

Article (refereed) – Published version

Painter, Stuart C.; Patey, Matthew D.; Forryan, Alexander; Torres-Valdes, Sinhue. 2013 Evaluating the balance between vertical diffusive nitrate supply and nitrogen fixation with reference to nitrate uptake in the eastern subtropical North Atlantic Ocean. *Journal of Geophysical Research: Oceans*, 118 (10). 5732-5749.
[10.1002/jgrc.20416](https://doi.org/10.1002/jgrc.20416)

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Evaluating the balance between vertical diffusive nitrate supply and nitrogen fixation with reference to nitrate uptake in the eastern subtropical North Atlantic Ocean

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Received 17 June 2013; revised 16 August 2013; accepted 25 September 2013; published 22 October 2013.

[1] The balance between N_2 fixation and diffusive NO_3^- supply is a key determinant for assessing the importance of both processes for new production in subtropical waters. Here we report observations of integrated N_2 fixation rates from the eastern subtropical North Atlantic Ocean with coincident estimates of diffusive NO_3^- supply. We find the average rate of N_2 fixation is equivalent to 62% of the diffusive NO_3^- supply, though N_2 fixation could exceed the diffusive flux at individual stations. Turbulent diffusivity measurements across the nitracline indicate a mean diffusivity of $0.077 \text{ cm}^2 \text{ s}^{-1}$. If approximations for methodological underestimates in the dominant N_2 fixation technique are considered, the magnitude of N_2 fixation is shown to represent 100% of the NO_3^- flux on average, and can be almost threefold higher at individual stations. As the study site is characterized by low rates of N_2 fixation compared to other sectors of the North Atlantic this confirms N_2 fixation as a major source term across the subtropical North Atlantic. The seasonal context of our observations suggests environmental factors underlie the in situ variability in observed N_2 fixation rates, and may well explain lower previous assessments of the importance of N_2 fixation relative to diffusive NO_3^- supply in this region. The diffusive NO_3^- supply provides <20% of measurable NO_3^- uptake with the remainder supplied via other mechanisms, most notably nitrification. The mean integrated rate of N_2 fixation equates to just 8% of the NO_3^- consumed on a daily basis by the phytoplankton community.

Citation: Painter, S. C., M. D. Patey, A. Forryan, and S. Torres-Valdes (2013), Evaluating the balance between vertical diffusive nitrate supply and nitrogen fixation with reference to nitrate uptake in the eastern subtropical North Atlantic Ocean, *J. Geophys. Res. Oceans*, 118, 5732–5749, doi:10.1002/jgrc.20416.

1. Introduction

[2] The supply of externally sourced nutrients to the surface waters of subtropical oligotrophic gyres is a key control on overall levels of new production within such ecosystems. Historically, the presence of persistent water column stratification and the absence of strong convective mixing focused attention upon the vertical diffusive flux of NO_3^- as the primary pathway by which NO_3^- (and hence new nutrients) reached the surface ocean. The validity of this statement appeared confirmed when *Lewis et al.* [1986]

found a general agreement (to within an order of magnitude) between rates of integrated NO_3^- uptake and of diffusive NO_3^- supply across the nitracline. In the period since this important study was published it has been recognized that the diffusive flux of nutrients is augmented by other sporadic inputs such as those provided by mesoscale eddies, internal waves, and atmospheric deposition (reviewed by *Lipschultz et al.* [2002]). The identification and quantification of these multiple nutrient supply pathways remains a major research objective, particularly given existing imbalances between rates of production that can be supported by directly measured nutrient supply terms and the higher rate of production indirectly inferred by geochemical methods [*Jenkins*, 1982].

[3] Alongside developments in our understanding and quantification of the physical mechanisms that supply nutrients to the surface ocean has been improved recognition of the role of diazotrophy in injecting new nitrogen into the surface ocean. N_2 fixation has now moved from being viewed as a relatively unimportant process [*Dugdale and Goering*, 1967] to one that occupies a central role in our conceptual view of nutrient cycling within the ocean [*Gruber*, 2008]. In particular, the importance of diazotrophic organisms for new and export production is increasingly recognized [*Karl et al.*, 1997, 2002, 2012; *Capone et al.*, 2005;

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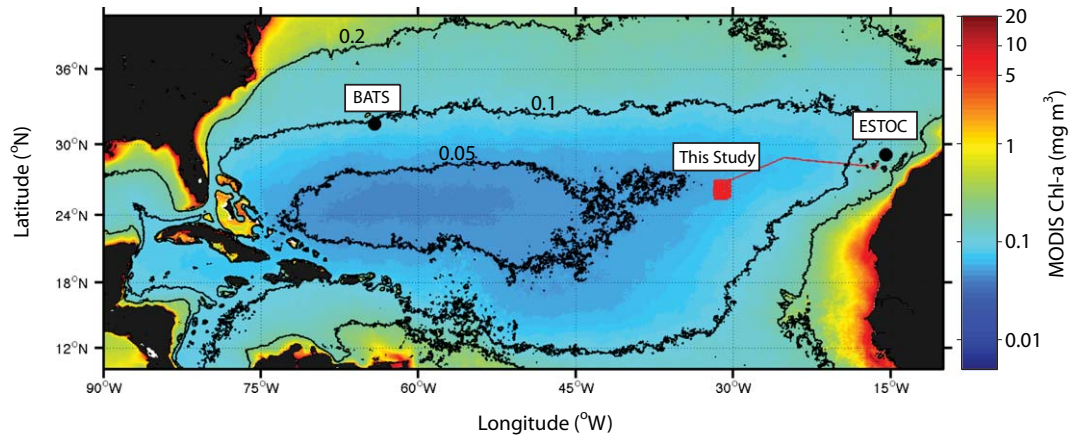


Figure 1. Cruise transect and survey region (red square) shown in relation to the wider subtropical North Atlantic and established time-series study sites including Bermuda Atlantic Time Series (BATS) and the European Station for Time Series in the Ocean (ESTOC). The underlying surface chlorophyll image is the MODIS Aqua mission average for the period 2002–2011. Selected contours (as labeled) are presented to show the long-term mean position of the North Atlantic subtropical gyre.

Carpenter and Capone, 2008; Karl and Letelier, 2008]. At the global scale active debate over the balance between N₂ fixation and denitrification, the major oceanic loss term for fixed N, suggests that the marine N cycle is largely balanced [Gruber and Sarmiento, 1997; Capone et al., 2005; Deutsch et al., 2007; Gruber, 2008; DeVries et al., 2012]. However, the relevance of regional variability in N₂ fixation rates is still not properly understood.

[4] Areal estimates of N₂ fixation rates have been shown to rival or exceed in magnitude the diffusive flux of NO₃[−] [Capone et al., 2005; Mahaffey et al., 2005]. The balance between these two supply processes, however, is spatially and temporally variable. Consequently, direct measurement of both processes is required to validate such conclusions and to identify regions of the ocean where N₂ fixation may be quantitatively more important. Surprisingly, it appears that only recently have the rates of diffusive NO₃[−] supply and of N₂ fixation been compared simultaneously [Mourino-Carballido et al., 2011]. Previous comparisons, particularly for basin scale syntheses, have often been made on the basis of an assumed turbulent diffusivity (K_z), such as that obtained by Ledwell et al. [1993] (i.e., K_z = 0.11 cm² s^{−1}), which is assumed to be broadly constant in space and time. Nevertheless, even with this assumption in place areal rates of N₂ fixation have been shown to represent significant inputs of new nitrogen, which suggests that unless diffusivities are considerably more heterogeneous than currently thought the general conclusion that N₂ fixation can exceed diffusive NO₃[−] supply is indeed valid in certain cases. However, limited understanding of the spatiotemporal variability in N₂ fixation rates away from the tropical belt of maximum N₂ fixation means that it is unclear if the dominance of N₂ fixation over the diffusive NO₃[−] supply can be considered typical.

[5] Despite general conclusions regarding the importance of N₂ fixation, Mourino-Carballido et al. [2011] found that N₂ fixation represented between 2% and 44% of the diffusive NO₃[−] flux in the subtropical North and South Atlantic, respectively. The low contribution for the North Atlantic seems to be at odds with the broader distribution

and importance of diazotrophy across the North Atlantic [Capone et al., 2005; Carpenter and Capone, 2008] and its implied impact upon the distribution of N* [Gruber and Sarmiento, 1997; but see also Singh et al., 2013]. This may indicate an important spatial decoupling between the two supply processes, indicate some facet of regional or seasonal variability that we poorly understand or possibly indicate that the true significance of N₂ fixation for new production is equally poorly understood [Singh et al., 2013].

[6] The eastern subtropical North Atlantic is characterized by relatively low rates of N₂ fixation compared to the western North Atlantic and equatorial regions [Capone et al., 2005; Carpenter and Capone, 2008; Subramaniam et al., 2013] and detailed assessments of the importance of N₂ fixation versus diffusive NO₃[−] supply for overall productivity levels are lacking. Here we address these omissions and present measurements of primary production, NO₃[−] uptake, and N₂ fixation from the eastern subtropical North Atlantic along with simultaneous estimates of the vertical diffusive NO₃[−] supply to critically evaluate the importance of N₂ fixation relative to the NO₃[−] supply.

2. Methods

2.1. Cruise Overview and Environmental Characterization

[7] All data were collected between 11 August and 12 September 2011 at a nominal location of 26.5°N, 29.5°W (Figure 1). All CTD casts were located within a 160 km radius around this position and conducted during repeated mesoscale surveys of the area. Water sample collection and environmental characterization were performed using a Seabird 9/11+ CTD-Niskin rosette package. Mixed layer depths were estimated using a density threshold criterion of 0.03 kg m^{−3} relative to the density at 10 m [de Boyer Montegut et al., 2004]. Mixed layer depths were generally deeper (20–60 m) between 11 August and 25 August than they were subsequently (<30 m), and the cruise mean mixed layer depth was 28 ± 15 m.

[8] The depth of the euphotic zone (0.1% surface irradiance) was derived from knowledge of the water column attenuation coefficient (K_d), which was estimated from a regression of log normalized irradiance intensity (PAR; photosynthetically active radiation) against depth. PAR data were collected from a CTG 2 π PAR sensor fitted to the CTD frame and only casts conducted around local noon (± 3 h) were used in this assessment, which resulted in an average K_d for the cruise of 0.044 ± 0.002 (range 0.041–0.048) and produced a cruise average euphotic zone depth of 158 ± 7 m; comparable to the deepest chlorophyll maximum (DCM) we observed.

2.2. Biological and Chemical Variables

[9] Discrete chlorophyll concentrations were estimated fluorometrically from 250 ml or 500 ml seawater samples filtered onto 25 mm Whatman GF/F (0.7 μ m pore size) glass fiber filters and extracted in 90% acetone at 4°C overnight (16–20 h). Chlorophyll extracts were measured following the method of *Welschmeyer* [1994] using a Turner Trilogy fluorometer calibrated against a pure chlorophyll-*a* standard (spinach extract—Sigma Aldrich). CTD fluorescence profiles were calibrated against the discrete chlorophyll samples using linear regression analysis. The calibration equation $y = 2.5766x - 0.0124$ ($R^2 = 0.93$, $n = 196$), where y is the calibrated fluorescence profile (mg m^{-3}) and x is the measured chlorophyll concentration (mg m^{-3}), was applied to all fluorescence profiles.

[10] Water samples for the determination of total nitrate ($NO_3^- + NO_2^-$, hereafter NO_3^-), phosphate (PO_4^{3-}) and orthosilicic acid ($Si(OH)_4$) concentration were drawn directly from CTD Niskin bottles into clear polystyrene vials and analyzed immediately or stored in the dark at 4°C whilst awaiting analysis. All concentrations were determined colorimetrically using a 3 channel Skalar Sanplus autoanalyser following the methods described by *Kirkwood* [1996]. Detection limits of better than 0.1 μ mol L^{-1} for NO_3^- and $Si(OH)_4$ and 0.01 μ mol L^{-1} for PO_4^{3-} were achieved throughout. As these detection limits were insufficient for the oligotrophic waters of the study site, we also measured NO_3^- and PO_4^{3-} concentrations at nanomolar levels using the liquid capillary waveguide methodology described by *Patey et al.* [2008, 2010]. This provided a limit of detection of around 0.5 nmol L^{-1} for both NO_3^- and PO_4^{3-} .

2.3. ^{13}C , $^{15}NO_3^-$, and $^{15}N_2$ Fixation Rate Measurements

[11] Water samples were collected from five depths across the euphotic zone for the measurement of primary production, NO_3^- uptake, and N_2 fixation. Sampling depths were chosen based on irradiance levels and represented 97, 55, 33, and 14% of surface irradiance intensity with an additional sample from the depth of the DCM ($< 1\%$ PAR). For the measurement of primary production and NO_3^- uptake, duplicate 2 L water samples from each depth were carefully measured into clean Nalgene polycarbonate bottles and dual labeled with $Na^{13}CO_3^-$ and $K^{15}NO_3^-$. The ^{13}C label (99% enriched, CIL Laboratories) was added at a fixed concentration of 105 μ mol L^{-1} , equating to 5% of the available dissolved inorganic carbon (DIC) pool (~ 2100 μ mol L^{-1}), whereas the $^{15}NO_3^-$ label (99.5% enriched)

was added at a variable concentration of between 1 and 5 nmol L^{-1} to take into account both spatial and vertical changes in the ambient NO_3^- concentration. These additions represent on average $16 \pm 9\%$ of the ambient NO_3^- concentration but in individual bottles the tracer addition varied from < 1 to 50% of the actual NO_3^- concentration; thus whilst our results broadly reflect natural conditions they are a combination of trace and saturated uptake rates.

[12] Samples from the 97–14% irradiance depths were incubated using on-deck incubators shielded to appropriate irradiance levels with optical filters (210 0.6 Neutral Density and 061 Mist Blue; Lee Filters) and flushed with running seawater collected from ~ 5 m depth ($\sim 24^\circ C$). Bottles from the deepest sampling depths were incubated using a FytoScope FS130 plant growth chamber (Photon Systems Instruments, www.psi.cz), which allowed precise temperature and irradiance control (cool white LEDs). The temperature within the growth chamber was set to $17^\circ C$ reflective of in situ temperatures, whilst the irradiance level was set to ~ 7 μ mol photons $m^{-2} s^{-1}$ which was previously determined to approximate peak daily irradiance intensities at the depth of the DCM. Confirmation of this approximation was gained from a retrospective analysis of ship measured total daily incident PAR, which ranged from 22.5 to 60 mols photons $m^{-2} d^{-1}$ and averaged 50.6 ± 10.6 mols photons $m^{-2} d^{-1}$ over the period of the cruise. The daily irradiance cycle at the DCM was estimated following *Kirk* [2010] using a transmission coefficient of 0.98 and the cruise average attenuation coefficient of $0.044 m^{-1}$. Calculated in this way, the average daily maximum irradiance intensity at the DCM was 6.8 ± 1.2 (range 3.2–8) μ mol photons $m^{-2} s^{-1}$ (Figure 2), whereas the daily average PAR experienced by the DCM was 1.9 ± 0.4 μ mol photons $m^{-2} s^{-1}$ (Figure 2). The daily integrated PAR received by the DCM averaged 0.17 ± 0.04 (range 0.08–0.2) mols photons $m^{-2} d^{-1}$.

[13] All samples were incubated for 4–6 h before gentle filtration onto ashed (> 6 h $450^\circ C$) 25 mm Whatman GF/F filters (0.7 μ m pore size). Filters were stored frozen at $-20^\circ C$ onboard. Once back in the laboratory all filters were dried overnight at $40^\circ C$, pelleted into tin capsules using a laboratory press and analyzed by mass spectrometry for particulate and isotopic N and C content. Uptake rates (units nmol $L^{-1} h^{-1}$) were calculated using the equations of *Dugdale and Wilkerson* [1986]. ^{13}C uptake rates were calculated assuming an ambient DIC concentration of 2100 μ mol C L^{-1} ($= 25.2$ mg C L^{-1}) and using a ^{13}C natural abundance of 1.0790%, which was the mean value obtained from the isotopic analysis of independent samples collected for measurements of particulate organic carbon (POC)/particulate organic nitrogen (PON) concentration and T_0 initial isotopic conditions as part of the N_2 fixation experiments.

[14] N_2 fixation rates were measured using the method described by *Montoya et al.* [1996] for both total and < 8 μ m size fractions (samples were prefiltered). The < 8 μ m size fraction was included to provide an indication of diazotrophy by the unicellular component of the plankton community. At each sampled depth and for each size fraction, a 4.2 L polycarbonate bottle was filled to the brim and closed with a gas tight septum cap taking care to remove all air bubbles first. Once sealed, 4 ml of 99% $^{15}N_2$ gas were injected through the septum and the bottle

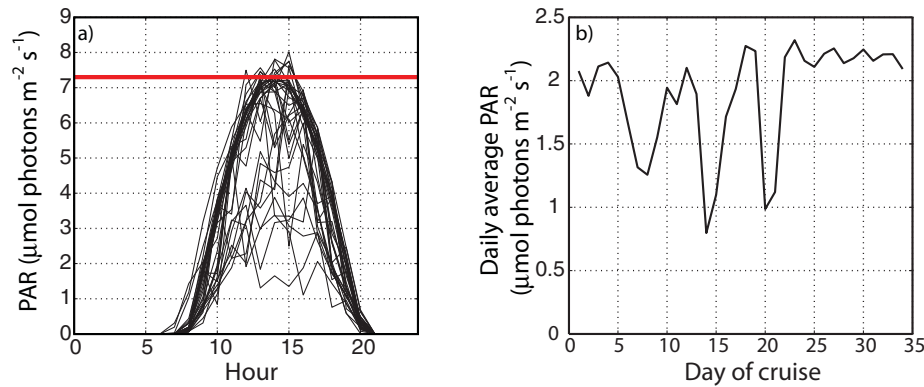


Figure 2. (a) Calculated daily irradiance cycles for the deep chlorophyll maximum where peak daily irradiance intensity averaged $6.8 \pm 1.2 \mu\text{mol photons m}^{-2} \text{s}^{-1}$. The red line indicates the irradiance intensity setting of the Fytoscope incubator ($\sim 7 \mu\text{mol photons m}^{-2} \text{s}^{-1}$) used in this study for samples collected from the deep chlorophyll maximum. (b) Daily average irradiance at the deep chlorophyll maximum with a cruise average of $1.9 \pm 0.4 \mu\text{mol photons m}^{-2} \text{s}^{-1}$.

rolled from side-to-side for several minutes to aid equilibration of the tracer with the dissolved N₂ gas content of the sample. All bottles were incubated for 24 h using the same incubators as described above, though incubators were covered/LED's turned off at night to protect samples from the ship's lights. All samples were filtered under gentle vacuum pressure onto ashed 25 mm Whatman GF/F filters, which were then stored frozen at -20°C for return to the laboratory where filters were dried and pelleted for analysis. N₂ fixation rates ($\text{nmol L}^{-1} \text{d}^{-1}$) were calculated using the mass balance equations detailed in Montoya *et al.* [1996].

[15] To provide daily integrals of carbon fixation and of NO₃[−] uptake from our subdaily incubations to compare to the daily rates of N₂ fixation, we have scaled the hourly carbon uptake rates by the standard solar day length (12 h) and hourly NO₃[−] uptake rates by 18 h to account for dark NO₃[−] uptake, which may be significant [Mulholland and Lomas, 2008]. In doing this, we have assumed that dark NO₃[−] uptake occurs at a rate equivalent to 50% of that observed in the daylight [Mulholland and Lomas, 2008].

2.4. Turbulence Measurements and Diffusive Nitrate Fluxes

[16] Turbulent kinetic energy dissipation (ϵ) was measured at 13 stations during the cruise from microstructure velocity shear measurements collected using an MSS90L free-fall microstructure shear profiler produced by Sea and Sun Technology GmbH and ISW Wassermesstechnik using the methodology described by Forryan *et al.* [2012, 2013]. Briefly, turbulent diffusivity is related to the dissipation of turbulent kinetic energy [Osborn, 1980; Moum *et al.*, 1995; Rippeth *et al.*, 2003] and under conditions of isotropic turbulence [Yamazaki and Osborn, 1990] the rate of turbulent kinetic energy dissipation can be calculated from the variance of the measured vertical microstructure shear by integration of the vertical microstructure shear power spectrum [Moum *et al.*, 1995; Rippeth *et al.*, 2003; Stips, 2005]. The microstructure shear power spectrum was estimated using the modified periodogram method [Welch, 1967] from the vertical microstructure shear fluctuations. This power spectrum was then used to calculate the rate of

turbulent diffusivity (K_z) as described by Forryan *et al.* [2012, 2013].

[17] Due to the limited number of stations at which turbulent diffusivity was measured (13) we have calculated the geometric mean diffusivity profile that we use to estimate diffusive NO₃[−] fluxes. This was achieved by first log transforming the diffusivity data prior to averaging. We then subsequently averaged the diffusivity measurements between 90 and 160 m depth (i.e., across the nitracline) to obtain the diffusivity term K_z . Vertical diffusive NO₃[−] fluxes were calculated at the top of the nitracline using a concentration of $100 \text{ nmol NO}_3^- \text{ L}^{-1}$ to signify this position and at the depth of the maximum nitracline gradient using a common methodology. In both cases, the local NO₃[−] gradient ($\frac{\partial \text{NO}_3}{\partial z}$) was obtained via linear interpolation, and combined with the mean diffusivity term (K_z) via the equation ($K_z \frac{\partial \text{NO}_3}{\partial z}$) to produce diffusive flux estimates.

3. Results

3.1. Environmental Context

[18] We present in Figure 3 a time series of observations for the upper 200 m of the water column under the assumption that spatial differences due to the geographic positioning of individual CTD casts is less influential than the regional temporal trend. The hydrographic conditions characterized by the contour plots of temperature, salinity, and density reveal that near surface waters ($<30 \text{ m}$) increased in temperature by $\sim 2.5^\circ\text{C}$ from $\sim 23.1^\circ\text{C}$ to $>25.7^\circ$, surface salinity increased by 0.35 from 37.24 to 37.59 and there was a coincident decrease in surface density of 0.58 kg m^{-3} from 25.59 to 25.01 kg m^{-3} over the period of observation. Below 50 m depth there was less temporal variability but a number of shallower isotherms, isohalines and isopycnals progressively deepen suggesting that the changes we observe slowly spread throughout the euphotic zone. Seasonal heating may indeed play a role in some of these near surface changes (see section 3.5) but the magnitude of change is often far larger than would be expected by this process alone thus the most likely explanation must also involve a degree of spatial variability. Within the

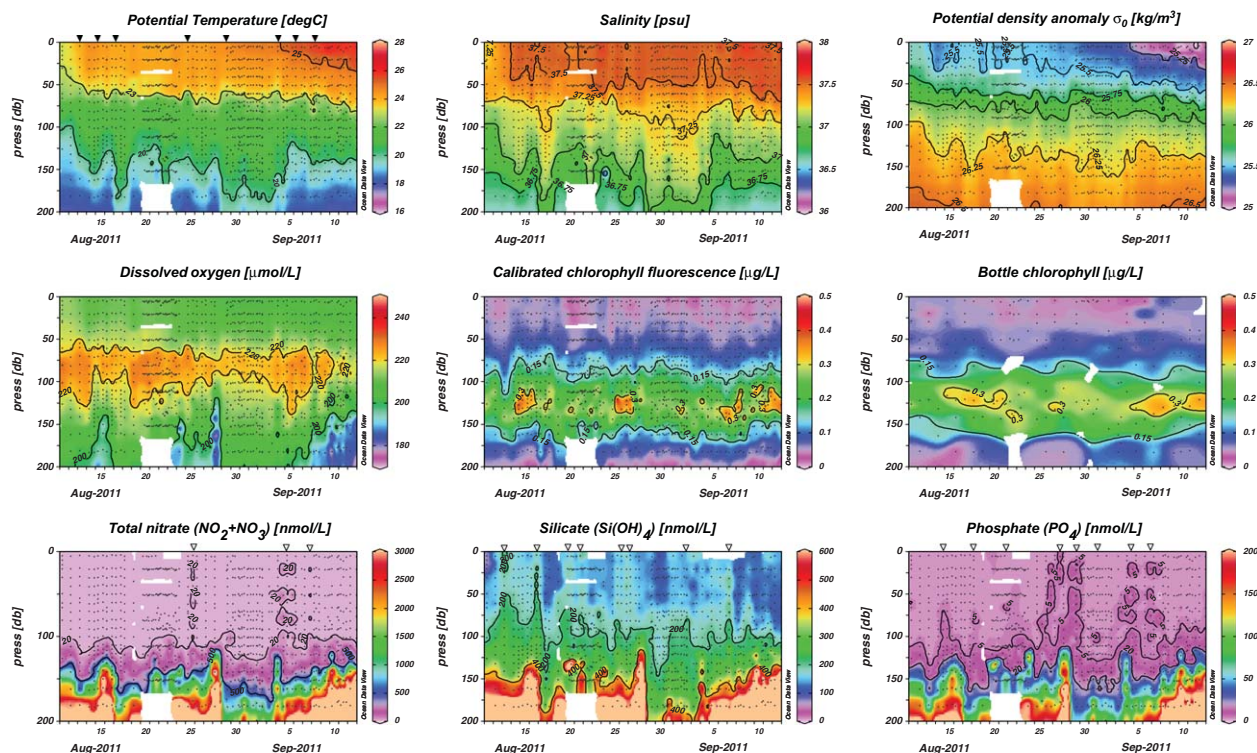


Figure 3. Time-series plots of environmental conditions during the cruise (a) potential temperature, (b) salinity, (c) density, (d) dissolved oxygen, (e) calibrated chlorophyll fluorescence, (f) bottle chlorophyll, (g) total nitrate, (h) orthosilicic acid (“silicate”), and (i) phosphate. Note that the hydrographic data are based upon a reduced data set of bottle stop observations so as to present a similar vertical resolution as the other parameters. The black triangles along the top of the potential temperature section indicate sampling periods for the rate measurements reported here. The white triangles along the top of the nutrient sections indicate nutrient injection events.

thermocline (~ 150 m) isotherms can be seen to oscillate vertically by up to 50 m, which is most likely due to undulations within the thermocline in response to physical forcing (instabilities, internal waves, etc).

[19] Dissolved oxygen profiles, which highlight an important region of primary production, show a persistent subsurface oxygen maximum located at a mean depth of 80 ± 8 m (range 53–99 m) (Figure 3). Maximum oxygen concentrations on individual profiles varied from 226 to $241 \mu\text{mol L}^{-1}$ (mean $233 \pm 3 \mu\text{mol L}^{-1}$) and whilst peak oxygen concentrations appear to decrease with time, the subsurface oxygen maximum remained clearly visible in all individual profiles.

[20] We also present in Figure 3 calibrated fluorescence and bottle chlorophyll data. The depth of the DCM was identified from calibrated chlorophyll fluorescence profiles and found to vary in depth from 105 to 159 m with a cruise average of 129 ± 11 m, some 50 m deeper than the mean depth of the oxygen maximum [as is typical for subtropical waters Hayward, 1991, 1994]. Maximum chlorophyll concentrations at the DCM were $0.3\text{--}0.4 \mu\text{g L}^{-1}$, though there could be considerable variability in the maximum chlorophyll concentration observed between profiles indicating significant heterogeneity within the DCM. Chlorophyll concentrations $>0.15 \mu\text{g L}^{-1}$ were observed between 75 and 175 m depth and elevated relative to surface concentrations ($\sim 0.05 \mu\text{g L}^{-1}$), resulting in a rather broad feature but

at its core the DCM was a rather more abrupt feature usually <10 m thick. Despite repeated sampling only a weak and insignificant correlation between the depth of the DCM and of the 1% irradiance depth was identified. This correlation was far weaker than is widely assumed and resulted from the DCM occupying a deeper position relative to the depth of the 1% isolume at most stations. Using the average attenuation coefficient for the cruise (0.044 m^{-1}), we calculate that the DCM was located at a mean irradiance intensity of 0.39% of surface irradiance and comparable to the 0.26% level identified by Venrick *et al.* [1973] in the North Pacific.

[21] Nutrient concentrations are also shown in Figure 3. NO_3^- concentrations were consistently at nanomolar levels ($<10 \text{ nmol L}^{-1}$) throughout the upper 100 m of the water column (mean concentration $7.1 \pm 1.8 \text{ nmol L}^{-1}$) and increased below 100 m depth signifying the position of the nitracline. The nitracline, defined here by a concentration of 100 nmol L^{-1} , was located at a mean depth of 129 ± 13 m but found to vary from 95 to 160 m on individual profiles. Silicate concentrations were typically $<200 \text{ nmol L}^{-1}$ in the upper 100 m of the water column (mean $164 \pm 87 \text{ nmol L}^{-1}$), but Si concentrations proved to be rather variable resulting in a patchy distribution and regions with Si concentrations $<100 \text{ nmol L}^{-1}$ could be discerned. The patchy nature of surface silicate concentrations is reflected in the position of deeper isopleths, which are shoaler when

Table 1. Summary of Station Positions, Integration Depths, and Integrated (Trapezoidal Method) State Variables and Biological Uptake Rates

Station	Date	Latitude (N)	Longitude (W)	Integration Depth (m)	Integrated chl- <i>a</i> (mg m ⁻²)	Integrated NO ₃ ⁻ (mmol NO ₃ ⁻ m ⁻²)	Integrated Production (mmol C m ⁻² d ⁻¹)	Integrated ρNO ₃ ⁻ (mmol N m ⁻² d ⁻¹)	Total Integrated N ₂ Fixation (μmol N m ⁻² d ⁻¹)	<8 μm Integrated N ₂ Fixation (μmol N m ⁻² d ⁻¹)
1	13/08/11	27.0022	30.2984	121	9.27	1.00	14.86	0.13	15.35	10.29
2	15/08/11	25.7964	31.1772	116	16.86	3.15	32.47	0.57	27.74	11.07
3	17/08/11	26.4469	31.0652	120	17.63	0.72	21.66	0.38	21.98	8.36
4	25/08/11	26.9946	30.3484	144	16.81	10.49	24.57	0.77	33.13	8.99
5	29/08/11	26.6028	31.3435	123	15.11	1.48	28.42	0.62	16.78	
6	04/09/11	26.0977	31.8269	111	13.07	1.26	19.36	0.40	95.62	
7	06/09/11	26.7296	30.9711	144	20.42	17.95	22.53	0.74	20.45	
8	08/09/11	25.3992	31.2602	130	20.18	5.57	26.95	1.18	63.30	

surface concentrations are higher and are deeper when they are lower, suggesting that this variability in surface Si concentrations is related to activity within the thermocline [Bibby and Moore, 2011]. Finally, PO₄³⁻ concentrations were typically <5 nmol L⁻¹ in the upper 100 m (mean 3.2 ± 0.9 nmol L⁻¹) and a little higher than reported for the western subtropical North Atlantic (0.2–1 nmol L⁻¹) [Wu *et al.*, 2000], but increased with depth. Concentrations >10 nmol L⁻¹ were typically encountered below 120 m depth, but there is clearly some variability between stations including what may very well be a number of eutrophication events when pockets of water with PO₄³⁻ > 5 nmol L⁻¹ were identified within the upper 100 m (Figure 3). We have used an arbitrary concentration of 100 nmol PO₄³⁻ L⁻¹ to signify the position of the phosphocline, recognizing that this may overestimate the actual depth given the generally lower concentration of PO₄³⁻ relative to NO₃⁻. In this way, we obtain a mean phosphocline depth of 177 ± 27 m, roughly 40 m deeper than the corresponding nitracline but supportive of the general observation of a vertical decoupling between the nitracline and phosphocline within the Atlantic Ocean [Longhurst and Harrison, 1989].

3.2. Rate Measurements

[22] Rates of carbon fixation varied from 0.6 to 29.3 nmol C L⁻¹ h⁻¹ (0.007–0.35 mmol C m⁻³ d⁻¹) with maximum rates located in surface waters (upper 30 m) and decreasing below this. Daily integrated production (Table 1) ranged from 14.9 to 32.5 mmol C m⁻² d⁻¹, very similar to previous measurements of primary production rates in these waters [Maranon *et al.*, 2000; Poulton *et al.*, 2006; Perez *et al.*, 2006].

[23] Rates of NO₃⁻ uptake (ρNO₃⁻) varied from 0.03 to 1.17 nmol N L⁻¹ h⁻¹. In general, ρNO₃⁻ rates remained relatively uniform or decreased with depth, which is consistent with both the NO₃⁻ impoverished nature of these waters and the reduction in irradiance intensity with depth. Given the sampling resolution, there is minimal evidence for any enhancement in ρNO₃⁻ at depth around the nitracline as reported previously [Le Bouteiller, 1986; Eppley and Koeve, 1990; Harrison 1990; Painter *et al.*, 2007]. Daily-integrated ρNO₃⁻ rates ranged from 0.13 to 1.18 mmol N m⁻² d⁻¹ (Table 1) and were very comparable to previous observations from this region (see section 4).

[24] Rates of N₂ fixation were generally <0.5 nmol N L⁻¹ d⁻¹ throughout the water column with isolated enhancements of up to 2.8 nmol N L⁻¹ d⁻¹ at the shallowest sampled depths. N₂ fixation rates in the <8 μm fraction were noticeably lower peaking at just 0.24 nmol N L⁻¹ d⁻¹, and unlike the vertical profiles for the total fraction, a small yet distinct peak in N₂ fixation by the <8 μm fraction was evident between 20 and 30 m. Daily-integrated N₂ fixation rates ranged from 15.4 to 95.62 μmol N m⁻² d⁻¹ for the full size fraction and from 8.36 to 11.07 μmol N m⁻² d⁻¹ for the <8 μm fraction (Table 1). The composition of the diazotrophic community, particularly in the <8 μm fraction, is unknown but a large and exceptional bloom of *Trichodesmium* was encountered during this cruise which was clearly visible from the ship and we consider the larger diazotroph fraction to have been *Trichodesmium* dominated.

Table 2. Comparison of Turbulent Diffusive NO₃[−] Supply Calculated at the Top of the Nitracline (=100 nmol NO₃[−] L^{−1}) and at the Depth of the Maximum Nitracline Gradient With Local Integrated NO₃[−] Uptake

Station	NO ₃ [−] Uptake (μmol N m ^{−2} d ^{−1})	Nitracline Gradient (μmol m ^{−4})	Kz (cm ² s ^{−1}) = 0.077			
			NO ₃ [−] Flux (μmol m ^{−2} d ^{−1})	Flux as % of Uptake	N ₂ Fix as % of Flux	N ₂ Fix as % of ρNO ₃
<i>Top of Nitracline</i>						
1	134.3	18.76	12.48	9.3	123	11
2	574.0	12.64	8.41	1.5	330	5
3	383.3	5.48	3.65	1.0	602	6
4	773.3	16.95	11.28	1.5	294	4
5	618.1	11.12	7.40	1.2	227	3
6	399.6	44.50	29.61	7.4	323	24
7	735.5	26.01	17.30	2.4	118	3
8	1175.3	12.37	8.23	0.7	769	5
Mean ± standard deviation	599.2 ± 312.7	18.5 ± 12.1	12.3 ± 8.1	3.1 ± 3.3	348 ± 228	8 ± 7
<i>Maximum Nitracline Gradient</i>						
1	134.3	129.42	86.10	64.1	17.8	
2	574.0	98.84	65.75	11.5	42.2	
3	383.3	54.25	36.09	9.4	60.9	
4	773.3	84.14	55.98	7.2	59.2	
5	618.1	82.86	55.13	8.9	30.4	
6	399.6	94.52	62.88	15.7	152.1	
7	735.5	67.46	44.88	6.1	45.6	
8	1175.3	111.84	74.40	6.3	85.1	
Mean ± standard deviation	599.2 ± 312.7	90.4 ± 23.9	60.2 ± 15.9	16.2 ± 19.6	61.7 ± 41.9	

3.3. Vertical Nutrient Gradients

[25] All nutrient concentrations increased significantly below ~100 m depth though there were small and important vertical offsets in the depth at which each nutrient began to increase in concentration. In general, silicate concentrations begin to increase at 100 m depth, NO₃[−] concentrations between 100 and 125 m depth, and PO₄^{3−} concentrations between 125 and 150 m depth. This vertical separation between the silicicline, nitracline, and phosphocline is not precise given the subjective criteria used to identify them but such depth offsets are recognized as important for the relative availability of different nutrients at the base of the euphotic zone [Longhurst and Harrison, 1989].

[26] Gradients in NO₃[−] concentration at the top of the nitracline (the depth range over which concentrations begin to consistently increase) for all profiles measured during this cruise ranged from 3 to 190 μmol N m^{−4}, with a mean gradient of 25 ± 27 μmol N m^{−4}. These gradients are comparable to the mean gradient reported by Mourino-Carbalido *et al.* [2011] of 62 ± 90 μmol N m^{−4} and to the range of gradients reported by Painter *et al.* [2008] of 33–124 μmol N m^{−4} for this region of the North Atlantic, though our mean gradient is somewhat lower than reported previously. We suspect that this is due to differences in where the nutrient gradient is calculated and sampling resolution between studies. To demonstrate this we present in Table 2 estimates of the nitracline gradient based on measurements taken at the top of the nitracline and on the basis of the maximum gradient in a particular profile, which was usually deeper than the top of the nitracline. Gradients at the top of the nitracline for the subset of eight stations we focus on here ranged from 5.48 to 44.5 μmol m^{−4} (mean 18.5 ± 12.1 μmol m^{−4}), whereas the maximum gradient ranged from 54.25 to 129.42 μmol m^{−4} (mean 90.4 ± 23.9

μmol m^{−4}). Both estimates are entirely consistent with previous observations but the notable increase in magnitude for the maximum gradient may help explain the ambiguity in interpreting previous nutrient flux estimates due to inconsistencies in the derivation of nitracline gradients (Figure 4; see section 4).

3.4. Ambient Diffusivity and Diffusive NO₃[−] Fluxes

[27] Profiles of the Brunt-Väisälä buoyancy frequency, turbulent kinetic energy, and turbulent diffusivity from all 13 stations are presented in Figure 5 along with geometrically calculated (log transformed) cruise mean profiles. In general, the individual profiles were very similar to one another indicating minimal spatial or temporal variability. Over the depth range at which the nitracline was located (95–160 m) turbulent diffusivity within individual 8 m depth bins varied from 0.017 to 0.345 cm² s^{−1}. For an individual profile, the mean diffusivity calculated over the 95–160 m depth interval varied from 0.035 to 0.187 cm² s^{−1}. The cruise mean diffusivity over the 95–160 m depth interval was 0.077 ± 0.01 cm² s^{−1} (95% confidence interval limits 0.059–0.101 cm² s^{−1}) and comparable to existing estimates of open ocean diffusivity (e.g., 0.11 ± 0.02 cm² s^{−1}) [Ledwell *et al.*, 1993]. This mean value appears to be 32% lower than that reported by Ledwell *et al.* [1993] but as the upper 95% confidence limit (0.101 cm² s^{−1}) approaches the value obtained by Ledwell *et al.* [1993] the two values are likely indistinguishable. In addition to estimating NO₃[−] fluxes at the top of the nitracline we also define fluxes on the basis of the maximum nitracline gradient as this should provide an upper estimate of the magnitude of NO₃[−] supply via turbulent diffusion. This is an important distinction as much of the ambiguity in flux estimates between different studies can be traced back to the depth, gradient, or approach used, which can broaden the

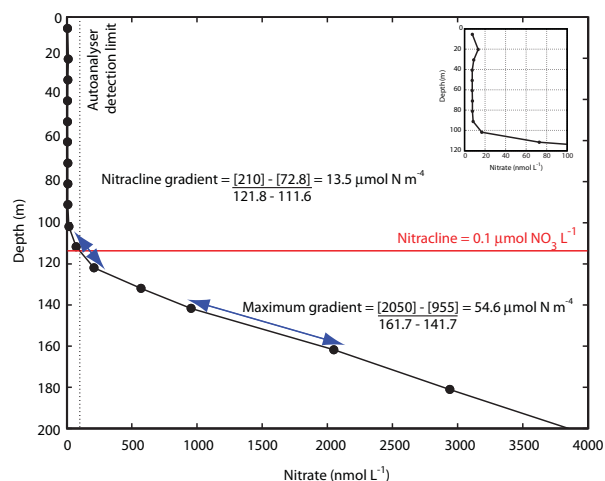


Figure 4. Schematic demonstrating the difference in nitracline gradient that may be calculated using different criteria. The gradient at the depth of the (arbitrarily) defined nitracline is almost four times smaller than the maximum gradient that may be derived from the same nutrient profile. Consequently, considerable variability may be introduced into the estimated nitrate flux. Interestingly, the maximum gradient is typically found near the base of the euphotic zone (0.1% surface PAR). Inset image shows nanomolar nutrient concentrations on a rescaled axis.

overall range of estimates because comparisons are not on a like-for-like basis.

[28] Turbulent diffusive NO₃[−] fluxes at the top of the nitracline ranged from 3.65 to 29.61 μmol N m^{−2} d^{−1} and represent on average ~3% of the measurable NO₃[−] demand in the overlying waters (Table 2). NO₃[−] fluxes calculated using the maximum gradient ranged from 36.09 to 86.10 μmol N m^{−2} d^{−1} and, whilst larger, still only repre-

sented ~16% of the mean NO₃[−] demand. With the exception of station 1 where the diffusive flux exceeded 60% of the measurable NO₃[−] uptake, the diffusive flux of NO₃[−] was typically <20% of the measured NO₃[−] uptake (Table 2).

[29] The significance of N₂ fixation as a supply term was evaluated against the diffusive NO₃[−] supply terms (Table 2). Depending upon how the nitracline gradient is defined our N₂ fixation observations could represent between 62% and 348% on average of the diffusive NO₃[−] supply. Our preference is to utilize the maximum nitracline gradient as this will produce the largest potential flux of NO₃[−]. Under this scenario, N₂ fixation represented between 18 and 152% of the diffusive NO₃[−] flux and 62% on average. In other words, N₂ fixation was generally not as large a source of new nitrogen to the surface ocean as the diffusive NO₃[−] flux, but neither was it an insignificant source.

3.5. Seasonal Context

[30] Modis Aqua sea surface temperature (SST) and surface chlorophyll data were analyzed to obtain the mean annual cycle for the eastern subtropical North Atlantic and to provide a seasonal context for our observations. In these waters, surface chlorophyll concentrations are typically <0.1 mg m^{−3} all year round but a weak seasonal cycle with a 2–3 fold variation in concentration is evident. Surface chlorophyll concentrations reach a typical annual maximum of ~0.08 mg m^{−3} during January to February, and a minimum concentration of ~0.04 mg m^{−3} in July to August (Figure 6). The mean MODIS chlorophyll concentration for the cruise period was 0.04 mg m^{−3} and was almost identical to the mean surface (5 m) chlorophyll concentration of 0.05 mg m^{−3} (range 0.03–0.08 mg m^{−3}) measured during the cruise. Based on the mean annual cycle shown in Figure 6, the annual mean surface chlorophyll concentration for these waters is 0.05 mg m^{−3}.

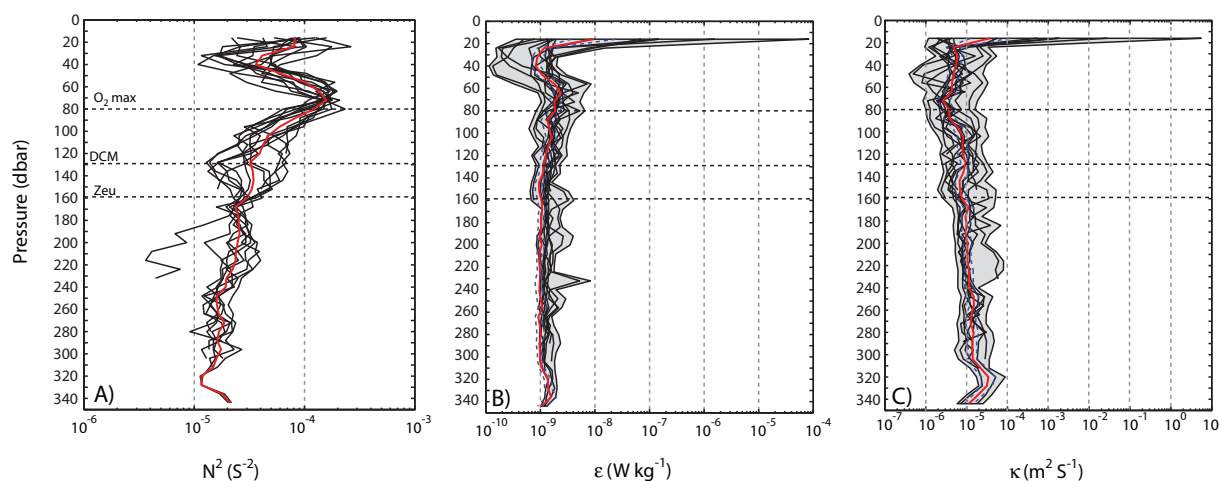


Figure 5. Vertical profiles (black lines) of (a) Brunt-Väisälä buoyancy frequency, (b) turbulent kinetic energy (ϵ), and (c) turbulent diffusivity (κ) collected from 13 stations in the eastern subtropical North Atlantic (~26.25°N, 31.1°W; see text for details). Gray shaded regions represent the envelope of 95% confidence intervals for the respective measurements. Thick red lines represent the cruise mean profile for each parameter and blue dashed lines the 95% confidence interval for the mean profile. Black dashed horizontal lines demarcate the cruise mean depth of the deep chlorophyll maximum (DCM), euphotic zone depth (Zeu; 0.1% surface irradiance), and oxygen maximum (O₂ max).

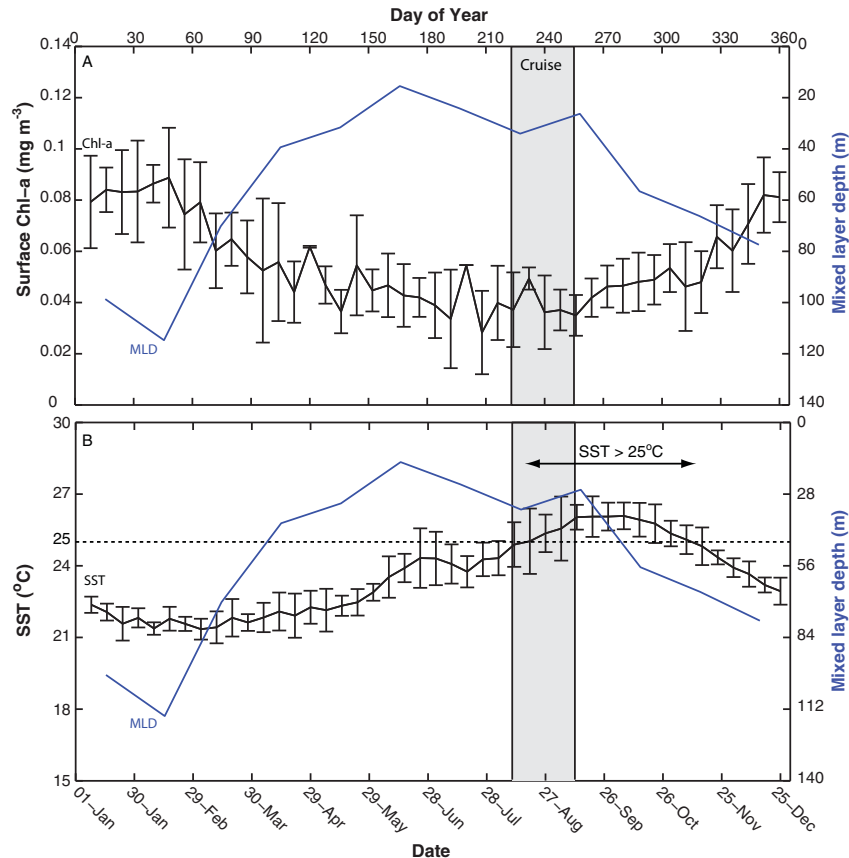


Figure 6. Annual cycle of (a) chlorophyll and (b) sea surface temperature for the eastern subtropical North Atlantic (based on observations centered around 26.25°N, 31.1°W). The annual chlorophyll cycle is calculated from Level 3 MODIS Aqua 8 day averaged chlorophyll data for the period 2002–2011 with error bars that represent the standard deviation of the mean calculated for each 8 day period from all available data points for that particular date over the data record. The annual SST cycle is derived from a similar MODIS Aqua Level 3 SST product for the same period with error bars representing the standard deviation of the mean. In both plots, the mixed layer depth (blue line) is derived from the ARGO based mixed layer climatology of Hosoda *et al.* [2010] and utilizes a density difference criteria of 0.03 kg m⁻³ relative to in situ density at 10 m. Note that Hosoda *et al.* provide monthly estimates of the mixed layer depth based on each full month of data and that in this figure the date of each mixed layer depth estimate has been reset to the middle of each month. The gray shaded region indicates the period of the present study.

[31] The annual cycle in SST is also shown in Figure 6. Maximum SSTs in excess of 26°C are typical from mid-September to early October, though temperatures generally exceed 25°C from mid-August through to mid-November. SST minima of ~22°C typically occur from late January through to late March. The climatological annual average SST for these waters is 23.5°C, which is warmer than the typical threshold temperature for *Trichodesmium* spp. of 20°C [Capone *et al.*, 1997], although below the 24–30°C temperature range over which maximum growth rates are observed [Breitbarth *et al.*, 2007]. The mean climatological MODIS SST for the cruise period was 25.3°C suggesting that ambient SSTs are usually optimal for maximum *Trichodesmium* growth at this time of year.

[32] In addition to mixed layer depth estimates calculated from cruise CTD casts, we also obtained the annual cycle in mixed layer depth for our study area from the

ARGO mixed layer climatology of Hosoda *et al.* [2010]. The mixed layer depth was estimated using the same density threshold criteria of 0.03 kg m⁻³ relative to the density at 10 m [de Boyer Montegut *et al.*, 2004]. The maximum mixed layer depth of ~120 m occurs in February/March, whilst the minimum mixed layer depth of ~20 m occurs in June. Despite the coincident timing of seasonal increases in mixed layer depth and in surface chlorophyll concentrations, it is still debated whether convective mixing during the cooler months is sufficiently deep enough to entrain significant quantities of nutrients to drive this change in chlorophyll concentration [Beers *et al.*, 1982; Hayward *et al.*, 1983; Hayward, 1987; Dore *et al.*, 2008]. An alternative explanation of seasonal change in cellular chlorophyll content in response to seasonally changing irradiance intensities has also been suggested as a causal mechanism for the seasonal cycle in surface chlorophyll concentrations [Winn *et al.*, 1995].

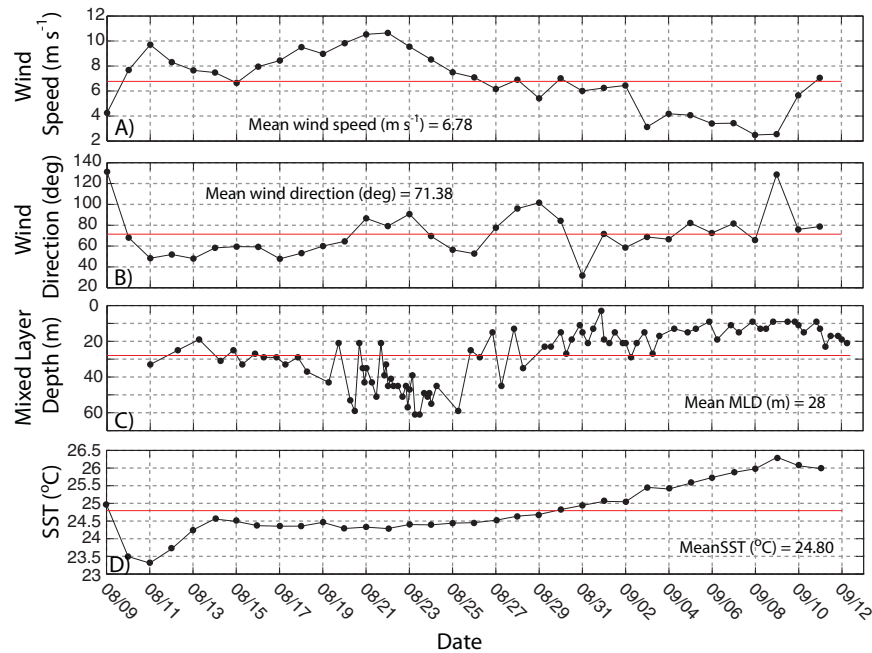


Figure 7. Local meteorological and surface ocean conditions. (a) Mean daily wind speed, (b) mean daily wind direction, (c) mixed layer depth, and (d) sea surface temperature. A prolonged reduction in the mean daily wind speed was evident from 22 August to 9 September during which time the mixed layer shoaled by almost 50 m and sea surface temperature increased by 2°C. The mixed layer depth was calculated from CTD profile data using a density threshold difference of 0.03 kg m⁻³ relative to in situ density at 10 m depth. Horizontal red lines represent the cruise mean for each parameter.

[33] In summary, our cruise during August and September took place during a period of relatively shallow mixed layer depths, low-surface chlorophyll concentrations, and high sea surface temperatures; conditions which should be preferential for diazotrophic organisms.

3.6. Intracruise Variability

[34] Throughout the preceding we have assumed environmental homogeneity throughout the cruise period. In Figure 7, we present observations of lower atmosphere and upper ocean variables measured during the cruise that reveal some important temporal changes during the observation period. It can be seen in Figure 7 that consistent gradual changes occurred to wind speeds (decreased from >8 m s⁻¹ to <4 m s⁻¹), mixed layer depths (shoaled from >40 m to <20 m), and sea surface temperature (increased from <24.5°C to >26°C). The only variable for which there was no consistent change was wind direction, which was primarily east northeasterly (~71°). An analysis of air-mass back trajectories (Figure 8) confirms both the general wind direction, as well as also identifying key air mass source regions. In general, the air masses passing over the study site came from regions north of the study site and were typically oceanic in terms of their recent history. Two discrepancies to this generalization are apparent. First, between 10 and 14 August there appears to have been a shift in the air-mass source toward NW Europe, and second, between 25 and 29 August air with a distinctly African origin reached the study site.

[35] As both NW Europe and Africa can be important source regions for air-borne nutrient and iron deposition to

the ocean we have attempted to ascertain what impact, if any, changes in wind direction may have had for our study particularly as sporadic rainfall (wet deposition) events were encountered and observed at a distance from the ship. Although no formal record of rainfall or of its nutrient and iron loading was made, the importance of wet deposition as a mechanism for introducing Saharan dust with its associated P and Fe content to the surface ocean requires consideration in the interpretation of our N₂ fixation results due to the positive effect such inputs can have on N₂ fixation rates and on the diazotrophic community [Langlois *et al.*, 2012; Benavides *et al.*, 2013]. A retrospective assessment of precipitation patterns over the study site was made possible by analysis of the NOAA Climate Prediction Centre's pentad (5 day) merged analysis of precipitation data set [Xie and Arkin, 1997]. This product is a global data set with a 2.5° resolution and consists of both direct-derived and satellite-derived precipitation rates, and whilst known to contain some inaccuracies [Gruber *et al.*, 2000; Xie *et al.*, 2003; Yin *et al.*, 2004] is useful in the current context for demonstrating the timing of precipitation events. In Figure 9, we present a time series of the average daily precipitation rate for an area of the ocean approximating 23.75–28.75°N, 28.75–33.75°W, slightly larger than our study site. From this analysis, we identify a prominent precipitation event (Pmax) for the period of 19 August to 23 August, when precipitation reached 2.4 mm d⁻¹. This was followed by average daily precipitation rates of >0.8 mm d⁻¹ for the period 24 August to 2 September. Whilst the overall precipitation rates are low by global standards, the increased precipitation between 19 August and 2 September followed minimal

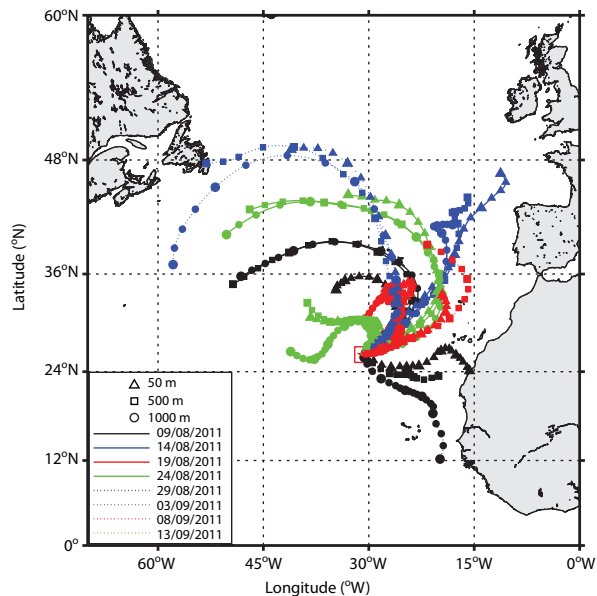


Figure 8. HYSPLIT air mass 5 day back trajectory results. Model runs (GDAS1 data set) were conducted at 5 day intervals to identify source regions that influenced the survey site (red box). Ship-based meteorological data indicate winds came from an average direction of 71° (ENE), comparable to HYSPLIT model results. The dominant source regions include W Atlantic, NW Europe, and NW Africa. Of interest here is the NW African source region identified as reaching the survey site on 29 August 2011, as air masses from this region have the potential to carry high dust loads to the remote open ocean.

precipitation (P_{min} ; $<0.1 \text{ mm d}^{-1}$) for the preceding 2.5 months and was coincident with air-mass trajectories coming from Africa (Figure 8). We speculate therefore that the increase in precipitation coincident with the presence of an air mass from Africa increased the likelihood of P and Fe injection to the surface ocean during the observation period and thus represents a significant and important event that could explain some of the variability in our N₂ fixation rates.

4. Discussion

[36] We have reported new observations of N₂ fixation, NO₃⁻ uptake and turbulent diffusive NO₃⁻ supply from the oligotrophic eastern subtropical North Atlantic Ocean during the annual period of maximum sea surface temperatures and under variable meteorological conditions. In assessing the importance of these processes, we find that on average the diffusive NO₃⁻ flux represents $<20\%$ of the measurable NO₃⁻ uptake, implying the requirement for additional NO₃⁻ supply terms to balance the biological demand. A similar estimate of 18% can be obtained from the data presented by Lewis *et al.* [1986]. We also find that whilst N₂ fixation may indeed exceed the diffusive NO₃⁻ flux at an individual station, on average it was a smaller supply term and represented 62% of the diffusive NO₃⁻ supply. Against this background, the integrated rate of N₂ fixation equates to just 8% of integrated NO₃⁻ uptake rates (Table 2). Thus,

whilst N₂ fixation may represent an important supply process for new nitrogen to the surface ocean the process of N₂ fixation itself is dwarfed in magnitude by the rate at which NO₃⁻ is consumed within these waters.

[37] To balance integrated rates of NO₃⁻ uptake an additional source (or sources) of NO₃⁻ other than turbulent diffusion must exist. Possible sources include lateral fluxes of NO₃⁻, atmospheric inputs, mesoscale eddies, and nitrification. Simple calculations based on individual bottle samples indicate that the residence time for NO₃⁻ in the upper ocean varied from as little as 7 h to as long as 223 h (~ 9.3 days) with residence times increasing to as much as 626 h (26 days) within the upper nitracline. Therefore in order to maintain surface NO₃⁻ concentrations around the cruise mean value of $\sim 7 \text{ nmol L}^{-1}$ there has to have been a process supplying NO₃⁻ on a time scale of hours to days. Whilst the impact of lateral NO₃⁻ fluxes over the time-scales required is considered minimal due to the lack of any significant horizontal gradients in NO₃⁻ concentrations within surface waters, the impact of mesoscale features is less obvious. Johnson *et al.* [2010] have highlighted the role of short-term input events in supplying NO₃⁻ to waters immediately above the nitracline but the mesoscale injection of NO₃⁻ was apparently not large enough to impact near surface ($<50 \text{ m}$) concentrations as NO₃⁻ was rapidly consumed subsurface. We observed significant undulations in the nitracline, silicicline, and phosphocline between 100 and 150 m (Figure 3) suggesting that there were important mesoscale factors at work. Intriguingly, whilst there is some evidence to suggest moderately elevated concentrations of NO₃⁻, Si(OH)₄, and PO₄³⁻ above the thermocline at several stations (Figure 3), there is little evidence to suggest that this supply mechanism was sufficient to impact near surface waters where NO₃⁻ residence times were typically $<24 \text{ h}$.

[38] Atmospheric inputs of nitrogen to the ocean are generally considered to be poorly constrained due to a lack of observational data but such inputs are potentially significant in nitrogen-limited systems. Baker *et al.* [2010] estimate a typical atmospheric input (dry plus wet deposition) for NO₃⁻ and NH₄⁺ of $21 \text{ } \mu\text{mol m}^{-2} \text{ d}^{-1}$ and $11.3 \text{ } \mu\text{mol m}^{-2} \text{ d}^{-1}$, respectively, for the region of our study. In the

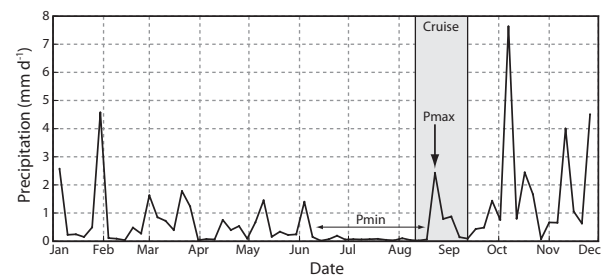


Figure 9. 2011 annual time series of precipitation for the region 23.75–28.75°N, 28.75–33.75°W. The period of the cruise is indicated by the gray box and the local maximum in precipitation (P_{max}) is identified by an arrow. P_{min} indicates the period of minimal precipitation prior to the cruise. This time series is based upon the NOAA Climate Prediction Centre Merged Analysis of Precipitation (CMAP) pentad product.

case of NO₃⁻, this would equate to 3% or less of the integrated NO₃⁻ pool (Table 1) suggesting that the atmosphere is a generally less important source than the deep ocean [Okin *et al.*, 2011]. However, depending upon the volume over which the atmospheric input is distributed the typical NO₃⁻ input suggested by Baker *et al.* [2010] may be locally significant. For example, if the total atmospheric input was evenly distributed over a 100 m water column this would equate to an input of 0.2 nmol L⁻¹ d⁻¹, approximately 3% of the mean NO₃⁻ concentration of 7.1 nmol L⁻¹ found over the upper 100 m. If, however, the atmospheric input was localized to a 10 m thick surface layer, the atmospheric input could equate to 2.1 nmol L⁻¹ d⁻¹ or ~30% of average in situ concentrations. Clearly, therefore, understanding how atmospheric inputs are distributed throughout the surface waters of the ocean is of particular importance for addressing any potential biological impact.

[39] Nitrification, the transformation of NH₄⁺ to NO₃⁻, is highly likely to be a major source of NO₂⁻ and NO₃⁻ in waters of the eastern subtropical North Atlantic but estimates of nitrification rates range widely [Yool *et al.*, 2007; Ward, 2008]. In an exploratory study from the Atlantic Ocean, Clark *et al.* [2007] reported NH₄⁺ and NO₂⁻ oxidation rates for a single irradiance depth (55% surface PAR) at a station located approximately 770 nmi north of our study site. The measured rates of NH₄⁺ oxidation (0.05 nmol L⁻¹ h⁻¹), and in particular of NO₂⁻ oxidation (0.45 nmol L⁻¹ h⁻¹), were found to be sufficient to provide 270% of the local NO₃⁻ demand (0.17 nmol L⁻¹ h⁻¹). In a more extensive study, Clark *et al.* [2008] reported NH₄⁺ and NO₂⁻ oxidation rates for two irradiance depths (55 and 1% surface PAR) along a transect of the Atlantic Ocean. Rates of NH₄⁺ oxidation were similar at both depth horizons ranging from 0 to 10 nmol L⁻¹ d⁻¹, whereas NO₂⁻ oxidation rates increased from 0.4 to 12 nmol L⁻¹ d⁻¹ at 55% PAR to 1–31 nmol L⁻¹ d⁻¹ at the base of the euphotic zone. These rates were sufficient to turnover all inorganic nitrogen pools (NH₄⁺, NO₂⁻, NO₃⁻) on a subdaily timescale.

[40] The mean NO₃⁻ uptake rate for our observations made between the surface and the deep chlorophyll maximum was equivalent to 4.6 nmol L⁻¹ d⁻¹, almost identical to the mean NO₂⁻ oxidation rate of 4.7 nmol L⁻¹ d⁻¹ reported by Clark *et al.* [2008] for depths equivalent to 55% PAR. Therefore, typical nitrification rates within the upper euphotic zone appear more than capable of providing the NO₃⁻ needed to support observed uptake rates and may easily account for the additional 80% needed to support our integrated uptake rates once the contribution from diffusion is accounted for. If, however, NO₂⁻ oxidation rates approach the higher mean value of 13.2 nmol L⁻¹ d⁻¹ as reported for the base of the euphotic zone by Clark *et al.* [2008], then nitrification alone could provide almost three times as much NO₃⁻ as is consumed on a daily basis. Clearly, under a steady-state assumption a sustained contribution of this magnitude over and above in situ demand would lead to an accumulation of NO₃⁻ in surface waters that may account for some of the variability in surface NO₃⁻ concentrations reported in previous studies [Painter *et al.*, 2008] but the breadth of existing nitrification observations is too small to explore this possibility further.

[41] Mourino-Carballido *et al.* [2011] reported simultaneous measurements of diffusive NO₃⁻ flux and N₂ fixation in the subtropical North Atlantic which indicated that N₂ fixation could provide only 2% of the nitrogen supplied by diffusive NO₃⁻ fluxes across the nitracline. This appears to be the result of an unusually high diffusivity term of 5 cm² s⁻¹, roughly 65 times larger than we observed, but also due to rates of N₂ fixation that were lower (11 ± 9 μmol N m⁻² d⁻¹) than those we report here. As the observations reported by Mourino-Carballido *et al.* [2011] were made in April and May, when sea surface temperatures can be 3–4°C cooler (Figure 6), it is perhaps not surprising that N₂ fixation rates were lower but the cause of the higher diffusivity term is less clear. Nitracline gradients reported by Mourino-Carballido *et al.* [2011] are similar to those we report here which indicates that the difference in diffusivity is a key factor behind the low supply from N₂ fixation reported by Mourino-Carballido *et al.* [2011]. A number of possible explanations for this exist including spurious data, temporal and spatial variability and differences in methodologies for the calculation of diffusivity, which may all singularly or collectively explain the disparity in the significance of N₂ fixation in these waters. Independent measurements of turbulent diffusivity from the wider subtropical North Atlantic tend to provide diffusivities closer to 0.1 cm² s⁻¹ [Lewis *et al.*, 1986; Ledwell *et al.*, 1993, 1998, 2008] but diffusivities >1 cm² s⁻¹ are known from studies in the subpolar North Atlantic [Law *et al.*, 2001; Jickells *et al.*, 2008]. It would appear therefore that the assessment of the importance of N₂ fixation reported by Mourino-Carballido *et al.* [2011] and in this study are not directly comparable due to both seasonal differences in N₂ fixation rates and important differences in the magnitude of the diffusivity term.

4.1. A Synthesis of Previous Observations

[42] Our principal observations of N₂ fixation, NO₃⁻ uptake and of the nitracline gradient are central to the question of how important N₂ fixation is as a supply mechanism of new nitrogen to the subtropical North Atlantic. More fundamentally, however, we must first assess how representative our observations are for the eastern subtropical North Atlantic.

[43] We present in Figure 10 an assessment of the maximum nitracline gradient calculated from 12 separate cruise data sets collected over a 13 year period (1995–2008) as part of the Atlantic Meridional Transect (www.amt-uk.org) [Robinson *et al.*, 2006]. Cruises were conducted in boreal spring or autumn months and show broad similarities in the maximum nitracline gradient between 24 and 36°N. There, gradients rarely exceeded 200 μmol m⁻⁴ and more typically were <100 μmol m⁻⁴, which suggests weak interannual variability. Differences in cruise track and occasional sampling within the NW African upwelling system (~20°N) can produce distinctly sharper nitracline gradients. Influence from upwelling near the equatorial region further south has a similar effect. Generally, all cruises show reduced variability in the nitracline gradient between 24 and 36°N. Three cruises shown in Figure 10 are more directly comparable to the present study (the others are influenced to differing degrees by the NW African upwelling system as a result of differences in cruise transect). These three cruises

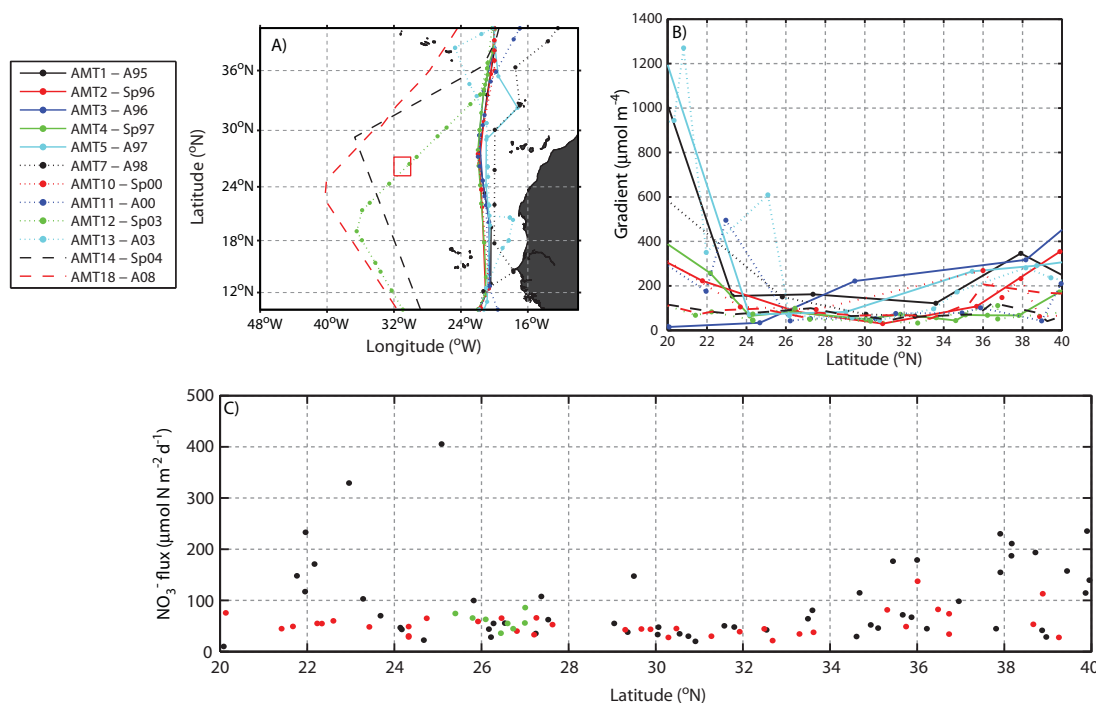


Figure 10. Observations of nitracline gradients and NO_3^- fluxes in the eastern subtropical North Atlantic from the Atlantic Meridional Transect (AMT) program made between 1995 and 2008. In (a) we present the various AMT cruise transects and station positions from which vertical profiles of NO_3^- were collected. In (b) we present the maximum nitracline gradients obtained for each profile. In (c) we present the resulting NO_3^- fluxes calculated using the mean diffusivity obtained in the present study (red box in Figure 10a; $0.077 \text{ cm}^2 \text{ s}^{-1}$). Note that differences in cruise transects and the influence of the NW African upwelling system is evident in the nitracline gradient data thus negating a truthful examination of seasonal or interannual variability in NO_3^- fluxes. However, three cruises (AMT12, AMT14, and AMT18) sampled the open waters of the eastern subtropical North Atlantic and are more appropriate comparisons to the present study; these cruises are indicated by the red dots in Figure 10c, the present study is shown in green. In the legend, the year of each cruise and the boreal season during cruise departure are abbreviated, thus A95 indicates Autumn 1995, Sp96 Spring 1996, and so on.

produce mean nitracline gradients of $63 \pm 24 \mu\text{mol NO}_3^- \text{ m}^{-4}$ (AMT12), $78 \pm 24 \mu\text{mol NO}_3^- \text{ m}^{-4}$ (AMT14), and $96 \pm 48 \mu\text{mol NO}_3^- \text{ m}^{-4}$ (AMT18) for stations between 24 and 36°N. These mean gradients are very comparable to the mean gradient of the present study ($90 \pm 24 \mu\text{mol NO}_3^- \text{ m}^{-4}$; Table 2) suggesting that nitracline gradients are relatively well constrained in this region to between 50 and $100 \mu\text{mol NO}_3^- \text{ m}^{-4}$. Using the individual gradients obtained from the AMT data set presented in Figure 10 in conjunction with the mean diffusivity of $0.077 \text{ cm}^2 \text{ s}^{-1}$ obtained in the present study, we thus obtain NO_3^- fluxes which are generally $<100 \mu\text{mol NO}_3^- \text{ m}^{-2} \text{ d}^{-1}$ (Figure 10).

[44] In Figure 11, we present a synthesis of NO_3^- uptake rates from previous studies in the eastern subtropical North Atlantic. Though there is quite often some considerable difference in the choice of integration depth between studies (examples range from 20 to 180 m) and in the sampling resolution (3–8 depths), there is nevertheless a broad agreement both in terms of the magnitude of the integrated NO_3^- pool and in terms of the integrated NO_3^- uptake rates. Estimates of the integrated NO_3^- pool vary from 0.2 to $166 \text{ mmol NO}_3^- \text{ m}^{-2}$ (mean $36 \text{ mmol NO}_3^- \text{ m}^{-2}$), with one exceptional estimate of $857 \text{ mmol NO}_3^- \text{ m}^{-2}$ reported by Varela *et al.* [2005] which appears to derive from the

location of this station being close to the NW African upwelling system and correspondingly high NO_3^- concentrations at depth. In the context of these previous observations, our estimates of the integrated NO_3^- pool of $0.72\text{--}17.95 \text{ mmol NO}_3^- \text{ m}^{-2}$ are comparable (Table 1 and Figure 11). The same general conclusion is true of our NO_3^- uptake rates ($0.13\text{--}1.18 \text{ mmol N m}^{-2} \text{ d}^{-1}$) (Table 1), which are within the range of previous estimates of $0.06\text{--}34 \text{ mmol NO}_3^- \text{ m}^{-2} \text{ d}^{-1}$ (mean $3.15 \text{ mmol NO}_3^- \text{ m}^{-2} \text{ d}^{-1}$) (Figure 11). Finally, we note a disparity between the common assumption for the seasonal increase in surface chlorophyll (Figure 3), which is considered due to the seasonal entrainment of NO_3^- from the nitracline when the surface mixed layer deepens in winter, and the data assembled here. There is no suggestion of seasonality in either the integrated NO_3^- pool or in NO_3^- uptake rates to support the assumption that increased NO_3^- concentrations underlie enhanced chlorophyll concentrations (Figure 11). There is, however, a significant temporal bias underlying the observations which cluster around year day 150 (late May) and year days 260–280 (September to October) which limits this temporal assessment and there appears to be surprisingly little information in the literature regarding NO_3^- uptake during the boreal winter months for this region.

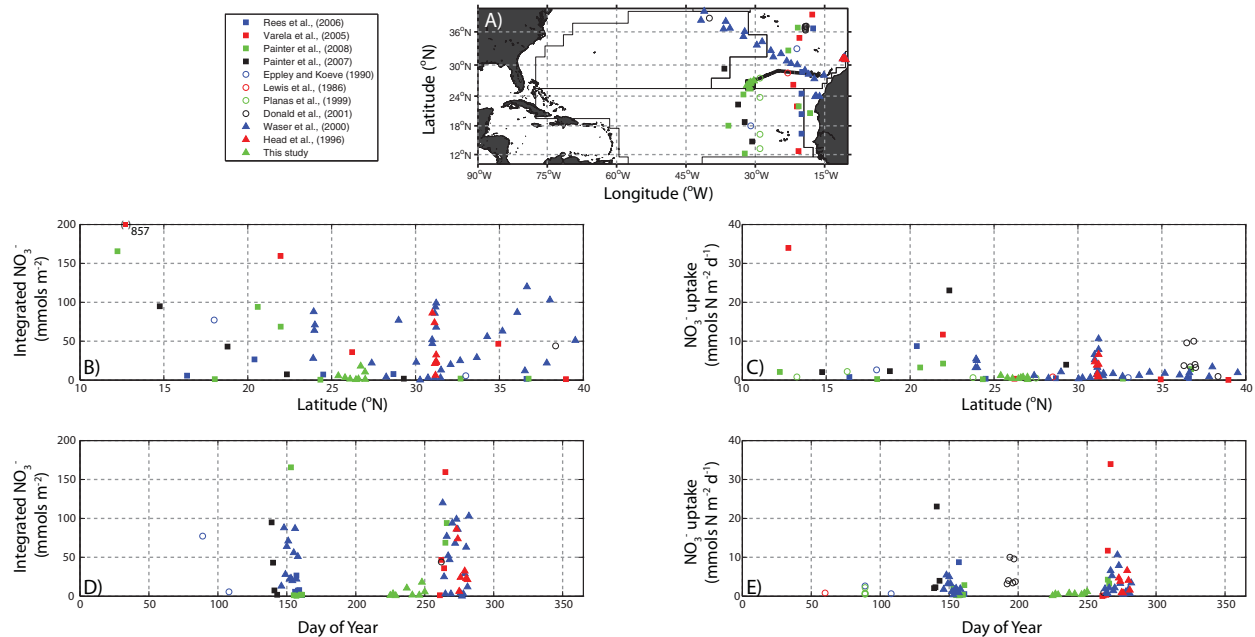


Figure 11. Synthesis of previous studies in the eastern subtropical North Atlantic (10–45°W, 10–40°N) where both integrated NO₃[−] pools and integrated NO₃[−] uptake rates have been obtained. (a) The position of previous observations used in this analysis along with the static boundaries of Longhurst Provinces [Longhurst, 1998]; the cruise track of the present study is represented by the black line. In (b) and (d) we show the range of integrated NO₃[−] pools observed by the various studies against latitude and day of year, respectively. Note that in Figure 11b one observation lies beyond the y axis limits and is denoted along with its concentration along the top axis. In (c) and (e) we show integrated rates of NO₃[−] uptake against latitude and day of year, respectively. It is evident that despite wide spatial coverage observations remain rather sparse through time and are typically clustered during the year with the majority of observations being obtained in June and September. Individual data sourced from Lewis *et al.* [1986], Eppeley and Koeve [1990], Head *et al.* [1996], Planas *et al.* [1999], Waser *et al.* [2000], Donald *et al.* [2001], Varela *et al.* [2005], Rees *et al.* [2006], and Painter *et al.* [2007, 2008].

[45] A global synthesis of N₂ fixation rates has recently been made available [Luo *et al.*, 2012] and we utilize this database to assess our observations. Our study site is located in the middle of a region of relatively low N₂ fixation rates (Figure 12) compared to the higher rates that are more typical nearer the equator [Moore *et al.*, 2009]. This is most likely due to relatively low surface iron concentrations in this region [Rijkenberg *et al.*, 2011, 2012] and more broadly to the impact iron has on rates of N₂ fixation [Mills *et al.*, 2004; Moore *et al.*, 2009]. Our measurements of surface N₂ fixation rates are comparable to previous observations (Figure 12), though our integrated rates can be higher. This suggests that general undersampling is responsible for an incomplete picture of the range and interannual variability of integrated N₂ fixation rates in this sector of the northeast subtropical Atlantic Ocean.

4.2. Methodological Caveats

[46] In the context of previous observations from the eastern subtropical North Atlantic there is no suggestion that our rate measurements or NO₃[−] flux estimates are atypical. The only parameter that we are unable to fully constrain is the ambient diffusivity, but the majority of existing observations tend to constrain this value closer to $\sim 0.1 \text{ cm}^2 \text{ s}^{-1}$ than to $1 \text{ cm}^2 \text{ s}^{-1}$. Thus, it is highly unlikely that the conclusions we reach above regarding the impor-

tance of N₂ fixation (62% of the diffusive flux) are significantly in error. However, despite its wide usage the ¹⁵N₂ “bubble” technique introduced by Montoya *et al.* [1996] has recently come under increased scrutiny amid concern that the technique underestimates N₂ fixation rates [Mohr *et al.*, 2010]. These concerns were subsequently investigated by Großkopf *et al.* [2012] using a technique similar to that first used by Dugdale *et al.* [1959, 1961] in which a preprepared seawater solution containing dissolved ¹⁵N₂ is added to a seawater sample as a solution rather than a gas. This approach removes the assumption that the injected bubble of ¹⁵N₂ equilibrates with the sample and harmonizes the technique to existing stable isotope methodologies in which the tracer is added as a solution. Using the method of Mohr *et al.* [2010], Großkopf *et al.* [2012] reported elevated N₂ fixation rates in the Atlantic Ocean compared to rates obtained with the bubble technique of Montoya *et al.* [1996]. In particular Großkopf *et al.* [2012] report an average increase in N₂ fixation rates of 62% compared to parallel results with the existing bubble technique. The full extent of a methodological underestimate of N₂ fixation rates on global nitrogen budgets requires a great deal of further observational study but Großkopf *et al.* [2012] estimate that the global N₂ fixation rate could increase by $\sim 70\%$ to $177 \pm 8 \text{ Tg N yr}^{-1}$. In a related study, Wilson *et al.* [2012] compared the N₂ bubble technique, the N₂ solution

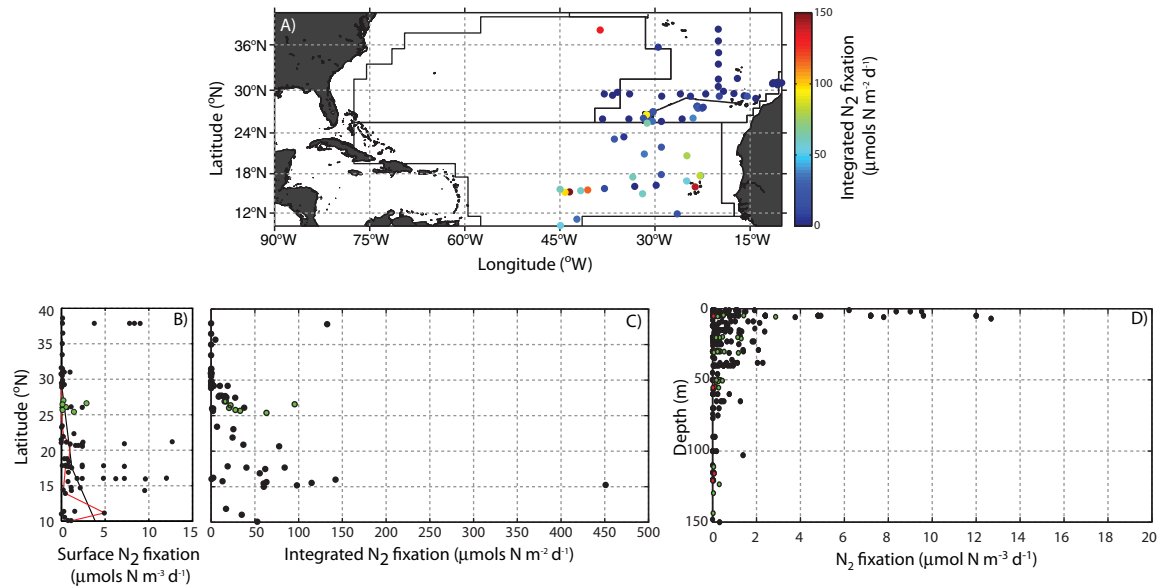


Figure 12. N₂ fixation measurements collated from the literature and as measured in this study. (a) The positions of profiles of N₂ fixation extracted from the database compiled by *Luo et al.* [2012] for the eastern subtropical North Atlantic (10–45°W, 10–40°N). The cruise track of the preset study is represented by the black line. (b) Surface (<10 m) N₂ fixation rates from the database and we highlight two recent meridional surveys [*Moore et al.*, 2009] black line and [*Mourino-Carballido et al.*, 2011] red line which indicate the low-latitude maxima in surface N₂ fixation rates; note that the x axis scale has been restricted to highlight the low N₂ fixation rates that are typical of the study region and that two results with fixation rates of $\sim 150 \mu\text{mol m}^{-3} \text{d}^{-1}$ are excluded from this figure. (c) Integrated N₂ fixation rates against latitude, which again highlights the low-latitude maxima that is typical of the North Atlantic. In (d), we present profiles of volumetric N₂ fixation rates highlighting the typical near surface maxima in N₂ fixation. In Figures 12b–12d, black dots represent literature observations, green (total) and red (<8 μm size fraction) dots represent the results of this study. All data extracted from the synthesis present by *Luo et al.* [2012] with individual data coming from studies reported by *Benavides et al.* [2011], *Capone et al.* [2005], *Fernandez et al.* [2010], *Mourino-Carballido et al.* [2011], *McCarthy and Carpenter* [1979], *Moore et al.* [2009], *Turk et al.* [2011], *Goebel et al.* [2010], *Rees et al.* (unpublished data), and *Mulholland et al.* (unpublished data).

technique, and the alternative acetylene reduction technique [*Stewart et al.*, 1967], and independently confirmed that the popular N₂ bubble technique does indeed lead to significant underestimates of N₂ fixation rates compared to the solution technique. Furthermore, *Wilson et al.* [2012] also highlight deficiencies in the acetylene reduction technique, particularly the use of a fixed conversion ratio between the amount of acetylene reduced and the amount of nitrogen fixed. Consequently, at the present time the historical record of N₂ fixation measurements must be viewed with caution and our best estimates of global N₂ fixation rates derived from these measurements as underestimates.

[47] In this study, we used the $^{15}\text{N}_2$ bubble technique and whilst our N₂ fixation results compare well to previous observations (Figure 12), the bulk of which are based on the N₂ bubble and acetylene reduction assays, growing recognition of methodological problems with the most commonly used method suggests that the historical record is less reliable than thought and that the level of uncertainty over regional and global N₂ fixation rates is probably higher than we would like. Although *Großkopf et al.* [2012] found that the makeup of the underlying diazotroph

community was critical in explaining the magnitude of the difference in the results of the two techniques, with *Trichodesmium* dominated communities producing smaller methodological differences, we cannot say with certainty what error is present in our results as we do not have a full description of the diazotrophic community. Our study region was located at the northern edge of the region recognized as receiving Saharan dust inputs (the Saharan dust belt 10°S–25°N), further north than the abundance maxima for *Trichodesmium* ($\sim 10^\circ\text{N}$) [*Moore et al.*, 2009], located in a region where SST exceeds 25°C only seasonally (Figure 6), and in a region where nitrogenase (*nifH*) expression suggests a diverse diazotrophic population is present [*Turk et al.*, 2011]. Consequently, it is highly probable that our N₂ fixation rates are low. The few size-fractionated N₂ fixation results we have, suggest that N₂ fixation by the <8 μm fraction, which was typically $\sim 10 \mu\text{mol N m}^{-2} \text{d}^{-1}$ (Table 1), represented between 27 and 67% of the total measurable N₂ fixation rate and thus any error may be significant. Interestingly, *Großkopf et al.* [2012] found that the methodological discrepancy was at a regional minimum at 25°N (their supplementary Figure 5), which may lower rather than increase the unquantified error in our results.

[48] Assuming our measured N₂ fixation rates are low by 62% (the average underestimate reported by *Großkopf et al.* [2012]) we have recalculated the importance of N₂ fixation relative to our NO₃[−] flux term (Table 2). The results suggest that for an individual comparison N₂ fixation could represent between 29 and 246% of the NO₃[−] flux with a mean value of 100%. Under this scenario, it would appear that N₂ fixation balances and frequently exceeds the diffusive flux of NO₃[−] in the eastern subtropical North Atlantic. Recognition of the increase in both the importance and magnitude of N₂ fixation rates will benefit on-going attempts at closing regional nitrogen budgets, particularly for the subtropical ocean where the discrepancy between the rates of nitrogen supply and geochemical estimates of nitrogen demand has remained unresolved for almost three decades. Combined with an important role for carbon export [Karl *et al.*, 2012], N₂ fixation is likely to continue to redefine our understanding of the marine nitrogen cycle in the years ahead.

[49] **Acknowledgments.** We thank C. M. Moore and D. Honey for technical advice, M. Bolshaw and D. Spanner (all University of Southampton) and A. Rees and L. Al-Moosawi (Plymouth Marine Laboratory) for assistance with the isotopic analyses, Y.-W. Luo (Woods Hole Oceanographic Institute) for providing access to the global diazotrophic database of Luo *et al.* [2012], S. van Gennip, V. Paba, R. Green, and J. Allen for assistance with the turbulence profiler at sea and the Natural Environment Research Council for supporting this work through the Oceans 2025 research program. We also thank the two reviewers for their constructive comments. This study benefited from access to the extensive AMT database (<http://www.amt-uk.org/>) and is a contribution to the international IMBER project and was supported by the UK Natural Environment Research Council National Capability funding to Plymouth Marine Laboratory and the National Oceanography Centre, Southampton. This is contribution 238 of the AMT program.

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