

**Controls on the phosphorus content of fine stream bed
sediments in agricultural headwater catchments at the
landscape-scale**

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2 There have been no landscape-scale assessments which quantify the relative importance
3 of the organic and mineral properties of BS (bed sediment) and associated catchment
4 characteristics (geology, land cover and mean topsoil phosphorus (P) content) for BSP
5 concentration. Mid infra red diffuse reflectance spectrometry was applied to estimate
6 the quantities of organic matter, dithionite extractable aluminium- (Al_d) and iron (Fe_d),
7 kaolinite, dioctahedral clay and mica (D&M) minerals in 1052 snapshot samples of fine
8 ($< 150 \mu m$) BS in small to medium-sized (5-55 km²) agricultural headwater catchments
9 across a large area (15 400 km²) of central England. Analyses included estimates of
10 BS specific surface area, cerium (Ce) concentrations (enriched in P-bearing apatite and
11 P-fertilisers), and catchment average topsoil P content.

12 Simple linear regression demonstrated that the proportion of variance in BSP
13 explained by specific components of BS across all catchments declined in the follow-
14 ing order: $Al_d > Fe_d > \text{topsoil P} = \text{kaolinite} = \text{residual iron} > \text{organic matter} = \text{Ce}$
15 $> \text{D\&M} > \text{mineral SSA}$. No single component accounted for more than 36% of the
16 variance in BSP. Multiple regression – including a classification of bedrock lithology
17 and proportions of arable and grassland by area – accounted for 61.9% of the vari-
18 ance in BSP. The proportions of arable and grassland by area in each catchment were
19 statistically significant but the coefficients – negative and positive, respectively – were
20 counter-intuitive. Across this large region – with widely differing geology and soils –
21 Fe_d in BS is more strongly associated with kaolinite than D&M minerals because iron-
22 oxyhydroxides and kaolinite form contemporaneously during pedogenesis. The SSA of
23 fine bed sediments is largely determined by catchment area, fitted accurately using a
24 power function.

25 **1. Introduction**

26 The dominant factors which control dissolved phosphorus (P) concentrations in head-
27 water streams are the quantities and types of P in their bed sediments (BS; House,
28 2003; van der Perk et al., 2006). Agricultural land use has been identified as a source
29 of enhanced inputs of particulate P (PP) to headwater bed sediments (Withers et al.,
30 2001) which can lead to elevated levels of dissolved P with profound implications for
31 water quality (Smith, 2003).

32 Transport and delivery of P to headwater sediments in agricultural catchments is
33 complex. There are differing diffuse and point sources which follow varying pathways
34 (erosive inputs, drains, bank slip) and delivery mechanisms (Beven et al., 2005). The
35 largest concentrations of P typically occur in the finer fractions of BS (Evans et al.,
36 2004) because they contain material with the largest P sorption capacities. Much is
37 known about the forms of PP (organic and inorganic) in soils and bed sediments and
38 chemical associations of inorganic P with specific mineral phases. For example, ad-
39 sorption and occlusion by Al and Fe oxyhydroxides (Hartikainen et al., 2010) , specific
40 and non-specific adsorption on clay minerals (Edzwald et al., 1976) and precipitation
41 as iron-phosphate phases (Emerson and Widmer, 1978), the occurrence of P in rare
42 earth element (REE) enriched apatite (Starinsky et al., 1982) or phosphate fertilisers
43 (Abdel-Haleem et al., 2001), or co-precipitation of phosphorus with calcite (Koschel
44 et al., 1978). The importance of these associations has been demonstrated in selected
45 catchments (e.g. Ballantine et al., 2009; Evans et al., 2004; Palmer-Felgate et al., 2009)
46 but their relative importance is hard to assess because of landscape-scale variations in
47 soil mineralogy (e.g. iron and aluminium oxyhydroxide and clay mineral composi-
48 tion), land use and management, geomorphology and geology. The latter – along with
49 hydrology – was identified as playing an important role in determining spatial and
50 temporal variation in storage of bed sediment P (Ballantine et al., 2009). In their
51 study, Palmer-Felgate et al. (2009) reported higher concentrations of bed sediment
52 phosphorus (BSP) at sites with the greatest agricultural impact across three, paired

53 headwater catchments in lowland England. No studies to date have investigated the
54 quantities of the different forms of P stored in bed sediment at larger, landscape scales.

55 One larger scale study (van der Perk et al., 2007) assessed the relative importance
56 of a range of BS element concentrations to explain variations in total bed sediment P
57 in headwaters of the moderate-sized (976 km²) Tamar catchment (SW England). They
58 showed that the total concentrations of five elements (Al, Fe, K,Ca and Mn) in fine
59 BS collected during the summer months accounted for 32% of BSP, but they did not
60 attempt to identify either the organic matter content or specific mineral phases in their
61 sediment samples. In the research described here, inferences are made regarding the
62 association between BSP and specific components of fine BS based on the amounts of
63 variation accounted for in the former by the latter. For example, much P is occluded
64 within the structures of iron-oxyhydroxides and P is also a constituent of organic
65 matter in sediment. So although these species (e.g Fe-oxyhydroxide occluded P) are
66 not measured directly, their presence is inferred because they account for the variation
67 in BSP. This approach can contribute to our understanding of the processes which
68 determine P delivery to BS and could help to enhance process-based models of sediment
69 P delivery to stream channels (e.g. Davison et al., 2007).

70 An improved understanding of the importance of the associations of P with spe-
71 cific organic and mineral components in BS samples can be captured by combining
72 the use of landscape-scale, snapshot sediment surveys (Johnson et al., 2005) with mid
73 infra red diffuse reflectance spectrometry (MIR-DRS). Analysis of sediments by MIR-
74 DRS is a rapid and cost-effective technique to accurately estimate the concentrations
75 of several important BS constituents. This technique is particularly effective in circum-
76 stances where mineral phases are both amorphous and crystalline. Diffuse reflectance
77 spectrometry has been widely applied in soil science, but less so in the analysis of
78 fluvial sediments (Bertaux et al., 1998; Martinez-Carreras et al.,2010). For accurate
79 quantification, some phases require statistical models to be developed between primary
80 measurements on sediment samples and their spectra before the models can be used

81 to estimate concentrations of these properties in a larger number of samples. In their
82 study, Martinez-Carreras et al. used the visible and near infra region, but more accu-
83 rate estimates of specific soil phases are often observed using the mid infra portion of
84 the spectrum (Madari et al., 2006).

85 Another property of sediments which may influence total P concentrations is
86 the specific surface area (SSA) of its mineral phases. In this paper, the term SSA
87 is used to denote measurements based on the BET isotherm (nitrogen adsorption)
88 which accounts for the surface texture and internal surfaces of certain mineral phases.
89 This is different to estimates of surface area (SA) based on particle size analyses with
90 assumptions regarding particle sphericity. No significant relationships between total
91 P and surface area measured in fine ($<63\mu\text{m}$) BS were reported for the Tern, Pang
92 and Lambourn catchments in the UK (Ballantine et al., 2009), nor between SSA and
93 total P for the Enborne catchment (Evans et al., 2004), but a significant positive
94 correlation was reported between SSA and P in the Lambourn (Evans et al., 2004).
95 Although measurements of SSA tend to be relatively costly, Rawlins et al. (2010)
96 recently showed how accurate estimates of mineral SSA of fine BS (following removal
97 of organic matter) could be made from the concentrations of four elements for a large
98 area of central England.

99 In this paper, statistical analyses are presented showing the relative importance
100 of ten components of fine BS and three catchment features which account for variations
101 in total BSP from 1052 headwater catchments for an area of 15 400 km² across central
102 England dominated by agricultural land use. Quantitative estimates of specific BS
103 components include organic matter, mineral SSA, dithionite extractable aluminium-
104 (Al_d) and iron-oxyhydroxides (Fe_d), residual iron, kaolinite and dioctahedral clay and
105 mica (D&M) minerals. A series of catchment-specific features are also included in the
106 statistical analysis: the variation in mean topsoil P concentrations, dominant lithology
107 and the proportions of arable and grassland. The quantities of organic matter, Al_d ,
108 and Fe_d were estimated accurately from statistical models developed between primary

109 measurements and MIR spectra, whilst the relative amounts of kaolinite and D&M
110 minerals in each BS sample were estimated from unique adsorption features. The
111 relationship between mineral SSA and headwater catchment area (5-55 km²) is also
112 investigated. The implications of the findings from this study for understanding P
113 storage in BS and for modelling the processes governing the transfer of P from soil to
114 aquatic ecosystems are discussed.

115 **2. Study region and methods**

116 *2.1. Geology, soils and land use.*

117 Bedrock in the study region ranges in age from Precambrian to Cretaceous with a wide
118 range of predominantly sedimentary lithologies including limestones, sandstones, silt-
119 stones, mudstones, shale, coal measures, marls, ironstones and chalk (Figure 1). There
120 are also extensive superficial deposits including glacial tills, glacial sands and gravels,
121 marine and river alluvium, river terrace deposits, and to the east of the region, peat de-
122 posits (Sylvester-Bradley and Ford, 1968). The soils are dominated by Brown Earths,
123 Surface Water Gleys, Pelosols, Ground Water Gley Soils (Soil Survey of England and
124 Wales, 1983a; Soil Survey of England and Wales, 1983b). The elevation range across
125 the region is between 20 and 255 m above sea level, with undulating topography; the
126 mean slope angle is around 2.5 °. The proportions of land use types across the entire
127 study region are arable (48%), grassland (21%), built-up areas (13%), woodland (7%)
128 and small areas of a range of other land use types (11%.)

129 *2.2. Bed sediment and topsoil surveys*

130 Figure 2 shows features of the study region – which covers around 15 400 km² of central
131 England – from which selected stream sediments samples were collected. The stream
132 sediment samples were collected by the G-BASE project of the British Geological
133 Survey (Johnson et al., 2005). The stream sediment sampling was undertaken in the
134 summers of 1997, 1998 and 1999 in rural and peri-urban areas.

135 Potential stream sampling sites were identified using Ordnance Survey maps.

136 Mainly first and second order streams were selected, either avoiding or located upstream
137 of obvious sources of contamination such as road intersections and farm buildings.
138 Where possible, sediment was collected from central areas of active stream beds after
139 removal of the upper layer of oxidised sediment. Between 15 and 25 kg of sediment
140 was wet-screened on site to collect the fraction finer than $150\mu\text{m}$ typically yielding a
141 final mineral mass of approximately 500 g. The finer than $150\mu\text{m}$ fraction was in part
142 chosen because it was shown to be most effective in discriminating between major and
143 trace element bed sediment geochemistry based on orientation surveys (Plant, 1971).
144 All samples were returned to a local field base for air-drying. The location of each
145 stream bed sampling site was recorded using a handheld GPS with an accuracy of
146 around 5 m. At site, the width of each stream, stream order and a classification of flow
147 conditions during sampling were recorded.

148 In total there were 5047 (not shown) stream sediment sampling sites across the
149 study region. The sites comprise a range of : i) stream orders (1st to 4th), ii) flow
150 conditions (no flow to bank full), and iii) stream channel sizes (widths of a few feet to
151 several metres). We wished to make comparisons of fine BS for streams of similar sizes
152 and flow conditions to avoid any bias such variation might introduce into our analysis.
153 We therefore restricted the number of stream sediment sampling sites to: i) first and
154 second order streams, ii) low to moderate flow conditions, and iii) streams with channels
155 of between 1 and 3 metres width. Of the original 5047 stream sediment sites, a total
156 of 1972 sites met these conditions. Of these sites, a total of 1052 were in catchments
157 where the bedrock geology was sufficiently widespread to form the dominant lithology
158 (see below) in at least 10 catchments across the study region.

159 Topsoil sampling sites were chosen from alternate kilometre squares across the
160 study area chosen by simple random selection within each square, subject to the avoid-
161 ance of roads, tracks, railways, urban land and other seriously disturbed ground. At
162 each site, surface litter was removed and soil was sampled to a depth of 15 cm us-
163 ing five holes at the corners and centre of a square with a side of length 20 m by a

164 hand auger and combined to form a bulked sample. All samples of soil were dried and
165 disaggregated and passed through a sieve (2 mm).

166 On return to the laboratory, all topsoil and BS samples were coned and quartered
167 and a 50-g sub-sample was ground in an agate planetary ball mill.

168 *2.3. Chemical analysis and estimation of specific surface area*

169 *2.3.1. Sediment geochemistry and specific surface area*

170 The total concentrations of a range of major (including Al, Fe, K, Ca, P and Mn) and
171 trace (including cerium (Ce), Lanthanum (La), molybdenum (Mo), vanadium (V) and
172 rubidium (Rb)) elements were determined in each BS and topsoil sample by wavelength
173 and energy dispersive X-ray fluorescence spectrometry (XRFS). Specific surface area
174 based on nitrogen adsorption was determined on a subset of 60 samples after removal
175 of organic matter. These 60 samples were selected by maximizing the range of total Al
176 concentrations, whilst maintaining a random component in the selection. All samples
177 were prepared using a Micromeritics Gemini VacPrep Degasser; the samples were de-
178 gassed overnight at 60 °C prior to SSA analysis. The BET specific surface area of each
179 sample was determined using a multi pressure point analysis using a Micromeritics
180 (Norcross, GA, USA) Gemini VI 2385C series physisorption system. A carbon black
181 standard was analysed with each batch of samples to monitor accuracy and precision.
182 With the exception of BS samples with the largest Mo concentrations, Rawlins et al.,
183 (2010) showed that SSA in BS from the study region could be accurately estimated
184 for all samples using the total concentration of four elements (V, Al, Rb, and Ca) in a
185 subset of 56 samples by multiple linear regression. This regression model was applied
186 to the data to estimate mineral SSA at the selected sites (n=1052; section 2.2).

187 *2.3.2. Extraction of iron and aluminium oxyhydroxides*

188 The 60 samples which had been selected for SSA determination were also used for the
189 determination of dithionite extractable iron (Fe_d) and aluminium (Al_d) oxyhydroxide
190 concentration. In common with other extractants, dithionite is non-specific; it extracts

191 into solution a range of soil constituents, but dominantly iron oxyhydroxides. Although
192 dithionite is not used routinely as an extractant for aluminium, it has been shown
193 to extract large quantities of non-crystalline aluminium phases from soils (Bera et
194 al., 2005). A mass of approximately 1 g of ground material was weighed using a
195 calibrated balance – the mass recorded – and placed into a 30 ml centrifuge tube. To
196 this, 20 ml of 25% (w/v) sodium citrate ($\text{Na}_3\text{C}_6\text{H}_5\text{O}_7 \cdot 2\text{H}_2\text{O}$) and 5 ml of 10 % (w/v)
197 sodium dithionite ($\text{Na}_2\text{S}_2\text{O}_4$) was added. The solutions were shaken overnight on a
198 bench-top shaking table and centrifuged the following day at 1370 **g** for 4 minutes. A
199 10 ml aliquot of solution was taken and the concentrations of total iron and aluminum
200 were determined by ICP-AES using matched standard and solution matrices. The
201 concentrations were converted into mass equivalents of Fe_d and Al_d (mg kg^{-1}).

202 2.3.3. *Total organic carbon*

203 Total organic carbon was determined for a total of 88 bed sediment samples; the 60
204 samples referred to above and a further 28 samples; their locations are shown in Fig-
205 ure 2. A 300 mg sub-sample was treated with a small quantity of HCl (5.7 M) to
206 remove any inorganic carbon and total organic carbon was estimated on the remain-
207 ing sample by combustion in a Costech ECS4010 Elemental Analyser (EA) calibrated
208 against an Acetanilide standard. Replicate analysis of well-mixed samples indicated
209 a precision of $\pm < 0.1\%$. In terrestrial sediments and soils, organic carbon comprises
210 close to 50% of the mass of organic matter (Prybil, 2010), so the concentrations of
211 OC reported from the analyses were used to represent the relative amount of organic
212 matter in each sediment sample.

213 2.4. *Landscape analysis*

214 2.4.1. *Catchment-based geological classes and land use types*

215

216 A 5 m resolution digital terrain model (DTM; Intermap, 2009) of the study region
217 and hydrological functions in ArcMap9.3TM(ESRI) were used to generate drainage

218 networks. Upon these were superimposed the locations of each of the 1052 sediment
219 sampling sites and these were snapped to the nearest stream, ensuring that where
220 two streams were close together, the correct stream had been selected by reference
221 to positions which had been recorded on hardcopies of Ordnance Survey maps during
222 sample collection. Hydrological functions and the DTM were used to delineate polygons
223 of the sub-catchment areas draining to each of the sampling sites and the catchment
224 sizes (km^2) were calculated for each.

225 Digital versions of the 1:50 000 maps of bedrock geology of England – part of
226 DigMap GB of the British Geological Survey (2006)– were used to determine the types
227 and proportions of each bedrock formation which outcrop in each catchment. The
228 catchment polygons were overlaid onto a layer of bedrock polygons with codes for each
229 class in a GIS system; the former were cut into sections using the latter. The GIS was
230 used to calculate the proportions of each bedrock type in each catchment based on the
231 codes for each bedrock formation polygon. Using the same procedure, digital versions
232 of the 25 m resolution Land Cover Map of Great Britain 2000 (Fuller et al., 2002) –
233 with codes for each dominant habitat type – were used to determine their proportions
234 in each catchment of the sediment sampling sites.

235 To determine the influence of bedrock type on BSP concentration, it was neces-
236 sary to identify those catchments which were dominated by one lithology and remove
237 any catchments which were anomalously large. Catchments where a single bedrock for-
238 mation accounted for more than 50% of its area – and where the sub-catchment area
239 was $< 55 \text{ km}^2$ – were selected; this left a total of 1052 catchments from 17 geological
240 classes for our analysis. For land use type, the proportions of arable and grassland
241 were determined in each catchment.

242 2.4.2. *Estimation of mean catchment topsoil P concentration*

243 To determine average catchment topsoil concentrations, a procedure which combined
244 the proportions of each parent material (PM) – bedrock geology or where present
245 overlying Quaternary deposits – in each catchment with the local concentrations of

246 total P in topsoil samples was used. This approach is described in greater detail by
247 Appleton et al. (2008). It provides optimal estimates of average catchment topsoil P
248 because the parent material classification accounts for a substantial proportion (33%)
249 of its variation based on an ANOVA analyses undertaken for the current study by the
250 author. In a GIS system, discrete polygons of soil parent material and codes associated
251 with them were created from 1:50 000 scale digital versions of bedrock and Quaternary
252 deposits. Based on location, each topsoil sample P concentration was linked to a
253 unique PM polygon and its associated code. The concentration of topsoil P for each
254 PM polygon was calculated as a distance weighted average for the five nearest samples
255 over the same PM polygon. The catchment boundaries were then overlaid on the
256 parent material polygons to determine the proportion of each unique parent material
257 polygon in each catchment. A final average topsoil P concentration for each catchment
258 was calculated as the area-weighted average of the P concentrations in each of the PM
259 polygons in the catchment.

260 2.5. *Mid infra red diffuse reflectance spectrometry (MIR-DRS)*

261 2.5.1. *Diffuse reflectance measurements.*

262 A Biorad Excalibur series (GS3000MX) fourier transform infra red spectrometer was
263 used for measuring the diffuse reflectance of each sediment sample with a Pike Techolo-
264 gies (Madison, WI) EasiDiffTM accessory. Background spectra were collected using a
265 powdered KBr sample (Pike Technologies, Madison, WI). Each sample was scanned 40
266 times at a resolution of 4 cm⁻¹ in the range 400–4000 cm⁻¹.

267 2.5.2. *Chemometrics.*

268 The spectral reflectance data were analysed using the *pls* package (Mevik and Wehrens,
269 2007) in the R environment (R Development Core Team, 2010) to form partial least
270 squares regression (PLSR) models based on the orthogonal scores algorithm. Cross
271 validation was applied to select the optimum number of components from which to
272 form the PLSR models; prior to forming each model, 10% of the samples were selected

273 randomly and were not used in model fitting. These samples were then used to as-
 274 sess the model performance by forming predictions and calculating the coefficient of
 275 determination (R^2) and root-mean-squared-error of cross validation (RMSE-CV) for
 276 between 1 and 12 model components. In the case of Fe_d , the quantity of residual iron
 277 in each sample – comprising a range of iron-bearing mineral phases – was calculated
 278 as the difference between total Fe (measured by XRF) and Fe_d .

279 To determine the significant wavelengths for prediction of the Al_d , Fe_d and OC,
 280 both the Variable Importance in the Projection (VIP) (Chong and Jun, 2005) and the
 281 PLS regression coefficients (b-coefficients; Haaland and Thomas, 1988) were used. For
 282 an observed variable y , the VIP was calculated by:

$$VIP_k(a) = K \sum_a w_{ak}^2 \left(\frac{SSY_a}{SSY_t} \right) \quad (1)$$

283 where $VIP_k(a)$ gives the importance of the k th predictor variable based on a model with
 284 a factors, w_{ak} is the corresponding loading weight of the k th variable in the a th PLSR
 285 factor, SSY_a is the explained sum of squares of y by a PLSR model with a factors,
 286 SSY_t is the total sum of squares of y , and K is the total number of predictor variables.
 287 The wavelength is considered important if the values of both the b-coefficients and VIP
 288 are sufficiently large. In this study, thresholds for VIP were set to 1 (Chong and Jun,
 289 2005) and the standard deviation of the b-coefficients was applied as their threshold.

290 The relative quantities of kaolinite and D&M minerals were estimated using spe-
 291 cific adsorption features in the MIR spectra. Kaolinite has a sharp, dominant adsorp-
 292 tion feature around wavenumber 3700 cm^{-1} (Ferraro, 1982) and D&M minerals around
 293 3625 cm^{-1} . The peak position of these adsorption features varies according to substi-
 294 tution of the mineral lattice (Besson and Drits, 1997) so to estimate their quantities it
 295 was decided to use peak area rather than peak height. The area of the D&M peak was
 296 between 3600 and 3640 cm^{-1} , whilst the kaolinite peak was between 3683 and 3710
 297 cm^{-1} . Ground reference materials were used to calibrate the kaolinite (Metropolitan
 298 Vickers Electrical Co., Manchester, UK) and the D&M (muscovite mica: Hydrite Flat
 299 D, Imerys, Georgia, USA) adsorption features. To assess the accuracy of MIR-DRS

300 for estimating the quantities of kaolinite and D&M minerals in sediment samples, both
301 of these reference materials were added as weighed amounts to a single sediment sam-
302 ple in which both adsorption features were minimal or absent. The addition of the
303 reference materials represented 1, 2, 4, 8, 16 and 20% of the total sample mass on a
304 weight-for-weight basis. Each sample was scanned using the procedure described above
305 and the area under each of the two adsorption features was estimated using a bespoke
306 procedure in the R environment. The percentage quantities of reference material were
307 plotted against the area under the adsorption feature; linear regression models fitted
308 by ordinary least squares to these pairs of points had a coefficient of determination (ad-
309 justed R^2) of 0.84 (kaolinite) and 0.81 (D&M) minerals. This was considered sufficient
310 for the purpose of estimating the relative quantities of these two groups of minerals in
311 all sediment samples.

312 **3. Statistical analysis**

313 *3.1. Multiple linear regression*

314 In this study the following list of ten continuous predictors were selected which could
315 account for variation in BSP concentrations: estimates of BS organic carbon (1), total
316 Fe (2), Fe_d (3), residual Fe (4), Al_d (5), the relative amounts of kaolinite (6) and D&M
317 (7) minerals, total Ce (a REE enriched in P-bearing apatite and P-fertilisers) (8), min-
318 eral SSA (9); and average catchment topsoil P concentration (10). Total Fe was only
319 included for comparison with the other two Fe components (Fe_d and residual Fe) but
320 total Fe was not included in the subsequent multiple regression analysis. The frequency
321 distribution of BSP was strongly positively skewed (skewness coefficient=3.4), so all
322 linear regression analyses were undertaken on log transformed values (skewness coeffi-
323 cient=0.2). As part of an exploratory analysis, simple linear regression using ordinary
324 least squares was used to assess the relative importance of these ten predictors and
325 simple scatterplots were created to assess the degree of correlation between them. As
326 might be expected, these plots showed that some predictors were significantly corre-

327 lated. If these were included in a multiple regression analysis there would be substantial
328 multi-collinearity. To account for this, a principal component analysis was undertaken
329 to elucidate the relationships between a set of orthogonal (uncorrelated) predictors; the
330 scores of the first nine principal components were extracted. To understand how each
331 of the variables relate to each principal component, the eigenvectors were converted to
332 correlation coefficients between the component scores and the original variates using
333 the following equation:

$$c_{ij} = a_{ij} \sqrt{v_j / \sigma_i^2}, \quad (2)$$

334 where a_{ij} is the i th element of the j th eigenvector, v_j is the j th eigenvalue, and σ_i^2 is
335 the variance of the i th original variable.

336 These correlations can then be used to examine which variables dominate the vari-
337 ation in each of the principal components. The nine, orthogonal principal components
338 could then be used in a traditional multiple regression analysis – avoiding problems
339 associated with multi-collinearity – based on the interpretation of the variables which
340 dominate the components.

341 In addition to the nine continuous variables (all excluding total Fe), two other
342 types of predictor were included in the statistical analysis. The first was dominant
343 catchment lithology; which was added to the multiple regression model as a factor
344 with 17 classes. The second was the proportion of arable land and grassland in each
345 of the catchments. These two classes on average accounted for 60% of the land cover
346 in each catchment. Their influence on BSP can be assessed by converting them to
347 compositional variates by taking their additive log ratios (alr; Aitchison, 1986). There
348 are V variables – here the proportions of different land use types within a catchment –
349 which sum approximately to 1 (or 100%). From these are chosen any $V - 1$ variables
350 with values for each unit z_1, z_2, \dots, z_{V-1} . These are transformed to:

$$q_i = \ln\left(\frac{z_i}{z_V}\right) \quad \text{for all } i = 1, 2, \dots, V - 1, \quad (3)$$

351 where z_V is the value of the remaining V th variable. The resulting values over all
 352 units have by definition a logistic normal distribution, and can be analysed by linear
 353 regression – as any other multivariate normal data. So the proportions of land use types
 354 (arable or grassland) were converted to their additive log ratios using the reciprocal
 355 proportion to close the composition (e.g. 100% minus arable or pasture) and each of
 356 the two alr's were included as continuous predictors in the regression analysis.

357 Multiple regression based on ordinary least squares was preformed with GEN-
 358 STAT v12.1 (Payne, 2008) using the FIT directive and the ACCUMULATED option. The
 359 latter accumulates the analysis of variance and sequentially tests the addition of each
 360 predictor to the model to identify whether each predictor is statistically significant.
 361 Each of the first 9 principal components (see above) were added to the model, followed
 362 by the land use types and the geological classification as a grouped variable. In the
 363 regression model, interactions between the geological classification and the nine prin-
 364 cipal components were also included to test for their significance; where these were
 365 statistically significant they were retained.

366 *3.2. Comparing specific surface area and catchment area*

367 Previous research has shown that mathematical functions (e.g. power, exponential or
 368 linear) can accurately describe relations between river length and features of particle
 369 size distributions (psd) (Morris and Williams, 1999 and references therein). In the case
 370 of headwater catchments, river (or stream length) can be hard to establish because some
 371 may be ephemeral, so in this study, catchment area (section 2.5.1) is used in preference
 372 to river length. Given that mineral SSA is strongly related to the psd of sediments
 373 and the samples in this study span a range of catchment sizes, it seemed plausible that
 374 a mathematical relationship could be established between SSA and catchment size.
 375 These two variates were plotted against one another for all 1052 sites and power and

376 exponential functions were fitted to them using the nonlinear least-squares function
377 (nls) from the MASS library in the R environment (R Development Core Team, 2010).
378 The power function has the form:

$$SSA = a + b \times A^c \quad (4)$$

379 where in this case A is catchment area (km²) and a, b and c are coefficients.

380 **4. Results and their interpretation**

381 The summary statistics for total BSP concentrations across the study region show that
382 the maximum and minimum concentration varies more than 40-fold (Table 1) reflecting
383 both natural variations in the spatial distribution of native soil P (inherited from parent
384 material), differing rates of fertiliser-P and/or manure application and variations in the
385 efficacy of mechanisms of P delivery to the drainage network. The median fine BSP
386 concentration across the study region (1266 mg kg⁻¹) is consistent with the ranges of
387 values reported for other agricultural catchments in the UK (see Table 6 in Ballantine
388 et al., 2009); however the maximum value (11 480 mg kg⁻¹) is far greater than has
389 been reported elsewhere. Given that the snapshot sampling of BS was undertaken in
390 the summer months, it is likely that the concentration and storage of P is likely to be
391 near its annual maximum (Ballantine et al., 2006).

392 The mean value of OC measured in 88 samples was 3% (6% organic matter).
393 Direct comparisons with published values for organic carbon in topsoils and subsoils
394 to assess their relative contributions to BS across the region are problematic because
395 of the differences in the size fractions analysed; < 2mm in the case of soil and <150µm
396 in BS. Total BS iron concentrations were generally large (median 4.9%) reflecting the
397 larger than average concentrations of soil Fe inherited from parent material across this
398 region of England (McGrath and Loveland, 1992).

399 *4.1. Estimating Fe_d, Al_d and organic carbon by MIR-DRS*

400 The performance of the PLSR models for the estimation of Fe_d, Al_d and OC by MIR-
401 DRS are summarised in Table 2 and graphically in Figure 3 . For all three variables,

402 the optimal models consisted of 5 or 6 orthogonal components and in each case the
403 coefficients of determination were large; in the case of model formulation (adj R^2 ;
404 range 0.8 – 0.89) and for cross-validation (adj R^2 ; range 0.69 – 0.80). The significant
405 adsorption bands (Figure 3) are consistent with published values of specific features in
406 the infra red spectra of organic carbon compounds (Baes and Bloom, 1989), amorphous
407 and crystalline iron oxides (including ferrhydrite, hematite and goethite; Cornell and
408 Schwertmann, 2003) and the OH-stretching region of aluminium oxides (3400 – 3600
409 cm^{-1} ; Elderfield and Hem, 1973).

410 The cross-validation errors (Table 2) were considered to be sufficiently small to
411 apply the models to estimate the concentrations of each of the three variables in all
412 sediment samples. Summary statistics of the measured and estimated values for each
413 of the three variables (Table 1) show that – with the exception of a few large OC values
414 (cf. Figure 4) – the range of the latter are only slightly larger than the former, although
415 the difference is not sufficient to be of great concern. This might be the case if the
416 PLSR models were extended much beyond the ranges for which they were developed.
417 The median values for the measured and estimated distributions are also quite similar
418 for each of the three variables which suggests that the former are representative of the
419 latter.

420 4.2. *Simple linear regression*

421 The results of simple linear regression in which each of the selected predictors (including
422 total Fe and average topsoil P) were used as explanatory variables for the estimation
423 of the log BSP concentration are presented in Table 3. Scatter plots between nine BS
424 properties (including log P concentration) are shown in Figure 5. The BS properties
425 which explain the largest proportion of the variation in log BSP are the concentrations
426 of log Al_d and log Fe_d , these minerals typically have the largest sorption capacities for
427 P in soils (Hartikainen et al., 2010). The concentrations of log Fe_d and log Al_d are
428 strongly correlated (see Figure 4) because they form in similar pedogenic environments
429 and are subsequently transported together. The proportion of variance explained in log

430 BSP by $\log Fe_d$ is substantially larger (27.7%) than either total Fe (20.9%) or residual
431 Fe (16.6%) suggesting that more P is associated with oxyhydroxides of iron than other
432 iron-bearing mineral phases. It is noteworthy from figure 5 that \log (total) Fe in BS
433 has strong positive linear correlations with both $\log Fe_d$ (Pearson $r = 0.84$) and $\log Al_d$
434 (Pearson $r = 0.76$).

435 The relative concentration of kaolinite mineral phases account for substantially
436 more of the variation in BSP than D&M minerals; 17.4 and 4.7% respectively. Around
437 twelve percent of BSP was accounted for by the occurrence of organic forms of P
438 which occurs in organic matter. The concentration of the rare-earth element cerium
439 (Ce) also accounts for a substantial portion (10.4%) of the variation in BSP suggesting
440 that a significant quantity of P in BS is associated with REE-enriched apatite or
441 phases derived from the addition of REE-enriched P fertiliser. Mineral SSA did not
442 by itself account for a large proportion of the variance in BSP (1.4%), although it is a
443 statistically significant predictor. Mineral SSA has a strong positive correlation with
444 $\log Fe_d$ (Pearson $r = 0.6$) because amorphous forms of the latter commonly account for
445 a large fraction of the surface area of aquatic sediments (Wang et al., 1997). However,
446 as SSA only accounts for a small proportion (0.8%) of the variation in \log BSP, BS
447 fractions $>$ around $10\mu\text{m}$ in diameter must also contain substantial concentrations
448 of P. The scatter plots (Figure 5) demonstrate substantial correlations (linear and
449 log-transformed) amongst some of the bed sediment properties which account for the
450 variation in BSP, specifically kaolinite, Fe_d , Al_d , SSA and Ce. This confirms that it was
451 necessary to undertake multiple linear regression analysis based on the uncorrelated,
452 principal component scores of the significant predictors.

453 Summary statistics for the proportions of arable and grassland in 547 catchments
454 are shown in Table 4. It is noteworthy that neither of the predictors of the propor-
455 tions of land use in each catchment (arable or grassland) are statistically significant
456 – they accounted for only small proportions of the variance in BSP (0.1 and 0.45 %
457 respectively). Their influence may be more complex than can be represented by simple

458 estimates of their proportions in each catchment. For example, more information on
459 the intensity of agricultural inputs, land management practices and connectivity be-
460 tween the hillslopes and channels may be required for the effects of land use type to
461 be fully evaluated.

462 4.3. *Principal component analysis*

463 The correlation between the nine variables and their PC scores (Equation ??) are sum-
464 marised in Table 5 along with the magnitude of the latent roots and their variances.
465 The first three PCs account for 80% of the variance in the set of nine predictors. The
466 correlations between nine predictors and the first three PC scores are summarised in
467 Figure 6. The first PC is dominated by Fe_d , Al_d and kaolinite. The close association
468 between kaolinite and Fe_d in Figure 6a is particularly strong, although the concentra-
469 tions of Al_d are also correlated strongly with the other two components. The former
470 can be explained by the stable association between kaolinite and iron oxyhydroxide
471 minerals in pedogenic environments formed by the release of structural iron from the
472 weathering of silicate minerals (Jackson, 1968).

473 The second and third PCs are dominated by OC and SSA, respectively. Cerium
474 (Ce) has the largest absolute correlation co-efficient with the fourth PC and average
475 catchment topsoil P is the largest in the fifth PC. The largest correlations in component
476 six is for D&M minerals, whilst in component seven it is the combination of residual
477 iron and Al_d . Finally, the largest correlations in the eighth and ninth components are
478 for kaolinite and Fe_d . Identification of the dominant features in each of the nine PCs
479 is of importance when interpreting the results of the multiple regression analysis.

480 4.4. *Multiple linear regression*

481 Results from the multiple linear regression analysis are presented in Table 6 in the form
482 of an analysis of variance reporting on the addition of predictors to the regression model.
483 The nine sediment-property derived predictors plus the geological classification – and a
484 set of interactions between them – accounted for a large proportion of (61.9%; adjusted
485 R^2) of the variance in log BSP. The first eight of the nine PCs were significant predictors

486 of BSP ($P < 0.05$) ; the relative importance of each PC can be inferred from their
487 variance ratios (VR) and the increase in adjusted R^2 after addition of each predictor
488 to the model. Larger VRs denote greater predictive power. The most significant
489 predictor (VR=663) is the first PC which was dominated by Fe_d , Al_d and kaolinite;
490 much BSP is sorbed to the surfaces of iron and aluminium oxyhydroxides and these
491 phases appear to be strongly associated with kaolinite.

492 Both PC five (VR=281) and PC four (VR=270) are of similar predictive power.
493 The fourth PC was most strongly correlated with Ce – indicative of primary P-bearing
494 apatite or REE enriched P-fertiliser phases – suggesting that these make a significant
495 contribution to BSP. The fifth PC was dominated by average catchment topsoil P
496 indicating that – as might be expected – its variations between catchments has a
497 substantial impact on BSP content. Organic matter dominated PC 2 which is the
498 next most significant predictor (VR 156) and denotes the occurrence of organic forms
499 of P in BS. Component 7 also has a large VR (121) which appears to be related to
500 residual iron phases and non-crystalline aluminium. Scanning electron microscopy with
501 energy-dispersive X-ray microanalysis of selected BS samples (not shown) confirmed
502 that Fe-phosphate phases were present in some cases. The remaining PCs (3,6 and 8)
503 have relatively modest VRs (range 7 – 52) by comparison with those reported above.
504 The third PC, which was dominated by mineral SSA, has a relatively small VR (15.2)
505 confirming that its role in influencing BSP is relatively minor.

506 A non-paired, two-sided t -test showed there was no statistically significant differ-
507 ence at the 5% confidence level ($P=0.058$) between total topsoil P at arable (n=4310)
508 and grassland (n=1836) sites across the region based on the dominant land use recorded
509 at the soil sampling location. However, the inclusion of the proportions of arable and
510 grassland in each catchment showed that these were both significant predictors with
511 relatively small VRs (8 and 9). The model co-efficients are negative for arable (-0.004)
512 and positive (0.008) for grassland; this may be attributable to the transportation of
513 larger particle size fractions – which contain smaller quantities of P – from arable

514 catchments, thus diluting BSP.

515 The geological classification was also a significant predictor (VR 6.4); it accounts
516 in part for the processes controlling the transport and delivery of P to BS. The sig-
517 nificance of the interactions between the 17 geological classes and the other PC-based
518 explanatory variables were also tested by adding them to the regression model; four
519 were found to be significant (PCs 1, 2, 3, and 7) although the VRs are modest (range
520 2.2–2.4). Stronger interpretations can be placed on the first three PCs because their
521 scores had larger correlation coefficients with specific sediment properties (see Table 4).
522 So the regression model includes 17 coefficients for geological class relating to each of
523 the following: PC1 (Fe_d , Al_d and kaolinite content), PC2 (organic carbon content),
524 PC3 (specific surface area) and PC7 (residual iron and $\log Al_d$).

525 4.5. *Specific surface area versus catchment area*

526 A scatter plot of catchment area versus BS mineral SSA based on the measured values
527 ($n=56$) and estimates from geochemistry (Rawlins et al., 2010) are shown in Figure 7.
528 The power function provided the best fit between catchment area and BS mineral SSA.
529 The values of the model coefficients a, b and c (Equation ??) and the standard errors
530 (in parenthesis) were: -32.0 (7.6), 15.9 (4.9) and 0.40 (0.05), respectively. The fitted
531 model had a residual standard error of 3.2 and accounted for 86% of the variance
532 in mineral SSA (un-adjusted $R^2=0.86$). In a previous study, Rawlins et al. (2010)
533 had shown that dominant catchment bedrock type accounted for 39% of the variance
534 in estimates of mineral SSA from the same region, albeit from a somewhat larger
535 dataset ($n=1236$). Catchment size – or channel length – is clearly the dominant factor
536 determining BS SSA (Figure 7) and the significance of bedrock type highlighted by
537 Rawlins et al. (2010) was due to the influence of dominant lithology on geomorphology
538 and hydrogeology which forms catchments with constrained size ranges. Although the
539 well-established phenomenon of downstream fining based on particle size (Morris and
540 Williams, 1999) has not previously been extended to encompass mineral SSA, it most
541 likely accounts for the relationship between mineral SSA and catchment size shown in

542 Figure 7. Considering that the location of BS sampling sites were not selected on the
543 basis of local stream flow regime (i.e. sampling was not biased towards pools or riffles),
544 the relationship between catchment size and fine BS SSA is remarkably consistent.

545 **5. Discussion**

546 The focus of this study was to determine the controls on BSP concentrations at the
547 landscape-scale using a range of quantitative sediment properties and catchment char-
548 acteristics to aid the understanding of processes relating to PP loss from catchments. A
549 widely-used, process-based model of phosphorus (P) and sediment mobilisation (PSY-
550 CHIC; Davison et al., 2007) has a range of spatial input data including information on
551 crops, P application rates, soil properties (including texture, organic carbon and bulk
552 density), climate data, slope and population density. In PSYCHIC, the finer fractions
553 of eroded soil are enriched in P (ratios 0.1 sand: 0.25 silt: 1 clay), but the model does
554 not account for the spatial distributions of Al and Fe oxyhydroxides which, as shown
555 in this study and previously (House et al., 1995), are the dominant phases with which
556 P is associated in BS across areas dominated by silicate bedrock.

557 Statistical analysis showed that both $\log Fe_d$ and Al_d were strongly correlated
558 with total $\log Fe$ in BS, so the latter could be used to estimate the former. In addi-
559 tion, there is a positive relationship between topsoil Fe concentrations and BSP across
560 the 1052 catchments (Pearson linear correlation $r=0.23$); a least squares regression
561 model shows that this relationship is statistically significant, accounting for 5.2% of
562 the variance in BSP. Geochemical survey data coupled with high-resolution maps of
563 soil parent material across England and Wales can now provide accurate estimates of
564 major element concentrations such as Fe in soil at scales finer than 1 km² (Appleton
565 et al., 2008). The spatial distribution of soil Fe – with a coefficient for P enrichment
566 – might effectively be incorporated into process-based models such as PSYCHIC to
567 improve estimates of PP transfer to BS in agricultural catchments in areas dominated
568 by silicate bedrock. In regions of carbonate bedrock – such as the Cretaceous chalk
569 dominated catchments of southern and eastern England – the importance of carbonate

570 phases in controlling forms of BSP may, in part, negate the role of iron oxyhydroxides
571 and non-crystalline aluminium phases. The spatial distribution of soil and BS Fe could
572 also aid understanding of where ecological risks associated with large concentrations
573 of bioavailable-P are likely to be mitigated by adsorption on the surfaces of Al and Fe
574 oxyhydroxides (Palmer-Felgate et al., 2009).

575 The importance of field drains as a pathway for rapid transfer of PP to stream
576 channels was not included in this study because accurate information on the distribu-
577 tion and functioning of drains was not available at a sufficiently fine spatial resolution
578 (Chapman et al., 2003). For the same reason, information on point source inputs of
579 phosphorus from sewage treatment works, septic tanks or agricultural point sources
580 such as farmyards were not incorporated into the analysis. No attempt was made to
581 investigate the connectivity of hydrological flow pathways from critical sources areas
582 and its impact on mobilisation and delivery of PP to catchment channels (Heathwaite
583 et al., 2005). Such approaches typically require high-resolution (lidar) topographic
584 data which is not available across all our study area. In subsequent analysis, models of
585 sediment mobilisation and delivery could be applied to a subset of the studied catch-
586 ments to investigate the significance of hydrological connectivity. Catchments could
587 be selected to maximise both the variation in hydrological connectivity from potential
588 contributing source areas and also the components which dominate BSP: i) kaolinite
589 and oxyhydroxides of Al and Fe, ii) organic matter and, iii) the rare earth signatures
590 of naturally occurring apatite or P-fertiliser/manure inputs.

591 Landscape-scale variation in bedrock geology was shown to be significant factor
592 controlling BSP concentrations based on statistical analysis. The approach of using of
593 a single, dominant bedrock class for each catchment in the model could be improved in
594 subsequent analysis. First, by accounting for the proportions of the different bedrock
595 types in each catchment and linking them to critical source areas. Second, by using par-
596 ent material distributions at 1:50 000 scale (i.e. including Quaternary deposits) rather
597 than just bedrock types across the catchments. In recently glaciated landscapes such

598 as those in central England, soil parent material class explains more of the variation of
599 soil properties than simple bedrock lithology (Rawlins et al., 2003).

600 Organic matter content had a significant positive correlation with BSP at the
601 landscape-scale; although this relationship has been reported previously (Ballantine et
602 al., 2009) it is not observed consistently (Palmer-Felgate et al., 2009). Where possible,
603 the snapshot sampling protocol adopted by the G-BASE project collects BS from active
604 parts of channels, avoiding areas with low flow velocities where sedimentation of fine
605 organic debris may be enhanced. It is therefore unlikely that this could lead to a positive
606 sampling bias in the reported median organic carbon values (3%; or 6% organic matter)
607 across the study region.

608 Fingerprinting of P in BS of agricultural catchments – either from apatite or REE-
609 enriched P-fertiliser based on their Ce content – has not, to the authors knowledge been
610 demonstrated previously. Quantifying the relative importance of these two P sources
611 (native P or fertiliser/manure-derived P) for BSP would be a considerable research
612 challenge. One potential means of assessing their relative contributions would be to
613 measure the REE profiles of sediment collected from soil erosion experiments in which
614 applied P and native P have markedly different REE fractionation. However, other
615 soil minerals would also contribute to the REE signature of runoff sediment and this
616 may limit the scope for discriminating between the native and applied-P derived REE
617 signatures.

618 Both organic and a range of inorganic mineral components in fine BS were es-
619 timated accurately using MIR-DRS based on unique adsorption features and using
620 chemometric approaches across the full mid-IR spectrum. As shown, BSP is closely
621 associated with these phases, so MIR-DRS – which when used with non-absorbing,
622 powdered KBr requires less than 0.1 g of sample – could be used to monitor changes in
623 their concentrations in suspended sediment samples through storm events to elucidate
624 PP dynamics. Based on measurements of total nitrogen (not reported) from the same
625 subset of 88 BS samples for which OC was determined, it was also possible to accu-

626 rarely estimate the former (adj R^2 0.79; RMSE-CV=0.06%) across a range of total N
627 from 0.1% to 0.51%. Landscape-scale data on the concentrations of total OC, N and
628 P in BS could enhance our understanding of nutrient cycling and losses from agricul-
629 tural systems (Quinton et al., 2010). Establishing the relative quantities of particulate
630 OC and mineral-stabilised OC in fine bed sediments using MIR-DRS (Zimmerman et
631 al., 2007) could also aid fingerprinting of inputs from topsoil and subsoil landscape
632 compartments (e.g. Fontaine et al., 2007).

633 To the authors knowledge, the very strong relationship reported between fine
634 (<150 μ m) BS mineral SSA and catchment area (range 5-55 km²) has not been ob-
635 served before. Bedrock in this region largely comprises a range of silicate-dominated
636 sedimentary sequences. Further investigation is required to assess the nature of this
637 relationship in larger catchments and whether it differs significantly in landscapes with
638 other bedrock types and geomorphic settings.

639 6. Conclusions

640 The main conclusions from this study are:

- 641 1. MIR-DRS is a cost-effective and rapid means to quantify the organic and mineral
642 composition of BS samples. Accurate estimates were made of five components
643 (Al_d , Fe_d , OC, kaolinite and D&M minerals) in 1052 BS from individual agricul-
644 tural headwater catchments.
- 645 2. Simple linear regression demonstrated that the proportion of variance in BSP
646 explained by specific components across all catchments declined in the following
647 order: $Al_d > Fe_d > \text{topsoil P} = \text{kaolinite} = \text{residual iron} > \text{organic matter} = \text{Ce} >$
648 $\text{D\&M} > \text{mineral SSA}$. There was significant correlation amongst these individual
649 predictors of BSP so their scores on nine PCs were used in a multiple regression
650 analysis to elucidate relationships with BS more clearly.
- 651 3. The multiple regression model accounted for 61.9% of the variance in BSP. The
652 first PC which was dominated by Al_d , Fe_d and kaolinite was the most important

653 predictor, but organic matter and average catchment topsoil P also accounted for
654 large proportions of the variance. There was a very strong positive linear corre-
655 lation between the relative quantity of kaolinite and Fe_d in BS which is related
656 to their contemporaneous formation and association in pedogenic environments
657 and subsequent joint transport to BS.

658 4. Cerium (Ce) content of BS is the dominant feature of one of the PCs and ac-
659 counts for a substantial proportion of the variation in BSP; it was inferred that
660 Ce denotes some combination of input from either REE-enriched native P or
661 fertiliser/manure-P, but the relative importance of these two sources cannot be
662 assessed based on the available data.

663 5. A classification based on dominant bedrock geology in each catchment – and
664 four interactions with other BS predictors – was statistically significant because
665 bedrock type partly accounts for the processes controlling the transport and
666 delivery of PP to headwater BS.

667 6. The proportions of arable and grassland by area in each catchment explained a
668 small, statistically significant proportion of the variation in BSP in the MRM.

669 7. Mineral SSA – estimated using four BS geochemical predictors – explained only a
670 small proportion of the variation in BSP. A power function fitted between esti-
671 mates of mineral SSA and catchment area (range 5-55 km²) explained 86% of the
672 variation in the former. This is largely due to the well-established phenomenon of
673 downstream fining, but this has not hitherto been extended to encompass mineral
674 SSA.

675 8. There was a positive relationship between topsoil Fe and BSP. Process-based
676 models of PP mobilisation – which do not currently account for the spatial dis-
677 tribution of soil Fe – might be improved by including estimates of topsoil Fe with
678 a coefficient for P enrichment using high-resolution maps of soil parent material

679 and geochemical survey data. Incorporating data on the distribution of soil Fe
680 into models of P mobilisation could enhance estimates of P delivery to BS.

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693 **References**

- 694 Abdel-Haleem, A.S., Sroor, A., El-Bahi, S. M., Zohny, E. 2001. Heavy metals and rare
695 earth elements in phosphate fertilizer components using instrumental neutron
696 activation analysis. *Appl. Radiat. Isotopes*. 55, 569–573.
- 697 Aitchison, J. 1986. *The Statistical Analysis of Compositional Data*. Chapman & Hall,
698 London.
- 699 Appleton, J.D., Rawlins, B.G., Thornton, I. 2008. National-scale estimation of po-
700 tentially harmful element ambient background concentrations in topsoil using
701 parent material classified soil:streamsediment relationships. *Appl. Geochem.* 23,
702 2596–2611.

- 703 Baes, A.U., Bloom, P.R., 1989. Diffuse reflectance and transmission Fourier-transform
704 infrared (Drift) spectroscopy of humic- and fulvic acids. *Soil Sci. Soc. Am. Proc.*
705 *53*, 695–700.
- 706 Ballantine, D., Walling, D., Collins, A. & Leeks, G. 2006. Phosphorus Storage in Fine
707 Channel Bed Sediments. *Water Air Soil Poll. - Focus.* *6*, 371–380.
- 708 Ballantine, D. J., Walling, D. E., Collins, A. L. & Leeks, G. J. L. 2009. The content
709 and storage of phosphorus in fine-grained channel bed sediment in contrasting
710 lowland agricultural catchments in the UK. *Geoderma.* *151*, 141–149.
- 711 Bera, R., Seal, A., Banerjee, M., Dolui, A. K. 2005. Nature and profile distribution
712 of iron and aluminum in relation to pedogenic processes in some soils developed
713 under tropical environment in India. *Environ. Geol.* *47*, 241–245.
- 714 Bertaux, J., Froehlich, F., Ildefonse, P. 1998. Multicomponent analysis of FTIR
715 spectra: quantification of amorphous and crystallized mineral phases in synthetic
716 and natural sediments. *J. Sediment. Res.* *68*, 440–447.
- 717 Besson, G., Drits, V. A. 1997. Refined relationships between chemical composition
718 of dioctahedral fine-grained micaceous minerals and their infrared spectra within
719 the OH stretching region .2. The main factors affecting OH vibrations and quan-
720 titative analysis. *Clay. Clay Miner.* *45*, 170–183
- 721 Beven, K., Heathwaite, L., Haygarth, P., Walling, D., Brazier, R. & Withers, P. 2005.
722 On the concept of delivery of sediment and nutrients to stream channels. *Hyd.*
723 *Proc.* *19*, 551–556.
- 724 British Geological Survey 2006. *Digital Geological Map of Great Britain 1:50 000*
725 *scale (DiGMapGB-50) data [CD-ROM] Version 3.14.* British Geological Survey,
726 Keyworth, Nottingham.
- 727 Chapman, A. S., Foster, I. D. L., Lees, J. A., Hodgkinson, R. J., Jackson, R. H.
728 2003. Sediment and phosphorus delivery from field to river via land drains in

- 729 England and Wales. A risk assessment using field and national databases. *Soil*
730 *Use Manage.* 19, 347–355.
- 731 Chong, I.-G., Jun, C.-H., 2005. Performance of some variable selection methods when
732 multicollinearity is present. *Chemometr. Intell. Lab.* 78, 103–112.
- 733 Cornell, R. M., Schwertmann, U. 2003. *The Iron Oxides: Structure, Properties,*
734 *Reactions, Occurrences and Uses: Chapter 3 – Characterization.* WILEY-VCH,
735 Weinheim, p 139–183.
- 736 Davison, P., Withers, P. J. A., Lord, E. I., Betson, M. J. and Stromqvist, J. 2007.
737 PSYCHIC - a process based model of phosphorus and sediment mobilisation
738 and delivery within agricultural catchments. Part 1: Model description and
739 parameterisation. *J. Hydrol.* 350, 290–302.
- 740 Edzwald, J. K., Toensing, D. C., Leung, M. C.-Y. 1976. Phosphate adsorption reac-
741 tions with clay minerals. *Env. Sci. Technol.* 10, 485–490.
- 742 Elderfield, H., Hem, J. D. 1973. The development of crystalline structure in alu-
743 minium hydroxide polymorphs on ageing. *Mineral. Mag.* 39, 89–96.
- 744 Emerson SM, Widmer G. 1978. Early diagenesis in anaerobic lake sediments II:
745 Thermodynamic and kinetic factors controlling the formation of iron phosphate.
746 *Geochim. Cosmochim. Ac.* 42, 1307–1316.
- 747 Evans, D.J., Johnes, P.J., Lawrence, D.S. 2004. Phsico-chemical controls on phospho-
748 rus cycling in two lowland streams. Part 2-The sediment phase. *Sci. Tot. Env.*
749 329, 165–182.
- 750 Ferraro, J. R. 1982. *The Sadtler infrared spectra handbook of minerals and clays.*
751 Heyden & Son Ltd. London. 440 pp.
- 752 Fontaine, S., Barot, S., Barre, P., Bdioui, N., Mary, B., Rumpel, C. 2007. Stability
753 of organic carbon in deep soil layers controlled by fresh carbon supply. *Nature.*

754 450, 277–280.

755 Fuller, R.M., Smith, G. M., Sanderson, J. M., Hill, R.A., Thomson, A.G. 2002. The
756 UK Land Cover Map 2000: Construction of a parcel-based vector Map from
757 satellite images. *Cartogr. J.* 39,15–25

758 Haaland, D.M., Thomas, E.V., 1988. Partial least-squares methods for spectral anal-
759 yses.1. Relation to other quantitative calibration methods and the extraction of
760 qualitative information. *Anal. Chem.* 60, 1193–1202.

761 Hartikainen, H., Rasa, K., Withers, P.J.A. 2010. Phosphorus exchange properties
762 of European soils and sediments derived from them. *Eur. J. Soil Sci.* DOI:
763 10.1111/j.1365-2389.2010.01295.x

764 Heathwaite, A.L., Quinn,P.F., Hewett, C. J. M. 2005. Modelling and managing Criti-
765 cal Source Areas of diffuse pollution from agricultural land using flow connectivity
766 simulation. *J. Hydrol.* 304, 446–461

767 House W.A., Denison F.H., Armitage P.D. 1995. Comparison of the uptake of in-
768 organic phosphorus to a suspended and streambed sediment. *Water Res.* 29,
769 767–779.

770 House, W. A. 2003. Geochemical cycling of phosphorus in rivers. *Appl. Geochem.*
771 18, 739–748.

772 Intermap, 2009. NEXTMap Britain. Intermap. <http://www.intermap.com/nextmapbritain>.
773 Accessed 6.12.09).

774 Jackson, M.L. 1968. Weathering of primary and secondary minerals in soils. *Trans.*
775 *Int. Congr. Soil Sci.* 4, 281–282.

776 Johnson, C. C., Breward, N., Ander, E. L., Ault, L. 2005. G-BASE: Baseline geo-
777 chemical mapping of Great Britain and Northern Ireland. *Geochem. Explor.*
778 *Environ. Anal.* 5, 1–13.

- 779 Koschel, R., Benndorf, J., Proft, G., Rechnagel, F. 1978. Calcite precipitation as a
780 natural control mechanism of eutrophication. *Arch. Hydrobiol.* 98, 380–408.
- 781 Madari, B. E., Reeves III, J. B., Machado, P. L. O. A., Guimares, C. M., Torres, E.,
782 McCarty, G. W. 2006. Mid- and near-infrared spectroscopic assessment of soil
783 compositional parameters and structural indices in two Ferralsols. *Geoderma.*
784 136, 245–259.
- 785 Martinez-Carreras, N., Krein, A., Udelhoven, T., Gallart, F., Iffly, J., Hoffmann,
786 L., Pfister, L., Walling, D. 2010. A rapid spectral-reflectance-based fingerprint-
787 ing approach for documenting suspended sediment sources during storm runoff
788 events. *J. Soils Sediments.* 10, 400–413.
- 789 McGrath, S. P., Loveland, P. J. 1992. *The Soil Geochemical Atlas of England and*
790 *Wales*, Blackie Academic and Professional, Glasgow.
- 791 Mevik, B.H. and Wehrens, R., 2007. The pls Package: Principal Component and
792 Partial Least Squares Regression in R. *J. Stat. Soft.* 18, 1–24.
- 793 Morris, P.H. and Williams, D.J. 1999. A worldwide correlation for exponential bed
794 particle size variation in subaerial aqueous flows. *Earth Surf. Proc. Land.* 24,
795 835–847.
- 796 Palmer-Felgate, E.J., Jarvie, H.P., Withers, P.J.A., Mortimer, R.J.G., Krom, M.D.
797 2009. Stream-bed phosphorus in paired catchments with different agricultural
798 land use intensity. *Agr. Ecosyst. Environ.* 134, 53-66.
- 799 Payne, R.W. (ed) 2008. *GenStat Release 11 Reference Manual. Part 2 Directives.*
800 VSN International, Hemel Hempstead.
- 801 Plant, J. A. 1971. Orientation studies on stream sediment sampling for a regional
802 geochemical survey in northern Scotland. *T I MIN METALL B*, 234–345.

- 803 Prybil, D. W. 2010. A critical review of the conventional SOC to SOM conversion
804 factor. *Geoderma*. 156, 75–83.
- 805 Quinton, J.N., Govers, G., Van Oost, K., Bardgett, R.D. 2010. The impact of agri-
806 cultural soil erosion on biogeochemical cycling. *Nature Geosci.* 3, 311–314.
- 807 R Development Core Team, 2010. R: A Language and Environment for Statis-
808 tical Computing., R Foundation for Statistical Computing, Vienna, Austria,
809 <http://www.R-project.org>.
- 810 Rawlins, B. G., Turner, G., Mounteney, I. & Wildman, G. 2010. Estimating specific
811 surface area of fine stream bed sediments from geochemistry. *Appl. Geochem.*
812 25, 1291–1300
- 813 Rawlins, B.G., Webster, R., Lister, T.R. 2003. The influence of parent material on top
814 soil geochemistry in eastern England. *Earth Surf. Proc. Land.* 28, 1389–1409.
- 815 Smith, V. H. 2003. Eutrophication of freshwater and coastal marine ecosystems: a
816 global problem. *Environ. Sci. Pollut. Res. Int.* 10, 126–139.
- 817 Soil Survey of England and Wales, 1983a. *Soils and their Use in Midland and Western*
818 *England*. Ordnance Survey for the Soil Survey of England and Wales, Southamp-
819 ton.
- 820 Soil Survey of England and Wales, 1983b. *Soils and their Use in Eastern England*.
821 Ordnance Survey for the Soil Survey of England and Wales, Southampton.
- 822 Starinsky, A., Katz, A., Kolodny, Y. 1982. The incorporation of uranium into diage-
823 netic phosphorite. *Geochim. Cosmochim. Ac.* 46, 1365–1374.
- 824 Sylvester-Bradley, P.C. and Ford, T.D. 1968. *The geology of the East Midlands*.
825 Leicester University Press, 400pp.

- 826 van der Perk, M., Owens, P. N., Deeks, L. K., Rawlins, B. G., Haygarth, P. M.,
827 Beven, K. J. 2007. Controls on catchment-scale patterns of phosphorus in soil,
828 streambed sediment, and stream water. *J. Environ. Qual.* 36, 694–708.
- 829 van der Perk, M., P.N. Owens, L.K. Deeks, and B.G. Rawlins. 2006. Sediment
830 geochemical controls on in-stream phosphorus concentrations during base flow.
831 *Water Air Soil Poll.* 6, 443–451.
- 832 Wang, F., Chen, J., Chen, J., Forsling, W. 1997. Surface properties of natural aquatic
833 sediments. *Wat. Res.* 31, 1796–1800.
- 834 Withers, P.J.A., Edwards, A.C., Foy, R.H., 2001. Phosphorus cycling in UK agri-
835 culture and implications for phosphorus loss from soil. *Soil Use Manage.* 17,
836 139–149.
- 837 Zimmermann, M., Leifeld, J., Fuhrer, J. 2007. Quantifying soil organic carbon frac-
838 tions by infrared-spectroscopy. *Soil Biol. Biochem.* 39, 224-231.

839 **Figure captions**

840 **Fig.1** Simplified geological map of the study region.

841 **Fig.2** Map of the study region showing topography, urban areas and major rivers.

842 **Fig.3** Wavelengths for which both variable importance in the projection (VIP) scores
843 and regression (beta) coefficients are significant (see text) in partial least squares
844 models of reflectance spectra for prediction of: a) Al_d , b) Fe_d and c) organic
845 carbon (OC).

846 **Fig.4.** Measured and predicted values for selected bed sediments: a) Al_d ($mg\ kg^{-1}$;
847 $n=60$), b) Fe_d ($mg\ kg^{-1}$; $n=60$), b) total organic carbon (%) ($n=88$) based on mid
848 infra red diffuse reflectance spectrometry.

849 **Fig.5.** Matrix of scatterplots for properties of fine bed sediments ($n=1052$) from
850 headwater catchments and catchment area: $\log P = \log$ phosphorus concentration
851 ($mg\ kg^{-1}$), cerium (Ce; $mg\ kg^{-1}$), OC = organic carbon (%), D&M minerals
852 (relative scale), kaolinite (relative scale), $\log Fe(d; mg\ kg^{-1})$, $\log Al(d; mg\ kg^{-1})$,
853 SSA=specific surface area ($m^2\ g^{-1}$), catch. area = catchment area (km^2).

854 **Fig.6** Projections of the correlations between variables and the principal component
855 scores in unit circles: a) component 2 against component 1; b) component 3
856 against component 1. Variable notation: SSA=specific surface area, OC=organic
857 carbon, k=kaolinite, D&M=dioctahedral clay and mica, Fe_d =dithionite ex-
858 tractable iron , Al_d =dithionite extractable aluminium , soil P= mean catchment
859 topsoil phosphorus, res Fe=residual iron.

860

861 **Fig.7** Fine bed sediment mineral specific surface area versus catchment area (km^2) at
862 1052 sampling sites. The dashed line is a power function fitted to the data (see
863 text).

Table 1 Selected statistics for measured and MIR-DRS estimated properties of fine bed sediments and topsoil across the study region.

Element concentrations in mg kg^{-1} .

	Sediments						Topsoil		
	*OC (%)	Fe _d	Al _d	†SSA	Fe	P	**P		
	meas.	meas.	meas.	meas.	meas.	meas.	meas.		
Min.	0.5	4974	265	249	5.98	1.60	16020	262	87.2
Max.	6.0	90140	4747	5430	50.5	63.3	213000	11480	11826
Median	2.9	18940	1114	944	20.1	23.2	49450	1266	829
Mean	3.0	20410	1192	1045	22.7	23.7	52910	1447	1008
St. Dev.	1.2	13041	690	538	10.1	8.61	19670	763	584
Skewness	0.32	2.63	2.6	2.3	0.54	0.37	1.54	3.43	3.93
Counts	88	987	60	1015	56	1015	1052	1052	7233

* Organic Carbon

** <2 mm size fraction

†units g m^{-2}

meas.=measured,est.=estimated

871 **Table 2** Selected features of optimal partial least squares regression models for esti-
 872 mation of Al_d , Fe_d and organic carbon in bed sediments.

	Property	^a n components	adj R^2	adj R^2 -CV	^b RMSEP-CV
873	^c Al_d	5	0.80	0.69	0.29
	^c Fe_d	5	0.89	0.80	0.25
	Org. Car.	6	0.88	0.80	0.53

874 ^a number of orthogonal PLSR components

875 ^b cross validation adjusted R^2

876 ^c root mean square error of cross validation (RMSEP-CV) given in log units

877

878 **Table 3** Summary statistics from simple linear regression by ordinary least squares for
 879 log bed sediment P (n=1052) for a range of explanatory variables.

Predictor	Estimate	Std. Error	P-value	adjusted R ² (%)
log Al _d	0.583	0.024	<0.001	35.6
log Fe _d	0.44	0.021	<0.001	28.4
log Fe	0.62	0.035	<0.001	23.0
^a residual Fe	-0.59	0.04	<0.001	16.4
kaolinite	241	16.1	<0.001	17.4
880 ^b topsoil P	0.53	0.035	<0.001	18.2
Organic carbon	0.135	0.01	<0.001	11.6
Ce	0.008	69.4 × 10 ⁻⁵	<0.001	10.4
^c D&M	-1.07	0.146	<0.001	4.7
SSA	0.006	0.001	<0.001	1.4
^d Arable	-0.008	0.006	0.163	0.1
^d Grassland	0.004	0.004	0.333	0.45

881 ^aresidual Fe is total Fe_d minus total Fe determined by XRFS (see text)

882 ^b average catchment topsoil P

883 ^c dioctahedral clay and mica minerals

884 ^d log ratios of the proportions of land use type:all other land use types

885

886 **Table 4** Summary statistics for the proportions of permanent grassland and arable
887 land in a subset of 547 catchments in which the sum of their proportions accounts for
888 more than 70% of the total catchment area. All proportions as percentages (%).

Statistic	Arable	Permanent grassland
Min	0.1	0.1
889 Median	35	24
Mean	38.9	22.7
Max	99.9	99.9
St. Dev.	23.8	21.8

890 **Table 5** Correlations between variables and their principal component scores for nine
 891 selected predictors. The latent roots and accumulated variance (%) are also reported.
 892 Selected large positive and negative coefficients in **bold**.

	Principal Component								
	1	2	3	4	5	6	7	8	9
SSA	0.64	0.04	0.58	0.41	0.27	-0.04	-0.11	0.006	0.028
OC	0.05	-0.94	0.23	-0.15	-0.04	0.21	-0.01	-0.03	-0.02
Fe _d	0.96	-0.04	0.02	-0.07	0.04	-0.16	0.06	0.07	-0.19
^a resid Fe	-0.82	0.22	0.35	0.17	-0.13	0.17	0.28	0.08	0.04
Soil P	0.73	0.08	0.05	0.22	0.63	0.04	-0.11	0.01	0.002
893 D&M	0.82	0.16	-0.29	0.22	0.16	0.29	0.09	-0.18	0.03
kaolinite	0.91	-0.06	-0.22	0.02	0.07	0.16	0.05	0.28	0.07
Al _d	0.89	-0.17	0.11	-0.11	-0.07	-0.22	0.27	-0.09	0.10
Ce	0.56	0.47	0.34	-0.56	0.002	0.17	-0.06	-0.02	0.01
Latent roots	5.18	1.22	0.77	0.65	0.53	0.29	0.19	0.13	0.05
acc. Var. (%)	57.5	71.0	79.5	86.7	92.6	95.8	97.9	99.4	100

894 ^a total Fe_d minus the total concentration of iron determined by XRFS (see text)

895

896 acc. Var. = accumulated variance

897

898 **Table 6** Analysis of variance summary following sequential inclusion of eight principal component predictors (Comp), the proportion
 899 of land use types and a dominant catchment geological classification (Geol) in a multiple linear regression model to predict log bed
 900 sediment P concentrations. Specific interactions (*) between the principal components and the geological classification are also listed.

901

Predictor	Deg. freedom	Sum squares	Mean square	Var. ratio	P-value	adjusted R ² (%)
Comp 1	1	50.8	50.8	663	<.001	24.1
Comp 2	1	11.9	11.9	156	<.001	29.7
Comp 3	1	1.17	1.17	15.2	<.001	30.2
Comp 4	1	20.7	20.7	270	<.001	39.9
Comp 5	1	21.5	21.5	281	<.001	50.2
Comp 6	1	0.499	0.499	6.52	0.011	50.3
Comp 7	1	9.23	9.23	121	<.001	54.7
Comp 8	1	3.96	3.96	51.8	<.001	56.6
Arable	1	0.688	0.688	8.99	0.003	56.9
Pasture	1	0.606	0.606	7.91	0.005	57.1
Geol	17	4.54	0.267	3.49	<.001	58.6
Comp 1*Geol	17	3.00	0.177	2.31	0.002	59.4
Comp 2*Geol	17	2.81	0.165	2.16	0.004	60.1
Comp 3*Geol	17	2.78	0.173	2.27	0.002	60.9
Comp 7*Geol	17	2.92	0.183	2.39	0.002	61.9
Residual	956	73.36	0.077			
Total	1051	210	0.200			

902

Figure 1:

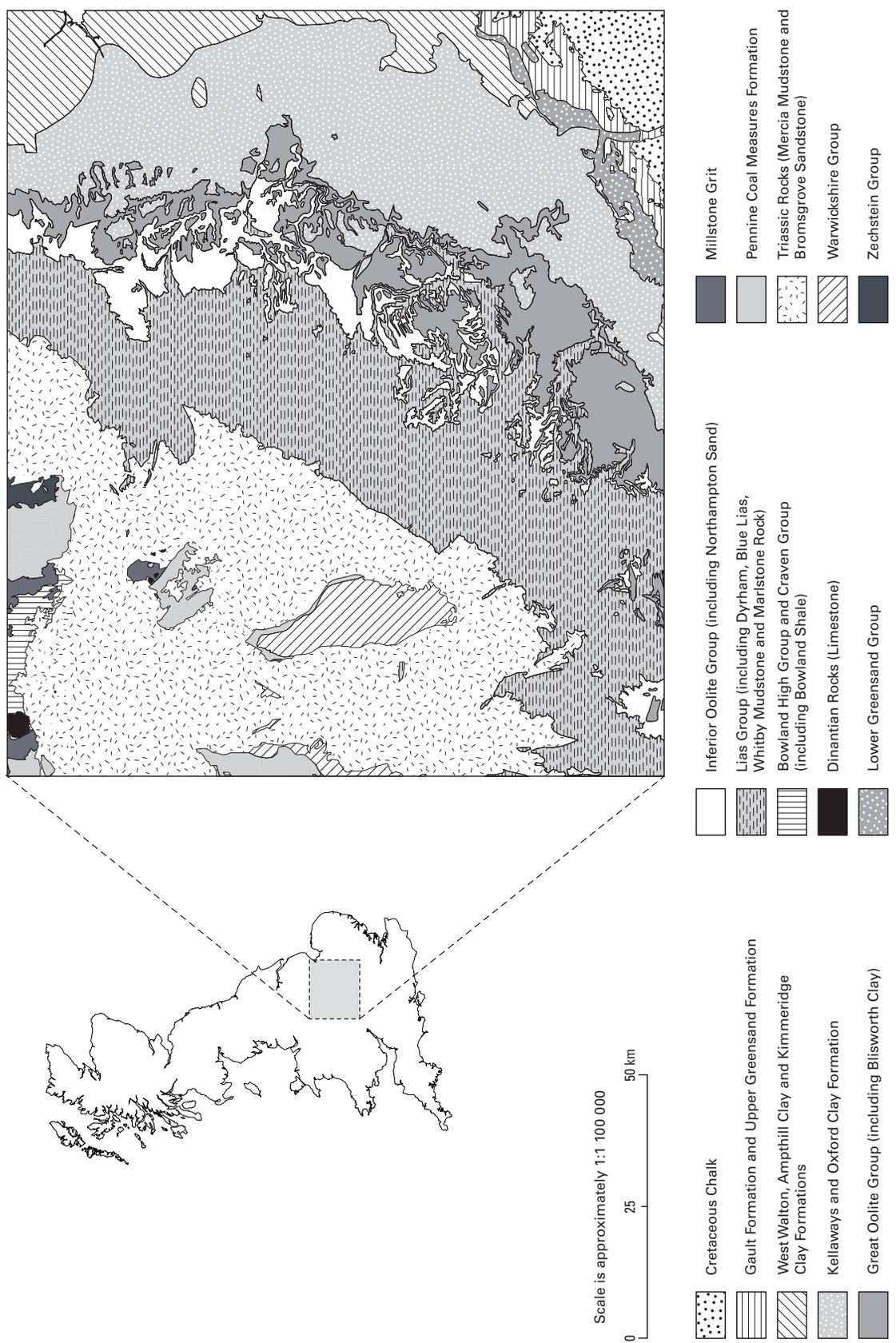


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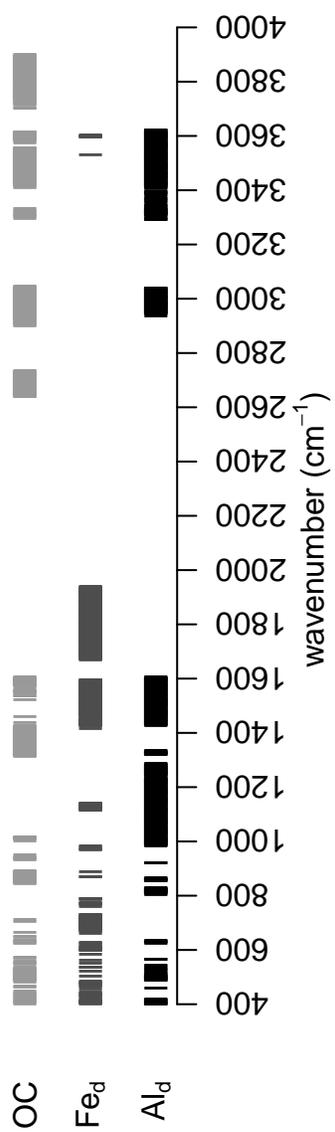


Figure 4:

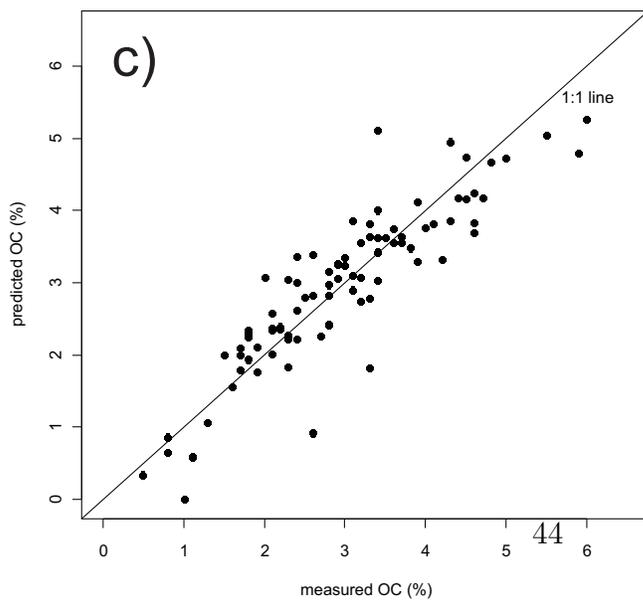
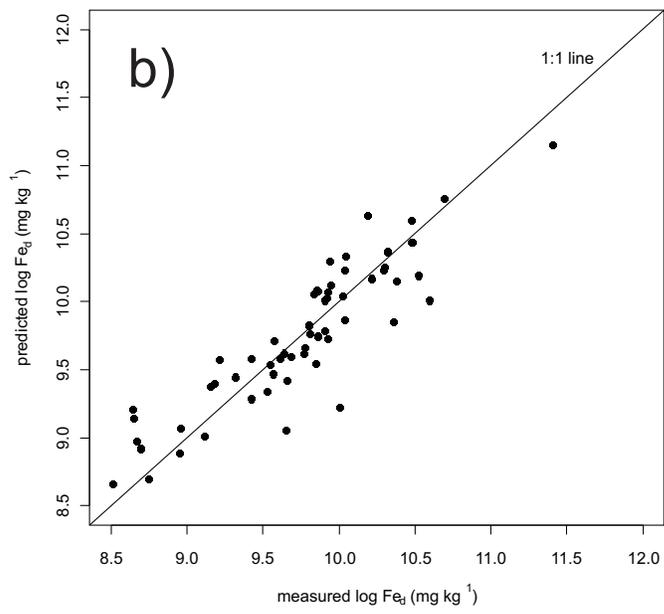
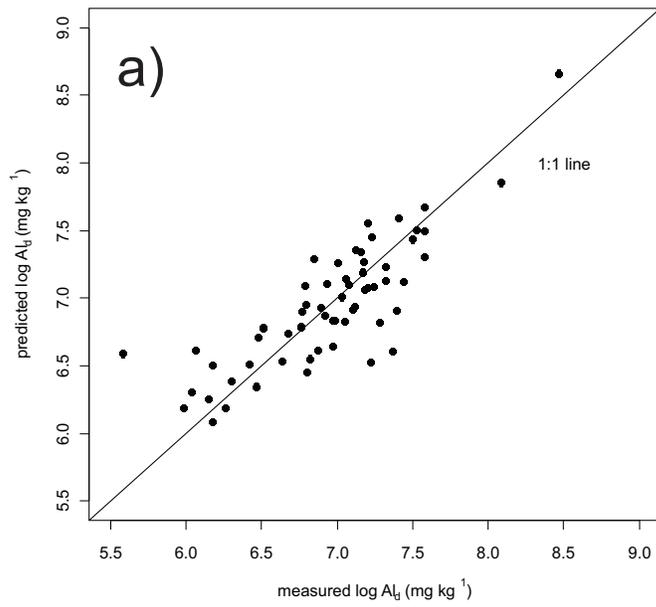


Figure 5:

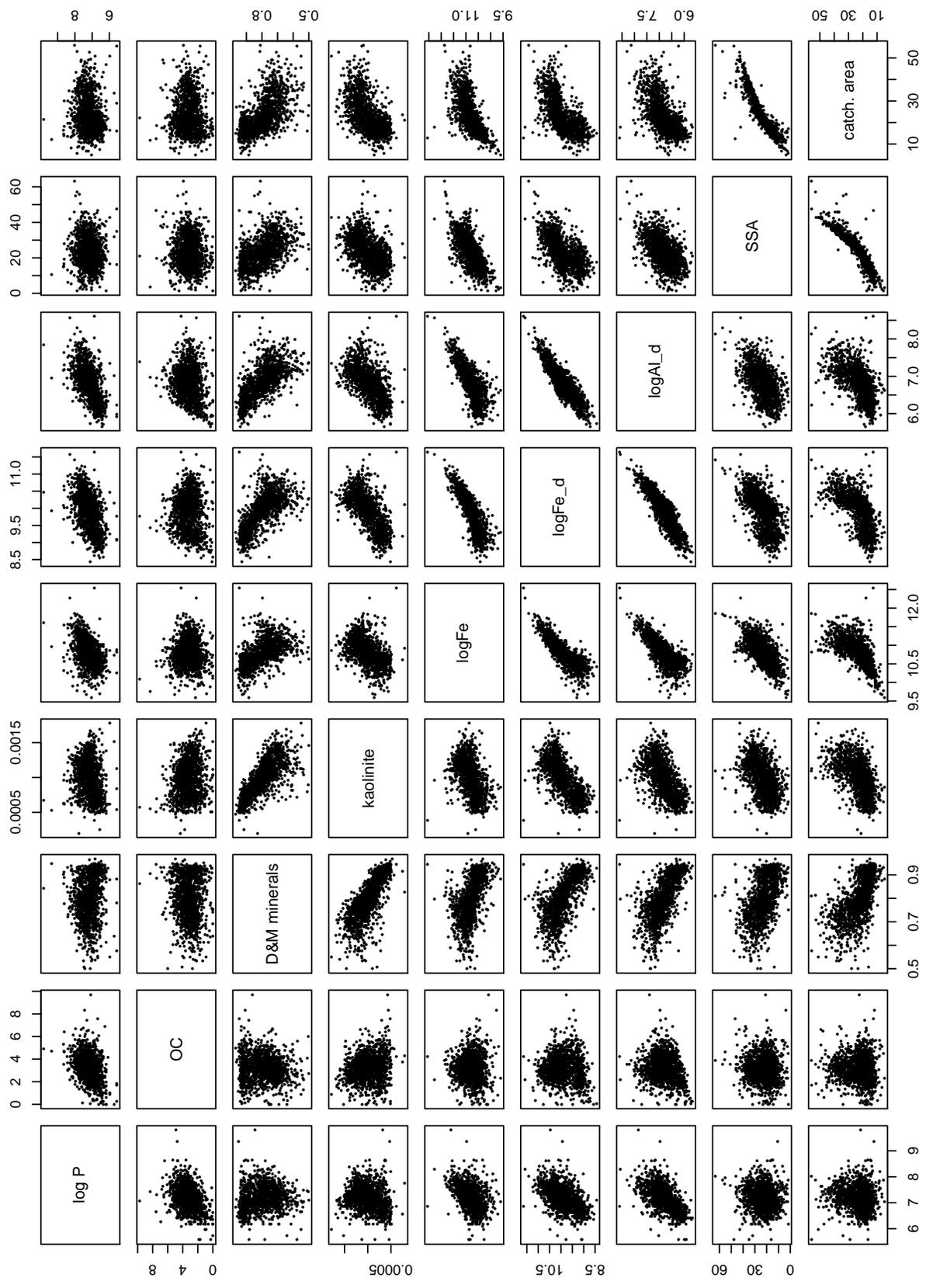


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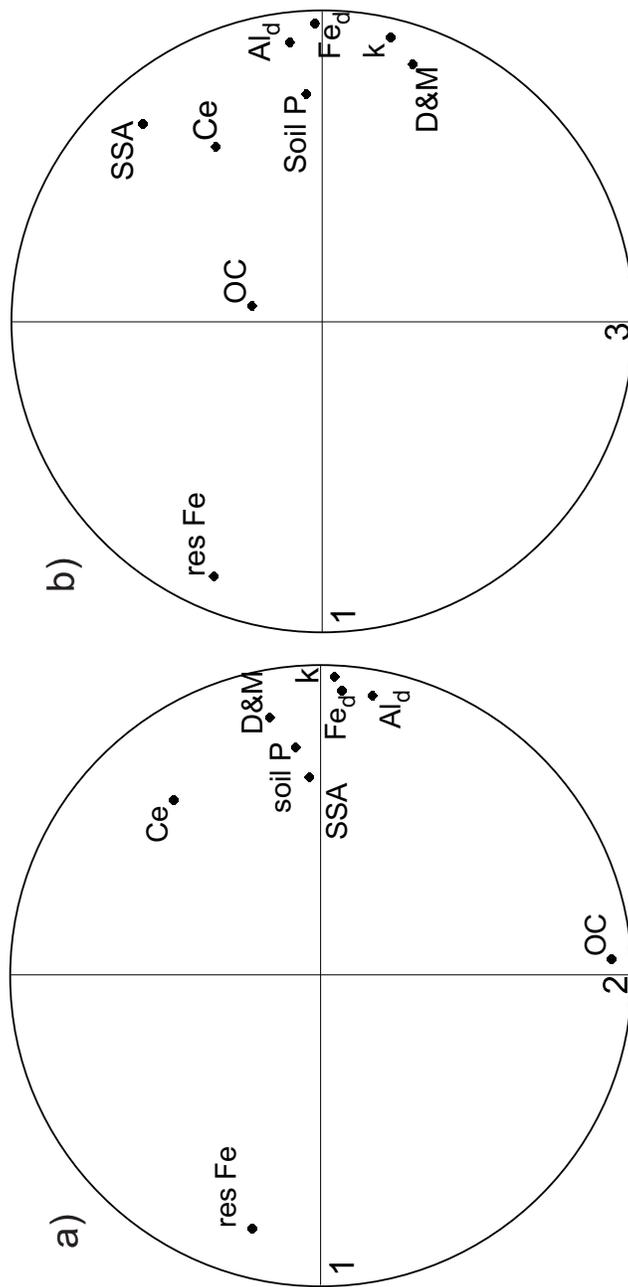
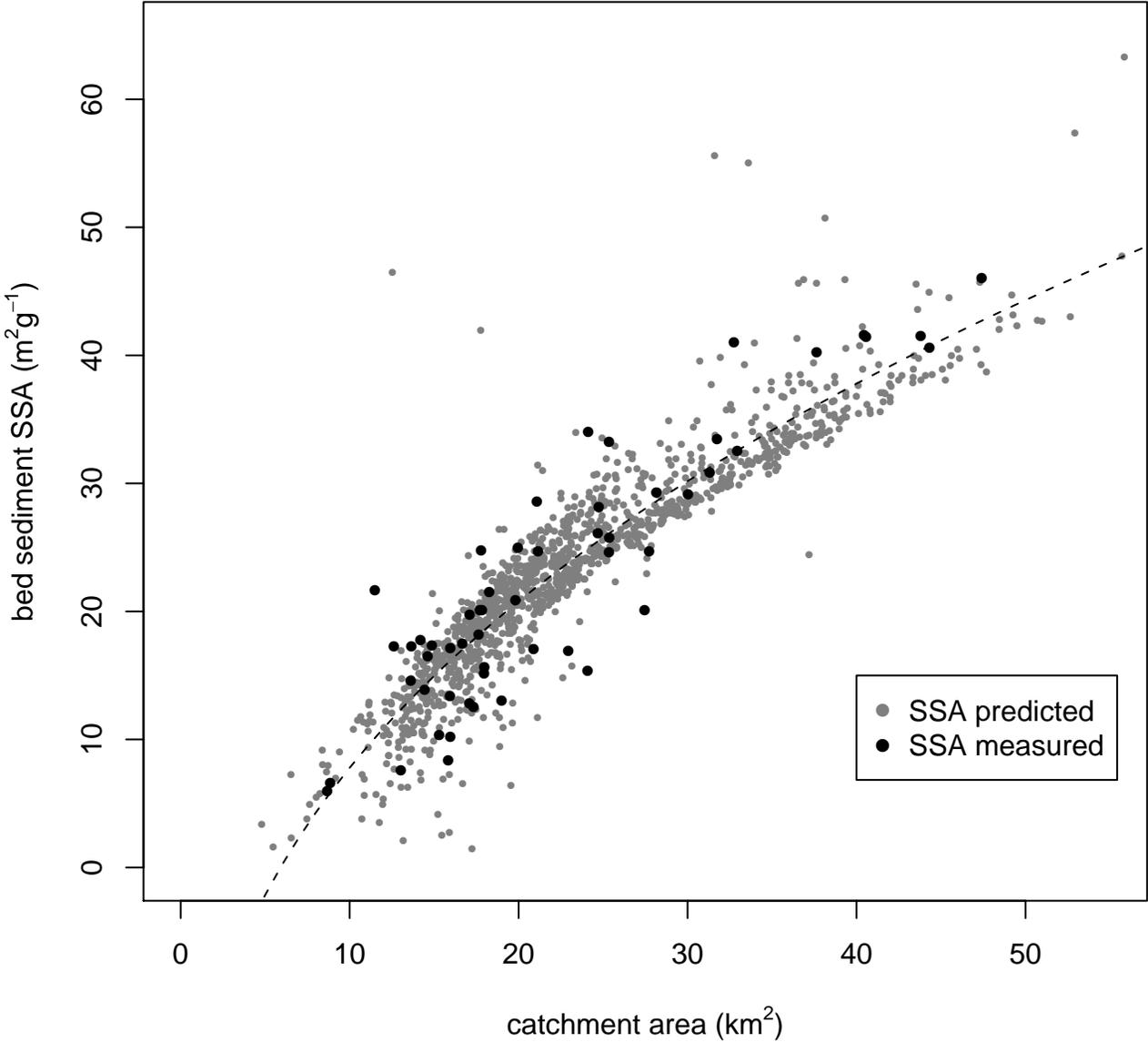


Figure 7:



vs CA.pdf