Geochronology (Re–Os and U–Pb) and fluid inclusion studies of molybdenite mineralisation associated with the Shap, Skiddaw and Weardale granites, UK

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Late Devonian magmatism in Northern England records key events associated with the Acadian phase of the Caledonian-Appalachian Orogen (C-AO). Zircon U–Pb and molybdenite Re–Os geochronology date emplacement and mineralisation in the Shap (405±1.8 Ma), Skiddaw (398.8±0.4 and 392.3±2.8 Ma) and Weardale granites (398.3±1.6 Ma). For the Shap granite, mineralisation and magmatism are contemporaneous, with mineralisation being directly associated with the boiling of CO2-rich magmatic fluids between 300 and 450°C, and 440 and 620 bars. For the Skiddaw granite, the Re–Os age suggests that sulphide mineralisation occurred post-magmatism (398.8±0.4 Ma) and was associated with the boiling (275 and 400°C and at 375–475 bars) of a non-magmatic fluid, enriched in N2, CH4, and S, which is isotopically heavy. In contrast, the co-magmatic molybdenite mineralisation of the Weardale granite formed from non-fluid boiling at 476 to 577°C at 1–1.7 kbars. The new accurate and precise ages indicate that magmatism and Mo-mineralisation occurred during the same period across eastern Avalonia (cf. Ireland). In addition, the ages provide a timing of tectonism of the Acadian phase of the C–AO in northern England. Based on the post-tectonic metamorphic mineral growth associated with the Shap and Skiddaw granite aureoles, Acadian deformation in the northern England continued episodically (before ~405 Ma) throughout the Emsian (~398 Ma).

Keywords: Rhenium, Osmium, U–Pb geochronology, Molybdenite, Zircon, Fluid inclusions, Shap, Weardale, Skiddaw, Acadian orogeny

Introduction

Granitic plutons represent a significant component of orogenic cycles and are often associated with one of the most important types of fluid flow regime in the crust. Commonly the magmatic-hydrothermal systems are genetically related to Cu–Mo±Au mineralisation. The Caledonian-Appalachian Orogen (C-AO) is marked by a ~50 m.y. period of granite intrusions with associated molybdenite mineralisation systems. This study combines Re–Os molybdenite and U–Pb zircon geochronology with fluid inclusion studies to investigate the timing and nature of molybdenite mineralisation associated with the Shap, Skiddaw and Weardale granites from the northern England sector of the Acadian phase of the C–AO (Fig. 1). These three granites represent three of six granite intrusions considered to be components of the Lake District batholith (Ennerdale, Eskdale, Threlkeld; Fig. 1). These intrusions reflect, in part, the evolution of the northern part of eastern Avalonia34–45 (and references therein) and the geochronology to be presented here can yield important information on the timing of tectonism during the Acadian phase of the C–AO in the northern England region. The results are further discussed in the context of granite-related molybdenite systems present in the western Ireland (Connemara) sector and in regard to the timing and duration of magmatism and also the tectonic setting of the C–AO.

Shap, Skiddaw and Weardale intrusions and associated mineralisation

The Shap granite

Intruded into Ordovician volcanics and Silurian sediments (Fig. 1), the Shap granite is a composite intrusion, representing three stages of magmatism. There is a progressive increase in grain size and K-feldspar megacrysts from stage one (early) to stage three (late), where the latter intrusion contains ~50 modal%
Simplified geology of the studied areas, Shap granite (modified after Wadge et al. 71), Weardale granite (modified after Dunham 20) and Skiddaw granite (modified after Shepherd et al. 59).

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WEARDALE GRANITE

A log of the Rookhope borehole B374-4278. Sample locations shown are SSKD576 (this study) and previous K-Ar and Rb-Sr studies (see text for details).

SKIDDAW GRANITE

Simplified geology of the Carrock deposit (modified after Shepherd et al. 1976). Sample locations shown are DS4-C and SK2 (this study) and previous K-Ar studies. SK1 is from Sineh Gill 6 km west of this area (54°58'22"N, 3°05'12"W; see text for details).

1 Simplified geology of the studied areas, Shap granite (modified after Wadge et al. 71), Weardale granite (modified after Dunham 20) and Skiddaw granite (modified after Shepherd et al. 59).
K-feldspar megacrysts. Stage two of the intrusion forms ~90% of the Shap granite body and comprises K-feldspar megacrysts (25–30 modal%, ~4 cm), with a matrix (1–2 mm grain size) of approximately equal amounts of quartz, orthoclase, plagioclase (oligoclase), together with lesser amounts of biotite. Accessory apatite, zircon, titanite are also present. All stages contain mafic enclaves. Sulphide mineralisation (pyrite–chalcopyrite–bornite ± molybdenite) in the Shap granite is both disseminated and quartz vein-hosted. Quartz veins are 1–2 cm thick, dip steeply to the east and post-date all stages of the Shap granite. The ore mineral assemblage consists of chalcopyrite–pyrite–bornite ± molybdenite. Molybdenite in some cases is the predominant ore mineral occurring with minor chalcopyrite, pyrite and bornite as both disseminations and in veins. No evidence was observed to determine the sulphide paragenesis. Quartz veins are bereft of an alteration selvage; however, matrix biotite and plagioclase are partially altered to chlorite and sericite respectively.

Several studies have presented determinations for the emplacement age of the Shap granite (Fig. 2). Uncertainties for all ages noted below are at the 2σ level. Whole-rock (stages one and two), feldspar and biotite Rb–Sr data yield an isochron age of 394 ± 3 Ma (n = 22). In contrast, Rundle reported a Rb–Sr age using whole-rock data (stages two and three intrusion) of 381 ± 8 Ma. A recent single crystal Rb–Sr feldspar study yields an age of 405 ± 2 Ma. K–Ar biotite analyses from the stage two granite yields a mean age of 397 ± 7 Ma; however, individual analyses range from 394 ± 12 to 403 ± 12 Ma (n = 3). A discordia of U–Pb data for three multi-grain zircon fractions yield an upper intersection age of 390 ± 6 Ma. Based on the agreement of this age with a mean of 21 Rb–Sr and K–Ar ages, the U–Pb age was suggested to reflect the age of emplacement for the Shap granite. Molybdenite from the Shap granite was used in one the first Re–Os molybdenite studies, which determined an age of 430 ± 20 Ma, using a decay constant of 1.66 × 10^-11 a^-1. Using the currently accepted decay constant (1.666 × 10^-11 a^-1), an age of 415 ± 19 Ma is calculated. Though all the age determinations overlap within uncertainty, there is a significant degree of disparity between the nominal ages (Fig. 2).

The Skiddaw granite

The granite is genetically and spatially related to the W mineralisation of the Carrock Fell deposit. Intruded into the early Ordovician metasediments of the Skiddaw group, the Skiddaw granite outcrops are restricted to three localities (Sinen Gill, Caldew Valley and Grainsgill). The latter represents the slightly smaller, though heavily mineralised and hydrothermally altered outcrop (Fig. 1). Unaltered, the Skiddaw granite is medium-grained and comprises plagioclase, orthoclase, biotite and quartz. Mineralisation is confined to the northern margin of the granite in Grainsgill and occurs in a series of steeply dipping N–S quartz veins typically ~0.5 m in thickness (Harding, Smith and Emerson veins, Fig. 1). Barren early quartz veins are cross-cut by the main ore-bearing N–S trending quartz veins that carry wolframite ± scheelite. Alteration associated with this mineralisation is minimal, with the exception of a greisen assemblage in the granite which also contains arsenopyrite ± pyrite ± apatite. Post-dating the latter is sulphide mineralisation (arsenopyrite, pyrite, chalcopyrite, molybdenite, sphalerite, bismuth) that occurs in quartz stringers in fractures of the veins and greisen zones. Calcite ± siderite ± fluorite veining post-dates sulphide mineralisation. Small E–W trending
quartz–galena–sphalerite veins cross-cut the Carrock Fell mineralisation and are thought to be part of the Caldbeck Fell Pb–Zn systems to the north.

Age determinations for the Skiddaw granite and its mineralisation are restricted to K–Ar mica and Rb–Sr fluid inclusion ages (Fig. 3). Magmatic biotite from Sinen Gill yield a mean K–Ar biotite age of 399 ± 6 Ma.9 Overlapping within uncertainty are slightly younger K–Ar ages (392 ± 4 Ma) for magmatic biotite from Sinen Gill and Caldew Valley, which are interpreted to record the final cooling of the Skiddaw granite.59 In accord with these ages is an Rb–Sr isochron age (398 ± 8 Ma) determined from K-Feldspar from altered granite; however, Rb–Sr whole-rock data yield a younger age of 383 ± 4 Ma.52 Younger ages are also reported for hydrothermal muscovite from greisen at Grainsgill and of the Harding vein (mean age of 387 ± 4 Ma),32 which overlaps, within uncertainty, with a Rb–Sr fluid inclusion age (392 ± 5 Ma) obtained from both early and late phases of quartz veining.62

The Weardale granite

This granite is a concealed body beneath the Carboniferous rocks. It is intersected at 390–5 m below the surface in the Rookhope borehole. The boundary between the Weardale granite and Carboniferous sediments is marked by a soil-like quartz–mica-rich mudstone, which is suggested to represent the weathered surface of the granite.20 (and references therein). The Carboniferous sediments are not metamorphosed, suggesting that the granite pre-dates the Carboniferous rocks. The granite is medium-grained and consists of muscovite, biotite, plagioclase, orthoclase and accessory magnetite-ilmenite, zircon and monazite. Quartz and mica define a flat lying fabric, which is suggested to represent the apex of a cupola.20 The granite contains veins of quartz pegmatite and aplite, a number of which are tourmaline-bearing.20 At 447 m, a 3 cm wide molybdenite bearing quartz pegmatite vein cuts the granite fabric. Molybdenite mineralisation is accompanied by chalcopyrite, minor bornite and pyrite (Fig. 1). No tourmaline is observed in the quartz and the relationship between tourmaline and molybdenite mineralisation is not known. An alteration selvage is absent; however, plagioclase and biotite show minor alteration to sericite and chlorite respectively.

Early age determinations have been presented from Rb–Sr and K–Ar muscovite analyses, which range from 356 ± 12 to 392 ± 6 Ma.16,26 A more recent Rb–Sr whole-rock analyses from 564 to 776 m was used to suggest that the emplacement age of the Weardale granite was 410 ± 10 Ma, intrusion of post-pegmatite vein apilte at 390 ± 8 Ma, and that hydrothermal activity continued to 365 ± 8 Ma.32

Samples and analytical protocols

Samples

Three molybdenite-bearing quartz veins were sampled from the Shap (DS1-06), Skiddaw (DS4-06) and Weardale (SSK0576) granites (Fig. 1). Suitably sized portions of each sample were used for Re–Os geochronology and fluid inclusion studies. An additional sample (DS6-06) of disseminated molybdenite from the Shap granite was also used to determine a Re–Os age. In the
Shap granite, molybdenite mineralisation occurs disseminated (DS6-06) and within about 1–2 cm quartz veins (DS1-06). Molybdenite is present as ~1 cm diameter rosettes, with individual grains of a 2–5 mm (Fig. 4), and is associated with minor chalcopyrite.

In the Skiddaw granite, the molybdenite sample (DS4-06) from the Carrock deposit was taken from the surface spoil near the adit to the Emerson vein (Fig. 4). The spoil consists largely of quartz vein and gresiened Skiddaw granite (Fig. 4) and is taken to represent the mineralisation of the Emerson vein. It is a 3 mm quartz molybdenite-bearing stringer that cross-cuts a greisen altered rock and is considered to represent sulphide mineralisation.3 Molybdenite is fine-grained (~1 mm) and occurs as single grains throughout the stringer with minor chalcopyrite and pyrite. Minor muscovite alteration is also associated with the quartz stringer, which occurs as 1–2 mm rosettes within and on the selvages of the stringer. Because the molybdenite post-dates the main W-bearing mineralisation, two ~8 kg samples of the Skiddaw granite were taken from separate localities at Sinen Gill (SK-1) and Caldew Valley (SK-2) for U–Pb zircon geochronology (Fig. 1).

In the Weardale granite, molybdenite has only been observed at 447 m in a quartz pegmatite vein in the Rookhope core (SSK0576, Fig. 4). Molybdenite is the predominant ore mineral in this pegmatite vein occurring as ~1 cm rosettes with 2–5 mm grains, and isolated grains (Fig. 4).

**Analytical protocols**

The Re–Os geochronometry was performed in the Northern Centre for Isotopic and Element Tracing (NCEIT) at Durham University. Using conventional mineral separation protocols (porcelain disc milling, FRANTZ magnetic separation, heavy liquid, water floatation) molybdenite (+44–210 µm fraction) was isolated from whole-rock and quartz vein samples.53 Isotope dilution, using a mixed tracer solution containing isotopically enriched $^{185}$Re and isotopically normal Os, in the form of a gravimetric and isotopic Os standard, was used to determine the isotopic abundance of $^{187}$Re and $^{187}$Os in analysed molybdenite. Dissolution of molybdenite and equilibration of sample and tracer Re and Os was conducted in inverse *aqua regia* using the Carius-tube method. Rhenium and Os were isolated from the inverse *aqua regia* solution using solvent
Zircon fractions were analysed by isolation dilution thermal ionisation mass spectrometry (ID-TIMS) at the NERC Isotope Geosciences Laboratory (NIGL). Analytical procedures are described in Ref. 47. Zircon crystals were separated from $< 355 \mu m$ whole-rock fraction of SK-1 and SK-2 using standard vibrating-table, specific gravity and magnetic techniques. Crystal fractions for analysis were picked by hand under a binocular microscope. Acicular zircons, typically $150 \times 50 \times 50 \mu m$ with visible melt inclusions, were preferentially selected in order to avoid crystals with inherited cores. A bulk zircon fraction was annealed at $850 \degree C$ in quartz glass beakers for 48 h. The zircon crystals were ultrasonically washed in 4 N HNO$_3$, rinsed in ultra-pure water, then further washed in warm 4 N HNO$_3$ before rinsing with distilled water to remove surface contamination. The annealed, cleaned bulk zircon fraction was then chromically abraded in 200 mL 29 N HF and 20 mL 8 N HNO$_3$ at 180°C for 12 h following a modified chemical abrasion technique to remove portions of grains which have suffered Pb-loss. Chemically abraded zircons were washed several times in ultra-pure water, in warm 3 N HCl for several hours on a hot-plate, rinsed again in ultra-pure water and 8 N HNO$_3$ and split into single grain fractions ready for dissolution. A mixed $^{205}$Pb-$^{235}$U tracer was used to spike all fractions. Representative fractions were checked in cleaned PMP beakers under fibre optic light to check for complete dissolution. Dissolved, spike-equilibrated samples were not subjected to ion-exchange procedures but were converted to chloride and loaded onto degassed rhenium filaments in silica gel following a procedure modified after. Isotope data were collected using a Thermo Electron Triton equipped with a new generation of MassCom secondary electron multiplier. A minimum of 100 ratios were collected for Pb and 60 for U. Pb ratios were scrutinised for any evidence of organic interferences which were determined to be negligible. Uncertainties were calculated using numerical error propagation. Isotope ratios were plotted using Isoplot version 3, and error ellipses and ages presented reflect 2$\sigma$ uncertainty. Total procedural blanks were between 0.2 and 0.6 pg for Pb and 0.1 pg for U. Samples were blank corrected for Pb, and any residual common Pb was corrected using a Stacey-Kramers common lead composition.

Fluid inclusion petrography and microthermometric analyses of the samples were conducted at the GeoFluids Research Laboratory, NUI, Galway. Heating and freezing runs on doubly polished fluid inclusion wafers (~100 $\mu m$ thick) were performed using a calibrated Linkam THMS600. Calibration with precision of $\pm 0.2 \degree C$ at $-56.6 \degree C$ and $\pm 1 \degree C$ at 300°C was conducted using synthetic H$_2$O and CO$_2$ standards. Following procedures outlined by Shepherd et al., the temperature of first ice melting $T_{FM}$, the temperature of last ice melting $T_{LM}$ and the temperature of homogenisation $T_H$ were measured in two-phase liquid-vapour inclusions hosted in quartz all wafers. Fluid salinities were calculated using $T_{LM}$ and the equations of Ref. 6. In addition, clathrate melting temperatures recorded in some of these two-phase inclusions were used with the equations of Ref. 18 to calculate their fluid salinities.

Laser Raman microspectroscopy (LRM) was used to determine the non-aqueous chemical species present in fluid inclusions. Analyses were conducted at CREGU.

Table 1  Re–Os isotope data for molybdenite from Shap, Weardale and Skiddaw granites

<table>
<thead>
<tr>
<th>Sample</th>
<th>Weight, mg</th>
<th>$^{187}$Re*, ppm</th>
<th>SD</th>
<th>$^{187}$Os*, ppb</th>
<th>SD</th>
<th>Age$^3$</th>
<th>SD</th>
</tr>
</thead>
<tbody>
<tr>
<td>HLP-5</td>
<td>11</td>
<td>1743 ± 0.8</td>
<td>637 ± 6</td>
<td>2.8</td>
<td>219 ± 2</td>
<td>0.9</td>
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</tr>
<tr>
<td></td>
<td>2</td>
<td>1746 ± 0.8</td>
<td>641 ± 4</td>
<td>2.9</td>
<td>220 ± 0</td>
<td>0.9</td>
<td></td>
</tr>
<tr>
<td></td>
<td>3</td>
<td>1735 ± 0.9</td>
<td>638 ± 6</td>
<td>2.9</td>
<td>220 ± 0</td>
<td>0.9</td>
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</tr>
<tr>
<td>Weardale</td>
<td>45</td>
<td>5930 ± 0.19</td>
<td>394 ± 8</td>
<td>1.0</td>
<td>398 ± 3</td>
<td>1.6</td>
<td></td>
</tr>
<tr>
<td>SSK0576</td>
<td>70</td>
<td>0.47 ± 0.0003</td>
<td>3.1</td>
<td>0.02</td>
<td>392 ± 3</td>
<td>2.8</td>
<td></td>
</tr>
<tr>
<td>Skiddaw</td>
<td>31</td>
<td>951 ± 0.03</td>
<td>641 ± 0.2</td>
<td>0.2</td>
<td>405 ± 5</td>
<td>1.7</td>
<td></td>
</tr>
<tr>
<td>DS4-06</td>
<td>21</td>
<td>961 ± 0.06</td>
<td>651 ± 0.2</td>
<td>0.2</td>
<td>405 ± 2</td>
<td>1.8</td>
<td></td>
</tr>
</tbody>
</table>

All uncertainties are quoted at the 2$\sigma$ level.

HLP-5 is an in-house control molybdenite powder (see text for discussion).

$^*$Re and Os concentrations and Re–Os age recalculated using uncertainties in Re and Os mass spectrometer measurements, standard and spike Re and Os isotopic compositions, and calibration uncertainties of $^{185}$Re and $^{187}$Os.

$^3$Age calculated using the decay constant $\lambda^{187}$Re = 1.666 x $10^{-11}$/year, without uncertainty.

$^4$Re–Os isotope data taken from Ref. 10.

$^5$Age established using $^{187}$Re = 1.612 x $10^{-11}$/year.

$^6$Age established using $^{187}$Re = 1.666 x $10^{-11}$/year.
and therefore the common lead correction.

Results

Re–Os molybdenite geochronology

Molybdenite from the Shap, Skiddaw and Weardale granites show considerable differences in 187Re and 187Os abundances (Table 1). The Weardale sample (SSK0576) has the highest 187Re (59 ppm) and 187Os (395 ppb) abundances, with Shap molybdenite (DS1-06, DS6-06) possessing significantly lower values of 9 ppm 187Re and 65 ppb187Os, and the molybdenite from the Carrock Fell deposit (Skiddaw, DS4-06) containing very low 187Re (500 ppb) and 187Os (3 \( \pm \) 1 ppb) contents.

The 187Re and 187Os systematics for all samples are used to calculate model ages with the decay constant of Ref. 58. Two molybdenite samples from Shap yield model Re–Os ages of 405 \( \pm \) 5 \( \pm \) 7 and 405 \( \pm \) 2 \( \pm \) 8 Ma (Table 1). The Weardale and Skiddaw molybdenite samples yield Re–Os ages of 398 \( \pm \) 3 \( \pm \) 6 and 392 \( \pm \) 3 \( \pm \) 2 \( \pm \) 8 Ma respectively (Table 1). All the Re–Os ages are outside the limits of uncertainty and therefore record distinct events in relation to the processes associated with the Shap, Weardale and Skiddaw granites.

Skiddaw granite U–Pb zircon geochronology

A total of eight single grain fractions, four from Sinen Gill and three from Caldew Valley, were utilised for U–Pb geochronology. These grains all possess high U abundances (549–4384 ppm; Table 2); seven grains overlap within error to give a concordia age of 398 \( \pm \) 8 Ma (concordance and equivalence: MSWD = 1.1, probability = 0.95; Fig. 5), with a single grain from the Caldew Valley (SK-2) sample slightly discordant due to Pb-loss (not shown). This is taken as the emplacement age of the granite. Previous attempts at dating the Skiddaw granite from these same localities were problematic in that multi-grain fractions of zircon showed signs of inheritance (J. Evans, personal communication, 2007; unpublished BGS data). In the case of this study, single acicular grains were selected and so inheritance was avoided. Furthermore, previous attempts at dating the Skiddaw granite using the U–Pb zircon technique utilised the air-abrasion technique to minimise the effects of Pb-loss. This would have inherently meant that delicate acicular zircons such as those used in this study would not have been preserved. Chemical abrasion has the effect of dissolving magmatic inclusions and so reducing the common lead content.42

<table>
<thead>
<tr>
<th>Sample</th>
<th>Fraction</th>
<th>Weight, mg</th>
<th>U, ppm</th>
<th>Cm–Pb, ppm</th>
<th>206Pb/204Pb</th>
<th>207Pb/206Pb</th>
<th>206Pb/238U</th>
<th>207Pb/235U</th>
<th>Rho (207Pb/206Pb)</th>
<th>Age, Ma</th>
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<tr>
<td>SK-1</td>
<td>Z1</td>
<td>2.6549</td>
<td>0.95</td>
<td>9.91</td>
<td>0.05468</td>
<td>0.09</td>
<td>0.06380</td>
<td>0.16</td>
<td>0.48178</td>
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<tr>
<td>SK-2</td>
<td>Z1</td>
<td>2.2389</td>
<td>1.03</td>
<td>9.67</td>
<td>0.05479</td>
<td>0.06</td>
<td>0.06375</td>
<td>0.17</td>
<td>0.48161</td>
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</tr>
<tr>
<td>SK-2</td>
<td>Z2</td>
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<td>0.57</td>
<td>3.78</td>
<td>0.05465</td>
<td>0.14</td>
<td>0.06380</td>
<td>0.17</td>
<td>0.48130</td>
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<tr>
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<td>Z3</td>
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<td>0.48161</td>
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<td>0.06</td>
<td>0.06380</td>
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<tr>
<td>SK-3</td>
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<td>0.06375</td>
<td>0.19</td>
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<td>0.06</td>
<td>0.06380</td>
<td>0.18</td>
<td>0.48195</td>
<td>402.7</td>
</tr>
</tbody>
</table>

All errors are 2\( \sigma \) (per cent for ratios; absolute for ages). Total common Pb in analysis corrected for spike and fractionation (0.05%/amu).

Table 2. U–Pb zircon ID-TIMS data for the Skiddaw granite.
Fluid inclusions

The fluid inclusion petrographic study adopted the concept of fluid inclusion assemblages (FIA) described by Goldstein,27 an approach that places fluid inclusions into assemblages interpreted to have been trapped at around the same time. Three types (types 1, 2 and 3) of aqueous fluid inclusion have been recognised in all three samples based on phases present at room temperature and their microthermometric behaviour (Table 3 and Fig. 6). Type 1 inclusions are vapour-rich, degree of fill $F$ is $0 \div 0.62$ ($F = \text{vol. liquid/\text{vol. liquid + vapour}}$). Type 2 are liquid rich, $F = 0.47 \div 0.87$. The majority of types 1 and 2 fluid inclusions form trails that do not crosscut crystal boundaries and are classified here as pseudosecondary based on the criteria of McCandless et al.40 In addition, a number of types 1 and 2 inclusions are hosted in clusters within the cores of quartz crystals. The authors interpret types 1 and 2 to constitute one FIA in each of the quartz vein samples from Shap and Skiddaw. In the Weardale pegmatite vein, the vapour rich type 1 inclusions are uncommon and type 2 inclusions invariably occur in annealed fractures without type 1.

The temperature of the first ice melting $T_{FM}$ in type 1 (only recorded in the Skiddaw sample) and type 2 inclusions occurs between $-19.5$ and $-23.7^\circ C$ indicating the dominant presence of NaCl in solution. The temperature of last ice melting $T_{LM}$ in types 1 and 2 inclusions (from $-0.1$ to $-5.2^\circ C$) was used to calculate fluid salinities in equivalent wt-%NaCl (eq. wt-%NaCl). The range of salinities for types 1 and 2 are broadly similar, i.e. $0.2 \div 8.1$ eq. wt-%NaCl (Table 3).

In both the Shap and Skiddaw vein samples (DS1-06 and DS4-06), clathrate melting between $+5.8$ and $+10^\circ C$ was observed in some type 1 (DS1-06 only) and type 2 inclusions. This indicates the presence of non-aqueous phases in types 1 and 2 inclusions (see the section on ‘Laser Raman microspectroscopy’), which can affect calculated fluid salinities.30 Clathrate melting temperatures and bulk fluid inclusion compositions were used to calculate fluid salinities of $3.3 \div 4.7$ eq.wt-%NaCl, within the range calculated from $T_{LM}$ values.

Homogenisation to the vapour phase takes place between $31.2$ and $44.5^\circ C$ in type 1 inclusions, whereas in type 2, homogenisation is to the liquid phase and takes place between $24.6$ and $42.5^\circ C$ (Fig. 7 and Table 3). On heating and before homogenisation, several types 1 and 2 inclusions in all samples decrepitated between $23.0$ and $43.8^\circ C$ (Table 3).

Type 3 fluid inclusions are liquid rich, $F = 0.71 \div 0.99$, and unlike types 1 and 2, occur along annealed fractures that cross-cut crystal boundaries and are considered to be secondary in origin. They occur in all three vein samples. Their $T_{FM}$ values range between $-19.4$ and $-25.2^\circ C$, thus are interpreted to reflect the dominance of NaCl in solution. $T_{LM}$ values range from $-0.2$ to $-4.9^\circ C$, yielding salinity values between $0.4$ and $7.7$ eq. wt-%NaCl (Fig. 8). Homogenisation to the liquid phase occurs between $86.6$ and $296.7^\circ C$.

Laser Raman microspectroscopy

Laser Raman microspectroscopy (LRM) of types 1 and 2 inclusions in the three vein samples from the Shap, Skiddaw and Weardale granites revealed the presence of...
Table 3 Fluid inclusion microthermometric data from the Shap, Skiddaw and Weardale granites

<table>
<thead>
<tr>
<th>Fluid inclusion type</th>
<th>Shap granite (DS1-06)</th>
<th>Skiddaw granite (DS4-06)</th>
<th>Weardale granite (SSK0576)</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Type 1: 4–30 μm: L+V (V-rich):</strong></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>occur in clusters in the core of quartz crystals (unknown origin) and trails that do not crosscut crystal boundaries (pseudosecondary): ellipsoidal or irregular shapes</td>
<td>Abundance: 23%</td>
<td>Abundance: 25%</td>
<td>Abundance: 3%</td>
</tr>
<tr>
<td></td>
<td>Fill: 0.18–0.54 (26)</td>
<td>Fill: 0.11–0.62 (22)</td>
<td>Fill: 0.43–0.61 (4)</td>
</tr>
<tr>
<td></td>
<td>( T_{FM} ): (-7) to (-4^\circ C) (9)</td>
<td>( T_{FM} ): (-20.1) to (-21^\circ C) (2)</td>
<td>( T_{LM} ): (-0.8^\circ C) (1) (1 eq. wt.%NaCl)</td>
</tr>
<tr>
<td></td>
<td>( T_{LM} ): (-4^\circ C) to (-2.3^\circ C) (9)</td>
<td>( T_{LM} ): (-0.4^\circ C) to (-2^\circ C) (9)</td>
<td>( T_{LM} ): (-0.8^\circ C) (1) (1 eq. wt.%NaCl)</td>
</tr>
<tr>
<td></td>
<td>( T_{LM} ): (-3.9^\circ C) eq. wt.%NaCl</td>
<td>( T_{LM} ): (-0.4^\circ C) to (-2^\circ C) (9)</td>
<td>( T_{LM} ): (-2.3^\circ C) (1) (1 eq. wt.%NaCl)</td>
</tr>
<tr>
<td></td>
<td>( T_{r\rightarrow V} ): 357.7–448.4(^\circ C) (19)</td>
<td>( T_{r\rightarrow V} ): 312.4–445.6(^\circ C) (21)</td>
<td>( T_{r\rightarrow V} ): 3577–448.4(^\circ C) (3)</td>
</tr>
<tr>
<td></td>
<td>( T_D ): 366.6–424.5(^\circ C) (7)</td>
<td>( T_D ): 408.3(^\circ C) (21)</td>
<td>( T_D ): 302–438.6(^\circ C) (24)</td>
</tr>
<tr>
<td>Abundance: 23%</td>
<td>Abundance: 25%</td>
<td>Abundance: 3%</td>
<td></td>
</tr>
<tr>
<td><strong>Type 2: 3–30 μm: L+V (L-rich):</strong></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>occur in clusters in the core of quartz crystals (unknown origin) and trails that do not crosscut crystal boundaries (pseudosecondary): ellipsoidal, irregular or negative crystal shapes</td>
<td>Abundance: 46%</td>
<td>Abundance: 37%</td>
<td>Abundance: 36%</td>
</tr>
<tr>
<td></td>
<td>Fill: 0.47–0.62 (53)</td>
<td>Fill: 0.51–0.87 (33)</td>
<td>Fill: 0.4–0.86 (41)</td>
</tr>
<tr>
<td></td>
<td>( T_{FM} ): (-19.9) to (-22.6^\circ C) (8)</td>
<td>( T_{FM} ): (-19.5) to (-22^\circ C) (5)</td>
<td>( T_{FM} ): (-21.2) to (-23.7^\circ C) (9)</td>
</tr>
<tr>
<td></td>
<td>( T_{LM} ): (-8.1) to (-5^\circ C) (24)</td>
<td>( T_{LM} ): (-0.3) to (-5.2^\circ C) (17)</td>
<td>( T_{LM} ): (-0.1) to (-3^\circ C) (24)</td>
</tr>
<tr>
<td></td>
<td>( T_{LM} ): (-4.7) eq. wt.%NaCl</td>
<td>( T_{LM} ): (-0.5) eq. wt.%NaCl</td>
<td>( T_{LM} ): (-0.5) eq. wt.%NaCl</td>
</tr>
<tr>
<td></td>
<td>( T_{r\rightarrow L} ): 250.7–418.9(^\circ C) (29)</td>
<td>( T_{r\rightarrow L} ): 251.4–397.4(^\circ C) (28)</td>
<td>( T_{r\rightarrow L} ): 246–425(^\circ C) (30)</td>
</tr>
<tr>
<td></td>
<td>( T_D ): 302–438.6(^\circ C) (24)</td>
<td>( T_D ): 302–438.6(^\circ C) (24)</td>
<td>( T_D ): 302.6–425(^\circ C) (10)</td>
</tr>
<tr>
<td>Abundance: 46%</td>
<td>Abundance: 37%</td>
<td>Abundance: 36%</td>
<td></td>
</tr>
<tr>
<td><strong>Type 3: 2–25 μm: L+V (L-rich):</strong></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>occur in trails that crosscut crystal boundaries (secondary): ellipsoidal or irregular shapes</td>
<td>Abundance: 31%</td>
<td>Abundance: 38%</td>
<td>Abundance: 61%</td>
</tr>
<tr>
<td></td>
<td>Fill: 0.87–0.96 (35)</td>
<td>Fill: 0.82–0.97 (34)</td>
<td>Fill: 0.71–0.99 (69)</td>
</tr>
<tr>
<td></td>
<td>( T_{FM} ): (-20.6) to (-25^\circ C) (8)</td>
<td>( T_{FM} ): (-19.4) to (-23.6^\circ C) (7)</td>
<td>( T_{FM} ): (-20.1) to (-25.2^\circ C) (12)</td>
</tr>
<tr>
<td></td>
<td>( T_{LM} ): (-6.7) to (-4.1^\circ C) (22)</td>
<td>( T_{LM} ): (-18) to (-3.4^\circ C) (21)</td>
<td>( T_{LM} ): (-0.2) to (-4.9^\circ C) (57)</td>
</tr>
<tr>
<td></td>
<td>( T_{LM} ): (-6.6) eq. wt.%NaCl</td>
<td>( T_{LM} ): (-3.1) eq. wt.%NaCl</td>
<td>( T_{LM} ): (-0.4)–(-7.7) eq. wt.%NaCl</td>
</tr>
<tr>
<td></td>
<td>( T_{r\rightarrow L} ): 101.6–201.8(^\circ C) (35)</td>
<td>( T_{r\rightarrow L} ): 86.6–250.3(^\circ C) (33)</td>
<td>( T_{r\rightarrow L} ): 140–236.7(^\circ C) (69)</td>
</tr>
<tr>
<td>Abundance: 31%</td>
<td>Abundance: 38%</td>
<td>Abundance: 61%</td>
<td></td>
</tr>
</tbody>
</table>

\( T_{FM} \): first ice melting; \( T_{LM} \): last ice melting; \( T_{r\rightarrow V} \): homogenisation temperature to vapour; \( T_{r\rightarrow L} \): homogenisation temperature to liquid; \( T_D \): decrepitation temperature; \( F \): degree of fill. Round brackets indicate the number of measurements. The degree of fill was measured by estimating the proportions of liquid and vapour at 25\(^\circ C\) and comparing to published reference charts.61 The criteria of Ref. 27 was used to determine the genetic relationship between inclusions and their host quartz crystals.
the non-aqueous phase, of CO₂, CH₄, N₂ and H₂S (Table 4 and Fig. 9). Types 1 and 2 inclusions in the vein quartz sample from the Shap granite have broadly similar and dominant abundances of CO₂ (about 90–99 mol.-%) with subordinate to trace amounts of N₂, CH₄ and trace H₂S. In contrast, types 1 and 2 inclusions hosted in vein quartz from the Skiddaw granite contain less CO₂ (about 41–85 mol.-%) coupled with higher abundances of CH₄ (about 3–15 mol.-%) and N₂ (11–44 mol.-%). Laser Raman microspectroscopy (LRM) of type 2 inclusions hosted in pegmatite vein quartz from the Weardale granite did not detect the presence of CO₂, CH₄, N₂ or H₂S (Table 4).

Discussion

Re–Os molybdenite and U–Pb zircon geochronology

Numerous studies have shown that the Re–Os systematics in molybdenite are robust, being unaffected by post-ore hydrothermal, metamorphic and magmatic activity. As a result, Re–Os molybdenite dates yield the timing of sulphide mineralisation. In mineral systems

7

Photomicrographs of fluid inclusion types

![Photomicrographs of fluid inclusion types](image)

6 Photomicrographs of fluid inclusion types

![Photomicrographs of fluid inclusion types](image)

5

<table>
<thead>
<tr>
<th>Shap (DS1-06)</th>
<th>Skiddaw (DS4-06)</th>
<th>Weardale (SSK0576)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Type 1 (n = 19)</td>
<td>Type 1 (n = 21)</td>
<td>Type 1 (n = 3)</td>
</tr>
<tr>
<td>Type 2 (n = 29)</td>
<td>Type 2 (n = 28)</td>
<td>Type 2 (n = 30)</td>
</tr>
<tr>
<td>Type 3 (n = 35)</td>
<td>Type 3 (n = 33)</td>
<td>Type 3 (n = 69)</td>
</tr>
</tbody>
</table>

7 $T_h$ histograms for types 1–3 fluid inclusions in northern England
were considered reliable. Subsequently, the Rb–Sr data can provide a minimum age for magmatism.

The crystallisation of plagioclase, supporting magma and interacted with a less evolved, mafic magma during K-feldspar formed in a more evolved granitic magma at 397 Ma of the K-feldspar (0.70739). This feature was considered to reflect that the initial Sr isotope composition is less isotopic heterogeneity in the magmas associated with the Skiddaw granite and that the initial Sr composition is less radiogenic (more juvenile) than previous determinations (cf. 0.70767). This re-interpretation of the Rb–Sr data further stresses that the absolute age and its uncertainty for granite emplacement is critical when applying Rb–Sr data to calculate initial isotopic compositions to evaluate magma sources and processes in ancient systems.

**The Shap granite**

The Re–Os ages for disseminated (405±2 ± 1-8 Ma) and vein-hosted (405-5±1-7 Ma) molybdenite are older and outside uncertainty of previous U–Pb, Rb–Sr and K–Ar age constraints (Fig. 2). The Re–Os ages suggest that magmatism and mineralisation occurred contemporaneously and ~5 m.y. earlier than previously considered (Fig. 2). The previous U–Pb, Rb–Sr and K–Ar age determinations likely reflect disturbance to the isoype systematics. The latter maybe accounted for by the alteration of matrix magmatic, plagioclase and biotite to sericite and chlorite respectively, which is not observed in megacrysts within enclaves. Remarkably, given the employed analytical protocols at the time, the recalculated 1964 Re–Os molybdenite age is within uncertainty of the new Re–Os age (Fig. 2).

The new Re–Os molybdenite age (405 Ma) allows for the reinterpretation of a recent Rb–Sr dataset from a single plagioclase-rimmed K-feldspar megacryst within a mafic inclusion (enclave). The isochronous behaviour of the Rb–Sr data (405±2 Ma), which requires the sample set to possess identical initial Sr isotope compositions, was not considered to reflect the timing of emplacement of the granite because the K–Ar age constraints were considered reliable. Subsequently, the Rb–Sr data were used to show that the initial Sr isotope composition at 397±7 Ma and U–Pb (390±6 Ma) determinations were considered reliable. Subsequently, the Rb–Sr data were used to show that the initial Sr isotope composition at 397±7 Ma and U–Pb (390±6 Ma) determinations were considered reliable. Subsequently, the Rb–Sr data were used to show that the initial Sr isotope composition at 397±7 Ma and U–Pb (390±6 Ma) determinations were considered reliable. Subsequently, the Rb–Sr data were used to show that the initial Sr isotope composition at 397±7 Ma and U–Pb (390±6 Ma) determinations were considered reliable.

**The Weardale granite**

The Re–Os age for disseminated (405±2 ± 1-8 Ma) and vein-hosted (405-5±1-7 Ma) molybdenite are older and outside uncertainty of previous U–Pb, Rb–Sr and K–Ar age constraints (Fig. 2). The Re–Os ages suggest that magmatism and mineralisation occurred contemporaneously and ~5 m.y. earlier than previously considered (Fig. 2). The previous U–Pb, Rb–Sr and K–Ar age determinations likely reflect disturbance to the isoype systematics. The latter maybe accounted for by the alteration of matrix magmatic, plagioclase and biotite to sericite and chlorite respectively, which is not observed in megacrysts within enclaves. Remarkably, given the employed analytical protocols at the time, the recalculated 1964 Re–Os molybdenite age is within uncertainty of the new Re–Os age (Fig. 2).

The new Re–Os molybdenite age (405 Ma) allows for the reinterpretation of a recent Rb–Sr dataset from a single plagioclase-rimmed K-feldspar megacryst within a mafic inclusion (enclave). The isochronous behaviour of the Rb–Sr data (405±2 Ma), which requires the sample set to possess identical initial Sr isotope compositions, was not considered to reflect the timing of emplacement of the granite because the K–Ar age constraints were considered reliable. Subsequently, the Rb–Sr data were used to show that the initial Sr isotope composition at 397±7 Ma and U–Pb (390±6 Ma) determinations were considered reliable. Subsequently, the Rb–Sr data were used to show that the initial Sr isotope composition at 397±7 Ma and U–Pb (390±6 Ma) determinations were considered reliable.
Table 4 Chemical composition of the non-aqueous phase of types 1 and 2 inclusions in the Shap, Skiddaw and Weardale granites, obtained from Raman microspectrometry: bulk fluid inclusion parameters calculated for inclusions on which full microthermometric analysis was obtained.

<table>
<thead>
<tr>
<th>Host granite</th>
<th>F.I. no.</th>
<th>F.I. type</th>
<th>CO₂</th>
<th>CH₄</th>
<th>N₂</th>
<th>H₂S</th>
<th>Fill</th>
<th>T_LM</th>
<th>T_MC</th>
<th>T_H</th>
<th>XH₂O</th>
<th>XCO₂</th>
<th>XCH₄</th>
<th>XN₂</th>
<th>XNaCl</th>
<th>Molar volumea, cm³ mol⁻¹</th>
</tr>
</thead>
<tbody>
<tr>
<td>Shap (DS1-06)</td>
<td>1 1</td>
<td>99</td>
<td>0.3</td>
<td>&lt;1</td>
<td>&lt;0.1</td>
<td>0.24</td>
<td>5.9</td>
<td>383.4 (V)</td>
<td></td>
<td></td>
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<td></td>
</tr>
<tr>
<td></td>
<td>2 1</td>
<td>95.5</td>
<td>&lt;0.3</td>
<td>4</td>
<td>0.4</td>
<td>0.22</td>
<td>−1.1</td>
<td>381.5 (V)</td>
<td></td>
<td></td>
<td></td>
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<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>3 1</td>
<td>90</td>
<td>&lt;0.3</td>
<td>10</td>
<td>n.d.</td>
<td></td>
<td></td>
<td></td>
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<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>4 2</td>
<td>91.5</td>
<td>2.5</td>
<td>6</td>
<td>0.4</td>
<td>0.71</td>
<td>−2.5</td>
<td>7.2</td>
<td>344.6 (L)</td>
<td>93.32</td>
<td>3.75</td>
<td>0.04</td>
<td>0.09</td>
<td>2.80</td>
<td>24.96</td>
<td></td>
</tr>
<tr>
<td></td>
<td>5 2</td>
<td>96</td>
<td>&lt;0.3</td>
<td>4</td>
<td>n.d.</td>
<td>0.72</td>
<td>−1.3</td>
<td>329.4 (L)</td>
<td></td>
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<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>6 1</td>
<td>97</td>
<td>&lt;0.3</td>
<td>3</td>
<td>n.d.</td>
<td>0.51</td>
<td>7</td>
<td>357.2 (V)</td>
<td></td>
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<tr>
<td></td>
<td>7 2</td>
<td>95.5</td>
<td>&lt;0.3</td>
<td>4</td>
<td>0.1</td>
<td></td>
<td></td>
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<tr>
<td></td>
<td>8 2</td>
<td>96</td>
<td>&lt;0.3</td>
<td>3.5</td>
<td>&lt;0.1</td>
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<td>−2.8</td>
<td>6</td>
<td>369.1 (L)</td>
<td>93.31</td>
<td>3.79</td>
<td>0.01</td>
<td>0.06</td>
<td>2.83</td>
<td>27.25</td>
<td></td>
</tr>
<tr>
<td></td>
<td>9 2</td>
<td>89.5</td>
<td>0.5</td>
<td>10</td>
<td>n.d.</td>
<td>0.59</td>
<td>−3.1</td>
<td>6.9</td>
<td>375.7 (L)</td>
<td>92.64</td>
<td>4.49</td>
<td>0.01</td>
<td>0.25</td>
<td>2.61</td>
<td>29.75</td>
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<tr>
<td></td>
<td>10 1</td>
<td>93.5</td>
<td>&lt;0.3</td>
<td>6</td>
<td>n.d.</td>
<td></td>
<td></td>
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<tr>
<td></td>
<td>11 1</td>
<td>98</td>
<td>&lt;0.3</td>
<td>18</td>
<td>&lt;0.1</td>
<td>0.26</td>
<td>−1.4</td>
<td>398.7 (V)</td>
<td></td>
<td></td>
<td></td>
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<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>12 1</td>
<td>94</td>
<td>&lt;0.3</td>
<td>5.5</td>
<td>&lt;0.1</td>
<td>0.54</td>
<td>−1.3</td>
<td>7.5</td>
<td>404 (V)</td>
<td>92.03</td>
<td>5.27</td>
<td>0.01</td>
<td>0.17</td>
<td>2.52</td>
<td>32.31</td>
<td></td>
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<tr>
<td></td>
<td>13 1</td>
<td>95</td>
<td>&lt;0.3</td>
<td>4.5</td>
<td>0.3</td>
<td>0.54</td>
<td>−1.4</td>
<td>392 (V)</td>
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<td></td>
</tr>
<tr>
<td></td>
<td>14 1</td>
<td>99</td>
<td>&lt;0.3</td>
<td>&lt;1</td>
<td>0.1</td>
<td></td>
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<tr>
<td>Skiddaw (DS4–06)</td>
<td>15 1</td>
<td>61</td>
<td>11</td>
<td>28</td>
<td>n.d.</td>
<td>0.36</td>
<td>−1</td>
<td>6.8</td>
<td>408* (D)</td>
<td>89.12</td>
<td>6.14</td>
<td>0.79</td>
<td>2.00</td>
<td>1.96</td>
<td>46.41</td>
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<tr>
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<td>16 2</td>
<td>41</td>
<td>15</td>
<td>44</td>
<td>n.d.</td>
<td>0.65</td>
<td>−1.7</td>
<td>230.4* (D)</td>
<td></td>
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</tr>
<tr>
<td></td>
<td>17 2</td>
<td>77</td>
<td>12</td>
<td>11</td>
<td>n.d.</td>
<td>0.59</td>
<td>−1.7</td>
<td>10.2</td>
<td>353.1 (L)</td>
<td>91.25</td>
<td>5.63</td>
<td>0.52</td>
<td>0.47</td>
<td>2.12</td>
<td>30.16</td>
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<tr>
<td></td>
<td>18 1</td>
<td>70</td>
<td>10</td>
<td>20</td>
<td>n.d.</td>
<td>0.28</td>
<td>−1</td>
<td>6.8</td>
<td>386.7 (V)</td>
<td>87.49</td>
<td>8.15</td>
<td>0.91</td>
<td>1.80</td>
<td>1.65</td>
<td>60.21</td>
<td></td>
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<tr>
<td></td>
<td>19 2</td>
<td>85</td>
<td>3</td>
<td>12</td>
<td>n.d.</td>
<td>0.67</td>
<td>−1.4</td>
<td>6.9</td>
<td>357.3 (L)</td>
<td>93.00</td>
<td>3.90</td>
<td>0.06</td>
<td>0.23</td>
<td>2.80</td>
<td>27.26</td>
<td></td>
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<tr>
<td>Weardale (SSK0576)</td>
<td>20 2</td>
<td>n.d.</td>
<td>n.d.</td>
<td>n.d.</td>
<td>n.d.</td>
<td>0.72</td>
<td>−0.6</td>
<td>390.8* (D)</td>
<td>99.35</td>
<td>0.65</td>
<td>39.67</td>
<td></td>
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<td></td>
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<tr>
<td></td>
<td>21 2</td>
<td>n.d.</td>
<td>n.d.</td>
<td>n.d.</td>
<td>n.d.</td>
<td>0.68</td>
<td>395* (D)</td>
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<tr>
<td></td>
<td>22 2</td>
<td>n.d.</td>
<td>n.d.</td>
<td>n.d.</td>
<td>n.d.</td>
<td>0.55</td>
<td>−1</td>
<td>387</td>
<td>96.92</td>
<td>1.08</td>
<td>37.45</td>
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Composition of the non-aqueous phase obtained from Raman microprobe spectrometry. F.I. = fluid inclusion; n.d. = not detected. T_LM = last ice melting temperature (°C); T_MC = clathrate melting temperature (°C); T_H = homogenisation temperature (°C). *Temperature of decrepitation. †Bulk fluid inclusion parameters calculated using the ICE-CLATHRATE program.
relationship with the Weardale magma. In addition, though within uncertainty, the reporting of two present day $^{87}\text{Sr}/^{86}\text{Sr}$ values (DH7, 11) may suggest some uncertainty in the Sr data reported by Holland and Lambert and subsequently the determined Rb–Sr age for the Weardale granite.

Fluids (types 1–3) in the Mo-bearing veins from Shap, Skiddaw and Weardale granites

The Mo-bearing veins from each of the three granites contain a similar range of fluid inclusion types, i.e. types 1–3. Microthermometry of all three types indicates similar fluid salinities that range between 0 and 8 eq. wt.%NaCl. $T_H$ histograms (Fig. 7) for types 1–3 fluid inclusions indicate a progressive decrease in homogenisation temperatures from type 1 ($\sim$450°C) through to type 3 ($\sim$150°C). Bivariate plots of $T_H$ versus salinity, however, show no obvious correlations between salinity and homogenisation temperatures (Fig. 8). Significant differences in fluid compositions between the three veins are revealed, however, by combining the LRM data with the microthermometric data.

Vapour-rich type 1 and liquid-rich type 2 inclusions in the Shap and Skiddaw veins are found in the same FIA and $T_H$ measurements from coexisting types 1 and 2 inclusions are identical ($\pm 10^\circ$C). This suggests that types 1 and 2 inclusions were trapped from immiscible fluids (i.e. boiling) during vein emplacement and Mo mineralisation.

The non-aqueous phases in types 1 and 2 inclusions from the Shap vein are dominated by CO$_2$ (89–99 mol.-%) with low abundances of N$_2$ (10 mol.-%) and CH$_4$ (1 mol.-%; Table 3 and Fig. 9). The presence of CO$_2$ combined with the low levels of N$_2$ in these fluids (mean $\pm 1$ equiv.) suggests that these fluids are magmatic in origin. A predominant magmatic origin for mineralising fluids is also supported by $^{34}$S values for sulphide minerals from 1 to 36%, consistent with I-type magmas.

The types 1 and 2 inclusions in the Skiddaw vein contain relatively high abundances of CH$_4$ (up to 15 mol.-%) and N$_2$ (up to 44 mol.-%) in comparison to the Shap vein (Table 3 and Fig. 8). The high N$_2$ content in these fluids indicates an influx of metamorphic fluids. Nitrogen enrichment is also a feature in the greisen altered Skiddaw Granite and this together with the recorded decrease in countryrock N$_2$ from >800 ppm (2.5 km away from the contact) to <410 ppm (0.5 km from the contact) is further evidence of countryrock involvement in fluid compositions. The latter may also be suggested by the sub-ppm Re concentrations (0.75 ppm 0.47 ppm $^{187}$Re, Table 1) in molybdenite from the Carrock deposit (DS4-06). For example, low Re-bearing (<5 ppm) molybdenite associated with W and Sn mineralisation is common to metamorphic-related environments and ore-fluids. However, molybdenite associated with magmatic-related mineral systems typically has high Re concentrations (tens to thousands of ppm) (cf. Refs. 40 and 56, and references therein and the Shap and Weardale granites of this study).
Type 1 inclusions are very rare in the Weardale granite’s Mo-bearing quartz pegmatite vein. The type 2 inclusions invariably occur in annealed fractures. Furthermore, the lack of obvious clathration in types 1 and 2 inclusions coupled with the LRM data for type 2 inclusion fluids indicates significant differences in their fluid composition to those recorded in the Shap and Skiddaw veins. However, the types 1 and 2 inclusions are found in pseudosecondary trails and represent fluids present during the crystallisation of this pegmatite. As type 1 inclusions make up only ~3% of the total fluid inclusion population, fluid boiling in the Weardale granite was relatively minor and this may be related to the lack of gaseous species within mineralising fluids (as indicated by LRM). Based on the petrographic setting of type 2 inclusions, they are interpreted as representing late-stage magmatic fluids containing little or no non-aqueous species.
The salinities of type 3 inclusion fluids from all three veins are similar to the salinities of types 1 and 2 inclusion fluids. However, their homogenisation temperatures are relatively low (Fig. 8). Furthermore, type 3 inclusions are confined to annealed fractures reflecting lower temperature fluid influxes that may be meteoric in origin and similar to meteoric fluids recorded in other late-Caledonian granites.12,24,49

**P–T modelling of fluid trapping conditions**

The presence of vapour-rich (type 1) and liquid-rich (type 2) inclusions in the Shap and Skiddaw veins may reflect fluid boiling. This may also be true in the case of the Weardale vein although type 1 inclusions are rare. If it is assumed that boiling occurred in the Shap and Skiddaw veins, then \( T_H \) equates to the true trapping temperature \( T_T \) of the fluids. In addition, trapping pressures of inclusions can be calculated using the volume and bulk composition of fluids (Table 3).

Trapping temperatures and pressures of types 1 and 2 inclusions in the Shap and Skiddaw veins were calculated18 and are plotted in Fig. 10a. Calculated trapping pressures (assuming a hydrostatic pressure gradient of 98.1 bar km\(^{-1}\)) for the Shap and Skiddaw veins’ inclusions are between 300 and 620 bars (3–6 km) and 375–475 bars (3.8–4.9 km) respectively. These emplacement depths are consistent with fluid inclusion data from wolframite-bearing quartz veins in the Skiddaw granite, which indicates that mineralisation occurred at depths >1.4 km.3

The rarity of vapour-rich type 1 inclusions in the Weardale vein is considered to reflect the absence of boiling and therefore, a pressure correction is required to determine the fluid trapping conditions. In Fig. 10b, isochores were constructed for individual type 2 FIA using the program FLUIDS.2 Assuming that the Weardale granite was emplaced at a similar crustal level to the Shap and Skiddaw granites (3–6 km), then type 2 inclusions were trapped at lithostatic pressures of 1–1.7 kbars and were trapped at progressively lower temperatures (about 650–350 °C) during isobaric cooling.

**Implications**

New Re-Os molybdenite ages indicate that emplacement and mineralisation of the Shap, Skiddaw and Weardale granites occurred during the upper Early Devonian (Emsian) and not during the lower Middle Devonian (Eifelian) (Fig. 11). However, mineralisation associated with the Carrock deposit (Skiddaw granite), in part, occurred during the Eifelian (Fig. 11). Additionally, the Shap granite has long been recognised to post-date deformation (main cleavage development) as shown by contact metamorphic minerals, at several localities, occurring in random orientation across the cleavage of the country rocks.7 In contrast, in the Skiddaw, aureole andalusite both pre- and post-dates the main cleavage.64 Previous attempts to constrain the timing of the Acadian tectonometamorphic events have been less definitive in northern England and Wales because of: the range in Rb–Sr, K–Ar and Ar–Ar ages; the behaviour of the isotopic systems; and the interpretation of the ages.17,21,22,43 Broadly, K–Ar and Ar–Ar ages suggest apparent contemporaneous deformation at 400 Ma.17,21,22,43,65 The Re–Os and U–Pb ages of this study establish the non-synchronous development of cleavage of the Acadian deformation front in the Lake District occurred before ~405 Ma (emplacement of the Shap granite) and continued episodically throughout the Emsian (~398 Ma, Skiddaw granite; Fig. 11).

The broad timing and fluid characteristics of the granite emplacement and Mo-mineralisation in the Shap, Skiddaw and Weardale granites coincide with that of the Irish sector of the C-AO (Fig. 11). For example, combined fluid inclusion data and U–Pb and Re–Os geochronology have shown that granite-related molybdenite mineralisation in the Connemara region was accompanied by aqueous–carbonic fluids in the Omey Granite at ~422 Ma and later, in the western sector of the Galway Batholith at ~410 Ma (Murvey) and ~407 Ma (Mace Head).23,28,49 Further magmatism and W–Sn–sulphide mineralisation is associated with the ~405 Ma Leinster granite, SE Ireland.49,50

The diachronous granite emplacement between northern England and Irish sector of the C-AO during the Acadian phase may relate to the oblique ‘soft’ collision plate convergence of eastern Avalonia with Laurentia15,66 associated with sinistral transtension during the Emsian.15,65 For example, in the Connemara region, granite-related Mo-mineralisation systems were initiated when orogen parallel structures (e.g. Great Glen and Southern Upland Faults) were active.15 They continued as part of the stitching Galway Batholith’s evolution (410–380 Ma) when these major lineaments had ceased to be active.25 In contrast, south of the Iapetus suture granite systems of northwest England and southeast Ireland were emplaced following the main Acadian deformation front.65

**Conclusions**

Molybdenite Re–Os and U–Pb zircon geochronology and fluid inclusion data presented in this study define the timing and fluid characteristics associated with sulphide mineralisation in the Lake District, and the nature and relationship between mineralisation and magmatism.

The Re–Os ages for mineralisation in the Shap Granite range are 405±2±1.8 Ma (disseminated molybdenite) and 405±5±1.7 Ma (vein-hosted molybdenite). These ages are older than previously published U–Pb, Rb–Sr and K–Ar age constraints and indicate that mineralisation and magmatism occurred during the Emsian. Fluid inclusion analysis shows that mineralisation in the Shap Granite is associated with the boiling of CO₂-rich late magmatic fluids between 300 and 450 °C, at hydrostatic pressures of between 440 and 620 bars.

Zircon U–Pb geochronology yields an emplacement age of the Skiddaw granite at 398±8±0.4 Ma. Molybdenite mineralisation that post-dates the main W mineralisation yields a Re–Os age of 392±3±2.8 Ma. The non-aqueous compositions of mineralising fluids and Re concentrations in molybdenite are consistent with a non-magmatic source for sulphide mineralisation, in contrast to earlier W mineralisation.3,59 Instead, mineralising fluids extensively interacted with the surrounding country rocks, becoming enriched in isotopically heavy sulphur, N₂ and CH₄. The elevated non-aqueous component of these fluids (up to 12.5 wt-%) facilitated boiling between 275 and 400 °C and at hydrostatic pressures of 375–475 bars.

Mineralisation in the Weardale granite is hosted in a molybdenite bearing quartz pegmatite vein, and yields a Re–Os age of 398.3±1.6 Ma. No evidence for fluid
boiling has been recorded, and mineralising fluids represent late stage magmatic fluids trapped at high temperatures (476–577 °C) and at lithostatic pressures of 1–1.7 kbars.

The new U–Pb and Re–Os ages reported in this study place constraints on the timing of main Acadian deformation in the Lake District, the eastern Avalonia sector of the C-AO. Acadian deformation was episodic and diachronous beginning before the emplacement of the Shap Granite at ~405 Ma, and continued episodically throughout the Emsian (~398 Ma, emplacement of the Skiddaw granite). This study confirms the notion that the timing of magmatism in northern England also coincides with that of the Irish sector of Avalonia.

Acknowledgements

D. Selby would like to acknowledge the Nuffield Bursary placement for secondary school students (SETPPOINT North East Centre for Lifelong Learning) through which the initial stages of this study were undertaken by S. Baggaley, F. McEvoy and M. Howe of the British Geological Survey are thanked for providing the molybdenite sample from the Weardale granite. T. Lhommme, at CREGU, Nancy, France, is thanked for conducting the Raman microspectroscopy analysis. N. Blamey, NUIG, is acknowledged for discussions on fluid evolution. J. LaFace and C. Thomson are thanked for their assistance in the field and for sample preparation. J. Evans is thanked for giving access to the previously collected Skiddaw granite samples, N. Boulton and A. Summer are thanked for technical assistance at NIGL. The U–Pb ID-TIMS portion of this study was funded by the BGS-NIGL programme (continental arc volcanism, fluid evolution and metallogenesis in the Lake District, project no. 40200). M. Bussell and an anonymous Applied Earth Science reviewer are thanked for their comments.

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Quentin Crowley graduated with a BSc degree (Hons) in Geology in 1993 and completed a PhD degree in 1997 in aspects of field, geochemical and tectonic evolution of the Galway Granite Batholith (both from the National University of Ireland Galway, NUIG). Following an NUIG lecturership, he took up a post-doctoral research position at Keele University, England, as part of a European-wide research network investigating the...
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Martin Feely graduated with a BSc degree in Geology from the National University of Ireland Galway (NUIG) in 1973, an MSc degree (1978) in Geology from University College Dublin and a PhD degree (1982) from NUIG. Following postdoctoral work at NUIG on radioelements in Irish granites, he joined the staff of NUIG’s Geology Department in 1985. He is currently a senior lecturer at NUIG. He was also adjunct Associate Professor of Earth Sciences at Boston University (1997–2005) and is currently adjunct Professor of Geological and Environmental Sciences at James Madison University, VA, USA. He is manager of the GeoFluids Research Group at NUIG.