 734 IPCC (2007) The physical science basis – contribution of working group I to the fourth assessment report of the  
735 IPCC. Cambridge University Press, Cambridge, New York.
- 736 Jagadeesh Babu, Y., Li, C., Frolking, S., Nayak, D.R. and Adhya, T. K. (2006). Field validation of DNDC model  
737 for methane and nitrous oxide emissions from rice-based production systems of India. *Nutrient Cycling*  
738 *Agroecosys.* 74, 157–174.
- 739 Kool, D.M., Wrage, N., Oenema, O., Dolfing, J. and van Groenigen, J.W. (2007). Oxygen exchange between  
740 (de)nitrification intermediates and H<sub>2</sub>O and its implications for source determination of NQ<sup>-</sup> and N<sub>2</sub>O: a  
741 review, *Rapid Commun. Mass Spectrom.* 21, 3569–3578.
- 742 Kool, D.M., Müller, C., Wrage, N., Oenema, O. and van Groenigen, J.W., (2009a). Oxygen exchange between  
743 nitrogen oxides and H<sub>2</sub>O can occur during nitrifier pathways. *Soil Biol. Biochem.* 41, 1632–1641.
- 744 Kool, D.M., Wrage, N., Oenema, O., Harris, D. and van Groenigen, J.W. (2009b). The <sup>18</sup>O signature of biogenic  
745 nitrous oxide is determined by O exchange with water. *Rapid Commun. Mass Spectrom.* 23, 104–108.
- 746 Kool, D.M., Wrage, N., Zechmeister-Boltenstern, S., Pfeiffer, M., Brus, D., Oenema, O. and van Groenigen, J.W.  
747 (2010). Nitrifier denitrification can be a source of N<sub>2</sub>O from soil: a revised approach to the dual isotope  
748 labelling method. NitroEurope Open Science Conference: Reactive Nitrogen and the European Greenhouse  
749 Gas Balance. February 3-4, 2010, Solothurn, Switzerland. Book of abstracts: pp. 12.
- 750 Kugler, Sz., Horváth, L. and Machon, A. (2008). Estimation of nitrogen flux between the atmosphere and  
751 aquatic/terrestrial ecosystems in Hungary. *Environ. Poll.* 154, 498–503.
- 752 Levy, P.E., Mobbs, D.C., Jones, S.K., Milne, R., Campbell C. and Sutton, M.A. (2007). Simulation of fluxes of  
753 greenhouse gases from European grasslands using the DNDC model. *Agric. Ecosys. Environ.* 121, 186–192.
- 754 Li, C., Frolking, S. and Frolking, T.A. (1992a). A model of nitrous oxide evolution from soil driven by rainfall  
755 events: 1. Model structure and sensitivity. *J. Geophys. Res.* 97, 9759–9776.
- 756 Li, C., Frolking, S. and Frolking, T.A. (1992b). A model of nitrous oxide evolution from soil driven by rainfall  
757 events: 2. Model applications. *J. Geophys. Res.* 97, 9777–9783.
- 758 Li, C. (2000) Modeling trace gas emissions from agricultural ecosystems. *Nutrient Cycling Agroecosys.* 58,  
759 259–276.
- 760 Li, C. (2007) Quantifying greenhouse gas emissions from soils: Scientific basis and modeling approach, *Soil Sci.*  
761 *Plant Nutrition* 53, 344-352.
- 762 Machon, A., Horváth, L., Weidinger, T., Grosz, B., Pintér, K., Tuba Z. and Führer, E. (2010). Estimation of net  
763 nitrogen flux between the atmosphere and a semi-natural grassland ecosystem in Hungary. *Eur. J. Soil Sci.*  
764 61, 631–639.
- 765 Machon, A., Horváth, L., Weidinger, T., Pintér, K., Grosz, B., Nagy, Z. and Führer, E. (2011) Weather induced  
766 variability of N-exchange between the atmosphere and a grassland in the Hungarian Great Plain. *Időjárás*  
767 115, 219–232.
- 768 Marner, B.B. and Harrison, R.M. (2004). A spatially refined monitoring based study of atmospheric nitrogen  
769 deposition. *Atmos. Environ.* 38, 5045–5056.

- 770 Massad, R. S., Nemitz, E. and Sutton, M. A. (2010). Review and parameterisation of bi-directional ammonia  
771 exchange between vegetation and the atmosphere. *Atmos. Chem. Phys. Discuss.* 10, 10359–10386.
- 772 Meixner, F.X., Fickinger, Th., Marufu, L., Serca, D., Nathaus, F.J., Makina, E., Mukurumbira, L. and Andreae,  
773 M.O. (1997). Preliminary results on nitric oxide emission from a southern African savanna ecosystem.  
774 *Nutrient Cycling Agroecosys.* 48, 123–138.
- 775 Meixner F.X. and Yang W.X. (2006). Biogenic emissions of nitric oxide and nitrous oxide from arid and semi-  
776 arid land, in: D’Odorico P., Porporato A. (eds.), *Dryland ecohydrology*. Springer, Dordrecht, Berlin,  
777 Heidelberg, New York. ISBN 1-4020-4260-7. Chapter 14, pp. 233–255.
- 778 Moldanová, J., Grennfelt, P., Jonsson, A., Simpson, D., Spranger, T., Aas, W., Munthe, J. and Rabl A., Nitrogen  
779 as a threat to European air quality (2011). in: Sutton, M.A., Howard, C.M., Erisman, J.W., Billen, G.,  
780 Bleeker, A., Grennfelt, P., van Grinsven, H. and Grizzetti B. (eds.), *The European Nitrogen Assessment*.  
781 Cambridge University Press. New York 2011, ISBN 978-1-107-00612-6. Chapter 18, pp. 405–433.
- 782 Müller, C., Martin, M., Stevens, R.J., Laughlin, R.J., K. Ammann, C., Ottow, J.C.G. and Jager H.J. (2002)  
783 Processes leading to N<sub>2</sub>O emissions in grassland soil during freezing and thawing. *Soil Biol. Biochem.* 34,  
784 1325–1331.
- 785 Nagy, Z., Pintér, K., Czóbel, Sz., Balogh, J., Horváth, L., Fóti, Sz., Barcza, Z., Weidinger, T., Csintalan, Zs.,  
786 Dinh, N.Q., Grosz, B. and Tuba, Z. (2007). The carbon-balance of semi-arid grassland in a wet and a dry  
787 year in Hungary. *Agric. Ecosys. Environ.* 121, 21–29.
- 788 Neufeldt, H., Schafer, M., Angenendt, E., Li, C., Kaltschmitt, M. and Zeddies, J. (2006). Disaggregated  
789 greenhouse gas emission inventories from agriculture via a coupled economic-ecosystem model. *Agric.*  
790 *Ecosys. Environ.* 112, 233–240.
- 791 Pathak, H., Li, C. and Wassmann, R. (2005). Greenhouse gas emissions from India rice fields: calibration and  
792 upscaling using the DNDC model. *Biogeosci.* 1, 1–11.
- 793 Pilegaard, K., Skiba, U., Ambus, P., Beier, C., Brüggemann, N., Butterbach-Bahl, K., Dick, J., Dorsey, J.,  
794 Duyzer, J., Gallagher, M., Gasche, R., Horvath, L., Kitzler, B., Leip, A., Pihlatie, M.K., Rosenkranz, P.,  
795 Seufert, G., Vesala, T., Westrate, H. and Zechmeister-Boltenstern, S. (2006). Factors controlling regional  
796 differences in forest soil emission of nitrogen oxides (NO and N<sub>2</sub>O). *Biogeosci.* 3, 651–661.
- 797 Pintér, K., Barcza, Z., Balogh, J., Czóbel, Sz., Csintalan, Zs., Tuba, Z. and Nagy, Z. (2008). Interannual  
798 variability of grasslands’ carbon-balance dependson soil type. *Commun. Ecol.* 9 (Suppl), 43–48.
- 799 Priemé, A. and Christensen, S. (2001). Natural perturbation, drying-wetting and freezing-thawing cycles and the  
800 emission of the nitrous oxide, carbon dioxide and methane from farmed organic soil. *Soil Biol. Biochem.*  
801 33, 2083–2091
- 802 Robertson, G.P. and Groffman P.M., (2007) Nitrogen transformation, in: Paul. E.A. (ed.), *Soil Microbiology,*  
803 *Biochemistry, and Ecology*. Springer, New York, New York, USA. Chapter 13, pp. 341–364
- 804 Saggar S., Giltrap D.L., Li C. and Tate K.R. (2007). Modelling nitrous oxide emissions from grazed grasslands  
805 in New Zealand. *Agric. Ecosys. Environ.* 119, 205–216.
- 806 Seinfeld, J.H. and Pandis, S.N. (1998). *Atmospheric chemistry and physics: From air pollution to climatic*  
807 *change*. John Wiley and sons Inc., New York, Chichester, Weinheim, Brisbane, Singapore, Toronto.
- 808 Skiba, U., Drewer, J., Tang, Y.S., van Dijk, N., Helfter, C., Nemitz, E., Famulari, D., Cape, J.N., Jones, S.K.,  
809 Twigg, M., Pihlatie, M., Vesala, T., Larsen, K.S., Carter, M.S., Ambus, P., Ibrom, A., Beier, C., Hensen, A.,  
810 Frumau, A., Erisman, J.W., Brüggemann, N., Gasche, R., Butterbach-Bahl, K., Neftel, A., Spirig, C.,  
811 Horvath, L., Freibauer, A., Cellier, P., Laville, P., Loubet, B., Magliulo, E., Bertolini, T., Seufert, G.,  
812 Andersson, M., Manca, G., Laurila, T., Aurela, M., Lohila, A., Zechmeister-Boltenstern, S., Kitzler, B.,  
813 Schaufler, G., Siemens, J., Kindler, R., Flechard, C. and Sutton, M.A. (2009) Biosphere–atmosphere  
814 exchange of reactive nitrogen and greenhouse gases at the NitroEurope core flux measurement sites:  
815 Measurement strategy and first data sets. *Agric. Ecosys. Environ.* 133, 139–149.
- 816 Smith, W.N., Desjardins, R.L., Grant, B., Li, C., Lemke, R., Rochette, P., Corre, M.D. and Pennock, D. (2002).  
817 Testing the DNDC model using N<sub>2</sub>O emissions at two experimental sites in Canada. *Can. J. Soil Sci.* 82,  
818 365–374.
- 819 Smith, K.A., Ball, T., Conen, F., Dobbie, K., Massheder, J. and Rey, A. (2003). Exchange of greenhouse  
820 gasesbetween soil and atmosphere: interactions of soil physical factors and biological processes, *Eur. J. Soil*  
821 *Sci.* 54, 779–791.
- 822 Smith, W.N., Grant, B.B., Desjardins, R.L., Worth, D., Li, C., Boles, S.H. and Huffmann, E.C. (2010). A tool to  
823 link agricultural activity data with the DNDC model to estimate GHG emissions factors in Canada. *Agric.*  
824 *Ecosys. Environ.* 136, 301–309.
- 825 Soussana, J. F., Allard, V., Pilegaard, K., Ambus, P., Amman, C., Campbell, C., Ceschia, E., Clifton-Brown, J.,  
826 Czóbel, SZ., Domingues, R., Flechard, C., Fuhrer, J., Hensen, A., Horvath, L., Jones, S. M., Kasper, G.,  
827 Martin, C., Nagy, Z., Neftel, A., Raschi, A., Baronti, S., Rees, R. M., Skiba, U., Stefani, P., Manca, G.,  
828 Sutton, M., Tuba, Z. and Valentini, R. (2007). Full accounting of the greenhouse gas (CO<sub>2</sub>, N<sub>2</sub>O, CH<sub>4</sub>)  
829 balance of nine European grassland sites. *Agric. Ecosys. Environ.* 121, 121–134.

- 830 Stolk, P.C., Jacobs, C.M.J., Moors, E.J., Hensen, A., Velthof, G.L. and Kabat, P. (2009). Significant non-  
831 linearity in nitrous oxide chamber data and its effect on calculated annual emissions. *Biogeosci. Discuss.* 6,  
832 115–141.
- 833 Sutton, M.A., Milford, C., Nemitz, E., Theobald, M.R., Hill, P.W., Fowler, D., Schjørring, J.K., Mattsson, M.E.,  
834 Nielsen, K.H., Husted, S., Erisman, J.W., Otjes, R., Hensen A., Mosquera, J., Cellier, P., Loubet, B., David, M.,  
835 Genermont, S., Neftel, A., Blatter, A., Hermann, B., Jones, S.K., Horvath, L., Führer, E., Mantzanas, C.,  
836 Koukoura, K., Gallagher, M., Williams, P., Flynn M., and Riedo, M. (2001). Biosphere-atmosphere interactions  
837 of ammonia with grasslands: Experimental strategy and results from a new European initiative. *Plant Soil* 228,  
838 131–145.
- 839 Sutton, M.A., Nemitz, E., Erisman, J.W., Beier, C., Butterbach-Bahl, K. and Cellier, P. (2007). Challenges in  
840 quantifying biosphere-atmosphere exchange of nitrogen species. *Environmental Pollution* 150, 125–139.
- 841 Sutton, M. A., Erisman, J. W., Dentener, F. and Möller, D. (2008). Ammonia in the environment: From ancient  
842 times to the present. *Envir. Poll.* 156, 583–604.
- 843 Sutton, M.A., Howard, C.M., Erisman, J.W., Billen, G., Bleeker, A., Grennfelt, P., van Grinsven, H. and  
844 Grizzetti B., (eds.) (2011). *The European Nitrogen Assessment*, Cambridge University Press. New York  
845 2011, ISBN 978-1-107-00612-6. [www.nine-esf.org/ENA](http://www.nine-esf.org/ENA)
- 846 Vitousek, P.M., Aber, J.D., Howarth, R.W., Likens, G.E., Matson, P.A., Schindler, D.W., Schlesinger, W.H. and  
847 Tilman, D.G. (1997). Human alteration of the global nitrogen cycle: sources and consequences. *Ecol. Appl.*  
848 7, 737–750.
- 849 Waldrip, H.M., Todd R.W. and Cole, N.A. (2003). Prediction of nitrogen excretion by beef cattle: A meta-  
850 analysis. *J. Anim. Sci.* 91, 4290–4302.
- 851 Watt, S.A., Wagner-Riddle, C., Edwards, G. and Vet, R.J. (2004) Evaluating a flux-gradient approach for flux  
852 and deposition velocity of nitrogen dioxide over short-grass surfaces. *Atmos. Environ.* 38, 2619–2626.
- 853 Wesely, M.L. and Hicks, B.B. (2000). A review of the current status of knowledge on dry deposition. *Atmos.*  
854 *Environ.* 34, 2261–2282.
- 855 Yamulki, S., Harrison, R.M., Goulding, K.W.T. and Webster, C.P. (1997). N<sub>2</sub>O, NO and NO<sub>2</sub> fluxes from a  
856 grassland: effect of soil pH. *Soil Biol. Biochem.* 29, 1199–1208.
- 857 Zhang, L., Wright, L.P. and Asman, W.A.H. (2010). Bi-directional air-surface exchange of atmospheric  
858 ammonia: A review of measurements and a development of a big-leaf model for applications in regional-  
859 scale air-quality models. *J. Geophys. Res.* D20310.
- 860 Xu-Ri., Wang, M. and Wang, Y. (2003). Using a modified DNDC model to estimate N<sub>2</sub>O fluxes from a semi-  
861 arid grassland in China. *Soil Biol. Biochem.* 35, 615–620.
- 862
- 863 Internet sources:
- 864 Cost 729 Action: <http://www.cost729.ceh.ac.uk> (Assessing and managing nitrogen fluxes in the atmosphere-  
865 biosphere system in Europe)
- 866 INI: <http://www.initrogen.org> (International Nitrogen Initiative, Optimizing nitrogen use in food and energy  
867 production and minimizing the consequent harm to humans and the environment)
- 868 NinE: <http://www.nine-esf.org> (Nitrogen in Europe, current problem and future solution)
- 869 NitroEurope: <http://www.nitroeuropa.eu>
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Table 1: Summary of the sensitivity test of DNDC

Parameters	Fluxes			
	N <sub>2</sub> O	NO	N <sub>2</sub>	NH <sub>3</sub>
	(kg N ha <sup>-1</sup> year <sup>-1</sup> )			
Original run	0.76	0.96	0.10	0.69
+1 °C	0.44	1.05	0.05	0.74
+2 °C	0.08	1.13	0.01	0.79
-1 °C	1.30	0.87	1.32	0.63
-2 °C	1.35	0.79	1.07	0.59
125% precipitation	0.83	1.10	0.10	0.73
75% precipitation	0.63	0.84	0.10	0.63
+10% SOC	0.95	1.04	0.14	0.69
-10% SOC	0.62	0.88	0.08	0.69
+10% clay content	0.68	0.98	0.08	0.69
-10% clay content	0.89	0.94	0.13	0.68
+ 0.5 pH	0.51	0.84	0.06	0.88
- 0.5 pH	0.96	1.04	0.14	0.58
Microbial activity index = 0.5	0.27	0.67	0.02	0.61
Without grazing	0.78	0.90	0.10	0.17

Table 2: Yearly average data for inferential estimation of dry fluxes

<b>N-form</b>	<b>Average concentration</b> ( $\mu\text{g N m}^{-3}$ )	<b>Mean deposition velocity</b> ( $\text{mm s}^{-1}$ )
HNO <sub>3</sub>	0.26	27 <sup>a</sup>
NO <sub>2</sub>	1.77	1.35 <sup>b</sup>
NO <sub>3</sub> <sup>-</sup>	0.52	5.0 <sup>c</sup>
NH <sub>4</sub> <sup>+</sup>	1.07	0.87 <sup>d</sup>
NH <sub>3</sub>	1.29	9.9 <sup>a</sup>

<sup>a</sup>Horváth et al. (1992; 2005)

<sup>b</sup>Hesterberg et al. (1996); Horváth et al. (2005); Marner and Harrison (2004); Yamulki et al. (1997); Watt et al. (2004)

<sup>c</sup>Borrell et al. (1997)

<sup>d</sup>Gallagher et al. (2002)

Table 3: Annual sum of N-exchange between the biosphere and the atmosphere and the main physical parameters

	2006	2007	2008	2009	2010	mean
<b>Exchange processes</b>	(kg N ha <sup>-1</sup> year <sup>-1</sup> )					
Wet deposition of NO <sub>3</sub> <sup>-</sup>	-2.26±0.11	-2.04±0.10	-2.19±0.11	-2.15±0.11	-2.34±0.12	-2.19±0.11
Wet deposition of NH <sub>4</sub> <sup>+</sup>	-1.12±0.06	-2.17±0.11	-3.01±0.15	-2.35±0.12	-3.43±0.17	-2.42±0.88
Total wet deposition	-3.38±0.17	-4.21±0.21	-5.20±0.26	-4.50±0.23	-5.77±0.29	-4.61±0.92
Dry deposition of HNO <sub>3</sub>	-1.57±0.16	-1.73±0.17	-2.64±0.26	-2.35±0.24	-2.95±0.29	-2.25±0.59
Dry deposition of NH <sub>3</sub>	-4.18±0.21	-4.54±0.23	-5.32±0.27	-4.27±0.21	-3.79±0.19	-4.42±0.57
Dry deposition of NO <sub>3</sub> <sup>-</sup>	-1.01±0.05	-0.62±0.02	-0.83±0.04	-0.86±0.04	-0.72±0.04	-0.81±0.15
Dry deposition of NH <sub>4</sub> <sup>+</sup>	-0.35±0.02	-0.23±0.01	-0.27±0.01	-0.30±0.02	-0.32±0.02	-0.29±0.05
Dry deposition of NO <sub>2</sub>	-0.75±0.04	-0.45±0.02	-0.80±0.04	-0.79±0.04	-0.95±0.05	-0.75±0.18
Total dry deposition	-7.86±0.37	-7.57±0.35	-9.86±0.62	-8.57±0.55	-8.73±0.59	-8.52±0.89
Total deposition	-11.2±0.54	-11.8±0.56	-15.1±0.88	-13.1±0.78	-14.5±0.88	-13.1±1.68
Soil emission of N <sub>2</sub> O	0.18±0.02	0.15±0.02	0.57±0.06	0.63±0.06	1.80±0.18	0.67±0.07
Soil emission of NO	1.63±0.16	0.79±0.08	1.19±0.12 <sup>a</sup>	1.18±0.12	0.78±0.08	1.11±0.35
Total soil emission (NO+N <sub>2</sub> O)	1.81±0.18	0.94±0.10	1.76±0.18	1.81±0.18	2.58±0.26	1.78±0.58
Total soil emission <sup>c</sup> (NO+N <sub>2</sub> O+NH <sub>3</sub> +N <sub>2</sub> )	2.21	1.48	1.99	1.67	2.91	2.08
<b>Net flux (surface-atmosphere)</b>	<b>-9.39±0.72</b>	<b>-10.9±0.66</b>	<b>-13.3±1.1<sup>b</sup></b>	<b>-11.3±0.96</b>	<b>-11.9±1.14</b>	<b>-11.3±1.4</b>
<b>Net flux<sup>c</sup> (surface-atmosphere)</b>	<b>-9.66</b>	<b>-11.7</b>	<b>-14.0</b>	<b>-12.7</b>	<b>-11.6</b>	<b>-11.9</b>
Biological nitrogen fixation	-2.9±1.0	-2.9±1.0	-2.9±1.0	-2.9±1.0	-2.9±1.0	-2.9±1.0
Output by grazing	22±9.0	22±9.0	22±9.0	22±9.0	22±9.0	22±9.0
Input by excreta	-16±5.0	-16±5.0	-16±5.0	-16±5.0	-16±5.0	-16±5.0
<b>Net balance (inc. BNF and grazing)<sup>c</sup></b>	<b>-6.6±9.0</b>	<b>-8.6±9.0</b>	<b>-10.9±9.0</b>	<b>-9.6±9.0</b>	<b>-8.5±9.0</b>	<b>-8.8±9.0</b>
<b>Physical parameters</b>	<b>2006</b>	<b>2007</b>	<b>2008</b>	<b>2009</b>	<b>2010</b>	<b>mean</b>
Temperature $T_{\text{air}}$ (°C)	10.1	11.1	11.0	11.2	11.0	10.9±0.4
Temperature $T_{\text{soil}}$ (°C)	10.2	11.5	11.1	11.4	10.9	11±0.5
Precipitation $p$ (mm)	524	446	467	486	967	578±219
Soil wetness $WFPS$ (%)	34.6±12	33.6±17	30.2±14	27.9±9.7	43±12	33.9±5.7

<sup>a</sup>No data for technical reason, calculated from the mean NO emission for 2006–2007 and 2009

<sup>b</sup>Estimated from the average of 2006–2010

<sup>c</sup>Soil flux estimated by DNDC modeling

Table 4: Measured and simulated C and N content in plant and soil and yearly grazed C and N estimated by the model

Year	C/N ratio in soil <sup>a</sup>		C/N ratio in plant <sup>b</sup>		C in plant	N in plant	Grazed C	Grazed N
	measured	simulated	measured	simulated				
	(%)				(kg ha <sup>-1</sup> year <sup>-1</sup> )			
2006	13.9	12.5	20.7	19.3	522	27	398	22
2007	13.0	12.5	21.2	19.1	458	24	302	17
2008	12.5	12.6	21.1	18.9	624	33	405	23
2009	14.9	12.6	20.5	18.9	435	23	270	15
2010	17.3	12.6	19.5	19.0	805	42	487	27
Average	14.3	12.6	20.6	19.0	568	30	372	21
SD	1.80	0.10	0.80	0.20	151	7.8	87	5.0

<sup>a</sup>average of 0–15 and 15–30 cm depths

<sup>b</sup>average of the below and above ground biomass

Table 5: Comparison of the measured and simulated (DNDC) soil N-fluxes

Year	I <sub>2</sub> O		NO		N <sub>2</sub>	NH <sub>3</sub>	Total
	measured	simulated	measured	simulated	simulated	simulated	simulated
	(kg N ha <sup>-1</sup> year <sup>-1</sup> )						
2002	0.86	0.48	–	0.76	0.41	0.55	2.20
2003	0.80	0.64	–	0.50	0.45	0.37	1.96
2004	0.74	1.00	–	0.62	0.38	0.45	2.45
2005	–	0.43	–	0.67	0.31	0.43	1.84
2006	0.18	0.38	1.63	1.12	0.37	0.36	2.21
2007	0.15	0.23	0.79	0.57	0.39	0.29	1.48
2008	0.57	0.52	0.42	0.59	0.46	0.42	1.99
2009	0.63	0.24	1.18	0.47	0.68	0.28	1.67
2010	1.80	1.04	0.78	0.70	0.66	0.50	2.91
Average	0.72	0.55	0.96	0.67	0.46	0.41	2.08
SD	0.51	0.29	0.46	0.19	0.13	0.89	0.43

Fig. 1: Distribution of organic and inorganic nitrogen in the precipitation

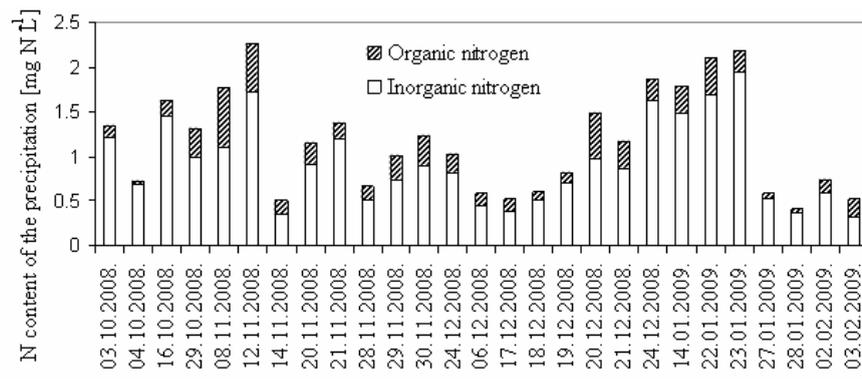


Fig. 2: Measured and simulated flux of N<sub>2</sub>O (a) and their monthly correlation (b)

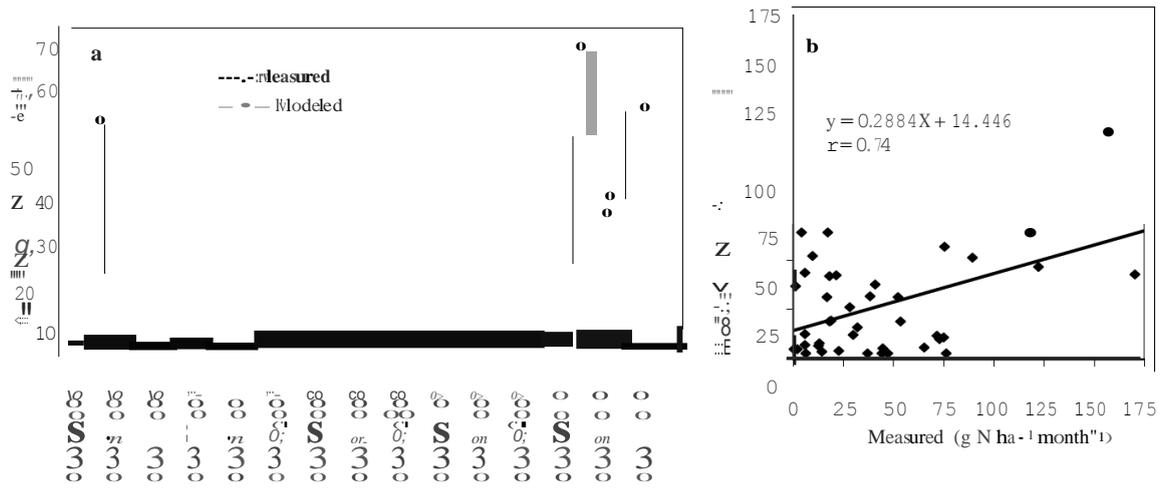


Fig. 3: Measured and simulated flux of NO (a) and their monthly correlation (b)

