- Iodine status in western Kenya: a community-based cross-sectional survey of urinary
 and drinking water iodine concentrations
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15 Abstract

Spot urinary iodine concentrations (UIC) are presented for 248 individuals from western 16 Kenya with paired drinking water collected between 2016 and 2018. The median UIC was 17 271 µg L⁻¹, ranging from 9 to 3146 µg L⁻¹, unadjusted for hydration status/dilution. From 18 these data, 12% were potentially iodine deficient (<100 µg L⁻¹), whilst 44% were considered 19 20 to have an excess iodine intake (>300 µg L⁻¹). The application of hydration status/urinary 21 dilution correction methods were evaluated for UICs, using creatinine, osmolality and specific gravity. The use of specific gravity correction for spot urine samples to account for 22 hydration status/urinary dilution presents a practical approach for studies with limited 23 24 budgets, rather than relying on unadjusted UICs, 24 hour sampling, use of significantly large 25 sample size in a cross-sectional study and other reported measures to smooth out the urinary dilution effect. Urinary corrections did influence boundary assessment for deficiency-26 27 sufficiency-excess for this group of participants, ranging from 31 to 44% having excess 28 iodine intake, albeit for a study of this size. However, comparison of the correction methods 29 did highlight that 22% of the variation in UICs was due to urinary dilution, highlighting the need for such correction, although creatinine performed poorly, yet specific gravity as a low-30 cost method was comparable to osmolality corrections as the often stated 'gold standard' 31 32 metric for urinary concentration. Paired drinking water samples contained a median iodine concentration of 3.2 µg L⁻¹ (0.2-304.1 µg L⁻¹). A weak correlation was observed between UIC 33 34 and water-I concentrations (R = 0.11).

35 **Keywords**: Urinary Iodine Concentrations; iodine excess; hydration status corrections

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44 Introduction

Approximately 2 billion people are estimated to be at risk of iodine deficiency disorders (IDD) 45 worldwide (WHO, 2004; Andersson et al. 2012), for which the most severe consequences 46 occur during foetal and early childhood development and can later manifest as goitre and 47 impaired cognitive development (Obican et al. 2012; Bath et al. 2013). The lodine Global 48 49 Network estimated that 19 countries remain at risk of IDD whilst 10 countries have excess iodine intake (IGN, 2017). Excess intake of iodine (I) can cause hyper- or hypo-thyroidism, 50 51 thyroid autoimmunity or euthyroid goitre (Burgi et al. 2010; Leung et al. 2014). Rohner et al. (2013) provided a comprehensive review of health consequences of I-deficient or excess 52 intake and an evaluation of biomarkers to define I-status. The I-status of populations was 53 54 traditionally quantified based on the prevalence of goitres, but the the use of urinary iodine concentration (UIC) measured in spot urine samples is now considered standard practice 55 given the subjective nature of goitre classification and the more widespread availability of 56 57 laboratories capable of urinary iodine analyses (UNICEF, 2007). In addition, over 90% of ingested iodine is rapidly excreted in the urine over 24-48 hours allowing for reliable 58 59 estimates of iodine intake, as well as providing ease and cost effectiveness of collection with 60 few limitations (Zimmerman et al. 2008; Zimmermann & Andersson 2012; Rohner et al. 2013; WHO 2013). UIC are normally expressed as a population median in μ g L⁻¹, since 61 assessment of individual status using single spot samples is not recommended due to high 62 intra-diurnal variations in iodine intake (Vejbjerg et al. 2009; Knudsen et al 2009; König et al. 63 2011; Zimmermann & Andersson 2012, Rohner et al. 2013). 64

The highest prevalence of iodine deficiency is in the African region, as measured using 65 66 school age children (SAC), at 39.3% (58 million) (Andersson et al. 2012). For example, in Malawi, 35% of the population (calculated from SAC) present a UIC of <100 µg L⁻¹, 67 representing a population risk of inadequate iodine intakes, whilst the same 2010 Malawi 68 Demographic and Health Survey (NSO, 2012) also reported that 62% of households were 69 consuming adequately iodised salt. Further, Farebrother et al. (2018) described the history 70 of the successful salt iodisation programme in Kenya which recorded a national median UIC 71 72 in SAC of 118 µg L⁻¹ in 2004, though regional variations in UIC were wide with values as high as 477 µg L⁻¹, which resulted in a reduction in salt-I concentration from 100 to 30-50 mg 73 kg⁻¹ in 2010 (Kenya Ministry of Health, 2011). Yet, the Kenyan National Micronutrient Survey 74 reported in 2011 a median UIC of 208 µg L⁻¹ in SAC, of which 30% were considered to have 75 excess-I intake at >300 µg L-1 (Kenya Ministry of Health, 2011). These examples show that 76 77 often the success of salt iodisation strategies is not maintained through lack of consistent and sustained monitoring of iodinated salt alongside UIC, which can be problematic for 78 79 assessment of both deficiency and excess l-intakes.

The Estimated Average Requirement (EAR) for iodine in children, adults and pregnant 80 women is 64, 107 and 143 µg *capita* d⁻¹, respectively (WHO, 2004). The Tolerable Upper 81 Limit (TUL) is defined as the level of iodine intake at which there is no demonstrable 82 83 evidence of toxicity and for adults is 600 and 1,100 µg capita d⁻¹ in the EU and USA, respectively (Zimmerman 2008; Leung et al 2015). Excess iodine intake can result in thyroid 84 disorders, but little data is available of the effects in vulnerable groups (Farebrother 2018), 85 although a healthy thyroid can tolerate I-intakes in excess of the EAR (Burgi et al. 2010; 86 Leung et al. 2015; Katagiri et al. 2017). Reports identifying excess I-intake are becoming 87 more common, with I-intake now being a problem of both deficiency and excess in many 88 regions often at a local scale. For example, in some regions in Argentina (Watts 2009), 89 Algeria (Barikmo et al. 2011), Somalia (Kassim et al. 2014), Malawi (Watts 2015) and Ghana 90

91 (Simpong et al. 2016; Abu et al. 2018), although Farebrother et al. (2018) reported a generally low prevalence of thyroid dysfunction across the study population in Kenya. On the 92 93 basis of I status defined as 'inadequate' if the median UIC is <100 µg L⁻¹, sufficient 100-300 μ g L⁻¹ and excess >300 μ g L⁻¹ (WHO/ICCIDD 2007), Watts et al. (2015) reported 12 and 94 33% of a study in Malawi to have a risk of moderate deficiency (50-100 µg L⁻¹) and excess, 95 96 similarly in Lesotho, 18 and 47% at risk of deficiency and excess, respectively (Sebotsa et al. 2005). The 2017 Iodine Global Scorecard (IGN 2017) classified 7 of 47 mainland African 97 countries as having 'more than adequate' I intake, 14 with no data and four as 'excessive' on 98 99 the basis of UIC in School Age Children (SAC). Such variances may be due to differences in dietary intake given that plant based foods are generally low in I content compared to the, 100 consumption of iodised salt, increasingly processed foods (e.g. stock cubes) or availability of 101 I-rich fish (Rohner et al. 2013; Watts et al. 2015; Abizari et al. 2017). Additional uncertainty 102 could be derived from cooking processes, for example, 6 to 51% of I is reportedly lost from 103 104 iodised salt when cooking (Rana & Raghuvanshi 2013). A review of cross-sectional studies by Katagiri et al. (2017) revealed the sources of excess I to be mainly from iodised salt or 105 drinking water. Rohner et al. (2013) also referred to studies reporting the contribution of I-106 intake from groundwater sources in China and Algeria (Andersen et al. 2009; Henjam et al. 107 108 2010).

109 Drinking water-I can vary in concentration according to the source; Reimann et al. (2003)

reported 0.3-961 μ g L⁻¹ (median 11 μ g L⁻¹) from spring, well and river sources along the Ethiopian Rift Valley. Aakre et al. (2015) reported a median water-I concentration of 102 μ g L⁻¹ (80-255 μ g L⁻¹) in Algerian refugee camps, which correlated with UIC, but also coupled with high milk-I and salt-I consumption resulted in thyroid dysfunction after excessive I-intake over several years. Farebrother et al. (2018) reported median water-I concentrations of 92 μ g L⁻¹ for Kenya, Tanzania and Djibouti although no correlation was observed with UIC.

Concentrations of spot urinary analyte concentrations, I included, require correction for 116 hydration-driven dilution variation (Middleton et al. 2016). However, corrections of UIC are 117 not commonly reported in the literature, with largley uncorrected UIC available for 118 comparison of datasets. The routine aproach to dilution correction in the wider biomonitoring 119 discipline is creatinine correction, the validity of which has been guestioned (Nermell et al. 120 2008). For a correction method to perform properly, it must satisfy a number of prerequisites. 121 Firstly, the metric used should be an accurate physicochemical marker of urinary dilution and 122 it has long been recognised that urinary osmolality, the proxy for which is specific gravity 123 (SG) is far more reliable that creatinine. Creatinine, the breakdown product of muscular 124 125 creatinine phosphate, is subject to appreciable variation, more so than osmolality (Yeh et al. 2015) from demographic factors, particularly in developing countries owing to varying intake 126 of protein and water and the existence of malnutrition, as well as age-sex cut-ff points 127 (Knudsen et al. 2000; Vejberg et al. 2009; Jooste et al. 2010; Cockell 2015). Secondly, the 128 application of the metric must be mathematically robust and representative of underlying 129 physiological changes in analyte excretion in relation to fluctuations in urinary flow rate 130 (Araki et al. 1986). Several criteria have previously been proposed (Middleton et al. 2016) to 131 assess the performance of urinary dilution corrections, one of which is the removal of dilution 132 variation from samples, crudely assessed by plotting corrected concentrations against the 133 measurement used to correct them. 134

This paper aims to: (1) evaluate a sample population for I status using UIC measurements in
western Kenya; (2) comparatively assess the use of creatinine, osmolality and specific
gravity as effective dilution corrections in comparison to uncorrected UIC (as is commonly
reported) and potential influence on boundary assessment for deficiency-sufficiency-excess,
and (3) explore whether drinking water influences UIC.

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142 Methods

143 Study setting

Sample collection was part of a wider project as described in Watts et al. (pending), which collected soil, crops, drinking water and a urine sample from households. Each household is shown in Figure 1, including Bomet, Bungoma, Elgoyo Marakwet, Kakamega, Kisumu, Nandi

147 Hills, Siaya, Uasin Gishu counties in Western Kenya.

148 Collection of urine

This paper describes the urinary iodine concentrations (UICs) and paired drinking water 149 iodine from each household. Adult and consenting volunteers at each site were requested 150 for a urinary sample following an explanation of the study and why samples were being 151 collected. In general, we attempted to collect from a minimum of 30 different sites that were 152 153 spread out evenly across each county, representing rural land-use, although the geographic size and accessibility resulted in slight variation in numbers per county. One sample was 154 generally collected from each household, a second volunteer provided a sample in <10% of 155 156 households. Ethical approval was obtained from the Institutional Research and Ethics Board of Moi University (000921). Volunteers above the age of 18 were requested. Pregnant and 157 lactating women excluded. Volunteers were asked to urinate into a 30 mL nalgene LDPE 158 bottle, which was hygienically capped and transported in a coolbox (Ca 4°C) and 159 subsequently filtered into an 8 mL nalgene LDPE bottle using a nylon 0.45 µm syringe filter 160 at the end of each day, followed by storage in a coolbox and freezing at -80°C on return to 161 the University of Eldoret laboratory in Kenya. Urines were transported frozen to the UK for 162 elemental analyses, including I and urinary dilution measurements for subsequent 163 164 corrections.

165 Collection of water

Generally, only drinking water was collected and filtered (nylon, 0.45 µm) on-site into a 30 166 mL nalgene LDPE bottle to be used for anion, organic carbon and pH/alkalinity 167 168 measurements and a 15 mL nalgene LDPE bottle to be acidified on return to the UK with 1% $HNO_3/0.5\%$ HCl for elemental analyses by inductively coupled plasma mass spectrometry 169 (ICP-MS) (Watts et al. pending), although iodine analyses were performed on the filtered 170 unacidified portion owing to the requirement of an alkaline matrix (Tetramethylammonium 171 Hydroxide-TMAH). Additional data was collected regarding the source (rain, river, borehole, 172 well), any treatment, reliability and field parameters, including conductivity, pH, temperature 173 174 and total dissolved solids.

175 ICP-MS analyses for iodine

176 Urine samples were analysed with a x20 dilution in 0.5% TMAH solution prior to analyses by

177 ICP-QQQ-MS (Agilent 8900) with the collision cell in no gas mode, Rf power 1550 W,

nebuliser flow rate of 0.4 ml min⁻¹, providing a limit of detection (LOD) of 0.2 μ g L⁻¹ (3SD

blanks). Tellurium was used as an internal standard to correct for minor signal drift.Measurements below the LOD were attributed a value half of the LOD. Water samples were

181 analysed separately to urines, but using the same method. Certified reference materials

were measured giving iodine concentrations of $114 \pm 2 \mu g L^{-1}$ (recovery: 109%; n = 15) for

183 SeronormTM Trace Elements Urine L-1 (Sero, Norway) and $39.7 \pm 3.0 \ \mu g \ L^{-1}$ (recovery: 99%;

184 n = 21) for a spiked SLRS-2 Riverine Water.

185 Urinary dilution corrections

Urinary creatinine was determined using a Randox liquid assay kit and a Randox RX Imola
chemistry analyser. Osmolality was measured by freezing-point osmometry using a
Gonotect Osmomat 030 (Gonetec, Germany). Specific gravity was measured with a PAL-10S digital refractometer (Atago, Japan) prior to filtration. Creatinine, SG and osmolality
corrections were performed using Equation 1:

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$$UIC_{cor} = UIC_{vol} \times (D_{ref})/(D_{meas}), \qquad (1)$$

where UIC_{cor} is dilution corrected urinary iodine concentration; UIC_{vol} is the measured, volume-based urinary iodine concentration (in µg/L); D_{ref} is the reference value to which UIC concentrations are scaled to and D_{meas} is that measured in the given specimen (note: D_{ref} -1 and D_{meas} -1 are used for SG correction). D_{ref} was 1 g L⁻¹ for creatinine – synonymous with the conventional division-based correction and yielding results in µg g creatinine⁻¹; and, for both SG and osmolality, the study group medians were selected: 1.017 (unitless) and 581 mOsm kg⁻¹, respectively.

- 200
- 201 Statistical analysis

202 Summary statistics (arithmetic mean, median, standard deviation (SD), minimum and

203 maximum) were calculated for UICs – uncorrected and corrected by various dilution metrics.

Pearson correlation coefficients (R) and significance tests (P-values) were calculated on

natural log-transformed variables due to the positively skewed distributions of UIC
 concentrations. All statisitcal analyses and graphics were performed in R version 3.4.3 and

concentrations. All statisitcal analyses and graphics were performed in R version 3.4
 the RStudio GUI.

208 Results & Discussion

209 Urinary iodine concentrations

Urinary iodine concentrations (UIC) for 248 adults are summarised in Table 1. Uncorrected 210 211 UIC values, which are commonly compared across studies provided a mean of $321 \pm 280 \mu g$ L^{-1} and median of 271 µg L^{-1} , with a wide range of UIC values from 9-3146 µg L^{-1} . Whilst only 212 12% of the measured population were considered to be iodine deficient (<100 μ g L⁻¹) 213 according to UIC, 44% were above 300 µg L⁻¹, indicating excessive iodine intake. These UIC 214 data are comparable to other studies reporting the prevalence of excess I intake, according 215 to UIC and other urinary biomarkers. For example, Farebrother et al. (2018) reported 216 uncorrected median UICs in non-pregnant, non-lactating women elsewhere in Kenya of 289 217 μg L⁻¹ (IQR 173, 458 μg L⁻¹), and coastal Tanzania, of 473 μgL-1 (IQR 321, 689 μg L⁻¹). 218 219 Median uncorrected UIC's reported in Malawi (Watts et al. 2015) were 221 µg L⁻¹ (141-344 μ g L⁻¹); in Port Sudan 464 and 561 μ g L⁻¹, Medani et al. (2012) and Hussein et al. (2012), 220 221 respectively. In Somalia non-pregnant women provided a median of 329 µg L⁻¹ (Kassim et al. 2014) and in Lesotho median UIC of 280 µg L⁻¹, with 21 and 47% of women considered 222 either deficient or excess following two years after introducing a salt iodisation programme 223 (Sebotsa et al. 2005). Additionally, the median uncorrected UIC of 271 µg L⁻¹ in this study is 224 225 high when compared to data summarised in the lodine Global Scorecard (IGN, 2017), median uncorrected UICs were 215, 175, 66 and 118 µg L⁻¹ for Lesotho, Malawi, Sudan and 226 Kenya, respectively in the general population. 227

- **Table 1**: Summary statistics for uncorrected and dilution corrected urinary iodine
- 229 concentrations (UIC), comparison to World Health Organisation (WHO) guidance values and
- to the Hays et al. (2018) Biomonitoring Equivalents (BE) displayed as number of
- individuals and in brackets as (%) of group below BE.

UIC (μg L ⁻¹)	Uncorrected	Osmolality-	Creatinine-	Specific
		corrected	corrected	corrected
n	246	242	230	246
Median	271	285	232	299
Arithmetic mean	321	325	395	321
SD	280	278	655	279
Range	9-3146	14-3870	6-4771	9-3149
Percentiles (25, 75)	163, 403	205, 382	149, 359	190, 404
Number	246	242	230	246
WHO Values, <i>n</i> (%)				
Extreme deficiency (<50 µg L ⁻¹)	9 (4)	1 (0)	9 (4)	8 (3)
Mild deficiency (<100 µg L ⁻¹)	30 (12)	6 (2)	23 (10)	30 (12)
Sufficient (100-300 µg L ⁻¹)	107 (43)	129 (53)	130 (57)	106 (43)
Excess (>300 µg L ⁻¹)	109 (44)	107 (44)	78 (34)	108 (44)
Hays et al. (2018) BE's, <i>n</i> (%)*				
EAR cut-off (60 µg L ⁻¹)	11 (4)	2 (1)	9 (4)	11 (4)
RDA (100 µg L ⁻¹)	30 (12)	6 (2)	23 (10)	30 (12)
UL (730 µg L ⁻¹)	11 (4)	6 (2)	17 (7)	11 (4)
MRL (450 µg L ⁻¹)	51 (21)	32 (13)	37 (16)	51 (21)

* Estimated Average Requirement (EAR), Recommended Dietary Allowance (RDA) and for

toxicity, Tolerable Upper Intake Level (UL) and Minimal Risk Level (MRL)

Hays et al. (2018) recently proposed new boundaries for determining iodine deficiency,

sufficiency and excess according to UIC by deriving Biomonitoring Equivalents (BEs).

236 Established exposure and guidance criteria were used, such as the Estimated Average

- 237 Requirement (EAR), Recommended Dietary Allowance (RDA) and for toxicity, Tolerable
- 238 Upper Intake Level (UL) and Minimal Risk Level (MRL). For UIC, BE's were derived for

adults as follows; 60,100, 730 and 450 µg L⁻¹, respectively. Zimmerman et al. (2016) also
 employed EAR and UL cutoff points to improve spot UIC to adjust I-intake distributions from

241 UIC surveys where single sample per subject is collected. Whilst these approaches have

not yet appeared in abundance for other studies in the literature, they are likely to gain

243 greater interest to improve the validity of UIC spot measurements as cost effective survey

tools. When considering Hays et al. (2018) BE's, for this study, using uncorrected UICs the

proportion of volunteers <EAR, <RDA, >UL and > MRL, were 4, 12, 4 and 21%, respectively.

Little difference in the proportion of UIC deficiencies results from these BE's, but excess is

- reduced from 44% for both the UL and MRL criteria.
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249 Influence of urinary dilution corrections

250 For UICs with dilution correction, creatinine is the most commonly reported correction

251 method, although has fallen out of favour (Cockell 2015). In this paper, creatinine corrected

- UIC (μ g g⁻¹) provided a median of 232 μ g L⁻¹, which was significantly higher than 203 μ g L⁻¹
- reported in Malawi (Watts et al. 2015), but presented a higher proportion of individuals with a

UIC in a range of sufficiency. Meanwhile, the number of volunteers exhibiting excess UIC
 reduced to 34 with creatinine adjustment from 44% (uncorrected) in this study.

Table 1 summarises the UIC data and influence of correction methods on the proportion of volunteers considered to have excess I-intake ranging between 31 to 44%, with 44% of volunteers in excess when using uncorrected UICs. No clear pattern can be discerned as to the influence of each method of correction on changes to numbers of individuals within each boundary for UICs for a study group of this size. Therefore, further analyses were undertaken to evaluate the validity of each correction method, in comparison to a UIC uncorrected for dilution.

263 Figure 2 shows each of the three dilution measurements in relation to one another. Expectedly, a strong positive correlation (R=0.82) was observed between osmolality and SG 264 265 (Figure 2A), albeit a number of possible outliers - interferences on SG measurement by refractometry are possible in the presence of large urinary solutes (i.e. proteinuria) (Imran et 266 al. 2010). Conversely, a weaker correlation was observed for creatinine in relation to both 267 osmolality (Figure 2B, R=0.47) and SG (Figure 2C, R=0.48). If, as is widely cited (Imran et 268 al. 2010), it is assumed that osmolality is the most robust measure of urinary dilution, 269 270 creatinine was a poor marker of urinary dilution in the present dataset, with only 22% of 271 variation explained by osmolality. This finding was markedly different when compared with two western populations: 48% variation in creatinine was explained by osmolality in a set of 272 2151 samples from the US NHANES survey (CDC, 2018) and 67 % in a UK survey of 202 273 274 adults (Middleton et al. 2016a). This may indicate the particularly poor utility of creatinine in an African population. 275

Figure 3 shows the correlations between both uncorrected (A-C) and corrected (D-F) UICs 276 277 and each dilution metric. Uncorrected concentrations showed positive correlations with each metric, demonstrating the necessity of correcting to remove hydration-driven dilution 278 variation. As evident in Figure 3B, almost a guarter of variation in UICs was attributable to 279 280 sample dilution alone. This was less evident for creatinine (Figure 3A) reiterating its poor reflection of urinary dilution. Figure 3 D-E show the efficacy (no correlation desired) of each 281 correction method in removing dilution variation from the sample set. No significant 282 correlation remained following correction by osmolality (Figure 3E) and SG (Figure 3F) - an 283 indication of their good performance. However, creatinine (Figure 3D) resulted in an 284 apparent over-correction, of a magnitude consistent with previous findings in relation to this 285 286 performance criterion (Middleton et al. 2016). It is possible to compensate for this overcorrection by modifying correction equations with a coefficient reflecting the disparity in 287 288 excretion slopes between creatinine and the analyte under investigation (Vij & Howell, 1998). However, the apparent variation in creatinine concentrations is not explained by dilution in 289 the present dataset and would make any such improvements misleading indicators of 290 291 performance.

292 Finally, given the relative paucity of studies utilising alternative correction methods to creatinine, there is a lack of consistency as to the reference values (D_{ref} in Equation 1) used 293 294 to normalise/scale datasets. This has profound implications when comparing between populations and to published referenced values. Firstly, comparing distributions of 295 uncorrected values between populations assumes that they do not differ by hydration status. 296 However, the median osmolality of the present study group was 581 mOsm kg⁻¹; compared 297 to 730 mOsm kg⁻¹ in the US NHANES data (Centers for Disease Control and Prevention, 298 299 2015). Secondly, using the median osmolality of the present study group as D_{ref} yielded a median UIC of 285 µg L⁻¹, whereas using the NHANES median yielded a median UIC of 358 300 μ g L⁻¹ – an increase of 25%. To ensure that studies are comparing like-for–like, standardised 301

302 guidelines on the proper use of dilution corrections, and the appropriate dilution reference

- 303 values, are urgently needed.
- 304

305 Drinking water and comparison to UIC

306 Drinking water concentrations for 268 samples provided a median of 3.2 μ g L⁻¹ (0.2-304.1 μ g L^{-1}) and mean of 12.3 ± 33.3 µg L^{-1} . Whilst the median water-I concentration was not high 307 compared to other studies, the range was comparable, for example, in Denmark <1.0 to 139 308 µg L⁻¹ (Pedersen et al. 1999) and 55 to 545 µg L⁻¹ in Algeria (Barikmo et al. 2011). Almost no 309 correlation of UIC (uncorrected) with drinking water was observed (R = 0.11), unlike other 310 311 studies that observed a much stronger correlation (Hussein et al. 2012; Aakre et al. 2015). Kassim et al. (2014) reported variations in Somalian UICs according to the source of drinking 312 water, with a greater association with borehole water, although the concentration of I in 313 water was not measured. In contrast to these studies, the maximum water-I concentration 314 reported in the USA was only 18 µg L⁻¹ (WHO, 2003). Given a typical water consumption of 315 1.8 L day⁻¹ (Beal et al. 2017), drinking water could assist in preventing I-deficiency, but may 316 317 also contribute to I-excess intake above upper tolerable limits of 1100 µg day⁻¹ (Leung et al. 2015) where water-I is high. For example, in this study the mean water-I concentration would 318 contribute an intake of 22 µg day⁻¹ and at the maximum water-I range, 547 µg day⁻¹. There is 319 320 currently no defined guideline for iodine in drinking water (WHO, 2003).

321

322 Conclusion

Whilst 24 hour urinary collection is cited as 'ideal' for UIC measurements, UIC single spot 323 analyses are generally used in studies for practical reasons in the field. Often the sample 324 size of 300 or 500 subjects is referred to for smoothing out differences in hydration 325 (Andersen et al. 2008; Konig et al. 2011; Rohner et al. 2013). Creatinine correction for 326 hydration status has been repeatedly reported as unsuitable for UIC in a developing country 327 scenario where malnutrition may be a problem, although there appears to be a lack of 328 329 published quantitative evidence to reinforce this widely held view in the literature. Correction for hydration status should be reconsidered, especially given the robustness, simplicity and 330 331 low cost of specific gravity measurements, which are comparable to the gold 332 standard/osmolality (Middleton et al. 2015). Use of specific gravity adjustments for spot urine samples to account for urinary dilution presents an opportunity for studies with limited 333 budgets and studies that are most commonly reported to have less than the 300+ 334 recommended number of subjects and for practical reasons where repeat sampling of 335 336 individuals may not be possible.

337 This study adds to the growing awareness of excess I-intake according to UIC since 338 Anderson et al. (2012) reported an increase from 5 to 11 countries between 2003 and 2011 in which the national median UIC was >300 µg L⁻¹. Whilst no correlation with drinking water 339 was observed in this study, further research to establish the association between water-I 340 concentration, physical and chemical parameters, alongside depth of boreholes, lithology 341 and extraction rates from aquifers is required. In combination with UIC, dietary and 342 343 physiological studies, a guideline value for water-I should be established to inform national programmes for monitoring and alleviating iodine deficiency and preventing any adverse 344 health effects secondary to chronic excessive iodine intakes. 345

346

347 Figure legends

348 **Figure 1** Map illustrating location of household collection points in western Kenya.

Figure 2 Scatterplots showing the relationship between urinary osmolality and creatinine (A),
 urinary osmolality and specific gravity (B) and urinary specific gravity and creatinine (C).
 Pearson correlation coeficients are significant to P<0.001.

Figure 3 Scatterplots of urinary iodine against dilution measurements both pre- (A-C) and

post- (D-F) correction by each method investigated. ** and * denote statistical significance to
 P<0.001 and <0.05, respectively.

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