- Status, source, ecological and health risk assessment of toxic metals and
 polycyclic aromatic hydrocarbons (PAHs) in street dust of Abadan, Iran
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- 8 Abstract

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9 The potential hazard to human health from exposure to toxic metals and polycyclic aromatic hydrocarbons (PAHs) in street dust was assessed using thirty street dust samples collected from Abadan, 10 an industrial city in Southwest Iran. The mean concentration of Pb, Zn, Cu, Cr, Cd, Ni, V, As and Co 11 12 in the street dust was: 59, 288, 113, 50, 0.5, 57, 36, 7.1 and 7.5 (mg/kg), respectively. The majority of 13 potentially toxic metals were elevated compared to background concentrations, with the exception of 14 Ni > Co > V, with a high enrichment value for Zn and Pb. Cadmium had a moderate potential ecological 15 16 risk compared to other toxic metals. Correlation coefficients and principal component analysis (PCA) 17 identified three principal groups as sources of toxic metals in Abadan street dust. Metals such as Cr, Ni, 18 Co and V were identified to originate from natural parent materials, Zn, Cu and Pb from anthropogenic 19 sources emitted from combustion/vehicular materials and Cd and As from other anthropogenic sources 20 such as the Abadan refinery and petrochemical complex. The sum of 13 PAHs ranged from 400 to 21 11766 (µg/kg), with a mean of 2436 (µg/kg). Approximately 68%–79% of ∑PAHs comprised of higher 22 molecular weight PAHs (HMWPAHs 4-6 rings), revealing a combustion origin. Principal component 23 analysis (PCA) for PAHs in Abadan street dust originated mainly from vehicular, petroleum emission 24 and fossil fuel. The toxic equivalency quantities (TEQs) of PAHs in the street dust ranges from 22 to

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25	951 (μ g/kg). High coefficients of correlation (r^2 = 0.98 to 0.83, p<0.01) among Pyrene, Benzo (a)
26	anthracene, Chrysene, Benzo (a) pyrene, Benzo (b) fluoranthene, Benzo (k) fluoranthene, and TEQ of
27	street dust showed that these PAHs were principal contributors to TEQ. Ecosystem risk calculations for
28	PAHs were high since the value of $RQ_{\Sigma PAHS_{(MPCS)}}$ was ≥ 1 and $RQ_{\Sigma PAHS_{(NCS)}}$ were ≥ 800 . The mean
29	values for $RQ_{\sum PAHs_{(MPCs)}}$ and $RQ_{\sum PAHs_{(NCs)}}$ were 8 and 1151, which indicated a relatively high
30	ecosystem risk of PAHs in street dust in Abadan. The total cancer risk was 8.64×10^{-3} for adults and
31	8.95×10^{-4} for children, suggesting a high potential cancer risk through both dust dermal contact and
32	ingestion pathways.
33	Keywords: Street dust, Toxic metal, PAHs, Pollution assessment, Health risk assessment, Abadan
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53 1. Introduction

Emissions from activities relating to rapid urbanisation, industrial plants and vehicular traffic in 54 55 megacities provide challenges for urban and residential environments. Urban street dust particles are 56 one of the indices for monitoring environmental urban contaminants (Brown and Peake, 2006; 57 Nazarpour et al., 2017b; Babaei et al., 2018), such as toxic metals and polycyclic aromatic hydrocarbons 58 (PAHs) that present potential harm to community health (Tang et al., 2013, Saeedi et al., 2012; Soltani 59 et al., 2015). Dermal and hand to mouth contact can result in direct exposure to such contaminated dust 60 particles, especially in the case of children in school grounds, parks and city side roads (Ghanavati et 61 al., 2018). The source of toxic metals in street dust can be petroleum, industrial sources, vehicle 62 emissions and geogenic/geochemical routes, such as weathering of parent materials, construction, 63 roadside and atmospheric dust accumulation (Gunawardana et al., 2012, Nazarpour et al., 2017b). Toxic 64 elements are not decomposable and reside in the soil/dust environment for a long time or can be re-65 suspended into the atmosphere, presenting a potential hazard to human health (Cook et al., 2005, 66 Zarasvandi et al., 2011; Ghanavati., 2018). PAHs comprise a cluster of carbon-based/organic complexes 67 with various aromatic rings and do not contain a heteroatom or carry substituents (Kamal et al., 2015). 68 They can present a major and permanent hazard to environmental security and human health, and for 69 these reasons they are of major concern in recent studies (Dong and Lee, 2009; Nazarpour et al., 2017b). 70 The sources of PAHs include natural processes (wood fires and volcanic eruptions) and anthropogenic 71 activities (e.g., straw and fuel wood combustion, waste incineration, vehicle exhaust, combustion of coal, spillage of petroleum, diesel oil and pyrolysis of organic matter) (Abdel-Shafy and Mansour, 2016; 72 73 Li et al., 2006). Sixteen parent PAHs are classified as significant contaminants and persistent toxic 74 substances (PTS) by the United States Environmental Protection Agency (US EPA) and United Nations Environment Program (UNEP) because of their durability, potential toxicity, exhibit mutagenic and 75 carcinogenic properties and are widely distributed in the natural environment in matrices such as dust 76 (UNEP, 2002; Kamal et al., 2015). In order to formulate sufficient measures to decrease environmental 77 78 risk of exposure to PAHs, it is essential to quantitatively assess contributions from several sources of 79 PAHs, such as pyrogenic or petrogenic consistent with the major PAH sources to the environment

80 (Wang et al., 2009). Commonly, PAHs resulting from petrogenic sources contain 3-4 ring structures and have petroleum products or crude oil as their main sources. Whereas, PAHs produced by pyrogenic 81 82 sources contain 4–6 ring constructions, formed when organic matters are exposed to high temperatures under low oxygen or no oxygen circumstances. The distillation of coal into coke and coal tar, or the 83 84 thermic cracking of petroleum sediment into lighter hydrocarbons are pyrolytic processes that occur in urban areas (Liu et al., 2007). PAHs present the main carcinogen and cancer risk to metropolitan 85 environments when present in urban dust, particularly when there are several anthropogenic sources. 86 87 PAHs with a high molecular weight have a greater carcinogenic toxicity compared to low molecular weight PAHs (Colombo et al., 2006; Szabová et al., 2008). Multivariate analysis including correlation 88 89 coefficients and principal component analysis (PCA) have been utilised to identify the source of toxic elements and PAHs (Long et al., 2013). Contamination of toxic metals and PAHs in urban dust have 90 91 been considered extensively in industrialised countries (Lee et al., 2011). Abadan is one of the most significant cities in Iran, owing to its existence as an industrial hub, with a large influence from the 92 93 Petrochemical Industries and Oil Refining Company. However, there is no published data about toxic 94 elements and PAH composition of street dust from Abadan and its potential health risk. Therefore, the 95 main objectives of this study were: (1) to determine the concentration and source of toxic metals and 96 PAHs in street dust samples collected from Abadan city and to evaluate their pollution level; (2) to 97 define their specifications and possible combustion sources (3) to evaluate the distribution of pyrogenic 98 and petrogenic PAHs in dust street samples; and (4) to evaluate potential ecosystem risk and the cancer 99 risk possibility of PAH exposure through dermal contact, ingestion and inhalation.

100 2. Materials and methods

101 2.1. Study Area

Abadan (30°20'40" N and 48°17'20"), as one of industrial cities of Khuzestan province with about
340000 inhabitants in 2014 is located in south west Iran on the border with Iraq, Kuwait and Saudi
Arabia, whose deserts are the main source of dust storm occurrences over Iran.

105 **2.2. Sample collection and preparation**

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In August 2016, thirty (30) street dust samples were collected from various locations in the city of Abadan (Fig.1). Samples were selected in the driest month of the year to avoid rain-washing away street dusts. Samples were collected nearby to: a refinery, petrochemical industries, an airport, and commercial and busy traffic areas of the city. Approximately 500 g of a composite street dust sample was collected by brushing an area of $2 \times 2 \text{ m}^2$ from the sidewalk. The samples were air-dried and passed through a $63\mu\text{m}$ mesh. Geographical coordination of sample locations was recorded at each sample point with a handheld GPS device.

113 2.3. Extraction and analysis of toxic metals

Toxic metals (Ni, As, V, Cr, Pb, Cu, Co, Zn and Cd) were analyzed by Inductively Coupled Plasma-Mass Spectrometry (ICP-MS; Perkin-Elmer Elan 9000). In order to measure toxic metals, a wet digestion process was adapted from Liang and Grégoire (Liang and Grégoire, 2000). Quality assessment and quality control contained analytical duplicate samples, a reagent blank, standard reference materials GSS-17 and GSS-25 (multi-element soil standard OREAS45EA and OREAS24P) and procedural blanks. Precision, specified by duplicate measurements, was < 5% relative standard deviation (RSD) and accuracy for the reference material, was between 95% and 110% for all metals studied.

121 2.4. Analysis of PAHs

Thirteen PAHs specified by the US EPA as precedence contaminants were analyzed, including 122 Phenanthrene (Phe), Anthracene (Ant), Pyrene (Pyr), Benzo (a) anthracene (BaA), Chrysene (Chr), 123 124 Benzo (b) fluoranthene (B(b)F), Benzo (k) fluoranthene (B(k)f), Benzo (a) pyrene (BaP), Indeno (1,2,3cd) pyrene (IcdP), Dibenzon (a,h) anthracene (DBA), Benzo (ghi) pyrylene (BghiP). The PAHs in street 125 dust samples were measured using gas chromatography/mass spectrometry system (GC/MS) in 126 127 Jundishapur University of Medical Sciences Comprehensive Laboratory (JUMSCL) pursuant to EPA 128 SW-846 procedure 3550C (US EPA, 2007) and EPA 8270D (US EPA, 1998). A sample of about 10 g was extracted using a solvent combination of dichloromethane and acetone (1:1 by volume, insecticide 129 130 grade, Merck). The samples were exposed to end-over-end tumbling (30 min, about 1 cycle/s) overnight. The extract was filtered via sodium sulphate followed by concentration by evaporation using 131

132 a Turbovap (Caliper Life Sciences) at 40 °C. The extracts were spiked with the USEPA8270 internal standard mix (Supelco, Sigma Aldrich, USA) and examined by gas chromatography mass spectrometry 133 (Agilent 5975, USA) with capillary column DB5ms (30 ms, 0.25 mm, 0.25 µm film, J&W, USA). The 134 GC/MS was operated in Selected Ion Mode with helium as the carrier gas (1 ml/min, BOC gases). The 135 136 oven temperature was held at 40°C for 1 min, increased to 310°C at 18°C/min and then held for 7 137 minutes at this temperature. Laboratory procedural blanks were analysed with each batch of samples, and limits of quantification were defined as three times greater than the blank levels. Duplicate samples, 138 139 laboratory control samples (LCS) and matrix spikes were analysed with each batch, with all analyses within ±20% relative percent difference (RPD) for duplicates (at the 10 for limits of Report (LOR's) 140 141 level), and accuracies of 95-117% for LCS.

142 **2.5 Quality control**

All 30 street dust samples were examined in duplicate. Filter blanks were lower than 10% of the sample quantity. The relative standard deviation was lower than 10% for all measured PAHs. Standard reference material of street dust (SRM 2585) from the National Institute of Standards and Technology (NIST) USA was applied for calibration and analytical control. PAHs were identified on the basis of retention times relative to five deuterated internal standards (d8-Nap, d10-Ace, d10-Phe, d10-Chr, and d12-Per). Recoveries of SRM and internal standards ranged from 75% (Pyr) to 133% (Nap).

149 **2.6.** Pollution assessment

150 2.6.1. Enrichment factor

An enrichment factor (EF) method was applied to determine the degree of toxic metal contamination
(Yuen et al., 2012) and define their sources, as either parent materials or human/anthropogenic sources
(Liu et al., 2014). Enrichment factor (EF) was estimated as follows:

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$$EF = \frac{\left(\frac{\text{Ci}}{\text{Cref}}\right)\text{sample}}{\left(\frac{\text{Ci}}{\text{Cref}}\right)\text{background}}$$
(1)

The $\left(\frac{\text{Ci}}{\text{Cref}}\right)$ is the ratio of value of toxic metal (mg/kg) to the value of target toxic metal in the selected 155 and background sample. In this paper, Al was used as the normaliser and contents of metals in the crust 156 were derived as background (Hsu et al., 2016; Yongming et al., 2006). Consistent with (Zhang and Liu, 157 2002), EF values $0.05 \le EF \le 1.5$ show that the toxic metal originated completely from crustal materials 158 159 or a natural source, although values greater than 1.5 would show anthropogenic source of toxic metals. 160 Degree of pollution can be categorised in five classes (Namaghi et al., 2011): (1) EF ≤ 2 (states deficiency to minimal enrichment); (2) $2 \le EF \le 5$ (moderate enrichment); (3) $5 \le EF \le 20$ (significant 161 enrichment); (4) $20 \le EF \le 40$ (very high enrichment) and (5) EF > 40 (extremely high enrichment). 162

163 2.6.2. Potential ecological risk (PER)

The Potential Ecological Risk (PER) of toxic metals was originally presented by Hakanson (1980) to
evaluate the degree of toxic metal contamination in street dust (Qiu, 2010; Sun et al., 2010). The PER
was estimated as follows:

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$$C_f = \frac{C_S}{C_n}$$
 (2)

$$E_r = T_r \times C_f \tag{3}$$

169
$$RI = \sum_{i}^{n} E_{r}$$
(4)

170 Where, RI is the sum of potential ecological risk for multiple toxic metals; Er is the potential ecological 171 risk of each toxic metal; Tr is the toxicity response factor of each toxic metal (Cd, Cu, Pb, Cr, As, Zn,V and Ni are 30, 5, 5, 2, 10, 1, 2 and 5) (Wang et al., 2011b); C_f is the contamination factor; C_S is the 172 concentration value of each toxic metal in the street dust and Cn is the corresponding background value 173 for toxic metals. Potential ecological risk values can be classified in the following order: Eri<40, low 174 potential ecological risk; $40 \le \text{Er}_i \le 80$, moderate potential ecological risk; $80 < \text{Er}_i \le 160$, considerable 175 potential ecological risk; $160 < Er_i \le 320$, high potential ecological risk; and $Er_i > 320$, very high 176 177 ecological risk (Wu et al., 2015; Zhao et al., 2014). Risk index values can be classified as follows: RI 178 < 150, low ecological risk; $150 \le RI < 300$, moderate ecological risk; $300 \le RI < 600$, considerable 179 ecological risk; and RI > 600, very high ecological risk.

180 2.6.3. Potential ecosystem risk

In order to estimate risk posed by certain PAHs in the Abadan, PAHs species values detected in street dust were compared against their corresponding quality values in the present study. The risk quotient (RQ) approach was used to estimate the potential ecosystem risk of PAHs in street dust (Kalf et al., 1997). The risk quotient (RQ) values were defined in Eq. 5.

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$$RQ = \frac{C_{PAHs}}{C_{QV}}$$
(5)

Where C_{PAHs} was concentration of certain PAH in street dust samples and C_{QV} was the corresponding quality values of certain PAHs in the medium. The negligible concentration of (NCs) and the maximum permissible concentrations (MPCs) were used as the quality values in the medium (Kalf et al., 1997). Therefore, $RQ_{(NCs)}$ and $RQ_{(MPCs)}$ were calculated in Eq. 6 and 7.

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$$RQ_{NC_S} = \frac{C_{PAH_S}}{C_{QV(NC_S)}}$$
(6)

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$$\mathrm{RQ}_{\mathrm{MPC}_{\mathrm{S}}} = \frac{C_{\mathrm{PAH}_{\mathrm{S}}}}{C_{\mathrm{QV}(\mathrm{MPC}_{\mathrm{S}})}}$$
(7)

192 Where CPAHs was concentration of certain PAH in street dust samples. RQ(NCS) and RQ(MPCS) were the quality values of the NCs and MPCs of PAHs in street dust. The RQ could only be used to evaluate the 193 194 ecosystem risk of 8 individual PAHs. In order to evaluate the ecosystem risk of other 5 individual PAHs [acenaphthylene (Acy), Fluorene (Fl), pyrene (Pyr), benzo (b) fluoranthene (B(b)F) and dibenzo (a,h) 195 196 anthracene (DBA), the toxic equivalency factors (TEFs)] (Nisbet and Lagoy, 1992) were used to infer that the NCs and MPCs of Acy, Fl, and Pyr (TEFs = 0.001) were similar to anthracene (Ant) (TEFs = 197 0.001), B(b)F(TEFs = 0.1) was similar to benzo(a)anthracene (BaA) (TEFs = 0.1) and dibenzo 198 (a,h)anthracene (DBA) (TEFs = 1) was similar to benzo(a)pyrene (BaP) (TEFs = 1). $RQ_{\Sigma PAHs}$, 199 $RQ_{\sum PAHS_{(NCS)}}$ and $RQ_{\sum PAHS_{(MPCS)}}$ were defined in Eq. 8-10 (Cao et al., 2010): 200

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$$RQ_{\Sigma PAHs} = \sum_{i=1}^{13} RQ_i \quad (RQ_i \ge 1)$$
 (8)

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$$RQ_{\Sigma PAHS_{(NCS)}} = \sum_{i=1}^{13} RQ_{(NCS)} \quad (RQ_{(NCS)} \ge 1)$$
(9)

203
$$RQ_{\sum PAHS_{(MPCS)}} = \sum_{i=1}^{13} RQ_{(MPCS)} \quad (RQ_{(MPCS)} \ge 1) \quad (10)$$

In principle, $RQ_{(NCs)} <1$ indicated that the single PAHs are likely of negligible concern, while RQ (MPCs)>1 would indicate that contamination of the single PAHs was of more concern and some control measures and remedial actions must be undertaken at once. Where $RQ_{(NCs)} >1$ and $RQ_{(MPCs)} <1$, contamination of single PAHs might be considered of medium concern and some control measures or remedial actions are required (Table 1).

209 2.6.4. Cancer risk assessment

210 PAH toxicity evaluations in the street dust samples were assessed applying relative toxicity values for individual PAH species. The PAH toxicities of street dust samples were calculated according to the 211 collection of toxicity equivalency factors (TEFs) for PAHs (Nisbet and LaGoy, 1992). For example, 212 213 BaP (a potent carcinogen in the PAH group) was identified as the reference chemical and assigned a 214 value of 1, whilst other PAHs have their specific TEF values according to their carcinogenic level relative to BaP (Fang et al., 2004). The carcinogenic potency of total PAHs is calculated using the toxic 215 benzo[a] pyrene equivalent (BaPeq) contents of each PAH (Lee and Dong, 2010). The toxic equivalent 216 217 quotient (TEQ) of each street dust sample was computed by collecting the products of each individual 218 PAH content. TEQs were calculated in Eq. 11 and 12.

$$BaPeq_i = PAH_i \times TEF_i$$
(11)

220
$$TEQ = \sum_{i}^{n} (PAH_i \times TEF_i)$$
(12)

221 Where PAH_i is the content of individual PAHs, and TEF_i is the corresponding toxic equivalency factor 222 and TEQ is the toxic equivalent of the compound. Humans are exposed to street dust throughout three major exposure routes including, dermal contact, ingestion and inhalation. The incremental lifetime cancer risk (ILCR) model was used to quantitatively calculate the exposure risk for environmental PAHs according to the U.S. EPA standard models (Chen and Liao, 2006; Peng et al., 2011). ILCR in terms of ingestion, dermal contact, and inhalation were calculated in Equations 13–15.

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$$LCRs_{Ingestion} = \frac{CS \times \left(CSF_{Ingestion} \times \sqrt[3]{\left(\frac{BW}{70}\right)}\right) \times IR_{Ingestion} \times EF \times ED}{BW \times AT \times 10^{6}}$$
(13)

229
$$LCRs_{Dermal} = \frac{CS \times \left(CSF_{Dermal} \times \sqrt[3]{\left(\frac{BW}{70}\right)}\right) \times SA \times AF \times ABS \times EF \times ED}{BW \times AT \times 10^6}$$
(14)

230
$$LCRs_{Inhalation} = \frac{CS \times \left(CSF_{Inhalation} \times \sqrt[3]{\left(\frac{BW}{70}\right)}\right) \times IR_{Inhalation} \times EF \times ED}{BW \times AT \times PEF}$$
(15)

231 Where CS is the sum of converted PAH levels according to toxic equivalents of BaP applying the toxic equivalency factor (TEF) (Fang et al., 2004; Nisbet and LaGoy, 1992). CSF is the carcinogenic slope 232 233 factor (mg/kg/day),;BW is body weight assumed to be 15 kg for children and 61.5 kg for adults (Shi et al., 2011); AT is the mean life span being $70 \times 365 = 2550$ (days) (Ferreira-Baptista and De Miguel, 234 235 2005); EF and ED are exposure frequency and duration, respectively. The EF value is 180 (day/year) 236 for children and adults (Ferreira-Baptista and De Miguel, 2005); ED value is 24 years for adults and 6 years for children (USEPA, 2001); IR Ingestion is the dust intake rate at 200 (mg/day) for children and 100 237 (mg/day) for adults (U.S. EPA 2001), IR_{Inhalation} is the inhalation rate at 10 (m³/day) for children and 20 238 (m^{3}/day) for adults (USEPA, 2001), SA is the dermal surface exposure; SA value is 2800 (cm²/day) for 239 240 children and 5700 (cm²/day) for adults, respectively. AF is the dermal adherence factor, AF which is 0.2 (mg/cm) for children and 0.07 (mg/cm) for adults (USEPA 2001); ABS is the dermal adsorption 241 fraction (unit less) being 0.13 (U.S. EPA 2001), and PEF is the particle emission factor defined as 242 1.36×10^9 (m³/kg) (USEPA 2001). The designation of CSF, according to the cancer-causing ability of 243 BaP; CSF_{Ingestion}, CSF_{Dermal} and CSF_{Inhalation} of BaP, are 7.3, 25 and 3.85 (mg/kg/day), respectively (Peng 244 245 et al., 2011). All the parameters used in these models for children (1 to 6 years old) and adults (7 to 31

years old) were according to the Risk Assessment Guidance of U.S. EPA and related publications(Wang et al., 2011a).

248 2.7. Evaluation of PAH sources

249 The ratios of designated isomeric PAHs have been extensively applied to recognise PAH sources (Ravindra et al., 2008). In this paper, the attribution of pyrogenic or petrogenic sources to the PAHs 250 251 found in the street dust fractions were based on two criteria: a) the ratio of Σ COMB to Σ PAHs $(\Sigma COMB/\Sigma PAHs)$ can be applied to recognise the source of PAHs, where $\Sigma COMB$ is the sum of the 252 253 main combustion specific compounds. These include Fl, BaA, Pyr, B(k)F, Chry, B(b)F, IcdP, BaP and 254 BghiP and Σ PAH as the sum of PAH concentrations (Kamal et al., 2015). Spearman's correlation analysis and principal component analysis (PCA) were carried out using the statistical software package 255 SPSS version 22.0 for windows. The Spearman's correlation coefficient was applied to measure the 256 strength of relationships between pairs of PAHs within samples. A significant and positive correlation 257 258 between PAHs often shows that the PAHs are derived from the same sources (Saeedi et al., 2012; Wang 259 et al., 2011a).

260 **3. Results and discussion**

261 **3.1.** Toxic metal concentrations

Statistical concentration data for toxic metals in street dust from Abadan is compared to 262 background/upper continental crust (UCC) values in Table 2. Mean concentrations for Pb, Zn, Cu, Cr, 263 264 Cd, Ni, V, As and Co in road dust were: 59, 288, 113, 50, 0.5, 57, 36, 7 and 8 mg/kg, respectively. With 265 the exception of Co and V, all metals were higher than their corresponding UCC, indicating possible 266 anthropogenic input of metals from vehicular traffic, industrial plants and construction activities. A Kolmogorov-Smirnov test indicated that all toxic metal concentrations were non-normally distributed 267 (significant level > 0.05) except for Zn, Cr and V (significant level < 0.05). Skewness values for all 268 toxic metals were positive, with only Cr, Ni and Co displaying normal distribution. The skewness for 269 270 Cu was 4.83 (highest value among all of the toxic metals), which shows the existence of extremely

271 polluted spots. Based on the coefficient of variation (CV), the toxic elements analysed can be categorised into two groups: Pb, Zn, Cu, Cd and As with a CV greater than 0.4; and Co, Cr, Ni and V 272 with a CV lower than 0.4. Elements dominated by a natural source exhibit a low CV, while those 273 274 affected by anthropogenic sources are expected to have a high CV (Wang and Lu, 2011; Yuan et al., 275 2014). There is no standardised protocol for the sampling and geochemical analysis of urban street dust, 276 and specifically such values do not exist in the study area of Abadan. The mean concentrations of 277 pollutants in street dust are generally compared to other urban environments rather than a defined 278 threshold value. Mean concentrations of metals in Abadan street dusts were compared with other cities 279 in the world in Table 3. The mean concentrations of Pb and Zn in Abadan street dusts in this study are higher than Ottawa and Beijing, but lower than Tehran, Shiraz, Xian, Hong Kong, Shanghai, Madrid 280 281 and Amman. The mean concentration of Cu is lower than Tehran, Shiraz, Shanghai, Madrid and Amman, and higher than Xian, Hong Kong, Ottawa and Beijing. Generally, every urban area has its 282 283 own typical signature for toxic metal combinations resulting from variations in anthropogenic and 284 natural geogenic sources.

285 **3.2.** Toxic metal contamination in street dust

Upper crust (UCC) values were applied as background concentrations for toxic metal enrichment 286 factors (EF) calculations, as reported in previous studies (Kartal et al., 2006; Reimann and de Caritat, 287 288 2005). In this regard, Al was taken as the reference element. The range of EF values for Pb, Zn, Cu, Cr, Cd, Ni, V, As and Co were 7–122, 20–157, 7–235, 8–14, 12–33, 17–24, 4–7, 6–43 and 5–6, respectively 289 (Fig. 2). The average EF values decreased as follows: Zn (68) > Pb (29) > Cu (28) > Cd (19) > As (11)290 > Cr (9) > Ni (8.7) > Co (5)> V (4). The mean EF values for Pb, Zn, Cu, Cd, and As in Abadan street 291 292 dust samples were > 10, suggesting anthropogenic sources for these metals in the study area, whereas 293 Ni, Cr, V and Co can be considered to originate primarily from natural geogenic sources, such as wind-294 blown soil minerals. Zinc exhibited the highest EF mean (> 40), suggesting extremely high enrichment. 295 These results demonstrate that the toxic metals reflect anthropogenic sources emitted from industrial 296 plants and vehicular materials, such as brakes and tyres (Kong et al., 2011).

297 In this study, the potential ecological risk (Er) of toxic elements was evaluated by applying the Hakanson technique (Hakanson, 1980). According to the trend of Er in dust samples: Cd (78) >Pb (20) 298 >Cu (19) >As (15) > Ni (14) >Zn (9)>Cr (3) > V (1). Mean Er values for Cu, Pb, As, Ni, Zn, Cr and V 299 were < 40, which suggested a low risk for these metals, thus indicating a low potential *Er*. In contrast, 300 301 Cd has a moderate potential Er compared to other toxic metals (Table 4). The potential ecological risk 302 indices (RI) values were evaluated to assess the risk of multiple-metals associated with street dusts. In 303 general, *RI* describes the sensitivity of the environment to potentially toxic metals providing an overall 304 indication of pollution. The RI shows that values for 57 % of all samples were at a low risk, 40% and 305 3% were at moderate and considerable risk, respectively (Table 5).

306 3.3. Toxic metal source identification

307 Inter-element relationships between elements provides useful information about the sources and pathways of toxic metals (Huang et al., 2009; Lu et al., 2010). Spearman's correlation coefficients of 308 309 toxic metals in Abadan street dusts are presented in Table 6. One group of toxic metals, Ni, Cr, Co and 310 V, show a very significant correlation with each other at p < 0.01: Co-V (0.90), Co-Ni (0.85), Ni-V 311 (0.83), Co-Cr (0.79), Cr-V (0.79) and Ni-Cr (0.70), suggesting a common source. A significant correlation (p < 0.01) was also found between a second group of toxic metals, Pb, Zn, Cu, Cd and As 312 including Pb-Cu (0.86), Cu-Zn (0.84), Pb-Zn (0.68), Pb-Cd (0.56), Cu-Cd (0.51), Cd-As (0.51), Pb-313 314 As(0.47) and Cu-As(0.47), which indicated a different source.

In this study, principal component analysis (PCA) was employed to further understand the relationship 315 between metals in urban dust and their possible sources for each factor. Bartlett's test of sphericity was 316 317 significant at p < 0.001 and 0.569 for the Kaiser–Meyer– Olkin Index, which confirmed that toxic metals associated with urban dust was appropriate for (PCA). The result of principle component 318 analysis (PCA) with varimax rotation was used for toxic metals concentrations. Three Principal 319 Components (PC) accounting for 83% of the total variance is presented in Table 7. The first principal 320 component (PC1) explains 42% of the total variance is dominated by Cr, Ni, Co and V. As well as, the 321 average values of EF for Cr, Ni, Co and V are essentially lower than for the other toxic elements. As a 322

result, these toxic metals likely originated from natural sources. The second component (PC2) accounting for 27% of total variance is characterised by Zn, Pb and Cu. In the loading plot (Fig. 3). A third (PC3) explains 14% of total variance, and is dominated primarily by As and Cd. However, the high values of EF for As and Cd proposes that anthropogenic sources are the main sources of these toxic metals.

328 3.4. Spatial distribution patterns of toxic metals

329 The spatial distribution of toxic metals concentrations is presented in Figure 4. The spatial distribution pattern of toxic metals concentrations can aid in identifying areas above UCC and to determine the 330 331 sources of contamination. Considering the activity of the two large refineries and petrochemical 332 complexes and the arrival and accumulation of a large number of toxic metal compounds, it was expected that the highest concentrations of toxic metals in street dust around the industrial zones. 333 However, spatial distribution of toxic metals and its adaptation to Abadan urban zoning showed that 334 relatively high concentrations of toxic metals occurred instead in dense residential areas with high 335 336 traffic volumes and urban shopping centres. According to the weather data, annual rainfall in Abadan 337 is about 157 (mm/year) and is mostly dry and frequented by prevailing winds from the northwest through all seasons, but with more intensity in the summer. The spatial distribution of metals in street 338 dust suggested that the prevailing wind has a significant influence on the distribution of toxic metals 339 340 across Abadan. As a result, toxic metals are transferred from the areas around the refinery and 341 petrochemicals to urban and residential areas.

342 3.5. PAH concentrations

Descriptive statistics for 13 PAHs in Abadan street dust samples are summarised in Table 8. Total PAH content (Σ 13PAHs) in Abadan street dust were in a range of 400–11766 (μ g/kg) with a mean of 2436 (μ g/kg), showing that PAHs are inclined to accumulate in street dust and that street dust can be applied as an index of urban environmental contamination. The mean value for PAH species in Abadan street dust decreased as follows: Pyr (753) > Phe (382) > Chr (278) > Fl (167) > Icdp (155) > B(b)F(153) (μ g/kg). Therefore, Pyr, Phe, Chr, Fl, Icdp, B(b)F are the major components. However, the lowest 349 measurable species in street dust are BaP and DBA with mean concentrations of 61 and 46 (μ g/kg). Combustion PAHs (ComPAHs, the sum of Pyr, Phe, BaA, Fl, Chr, BaP and B(b + k)F after (Xiao et al. 350 351 2014) make up a considerable proportion of the total content of PAHs in measured samples, a range of 0.73 to 0.80 (µg/kg). The ratio of carcinogenic PAHs— B(k)F, B(b)F, BaA, BaP and Chr (Rajput and 352 353 Lakhani, 2009) to total content of PAHs (CanPAHs / 213PAHs) range from 0.20 and 0.40. The ratio of non-carcinogenic PAHs to total content of PAHs (NCanPAHs / Σ 13PAHs) ranging from 0.60 and 354 355 0.80. The street dust content of total PAHs (2436 μ g/kg) in this study was greater when compared with 356 urban Tehran, Iran (330 µg/kg) (Saeedi et al., 2012), Bangkok, Thailand (1100 µg/kg) 357 (Boonyatumanond et al., 2007), Isfahan, Iran (1074 µg/kg) (Soltani et al., 2015), Hanoi, Vietnam (1500 μ g/kg) (Tue et al., 2014). However, concentrations of Σ 13PAHs in urban Abadan street dust was lower 358 359 compared with urban Sydney, Australia (2910 µg/kg) (Nguyen et al., 2014), Lanzhou, China (3900 360 µg/kg) (Jiang et al., 2014), Guangzhou, China (4800 µg/kg) (Wang et al., 2011a), Xian, China (10620 361 µg/kg) (Wei et al., 2015).

In general, PAHs in urban dust is classified with the number of aromatic rings, three-ring PAHs (Ant, 362 363 Phe, Fl and Acy) contributed 30 % of Σ PAHs, four-ring PAHs (BaA, Pyr and Chr) contributed 47 % of ∑PAHs, Among five-ring (B(b)F,B(k)f, BaP and IcdP) to six-ring (DBA and BghiP) PAHs, contributed 364 19%, and 5 %, respectively. The average content of lower molecular weight PAHs (LMW 3 rings) is 365 366 737 (μ g/kg), contributed 30% of Σ PAHs, although the mean content of higher molecular weight PAHs 367 (HMW 4-6 rings) was 1699 (µg/kg), contributed 70% of ∑PAHs, indicating a pyrogenic source 368 (Zakaria et al., 2002). The contribution of high molecular weight PAHs was significant and likely from 369 a common origin of vehicle emission (Hassanien and Abdel-Latif, 2008). In addition, this finding was 370 consistent with individual PAHs detected in gasoline engine soot enriched by high molecular weight 371 PAHs, indicating a prevailing effect of gasoline engine release (Kong et al., 2012).

372 **3.6. PAHs source identification**

373 Spearman's correlation coefficients among PAH species, \sum PAHs and TEQ in Abadan street dusts are 374 presented in Table 9. The majority of variables were significantly correlated (p < 0.05), although, stronger correlations characterised by r > 0.7 occurred for HMW species (p < 0.01). Although, the stronger correlations characterised by r-values (r>0.8) were detected only among HMW species group, and concurrently correlated with $\sum PAHs$. LMW compounds exhibited a negligible correlation with HMW compounds, indicating various sources for these two groups. Significant correlation coefficients between HMW and TEQs in Abadan street dust ($r^2 = 0.932, 0.433$ and 0.980, p < 0.01) show that LMW compounds tend to have lesser toxicity than HMW compounds (Anastasopoulos et al., 2012).

381 In this study, principal components analysis (PCA) with varimax rotation was applied to identify the 382 relationship among PAH compositions and possible chemical sources in urban street dust. Principal 383 components analysis (PCA) with the varimax rotation Bartlett's test of sphericity were significant at p < 0.001 and results of Kaiser-Meyer- Olkin Index of 0.689 extracted three principal components 384 385 (Factor 1, Factor 2 and Factor 3) and represented more than 87% of the total variance of Σ PAHs in the urban street dust (Table 10). Factor 1 (63% of the total variance) was characterised by high loadings of 386 PAHs with four to six rings, including Chr, BaA, Pyr, B(k)F, B(b)F, BaP, IcdP, DBA, DghiP, 387 representing vehicle-related sources (Boonyatumanond et al., 2007). BaP, Chr, and BaA are identified 388 389 as markers of gasoline emissions (Larsen and Baker, 2003), although B(k)F and B(b)F are identified as markers of fossil fuels combustion (Park et al., 2002). In summary, Factor 1 was assigned to vehicular 390 traffic contamination and petrogenic sources. Factor 2 (16% of the total variance) had high loadings of 391 392 PAHs with three rings, including Acy, Fl and Phe, representing combustion sources (Steinhauer and 393 Boehm, 1992). LMW PAHs, such as Phe, Acy, Fl, show the attendance of significant combustion 394 products of low-temperature pyrogenic processes (Jenkins et al., 1996). Factor 3 (7% of the total 395 variance) displayed high loading values of Ant, in which Ant represents oil combustion markers (Fig.5) 396 (Dong and Lee, 2009).

397 3.7. Spatial distribution patterns of PAHs

The spatial distribution of PAHs in Abadan street dust is shown in Fig. 6. The results of the spatial distribution pattern of PAHs concentration and its adaptation to urban zoning in Abadan showed that the highest concentration of PAHs with lower molecular weight PAHs (LMW 3 rings) occurred around 401 these industrial centers, with high residential density and traffic volume (Abadan-Khorramshahr road). 402 The highest concentration of PAHs with higher molecular weight PAHs (HMW 4–6 rings) was 403 observed close to a refinery and petrochemical complex. The spatial distribution pattern showed that 404 the highest concentrations of Σ PAHs and TEQ were around two refineries within a petrochemical 405 complex. The results show that the most important factor in the entry and accumulation of a large 406 amount of PAHs in Abadan street dust is due to the activity of industries (refinery and petrochemicals), 407 combustion of oil and gas and traffic.

408 **3.8.** Potential ecosystem risk

409 The mean RQ (MPCs) values of Fl and Pyr were all higher than 1, indicating a high level of risk to the 410 ecosystem, with severe toxicity (Table 11). The mean RQ (MPCs) and RQ (NCs) values of BghiP in street dust was less than 1, indicating a relatively low risk. The mean RQ (MPCs) values of other individual 411 PAHs, such as Acy, Phe, Ant, BaA, Chr, B(b)F, B(k)F, BaP, IcdP, and DBA in street dust were all lees 412 than 1 and RQ (NCs) higher than 1, indicating a moderate risk. These results indicated that 3- and 4-ring 413 PAHs contributed to the main ecosystem risk burden in street dust. $RQ_{\Sigma PAHS(MPCS)}$, were higher than 1 414 and $RQ_{\sum PAHS(NCS)}$ were higher than 800. These results indicated that the ecosystem was at a high risk 415 in Abadan. In conclusion, the results of the diagnostic ratios indicated that PAHs in Abadan originated 416 417 from the activity of industries, vehicular emission; fossil fuel and petroleum emission. The results of 418 ecosystem risk assessments indicated a high risk of Fl and Pyr in street dust. Therefore, several control measures and strict rules should be implemented by the government to decrease emissions. 419

420 **3.9.** Health risk assessments

Health risk assessments of the studied PAHs in the street dust samples through the three main exposure pathways including ingestion, dermal exposure, and inhalation was performed for children and adults (Peng et al., 2011). The toxic equivalency concentration (TEQ) of 13 PAHs in Abadan street dusts was assessed using the total BaPeq, each of which was computed using toxic equivalence factors (TEFs) (Tue et al., 2014). Results indicated that the dust exposure pathways of PAHs for both children and adults decrease as follows: dermal contact > ingestion >inhalation. The conclusion indicated that the

levels of cancer risk through ingestion and dermal contact range from 10⁻³ to 10⁻⁵ in all street dust 427 samples, which were 10^4 to 10^5 times greater than that via an inhalation pathway (10^{-9} to 10^{-8}) (Table 428 12). Consequently, inhalation of suspended particles via the nose and mouth was almost negligible, 429 when compared with the other pathways. The incremental life time cancer risk (ILCR) through 430 431 ingestion are greater for children than adults due to their hand-to-mouth activity, so that polluted street dust can be more easily ingested (Jiang et al., 2014), although risks through dermal contact and 432 433 inhalation are greater for adults than children. The greater incremental life time cancer risk (ILCR) for 434 adults via dermal contact can be described by the greater values of dermal exposure area (SA) and 435 exposure duration (ED) of adults (Wang et al., 2011a). As well as, primary expansion of organ, nervous, and immune systems could perhaps increase the carcinogens sensitivity in children (Maertens et al., 436 437 2008). Therefore, the potential health hazard for children exposed to street dust with associated PAHs was significantly higher than for adults (Wang et al., 2011a). The greater cancer risks through inhalation 438 for adults was due to their greater exposure period and inhalation rate. It is commonly supposed that 439 the incremental life time cancer risk (ILCR) value of $\leq 10^{-6}$ indicates a negligible risk under most 440 regulatory programmes, a value among 10^{-6} and 10^{-4} indicated a potential risk, whereas a value > 10^{-4} 441 is considered to be a high risk (Wang et al., 2011a). In this study, the total cancer risk is 8.64×10^{-3} for 442 adults and 8.95×10⁻⁴ for children. The total cancer risks for both adults and children in the Abadan urban 443 444 area indicated a high potential cancer risk through both dust dermal contact and ingestion pathways.

445 4. Conclusion

The concentration values, enrichment factors and ecological risk of the potentially toxic metals (Pb, Zn, 446 Cu, Cd, Cr, As, Ni, Co and V) in Abadan street dusts were determined, with all but Co and V in street 447 448 dust greater than their corresponding UCC, demonstrating potential anthropogenic sources. EF values in street dust decreased as follows: Zn > Pb > Cu > Cd > As > Cr > Ni > Co > V. Lead, Zn and Cu in 449 450 street dusts demonstrated a high EF mean enrichment value, suggesting that the dusts in this study area 451 were significantly affected by human activities (vehicular traffic, abrasion of tyres, asphalt pavement 452 and industrial plants). Lead, Zn, Cu, Cr, As, Ni, Co and V had a low PER, whereas Cd had a moderate PER value (77.90). Based on the correlation coefficients and principal component analysis (PCA), toxic 453

454 metals in street dust were grouped into three main sources: Cr, Ni, Co and V that originate mainly from natural sources, whilst Pb, Zn and Cu originate from anthropogenic sources emitted from vehicular 455 materials and Cd and As from wider anthropogenic activity. The total concentration of Σ 13PAHs 456 ranged from 400 to 11766 (μ g/kg), with a mean of 2436 (μ g/kg). Approximately 68%–79% of Σ PAHs 457 458 was comprised of higher molecular weight PAHs (HMW 4-6 rings), revealing a combustion origin. Principal component analysis (PCA) suggested that the activity of industries (refinery and 459 460 petrochemicals); vehicular emission; fossil fuel and petroleum emission are likely the main contributors 461 of PAHs in Abadan street dust. Ecosystem risk assessment indicated that Acy, Phe, Ant, BaA, Chr, 462 B(b)F, B(k)F, BaP, IcdP, and DBA widely showed middle level ecosystem risk and Fl and Pvr indicated a high level of risk to the ecosystem in street dust of Abadan. Incremental lifetime cancer risk (ILCR) 463 464 estimation showed that the PAH concentrations in Abadan street dusts had a high carcinogenic risk 465 through both dust ingestion and dermal contact pathways for children and adults.

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Figure captions

Fig. 1. Sampling site, refinery and petrochemical complex (yellow rectangle) locations in Abadan, Iran.

Fig.2. Box-plot of EF for toxic metals in Abadan street dust.

Fig. 3. Loading plots of the principal component analysis (PCA) for toxic metals in the street dust samples from Abadan.

Fig. 4. Spatial distribution of toxic metals concentration in street dust of Abadan

Fig. 5. Factor loading plots of factor analysis for PAHs in the street dust samples from Abadan.

Fig. 6. Spatial distribution of PAHs concentration in street dust of Abadan.











Fig 3.



233000 234000 235000 236000 237000 238000 239000 240000 241000





Fig 4.



Fig 5.



Fig. 6

Table captions

Table 1. Risk classification of individual PAHs and Σ PAHs.

- Table 2. Statistical summaries of element concentrations in street dusts from Abadan.
- Table 3. A comparison of toxic metals concentrations (mg/kg) in street dusts of Abadan and other cities.
- Table 4. Distribution of potential ecological risk factor *Er* of heavy metal.
- Table 5. Indices and grades of potential ecological risk.
- Table 6. Correlation matrix for toxic metal concentrations.
- Table 7. Matrix of principal component analysis loadings for toxic metal concentrations in Abadan street dust.
- Table 8. Descriptive statistics of PAHs in street dust of Abadan (μ g/kg).
- Table 9. Correlation coefficients among individual PAHs, **DPAHs** and **TEQ** in Abadan street dust samples.
- Table 10. PCA loadings for PAHs in street dust of Abadan.
- Table 11. Mean value of RQ (NCs) and RQ (MPCs) of PAHs in street dust of Abadan (µg/kg).
- Table 12. Risk of cancer in Abadan due to human exposure to PAHs through street dust.

Table 1					
Individual PAHs			ΣPAHs		
	RQ (NCs)	RQ _(MPCs)	RC	ΣPAHs (NCs)	RQ _{2PAHs} (MPCs)
Risk-free	< 1	< 1	Risk-free Low-risk	< 1 > 1 : < 800	<1 <1
Moderate-risk	≥ 1	< 1	Moderate-risk1 Moderate-risk2	≥ 800 ≤ 800	< 1 ≥ 1
High-risk	≥ 1	≥ 1	High-risk	≥ 800	≥ 1

Table 2	

Flore	Un:t	Street dusts (n=30)						
Element	Unit	Min-Max	Mean ± SD	Skewness	CV	K-S	Upper crust content ^a	
Pb	mg/kg	15 - 262	59±43	3.67	0.73	0.000	15	
Zn	mg/kg	116 - 701	288 ± 139	1.01	0.48	0.065	31	
Cu	mg/kg	31 - 982	113 ± 168	4.83	1.49	0.000	29	
Cr	mg/kg	31 - 72	50 ± 8	0.35	0.17	0.200	35	
Cd	mg/kg	0.43 - 0.75	0.52 ± 0.08	1.14	0.45	0.037	0.09	
Ni	mg/kg	35 - 88	57 ± 10	0.69	0.17	0.004	47	
V	mg/kg	28 - 48	36 ± 4	0.71	0.11	0.200	60	
As	mg/kg	4 - 29	7 ± 5	3.31	0.72	0.000	4.8	
Со	mg/kg	5 - 12	8 ± 1	0.84	0.18	0.000	10	

Rollinson, (2014).

Table 3										
City	Pb	Zn	Cu	Cr	Cd	As	V	Co	Ni	Reference
Tehran (Iran)	116	403	136.34	67	0.5	7	-	-	78	(Saeedi et al., 2012)
Shiraz (Iran)	257	873	225.3	34	11	-	-	-	35	(Keshavarzi et al., 2015)
Xi an (China)	231	421	94.98	167	-	11	-	-	-	(Yongming et al., 2006)
Hon Kong (China)	120	3840	110	124	-	67	37	10	29	(Yeung et al., 2003)
Beijing (china)	54	219	46	87	1.1	6	-	9	34	(Liu and Cen, 2007)
Shanghai(China)	237	753	258	264	1.0	8	-	-	66	(Shi et al., 2011)
Ottawa(Canada)	33	101	38	42	0.3	1	-	8	15	(Rasmussen et al., 2001)
Madrid (Spain)	1927	467	188	61	-	-	17	3	44	(de Miguel et al., 1997)
Amman (Jordan)	271	351	139	29	1.9	-	-	32	66	(Al-Momani, 2009)
Abadan (Iran)	59	288	112.97	50	0.52	7	36	8	57	This Study

Table 4			

			Er		
Element	Min	Max	Mean	SD	Skewness
Pb	5	87	20	14	4
Zn	4	23	9	5	1
Cu	5	169	19	29	5
Cr	2	4	3	0.48	0.35
Cd	65	113	78	11	1
Ni	9	22	14	2	0.69
V	0.93	2	1	0.13	0.71
As	8	61	15	11	3

Table 5

		RI			Number of sample					
	Min	Max	Mean	Number of sample	Low risk	Moderate risk	Considerable risk	High risk		
Street dust	115	330	159	30	17(% 56.66)	12(% 40)	1(% 3.34)	0		

Table 0									
Element	Pb	Zn	Cu	Cr	Cd	Ni	V	As	Co
Pb	1								
Zn	0.680**	1							
Cu	0.861**	0.840**	1						
Cr	0.366*	0.546**	0.511**	1					
Cd	0.556**	0.525**	0.508**	0.021	1				
Ni	-0.139	0.052	-0.020	0.699**	-0.300	1			
V	0.054	0.271	0.168	0.785**	-0.149	0.826**	1		
As	0.466**	0.513**	0.465**	0.385*	0.508**	-0.004	0.308	1	
Со	0.030	0.262	0.121	0.789**	-0.088	0.849**	0.900**	0.261	1

** Correlation is significant at the 0.01 level (2-tailed). * Correlation is significant at the 0.05 level (2-tailed).

Table 6

Table 7					
Element	Component		Communities		
	1	2	3		
Pb	-0.037	0.717	0.329	0.623	
Zn	0.249	0.869	0.248	0.878	
Cu	0.157	0.864	-0.134	0.789	
Cr	0.804	0.345	0.112	0.778	
Cd	-0.155	0.246	0.892	0.881	
Ni	0.932	0.050	-0.156	0.895	
V	0.939	0.112	-0.046	0.897	
As	0.088	0.040	0.884	0.790	
Co	0.973	0.009	0.044	0.949	
Initial Eigenvalue	3.815	2.428	1.237		
Total variance %	42.389	26.980	13.744		
Cumulative %	42.389	69.368	83.113		

Table 8						
Compounds	Aromatic ring	TEF	Mean	Min	Max	S.D
Acenaphthylene (Acy)	3	0.001	84	19	396	68
Fluorene (Fl)	3	0.001	167	29	399	81
Phenanthrene (Phe)	3	0.001	382	69	875	184
Anthracene (Ant)	3	0.001	104	11	796	137
Pyrene (Pyr)	4	0.001	753	169	3541	644
Benzo(a)anthracene (BaA)	4	0.100	102	8	1118	198
Chrysene (Chr)	4	0.010	278	42	1267	227
Benzo(b)fluoranthene (B(b)F)	5	0.100	153	19	1166	210
Benzo(k)fluoranthene (B(k)F)	5	0.100	85	7	602	112
Benzo(a)pyrene (BaP)	5	1.000	61	4	496	101
Indeno(1,2,3-cd)pyrene (IcdP)	5	0.100	155	6	576	197
Dibenzo(a,h)anthracene (DBA)	6	1.000	46	13	79	33
Benzo(ghi)pyrylene (BghiP)	6	0.010	66	3	456	102
3 ring %			30	32	21	21
4 ring %			47	55	50	47
5 ring %			18.6	9	24	27
6 ring %			4.61	4	5	6
∑PAHs			2436	400	11766	2295
LMW PAHs			737	128	2467	471
HMW PAHs			1699	272	9300	1825
COMPAHs/Σ13PAHs			0.74	0.77	0.80	0.73
NCANPAHs/Σ13PAHs			0.72	0.60	0.80	0.63
CANPAHs/Σ13PAHs			0.28	0.20	0.40	0.37
TEQ			162	22.	951	211
TEQ/ΣPAHs			0.07	0.06	0.08	0.09

 $\Sigma PAHs:$ Sum of individual mass content of detected PAHs .

LMWPAHs: Sum of low molecular weight 3 ring PAHs.

HMWPAHs: Sum of high molecular weight 4–6 ring PAHs.

COMPAHs: Sum of major combustion derived PAH content.

CANPAHS: Sum of carcinogenetic PAHs.

NCANPAHs: Sum of non-carcinogenetic PAHs.

TEF: Toxic equivalence factors (data from (Collins et al., 1998; Tsai et al., 2004).

TEQ: Toxic equivalency concentration.

	Acy	F1	Phe	Ant	Pyr	BaA	Chr	B(b)F	B(k)f	BaP	IcdP	DBA	BghiP	{PAH	LMW PAHs	HMW PAHs	TEQ
Асу	1.000																
Fl	0.733**	1.000															
Phe	0.407^{*}	0.749**	1.000														
Ant	0.312	0.526**	0.649**	1.000													
Pyr	0.212	0.387^{*}	0.696**	0.603**	1.000												
BaA	0.182	0.366*	0.614**	0.685**	0.838**	1.000											
Chr	0.170	0.242	0.638**	0.539**	0.830**	0.777^{**}	1.000										
B(b)F	0.157	0.246	0.611**	0.561**	0.838**	0.871**	0.940**	1.000									
B(k)f	0.075	0.223	0.595**	0.632**	0.846**	0.840^{**}	0.853**	0.913**	1.000								
BaP	0.162	0.340	0.552**	0.506**	0.783**	0.900^{**}	0.769**	0.878^{**}	0.836**	1.000							
cdP	-0.021	0.101	0.414*	0.348	0.575**	0.598**	0.631**	0.655**	0.664**	0.652**	1.000						
DBA	0.116	0.256	0.433*	0.387^{*}	0.433*	0.433*	0.433*	0.433*	0.433*	0.434*	0.619**	1.000					
BghiP	-0.056	0.211	0.451*	0.377^{*}	0.635**	0.653**	0.673**	0.773**	0.679**	0.798**	0.527**	0.441*	1.000				
ΣPAHs	.0319	0.531**	0.777**	0.782**	0.915**	0.800^{**}	0.822**	0.818**	0.863**	0.755**	0.596**	0.433*	0.594**	1.000			
LMW PAHs	0.548**	0.770**	0.826**	0.815**	0.580**	0.464**	0.484**	0.429*	0.515**	0.398*	0.331	0.432*	0.297	0.786**	1.000		
HMW PAHs	0.141	0.309	0.646**	0.603**	0.966**	0.895**	0.900**	0.928**	0.913**	0.875**	0.629**	0.433*	0.728**	0.906**	0.510**	1.000	
TEQ	0.163	0.327	0.584**	0.587**	0.836**	0.927**	0.834**	0.927**	0.899**	0.980**	0.658**	0.433*	0.782**	0.823**	0.453*	0.923**	1.

** Correlation is significant at the 0.01 level (2-tailed).

* Correlation is significant at the 0.05 level (2-tailed).

Table 10

	Factor 1	Factor 2	Factor 3
Acy	-0.123	0.783	-0.044
Fl	0.137	0.947	-0.059
Phe	0.355	0.635	0.038
Ant	0.116	0.004	0.991
Pyr	0.922	0.229	0.110
BaA	0.982	0.055	0.079
Chr	0.935	0.197	0.104
B(b)F	0.983	0.107	0.072
B(k)f	0.970	0.137	0.124
BaP	0.984	0.110	0.049
IcdP	0.978	0.051	0.071
DBA	0.953	-0.011	0.077
BghiP	0.955	0.085	0.044
Total variance %	63.606	16.365	7.501
Cumulative %	63.606	79.971	87.472

Tab	le	11	
1 uoi			

Compounds	TEE	NCs MPCs		RQ (NCs)	RQ (MPCs)			
Compounds	1 LT	$(\mu g kg^{-1})$						
Acy	0.001	1.2	120	70	0.7			
Fl	0.001	1.2	120	139	1.4			
Phe	0.001	5.1	510	75	0.7			
Ant	0.001	1.2	120	87	0.9			
Pyr	0.001	1.2	120	628	6.3			
BaA	0.100	2.5	250	41	0.4			
Chr	0.010	107	10700	3	0.0			
B(b)F	0.100	2.5	250	61	0.6			
B(k)f	0.100	24	2400	4	0.0			
BaP	1.000	2.6	260	23	0.2			
IcdP	0.100	59	5900	3	0.0			
DBA	1.000	2.6	260	18	0.2			
BghiP	0.010	75	7500	0.9	0.0			
ΣPAHs				1151	7.7			

Table 12

	Child				Adult				
Exposure pathways	ILCRsing	ILCRs _{der}	ILCRs _{inh}	Cancer risk	ILCRsing	ILCRs _{der}	ILCRs _{inh}	Cancer risk	
Mean	3.98E-04	4.97E-04	7.70E-09	8.95E-04	3.11E-04	5.52E-04	2.41E-08	8.64E-04	
Min	5.43E-05	6.77E-05	1.10E-09	1.22E-04	4.24E-05	7.53E-05	3.30E-09	1.17E-04	
Max	2.34E-03	2.91E-03	4.54E-08	5.26E-03	1.82E-03	3.24E-03	1.41E-07	5.07E-03	